Tank 241-T-111 Characterization Report

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Date Published September 1994

Prepared for the U.S. Department of Energy Office of Environmental Restoration and Waste Management



Hanford Operations and Engineering Contractor for the U.S. Department of Energy under Contract DE-AC06-87RL10930



Approved for Public Release

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Document Number:

WHC-EP-0806

Document Title:

Tank 241-T-111 Characterization Report

Release Date:

October 6, 1994

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EXECUTIVE SUMMARY

Single-shell tank 241-T-111 is a Hanford-Site High-Level Waste tank that was most recently sampled in late October and early November 1991. Analyses of materials obtained from tank 241-T-111 were conducted to support the Hanford Federal Facility Agreement and Consent Order¹ (Tri-Party Agreement) Milestone M-10-06 (now M-44-05). The waste in tank 241-T-111 is a complex material primarily made up of water and inorganic salts in a gel-like-matrix. The insoluble solids are a mixture of phosphates, silicates, oxides, and hydroxides in combination with lanthanum, calcium, manganese, iron, bismuth, uranium, and chromium. The soluble analytes are primarily sodium, nitrate, sulfate, and fluoride (see Table ES-1).

Exotherms over -215 cal/dry gram were detected in the first two segments from each core. As a result of these analyses, tank 241-T-111 has been placed on the Organic Watch List². The source of these exotherms remains under investigation; however, under present tank conditions, there is no possibility that a rapid exothermic reaction will occur. Another finding from the characterization analyses was that the present method for determining or measuring total organic carbon was not effective for the organic materials in this waste matrix. Further investigation of the method failure for this waste is underway and other assay methods are being considered.

¹Ecology, EPA, and DOE, 1994, Hanford Federal Facility Agreement and Consent Order, 2 vols. as amended, Washington State Department of Ecology, U.S. Environmental Protection Agency, and U.S. Department of Energy, Olympia, Washington.

²Wicks, J. H. 1994, *Tank T-111*, (Internal Memo 94-003 to D. A. Turner, March 3) Westinghouse Hanford Company, Richland, Washington.

Comparisons of the calculated bulk inventories for various analytes of concern show that tank 241-T-111 is within established operating safety requirements for heat-load, ferrocyanide, and plutonium inventory. From assessment of past and present surveillance data, tank 241-T-111 is considered an assumed leaking tank (see Table ES-2). Mitigation presently is taking place to remove the remaining free liquid in the tank in order to forestall any further leakage of tank contents to the environment. The free liquid is being transferred to tank 241-SY-102 as part of the overall single-shell tank stabilization effort³.

The results of the analyses have been compared to the dangerous waste codes in the "Washington Dangerous Waste Regulations" (WAC 173-303)4. This assessment was conducted by comparing tank analyses against dangerous waste characteristics ("D" waste codes) and against state waste codes. The comparison did not include checking tank analyses against "U," "P," "F," or "K" waste codes because application of these codes is dependent on the source of the waste and not on particular constituent concentrations. The results indicate that the waste in this tank is adequately described in the Dangerous Waste Permit Application for the Single-Shell Tank System; this permit is discussed in the Tank Characterization Reference Guide.5

³Jenkins, C. E. and D. B. Engleman, 1994, Engineering Report: Managing the Assumed Leak from Single-Shell Tank 241-T-111, WHC-SD-WM-ER-337, Westinghouse Hanford Company, Richland, Washington.

⁴WAC 173-303, "Dangerous Waste Regulations", Washington Administrative Code, as amended, Olympia, Washington.

⁵De Lorenzo, D. S., et al. 1994, Tank Characterization Reference Guide, WHC-SD-WM-TI-648, Rev. 0, Westinghouse Hanford Company, Richland, Washington.

Analysis of the process history of the tank provided valuable information about the likely physical and chemical condition of the waste. Direct comparisons with historical tank =========reviews underway at Los Alamos National Laboratory are not possible at this time because the Los Alamos National Laboratory effort has not progressed to encompass tank 241-T-111. However, estimates and comparisons using the available process knowledge⁶ have been made. Further comparisons with the Los Alamos National Laboratory database will be made when it becomes available in 1995. The available historical information, in combination with the analysis of the tank waste, supports the conclusion that a rapid exothermic reaction in tank 241-T-111 with the present tank conditions is not plausible because of the high moisture content of the waste and lack of any concentrated heat source. Therefore, the contents of tank 241-T-111 present no credible imminent threat to the workers at the Hanford Site, the public, or the environment. Because an exothermic reaction is not credible, the consequences of this accident scenario, as promulgated by the General Accounting Office, are not applicable. However, until the source and mechanism of the exotherm is further clarified, it was considered prudent to list the tank on the Organics Watch List, apply more rigorous access controls, and conduct further intrusive operations in tank 241-T-111 with greater care.

⁶Agnew, S. F., 1994, Hanford Defined Wastes: Chemical and Radionuclide Compositions, LA-UR-94-2657, Los Alamos National Laboratory, Los Alamos, New Mexico.

⁷Peach, J. D., 1990, Consequences of Explosion of Hanford's Single-Shell Tank are Understated, (Letter B-241479 to C. M. Synar, Chairman of Environment, Energy, and Natural Resources Subcommittee, Committee on Government Operations, House of Representatives), GAO/RCED-91-34, General Accounting Office, Washington, D.C.

Table ES-1. Single-Shell Tank 241-T-111 Concentrations and Inventories for Critical List Analytes.

Physical Properties						
Density 1.16-1.28 g/mL	H ₂ O 76.0%	1,650,000 kg				
Temperature 16 °C	Heat Load 81 w	(24 BTU/hr)				
pH 11.65	Total Waste Mass	2,171,000 kg				
Chemical Constituents	Average Concentration (wt%)	Bulk Inventory (kg)				
Calcium (Ca)	0.242	5,260				
Chromium (Cr)	0.198	4,290				
Iron (Fe)	1.85	40,200				
Manganese (Mn)	0.633	13,700				
Sodium (Na)	3.70	80,300				
Bismuth (Bi)	2.60	56,300				
Lanthanum (La)	0.422	9,200				
Silicon (Si)	0.567	12,300				
Uranium (U)	0.355	7,700				
Total Phosphate (PO ₄ -3)	3.23	70,100				
Sulfate (SO ₄ -2)	0.355	7,700				
Nitrate (NO ₃)	4.13	89,700				
Fluoride (F)	0.230	4,990				
Total Organic Carbon (TOC)	0.312	6,770				
Radionuclides	(μCi/g)	(Ci)				
Total Plutonium	0.304	660				
Am-241	0.0425	92.4				
Sr-90	5.41	11,800				
Cs-137	0.166	360				

Table ES-2. Tank 241-T-111.

Tank De	scription
Туре:	Single-Shell
Constructed:	1944
In Service:	1945
Diameter:	75 ft (22.9 m)
Usable Depth:	17 ft (5.2 m)
Operating Capacity:	530,000 gal
	(2.01E+06 L)
Bottom Shape:	Dished
Hanford Coordinates:	43.347° North
	75.737° West
Ventilation:	Passive
Tank Status: a	s of May 1994
Contents:	Non-Complexed Waste
Total Waste:	456,000 gal
	(1.73E+06 L)
Supernate Volume:	0 gal
	(0 L)
Drainable Interstitial	
Liquid:	51,000 gal (193,000 L)
Manual Tape Surface Level (under riser):	161.1 in. (408.9 cm)
Liquid Observation Well Level:	150 0 := /400 4
	158.8 in. (403.4 cm)
Integrity Category:	Assumed Leaker
Watch List Status:	Organic

PUMP PIT
O 10

OVERFLOW

PUMP PIT
O 10

N2

N3

N4

N5

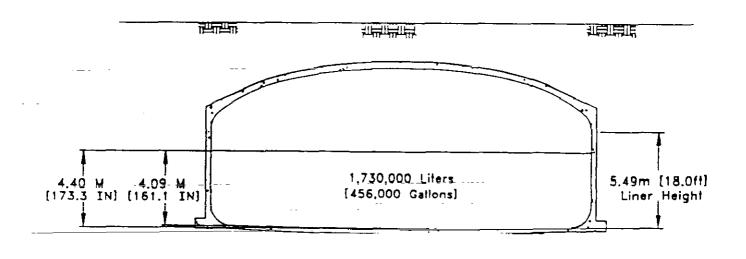
N5

N6

MH

N5

TC=Thermocouple



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---LIST-OF-ACRONYMS/ABBREVIATIONS

2C Second-Cycle Decontamination Waste From The Bismuth

Phosphate Process

224 Lanthanum Fluoride Waste

AA Atomic Absorption
ANOVA Analysis Of Variance
CI Confidence Interval

_DSC __Differential Scanning Calorimetry

GEA Gamma Energy Analysis IC Ion Chromatography

Inductively Coupled Plasma - Atomic Emission Spectroscopy

PNL Pacific Northwest Laboratory
RPD Relative Percent Difference

SST Single-Shell Tank

TGA Thermogravimetric Analysis
TIC Total Inorganic Carbon
TOC Total Organic Carbon

Tri-Party

Agreement Hanford Federal Facility Agreement And Consent Order

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TANK CHARACTERIZATION REPORT: TANK 241-T-111

1.0 INTRODUCTION

In late October and early November of 1991, single-shell tank (SST) 241-T-111 was sampled and analyses were conducted on the materials obtained to complete *Hanford Federal Facility*- Agreement and Consent Order (Tri-Party Agreement) Milestone M-10-00 (Ecology et al. 1992) to sample and analyze two cores from each tank. Other objectives that these measurements and inventory estimates support are as follows.

- Obtain estimates of both the concentration and total quantity of key analytes relating to safety issues, such as organics and radionuclides.
- Provide input to risk assessment-based disposal decisions for the waste.
- Implement physical property measurements, such as rheology, bulk density, and particle size. These measurements are necessary for the design and fabrication of retrieval, pretreatment, and vitrification systems.

1.1 PURPOSE

The purpose of the tank characterization report is to describe and characterize the waste in SST 241-T-111, based on information from various sources. This report summarizes the available information regarding the waste in tank 241-T-111, and arranges this information in a format useful to support management and technical decisions concerning this waste tank.

1.2 SCOPE

This report presents a broad background of information that was available before core sampling, which initially guided the development of the sampling and analysis program. This material includes process stream data, historical information about any previous characterization testing, transfer records, and observations from in-tank photographs. The results of tank 241-T-111 core-sample analyses are summarized and presented, along with a statistical interpretation of the data. The information obtained from historical sources will be compared and correlated with the actual waste measurements in this report. As characterization efforts proceed and additional information becomes available, this document will be revised periodically to reflect the new data set.

2.0 HISTORICAL TANK INFORMATION AND EVALUATION

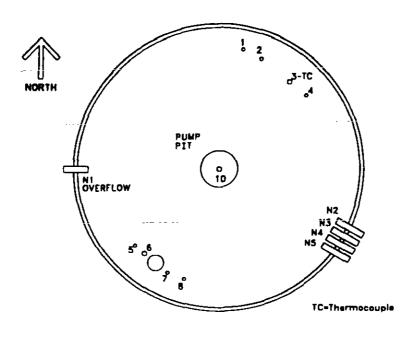
2.1 BACKGROUND

Radioactive wastes from defense operations have accumulated at the Hanford Site in underground waste tanks since the late 1940s. The original tank farms (B, C, T, and U) were built from 1943 to 1944. Tank 241-T-111 was placed into service in 1945. Groups of waste tanks that were physically located together and built at the same time are called tank farms. Each original tank has a diameter of 22.9 m (75 ft), an operating depth of 5.2 m (17 ft), and a nominal capacity of 2 million liters (530,000 gal). The basic design of a typical SST is shown in Figure 2-1. The tanks were constructed of reinforced concrete with a mild steel liner covering their bottoms and sides. The carbon steel liners were designed to receive and contain neutralized, mildly alkaline wastes. The tops of the tanks are concrete domes. Tanks such as 241-T-111 were all covered by at least 1.8 m (6 ft) of soil for shielding purposes (Anderson 1990). The tanks in the tank farms were connected in groups of three or four and overflowed from one to another in a configuration known as a cascade. Tank 241-T-111 is the middle tank in a cascade that includes 241-T-110 and 241-T-112. Cascades served several functions in Hanford-Site waste management operations. By cascading tanks, fewer-connections needed to be made during waste disposal. Consequently, all three tanks were usable without having to connect the active waste transfer line directly to each individual tank. This handling method reduced the likelihood for personnel exposure to the waste and diminished the chances for a loss of tank integrity because of overfilling. Another benefit of the cascades was clarification of the wastes. When used in this manner, most of the solids in the waste slurries routed to the tanks settled in the first tank (241-T-110), and the clarified liquids cascaded on to the other tanks in the series (241-T-111 and 241-T-112). Supernate from the final tank in a cascade series was sometimes routed to a disposal trench. Since most radionuclides are insoluble in aqueous alkaline media, clarification reduced the potential amount of radiological contamination to the environment. However, historical sources report that cascade lines routinely clogged (Anderson 1990). When clogging occurred, very little could be done to resolve the problem, other than rerouting the effluent stream directly to the disposal tank. Cascading was a common practice in the early-process history of the tanks, but became less frequent as time passed, virtually ceasing by the late 1950s.

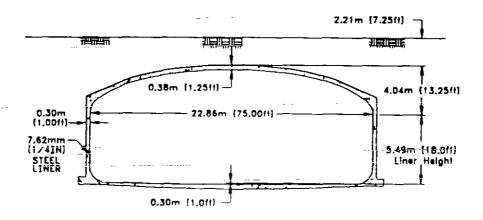
-2.2 TANK 241-T-H1-HISTORY

The first type of waste that tank 241-T-111 received and stored was second-cycle decontamination waste from the bismuth phosphate process (2C) (1945 to 1952). The tank was filled by a cascade of 2C waste in the fourth quarter of 1945. The supernatant in the tank was disposed to a crib in the third and fourth quarters of 1947. The tank was then refilled with 2C waste in the second quarter of 1948. After the cascade was filled again in late 1948, tank 241-T-111 remained in active service. From 1953 to 1955, tank 241-T-111 was

Figure 2-1. Typical Single-Shell Tank Diagram.



TANK RISER LOCATION



Note: Tank 241-T-111 has a dished bottom.

10 Notalia

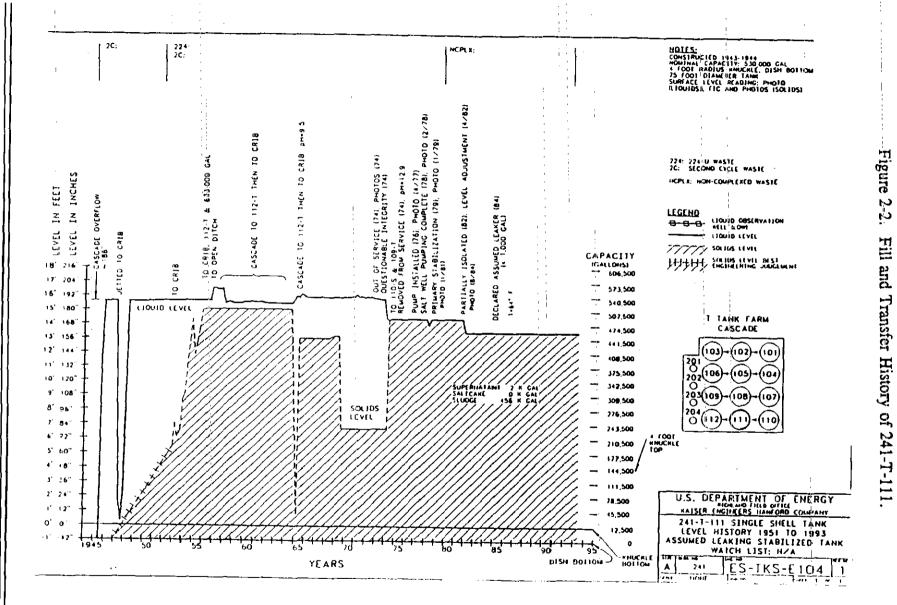
used to cascade 2C and lanthanum fluoride waste (224) from the LaF₃ finishing process in T Plant to a crib. In addition, Anderson (1990) reports that some 5-6 cell drainage out of B Plant was routed to the tank along with 224 waste in 1952.

After the end of the T-Plant cascade in 1955, the reported total waste volume remained relatively unchanged [between 1.98 and 2.12 million L (524,000 and 560,000 gal)] for the remainder of the tank's active service life. There was a residual heel of at least 1.85 million L (488,000 gal) left from previous waste management operations. Tank 241-T-111 remained in 2C service through the third quarter of 1956 (T Plant ran BiPO₄ until August 1956). Anderson (1990) notes that in 1952 the tank was also receiving 224 waste from the plutonium purification/concentration processes performed in the 224-T building.

T Plant initially was built as a bismuth phosphate processing plant, however that purpose was changed when it became an equipment decontamination facility. The tank would have also received miscellaneous decontamination chemicals from T-Plant decontamination operations performed in the 1960s. The records are not clear whether tank 241-T-111 always received waste as overflow from tank 241-T-110, or if waste was later routed directly to it. There is anecdotal evidence that the cascade overflow line plugged early in the service life of tank 241-T-110, and that a direct discharge line from T Plant was used to dispose of wastes to tank 241-T-111, however no reference confirming this fact can be found. Between 1964 and 1974, the reported solids volume fluctuated widely between 1.93 and 0.88 million L (510,000 and 233,000 gal). A highly anomalous reading of 150,000 L (40,000 gal) is not considered credible and is believed to be the result of a transcription error. In that time period, the total volume reported rose slightly, then decreased from 2.06 million L to 1.85 million L (541,000 to 488,000 gal) (Anderson 1990). In the second quarter of 1974, there were two small transfers out of Tank 241-T-111: one of 106,000 L (28,000 gal) to tank 241-S-110 and one of 53,000 L (14,000 gal) to tank 241-T-109.

In the third quarter of 1974; tank 241-T-111 was removed from active service. In the first and second quarters of 1976, two minor transfers of 30,300 L and 19,000 L (8,000 and 5,000 gal), respectively, were made out of tank 241-T-111. Saltwell pumping commenced in the third quarter of 1976 as part of the tank stabilization effort, and no further waste receipts were made. In 1979, the integrity of the tank was questioned and dry well 50-11-11 was drilled in the third quarter. As a result of an observed level drop, tank 241-T-111 was declared an assumed leaker in 1984 (Hanlon 1994). Figure 2-2 illustrates the fill and transfer history of 241-T-111.

The wide fluctuation in the early reported solids level makes it difficult to derive any firm conclusions regarding the stratification in the tank on a strictly historical basis. Overall sludge volume in the tank may have decreased somewhat between 1956 and 1974 with further settling and compaction from the weight of overlying solids. The amount of sludge added since the end of the T-Plant cascade activity probably is negligible, because the transfer history of the tank was so limited. Floating suction pumps do not transfer solids



readily, and the movement of more than 950,000 L (250,000 gal) of solids in a quarter seems unlikely. However, the reported solids measurement in the fourth quarter of 1956 [1.93 million L (510,000 gal)] and the reported solids measurement at the end of the tank's active service life in 1980 [1.85 million L (488,000 gal)] appears to be reasonable, a 4.5 percent difference, given the time and compaction processes ongoing in the tank, and the inherent uncertainties associated with early solids measurements in the tank farms.

2.3 PROCESS KNOWLEDGE

Process knowledge obtained from historical records can be used to predict the major constituents and some general physical properties of the waste matrix in the tanks. Tank 241-T-111 is expected to contain relatively soft sludge, which can be push-mode sampled. During its operating history, tank 241-T-111 was never subject to any of the various waste volume reduction or in-tank solidification processes. Consequently, there was no formation of hard salt cake on top of the sludge, as there was in the BY or TX Tank Farms (Anderson 1990). This expectation was supported by inspection of in-tank photographs that indicated a moist and pliant waste surface (see Figures 2-3 and 2-4). The effluents that were added to the tank during waste management operations were slurries consisting primarily of water (Schneider 1951). There was no mixing equipment in tank 241-T-111 to blend the layers of settled solids together and there were distinct differences in the composition of wastes directed to it over its operating life. Agnew (1994) predicts that 2C and 224 wastes are not identical in makeup, and analytically observable variations in composition are expected. Because of the lack of agitation or mixing in the tank, observable evidence of layering is expected in some of the segment-level analytical results.

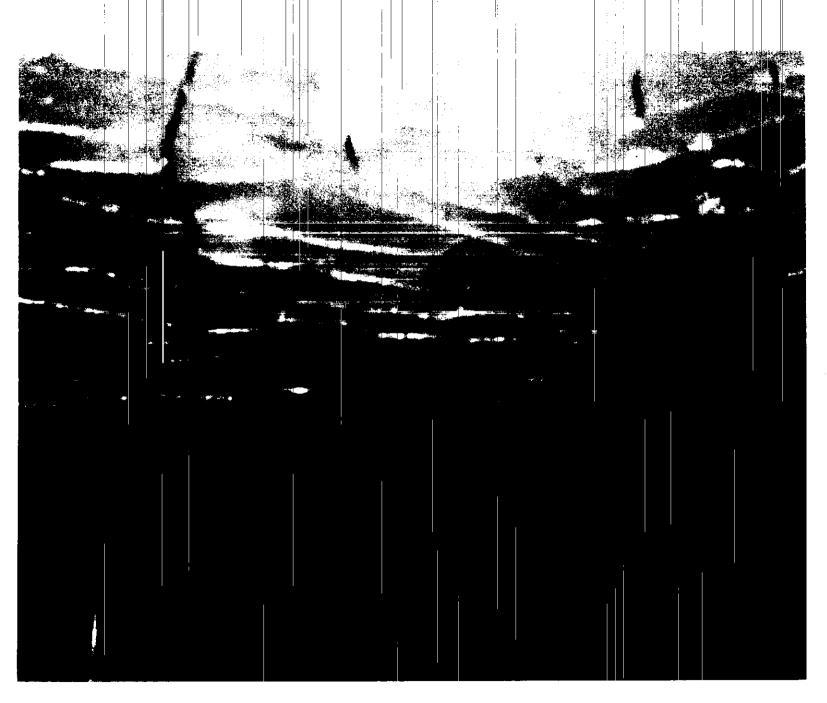
Previous analytical data on tank 241-T-111 is scant. The results of some liquid samples taken in 1974 are provided in Appendix C. Because the tank has been pumped several times since those samples were taken, they are not considered representative of the present waste matrix. The lower solids in tank 241-B-110 should be similar to the lower solids in tank 241-T-111 (Borsheim 1994). Both tanks were filled at least twice with 2C. Tank 241-B-110 was the first tank in the B-Plant 2C cascade while tank 241-T-111 was the second tank in the T-Plant 2C cascade. Further studies and comparisons of tank compositions will be forthcoming, pending additional analytical information on the various waste types and matrices that exist in the tank farms. The solids from the neutralized solution would have settled out in the tanks while the supernatant was disposed to cribs. The upper solids of tank 241-T-111 should have some similarity to the B and T 200-series tanks, because they all received 224 waste.

The estimated composition of neutralized 2C and 224 waste streams (i.e. unsettled) are given in Table 2-1 as determined from Schneider (1951). The Schneider (1951) process stream compositions are for the unsettled wastes being transferred from the separations plants. The

Overflow Nozzie

Figure 2-3. 241-T-111 Surveillance Photo - Collage.

Figure 2-4. 241-T-111 Surveillance Photo - Close up.



estimates using Agnew (1994) provide insight to the analyte concentrations of the settled sludge. How the waste settled, the solids content of the settled waste, and how it affected the waste inventory of the tank will be discussed further in Section 6.

Table 2-1. Typical Waste Stream Compositions and Expected Sludge Compositions for Selected Analytes (Wet Basis).

Analyte	BiPO (Schnei	itled 2C 4 waste der 1951)	(Agnet average and 19 compo	Settled 2C BiPO ₄ waste, (Agnew 1994) average 1944-1951 and 1951-1956 compositions		Unsettled BiPO ₄ -LaF ₃ 224 process waste (Schneider 1951)		Settled BiPO ₄ -LaF ₃ 224 process waste (Agnew 1994)	
Cation	Wt %	(μg/g)_	Wt %	(μg/g)	Wt %	(μg/g)	Wt %	(μg/g)	
Bi	0.12	1,200	2.55	25,500	0.11	1,100	2.07	20,690	
Cr	0.006	60	0.0033	33	0.016	160	0.0035	35	
Na*	3.5	35,000	6.46	64,600	3.39	33,900	8.50	85,000	
NH ₄ ***	0.16	1,600	-	-	0.011	110	-	-	
Fe	0.17	1,700	1.79	17,900	-	= -	0	0	
Mn	-	-	0	0	0.031	310	0.019	190	
K	-	-	0	0	0.79	7,900	0.779	7,790	
La		<u>.</u>	.0	·0- ·	0.05	500	2.67	26,700	
Anions	· · ·								
PO ₄ 3-	2.2	22,000	5.63	56,300	0.28	2,800	1.25	12,500	
SO ₄ 2	0.34	3,400 -	0.276	2,760	0.032	320	0	0	
NO ₃	5.8	58,000	4.28	42,800	3.91	39,100	6.16	61,600	
F:			0.494	4,940	0.52	5,200	4.66	46,600	
SiF ₆ ²	0.35	3,500	-	-	-		-	T -	
$C_2O_4^{-2}$	-		-	0	0.12	1,200	1.52	15,200	
		873,000	76.0	760,000	90.75	907,500	69.0	689,600	

^{*}Analytes listed in italics are mostly soluble.

^{**}NH₄+ probably has dissipated over time and is believed to be no longer present.

2.4 SURVEILLANCE DATA

The most recent waste inventory measurement for tank 241-T-111 reports approximately 1.73 million L (456,000 gal) of solid waste with an estimated 193,000 L (51,000 gal) of drainable liquids (Hanlon 1994). The most recent surveillance data shows a discernable downward trend over the past 18 months, nearly a 3.8 cm drop in surface level, after a very long, gradual, but minor upward trend over the last ten years (see Figure 2-5). This recent behavior possibly indicates further settling, or leaking has taken place. These figures translate to a waste depth of 408.9 cm (161.1 in.) underneath the riser and 440.2 cm (173.3 in.) at the tank centerline. The tank has not been fully interim stabilized yet, and is an assumed leaker. The recent waste temperature in tank 241-T-111, taken from a thermocouple tree is approximately 16 °C (60.5 °F) (Rios 1994), and the estimated heat load in the tank is less than 2.93 kW (10,000 Btu/hr).

2.5 TANK STATUS

Tank 241-T-111 was a non-watch-list SST, with no historical indication of any potential safety issue. However, during the review of the energetics data that was done in support of characterization, exotherms in excess of -125 cal/dry gram of waste were noticed in the top 3 segments of core 31 and the top 2 segments of core 33. Additional follow-up work has been done confirming the initial observation and as a result, 241-T-111 has been added to the Organic Tank Watch List (Wicks 1994). The exotherms were not predicted from the process history of the tank and the known characteristics of the 2C and 224 waste streams disposed there.

Figure 2-5. 241-T-111 Surveillance Data.

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3.0 TANK SAMPLING OVERVIEW

3.1 DESCRIPTION OF SAMPLING EVENT

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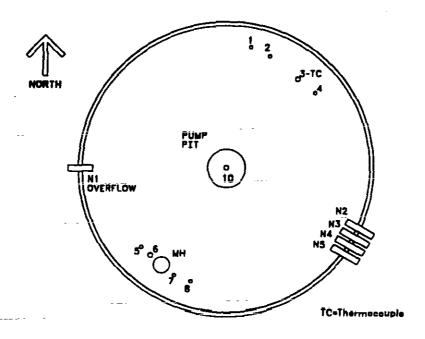
Tank 241-T-111 was push-mode core sampled through three risers during a period from October 22, 1991 to November 7, 1991. The core samples were obtained using a specially designed core sampling truck. Access to the interior of the tank is provided by various tank risers. Risers are pipes of various diameters leading into the tank dome from the ground. The riser configuration for tank 241-T-111 is given in Figure 3-1. A review of the tank farm operating records and a field inspection of the tank risers determine which risers can be used in the sampling operation. During sampling, a riser is opened and the truck is positioned over the riser. The sampler is lowered into the tank through the drill string and pushed into the waste. Further information regarding the core sampling operation can be found in Tank Farm Operating Procedure (Ross 1993). Nine segments were expected from each core sample. Each segment is approximately 48 cm (19 in) long. Core 31 was obtained from riser 6 on October 22, 1991. Core 32 was obtained from riser 2 from October 24, 1991 to October 25, 1991. Core 33 was obtained from riser 3 on November 5, 1991 to November 7, 1991.

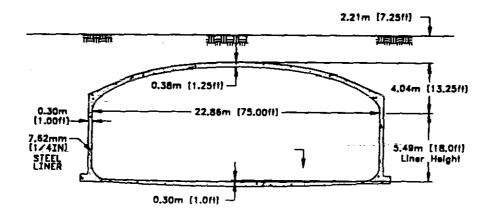
The sampler is constructed of stainless steel and is 48 cm (19 in) long, with a 2.2-cm (7/8-in) inside diameter, and has a volume of 187 mL (0.05 gal). A hydrostatic fluid of normal paraffin hydrocarbons, similar to kerosene, was used in establishing a head balance while taking these cores. Objections involving sample degradation and contamination have been raised regarding the use of this fluid, and the practice has since been discontinued. However, for cores 31 and 33, nearly full recovery was achieved in every case. There were little or no drainable liquids observed in the sample liners or in the samplers upon extrusion of the samples, and although hydraulic permeability measurements were not taken as part of the characterization effort, the waste did not appear porous. Thus, sample contamination from the hydrostatic fluid is not deemed to be a significant issue with the analysis of the sample-or the interpretation of the results.

The casks were transported to the 222-S Laboratory for characterization analysis. This facility is operated by Westinghouse Hanford Company in the 200 West Area of the Hanford Site. Further physical and radiochemical characterization was performed at the 325 Laboratory. Specific analyses aimed at identifying and resolving the unusual energetics observed (Bean 1994) are also being done at 325 Laboratory. That facility is operated by Battelle Pacific Northwest Laboratory (PNL), and is located in the 300 Area of the Hanford Site.

Before the most recent stabilization effort, a series of liquid grab samples were taken and analyzed to ensure waste compatibility with the tank receiving the 241-T-111 liquid waste. These grab samples were acquired using a sampling method known as "bottle-on-a-string," in which a weighted bottle with a shuttered mouth is lowered into the liquid waste. The bottle's

Figure 3-1. Tank 241-T-111 Riser Configuration.





WHC-EP-0806

mouth is opened, it fills with liquid, the mouth is closed, and the sample is retrieved and transported to the 222-S Laboratory. Winters et al. (1990a and 1990b) has a more detailed description of this sample method.

3.2 CHAIN OF CUSTODY

A chain-of-custody record was kept during the sampling event for each segment that was sampled. The chain-of-custody form is a one-page record that is used to ensure that (1) the sample is safely and properly transported from the field to the laboratory, and (2) the correct personnel are involved in the sampling operation and transportation of the sample to the laboratory.

One of the additional functions of the chain-of-custody record is to provide radiation survey data. This is a record of the radiation dose that is emitted from the shipping cask. The dose rates in mrem/hour are measured from the top, sides, and bottom of the cask. These values are recorded on the chain-of-custody record and represent the radiation being emitted directly from the sample. The last item recorded under the radiation survey data is the smearable contamination. Smearable contamination represents the radiation from waste material that is not sealed within the shipping cask; values greater than 100 mrem/hour are considered unsafe. Measurements are made both in the field and in the laboratory. No smearable contamination was found with these samples.

The chain-of-custody has several other important functions: (1) to provide a modest description of the cask, sampler, and the expected contents of the sampler (shipment, sample, and cask serial numbers for the specific sampling event); (2) to provide summary information about the analytical suite that the sample will undergo or reference the salient documentation; (3) to provide traceability for the sample during transport; and (4) to ensure sample integrity on arrival at the laboratory. This information is provided to ensure that each sample can be uniquely identified. A summary of the most pertinent data contained in the chain-of-custody forms for the tank 241-T-111 samples is presented in Tables 3-1 and 3-2.

Copies of the chain-of-custody forms are available in the full data package and through Hanford-Site Central Files. From inspection of the chain-of-custody records, there appear to be irregularities in the sampling and transport of tank 241-T-111 samples. Valve failures were observed in individual segments in all three core samples, and core 32 was considered to be completely compromised and non-representative. Each segment was almost entirely aqueous, containing at most a small amount of suspended solids. Liquid was also found in some of the liners surrounding the samplers from core 32, and was assumed to be leakage from the sampler. These irregularities merit a sampling concern, sample integrity concern, and potential safety concern (i.e., sample containment was compromised). However, the double-containment strategy employed in the handling of the samples was successful in preventing any excessive radiological exposure to personnel and no material escaped confinement. Further investigation and refinement of the sampling process, procedures, and sampler design is in progress.

Table 3-1. Core 31--Chain of Custody Summary.

Sample	Core 31	
Place Taken	241-T-111 Riser 6]
Date Taken	10/22/91	
Date Released	10/23/91	
Time Released	9:35 P.M.	
Sender	D. C. Hartley]
Receiver	V. Johansen	
Place Received	222-S Laboratory]
Time Received	10:00 P.M.	
Sample Number	Smearable Contamination	Dose Rate Through the Drill String
91-090 (Segment 1)	< DL alpha < DL beta-gamma	4.5 mR/hr
91-091 (Segment 2)	< DL alpha < DL beta-gamma	2.4 mR/hr
91-092 (Segment 3)	< DL alpha < DL beta-gamma	2.5 mR/hr
91-093 (Segment 4)	< DL alpha < DL beta-gamma	2.0 mR/hr
91-094 (Segment 5)	< DL alpha < DL beta-gamma	1.5 mR/hr
91-095 (Segment 6)	< DL alpha < DL beta-gamma	1.5 mR/hr
91-096 (Segment 7)	< DL alpha < DL beta-gamma	0.5 mR/hr
91-097 (Segment 8)	< DL alpha < DL beta-gamma	1.5 mR/hr
91-098 (Segment 9)	< DL alpha < DL beta-gamma	0.3 mR/hr

< DL = below detection limit

Table 3-2. Core 33Chain of Custody S	Summary
--------------------------------------	---------

Sample	Core 33		7
Place Taken	241-T-111 Riser 3	-	1
Dates Taken	11/5/91 to 11/6/9	1	1
Dates Released	11/6/91 to 11/7/9	1	1
Sender	D. C. Hartley		†
Receiver	V. Johansen	· · · · · · · · · · · · · · · · · · ·	4
Place Received	222-S Laboratory	 	
Times Received	See Below	-	
Sample Number/ Date Sampled	Date Released/ Time Released/ Time Received	Smearable Contamination	Dose Rate Through the Drill String
91-108 (Segment 1)	11/6/91 10:10	< DL alpha < DL beta-gamma	3 mR/hr
11/5/91	10:55	1 22 oom gamma	
91-109 (Segment 2) 11/5/91	11/6/91 10:10 10:55	< DL alpha < DL beta-gamma	2.5 mR/hr
91-110 (Segment 3) 11/5/91	11/6/91 10:10 10:55	< DL alpha < DL beta-gamma	10 mR/hr
91-111 (Segment 4) 11/5/91	11/6/91 14:10 14:30	< DL alpha < DL beta-gamma	5 mR/hr
91-112 (Segment 5) 11/6/91	11/6/91 14:10 14:30	< DL alpha < DL beta-gamma	< 0.5 mR/hr
91-113 (Segment 6) 11/6/91	11/6/91 14:10 14:30	< DL alpha < DL beta-gamma	2 mR/hr
91-114 (Segment 7) 11/6/91	11/7/91 10:10 10:35	< DL alpha < DL beta-gamma	1.5 mR/hr
91-115 (Segment 8) 11/6/91	11/7/91 10:10 10:35	< DL alpha < DL beta-gamma	1 mR/hr
91-116 (Segment 9) 11/6/91	11/7/91 10:10 10:35	< DL alpha < DL beta-gamma	l mR/hr

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4.0-SAMPLE HANDLING AND ANALYTICAL SCHEME

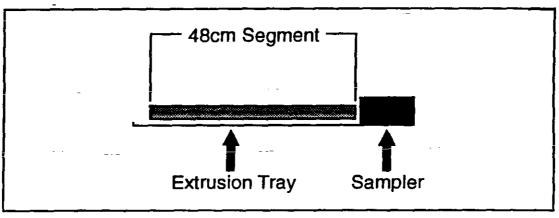
The primary objective for these waste analyses was to meet requirements of the Hanford Federal Facility Agreement and Consent Order (Ecology et al. 1992, Ecology et al. 1994). Two composites from each core were built and analyzed in accordance with the complete baseline case core composite scenario detailed in WHC-EP-0210 (Winters et al. 1990a, Winters et al. 1990b) and as amended by Hill et al. (1991). However, there are other concerns and interests for this data. First, this information will help to evaluate whether constituent concentrations are within safe operating limits by determining whether they are flammable or explosive. Second, analyte concentrations of interest to the various Hanford-Site technical programs can be determined.

Because tank 241-T-111 initially was identified as a non-watch-list tank, extensive analytical measurements were not specified to resolve any previously identified safety concerns associated with this tank. The analysis horizon for characterization was determined to be on the core composite level with selected analyses being performed on a segment-level basis. However, after the discovery of substantial exotherms in the top 100 cm (40 in) of the waste, additional testing on a segment level basis was done on those samples exhibiting reactive behavior. This additional testing, with an emphasis on providing resolution to the safety issues raised by the presence of the exotherms in the waste is still underway at the time of the writing of this report. This report will be updated to reflect the new data or root causes of the energetics when that information becomes available.

4.1 SAMPLE BREAKDOWN PROCEDURE

In order to obtain the sample, the sampler is shipped in a vertical position and removed from the shipping cask directly into the hot cell. At this time, the sampler must be placed in the horizontal position. The sample is then loaded into the mechanical extruder and removed by pushing it out from the back of the sampler with a piston. In this case, the sampler is pressed against a fixed piston, forcing the sample into the extrusion tray. If a full sample is captured, the material nearest the valve will be from a deeper part of the tank. The material near the piston is closer to the surface. The sample and any liquids are collected on a metal tray. Next, the mass of the segment and the approximate length are recorded. From this information, the gross bulk densities of the segments can be estimated until further physical properties work is performed. The sample volume is determined by measuring the length of the extruded sample using a linear unit volume of 9.85 mL/in, Figure 4-1 illustrates how the SST segment sample was extruded. Color photographs documenting the extrusions of each of the segments from tank 241-T-111 were taken and are on file at the 222-S Laboratory.

Figure 4-1. Typical Single-Shell Tank Segment Extrusion.



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Several different styles of nomenclature presently are used for distinguishing core samples, sample segments, and subsegments in the existing literature. Two major conventions are used in the documentation relating to core sampling in general. The first is designating the segment with the last two digits of the calendar year (92-) and then numbering the segments sequentially (-001, -002, etc.). This system resets itself every calendar year. The second system distinguishes the tank, core, segment (and subsegment, if necessary), with segment 1 being the at the top of the core sample and having the number increase as a function of depth in the tank so that segment 9 is at the bottom of the tank in 241-T-111. An example of this naming protocol for the second segment of the first core is 241-T-111-Core 31-Segment 2. Where no tank identification is given in this report, it is understood to mean tank 241-T-111.

4.2 TANK 241-T-111 CORE SAMPLE DESCRIPTION

The location of the risers, the dished bottom of the tank, and safety margins in the sampling protocol preclude obtaining samples from the entire waste depth in the tank. In addition, the sampling protocol establishes that segments will be calculated from the bottom up. Thus, depending on the waste depth, maximum recovery for the top segment from tank 241-T-111 is not necessarily going to be a full 48-cm (19-in) segment. However, for cores 31 and 33, sample recovery was excellent; overall recoveries were in excess of 80 percent. Segment recoveries were based on the maximum recoverable volume for the segment regardless of solid/liquid ratio. The core recoveries reported in the data package are determined based on a visual inspection of the sample length and apparent volume at the time the samples are extruded. Further study of the color photographs taken after extrusion can aid in clarifying a rough sample volume. Tables 4-1 and 4-2 present the initial measurements and observations regarding the core samples on extrusion, and an estimated range of the core recovery on a volume basis for cores 31 and 33.

Although samples for core 32 were taken from riser 2, the materials obtained at all levels appeared to be particulate suspended in an aqueous solution, with slight traces of normal paraffin-hydrocarbon contamination observed in a few samples. These samples did not correspond to the observed conditions in the tank and were considered non-representative.

The results of the core-32 sampling exercise were attributed to sampler failure, and because no acceptable samples were acquired, no assays were performed. Therefore, no results for core 32 will be reported. Valve failures were reported routinely for all three core samples at deeper positions in the tank. The full data package (McKinney et al. 1993) containing all of the assay results is available from the Hanford-Site Central Files.

General characteristics of tank 241-T-111 waste materials are as follows.

- Very little drainable liquid was associated with these samples either in the liner or in the extruder.
- Core samples generally were dark brown or black in color. The brown solids were streaked through with grey/white material.

Core number (Riser 6)	Segment	Core recovery (Vol. basis)	Total mass (g)	Comments
31	1	27%	64.0	Sampler was nearly empty; contained approximately 50 mL of black/brown low viscosity solids. Apparently homogeneous.
31	2	80-100%	182.8	Sampler was almost completely filled with solids. Again, the material is dark brown or black with a fluid or gel-like consistency, and appears to be homogeneous. A small amount of liner liquid was observed. The liquid was observed to be two phase (NPH and aqueous phases).
31	3	95-100%	162.2	Sampler: was almost completely filled with solids. The waste is dark brown with a thick, viscous consistency, and appears to be completely homogeneous.
31	4	80-100%	153.5	Sampler was almost completely filled with solids. The waste is dark brown with a thick, viscous consistency, and appears to be completely homogeneous. The top eighth contained waste material that appeared to be more fluid than the rest of the sample. No sampler liquid or liner liquid was observed.
31	5	100%	190.9	Similar to previous observations; no sampler or liner liquid.
31	6	0%	NA	Sampler empty.
31	7	90-100%	186.4	Sampler was almost completely filled with solids. The waste is dark brown with a thick, viscous consistency, and appears to be completely homogeneous.
31	8	100%	186.4	Similar to previous observations; no sampler or liner liquid.
31	9	100%	203.1	Sample was not homogeneous. Sample began as before (dark brown and viscous), but gradually became lighter as a function of depth. Sample was divided into two portions, a light end (133.4 g) and a dark end (69.7 g). Consistency of the sample remained the same throughout.

4-4

Table

4-2.

Tank

241-T-111

Core

33

Sample

Description

Summary

• The samples had a viscous or gel-like consistency. They were thick, relatively smooth sludges (swamp mud was the descriptive term used by the hot-cell ----observer). The core materials all appeared to be saturated with liquid, which did not drain.

4.3 HOLD TIME CONSIDERATIONS

All analytes have a predetermined maximum allowable holding time set by the Environmental Protection Agency (EPA 1986), during which the analysis should be completed. Completion of analysis during the maximum allowable holding time enhances the regulatory defensibility of the data. The length of the holding time varies for each analyte. For example, analyses performed on volatile and semivolatile organic compounds, many of which decompose or dissipate quickly, have shorter holding times. On the other hand, persistent analytes such as metals (except mercury), do not readily decompose or dissipate, and therefore have much longer acceptable holding times. Nearly all of the analyses of cores 31 and 33 exceeded their respective maximum allowable holding times. The only analyses that came close to meeting holding-time criteria were radiochemistry and metal analyses. Both of these analyses were completed about six months after sampling, and six months is the maximum hold time for these analyses. Although exceeding the maximum allowable holding times weakens the defensibility of the analytical results for some uses, it is anticipated that the overall effect on the analytical results for tank 241-T-111 waste relative to waste management and disposal information needs is minimal. Further discussion of holding times can be found in Winters et al. (1990a).

4.4 SAMPLE PREPARATION

Sample preparation procedures are conducted in order to optimize the recovery of each analyte of interest from the tank waste. Water digestion, acid digestion, and potassium hydroxide fusion commonly are used to extract metals and several radioisotopes from solid samples, and in some cases digestions are performed on liquid samples to improve analytical resolution. Many separations are specific to a particular analysis and are described within the corresponding analytical methods referenced in Section 4.5. In order to verify analyte recoveries resulting from separation techniques, laboratory control samples, carriers, spikes, tracers, and surrogates are analyzed concurrently with the characterization samples.

In-some-cases no sample preparation is necessary or desired. Direct analyses are assays performed on the sample matrix with little or no sample preparation. Several direct analyses were performed relating to the physical or energetic properties of the waste: density, thermogravimetric analysis (TGA), differential scanning calorimetry (DSC), and gravimetric weight percent water.

Water leach (or water digestion) analyses are assays performed after the sample matrix has been digested in distilled/deionized water. The water is then analyzed for soluble analytes. The soluble anions are determined by ion chromatography (IC). The primary anions analyzed in this manner are fluoride, chloride, nitrate, nitrite, phosphate, and sulfate. In addition, free cyanide and pH were also analyzed from water digestion samples. Note that IC assays use a 1 to 100 sample to water digestion, where pH measurements use a 1 to 1 sample to water ratio. Selected radionuclides were measured on some of the water digestion samples to determine the type and number of water soluble radionuclides. Inductively coupled plasma-atomic emission spectroscopy (ICP) and atomic absorption (AA) spectroscopy were also performed on some of the water digestion samples. These assays were performed to determine the amount of soluble metal cations (ICP) or arsenic, mercury, or selenium (AA). In many cases, these analytes were below the detection limits in the water digestion samples, suggesting that most of the analytes are not water soluble.

Acid digestion is a preparation method in which the sample is dissolved in a mixture of nitric and hydrochloric acids. This preparation brings most of the insoluble metals into a solution with a minimum amount of dilution, and usually is best for the detection of trace and some major metals. These properties are the reason that acid digestion generally is used as the sample preparation for the homogenization tests in SW-846-based environmental sampling. However, experience with Hanford-Site tank-waste matrices has shown that acid digestion does not always provide complete solubilization, and that a more rigorous dissolution preparation, such as fusion, may be necessary to get adequate quantitation. The analyses performed on this preparation were the ICP, gamma energy analysis (GEA), and AA analysis (the AA analysis used nitric acid only). IC analysis was not performed with the acid digestion preparation solution because that method introduces additional Cl or NO₃ anions, confounding the results for that sample.

Analyses that were performed on fusion-prepared samples were ICP and GEA for radionuclides. Fusion dissolution analyses are assays performed on the sample matrix after it has been fused with potassium hydroxide in a nickel crucible and dissolved in acid. This preparation dissolves the entire sample, whereas other sample preparation procedures may not completely dissolve the sample matrix. However, one significant disadvantage of fusion preparation is that large amounts of potassium hydroxide are required to bring a sample into solution. Because of this high dilution factor, trace elements are less likely to be quantified correctly, if they are detected at all. Another limitation of the preparation method is if the sample contains substantial quantities of potassium or nickel, these analytes will not be quantifiable because the procedure uses potassium hydroxide and a nickel crucible. This limitation can be overcome using alternate preparation methods if potassium or nickel are analytes critical to interpretation of the data. Elements that occur in abundance, such as major metals, or that are highly insoluble are likely to be detected better by the fusion results than by any other sample preparation.

Generally, fusion dissolution is the preferred method of analyzing radionuclide content, with the exception of ¹⁴C, ¹²⁹I, and ³H (tritium). However, the sample preparation specified in the test instructions for ¹⁴C (water digestion) probably is not best for the high-level waste

matrices. Difficulty with dissolving the sample with a water leach, and volatility associated with a fusion preparation, potentially will bias the ¹⁴C results low for both sample preparation types if they are associated with the water insoluble solid materials, and similar difficulties are encountered for the other radionuclides mentioned. However, none of these analytes are expected to be significant contributors to the radionuclide content of the waste.

Major metal components that were detected well with fusion ICP analysis for tank 241-T-111 were calcium, chromium, iron, manganese, sodium, bismuth, and lanthanum. Phosphorous, sulfur, and silicon are non-metallic analytes detected by ICP. In the case of these elements, the fusion result is the preferred method of analysis because it is believed to provide more complete dissolution of the waste, and therefore, more complete quantitation of the analytes. Comparisons of these results with the IC results can provide insight to the solubility characteristics of the waste. Some of the primary radionuclides that are measured using this sample preparation are ²³⁷Np, ^{239/240}Pu, ⁹⁰Sr, ¹³⁷Cs, and ⁹⁹Tc. A total alpha and total beta count were performed on the fusion dissolution samples as well.

As previously noted, chemical and radiological analyses were done largely on core composites, and in these previous characterization efforts, the core composites were built using quantities of segments based on a proportion of the total weight of sample for the core (Winters et al. 1990a, Winters et al. 1990b). This method assumed that the sample obtained is representative of what is in the tank. However, when partially filled segments are obtained, this procedure assumes that the tank does not contain any waste in this area. Incomplete recovery for a segment probably is the result of sampling problems rather than voids in the waste. The approach used in this analysis effort was to composite equal quantities of the homogenized segment material and assume that whatever is obtained in a partial segment is representative of a whole segment. Some inaccuracies may be introduced from this method because of density differences between segments. However, the inaccuracies introduced from density differences probably are small. In general, those deviations are minimal compared to the other errors inherent in core sampling and analysis. If full segments are obtained for the entire core, and the homogenization procedure is satisfactory, there will be little difference between the two approaches.

4.5 ANALYTICAL METHODS

This section briefly describes the analyses used to characterize the waste in tank 241-T-111. The analyses were split between the Westinghouse 222-S Laboratory and PNL. Several of the analytical tests performed on the composites were also performed on the segments, but on a much more limited scope. There were no free liquids from cores 31 and 33, thus there are no separate liquid core composite results. However, in March 1994, a series of liquid grab samples were taken and analyzed for compatibility considerations (Carothers et al. 1994) before the start of stabilization. These results will be presented with the water digestion results and comparisons and conclusions will be made.

4.5.1 Physical and Rheological Tests

Physical tests completed at the 222-S Laboratory included particle size analysis, TGA, DSC, specific gravity, and percent water analyses. Duplicates were performed for the percent water analyses. The physical properties measured at PNL included weight percent solids, settling behavior, and weight percent dissolved solids. Rheological testing on these samples were performed at PNL and included shear strength and shear stress as a function of shear rate. Rheological properties were measured in duplicate. Table 4-3 lists the analytical methods used for physical and rheological testing. Three segments from Core 31 (segments 2, 4, and 8) were selected to perform the full suite of rheological and physical measurements, in addition to the particle size assay done on each segment. Viscosity, settling properties, fluid behavior, and shear strength were some of the primary characteristics investigated. The samples tested for these properties were not homogenized before analysis.

Table 4-3. Analytical Methods for Physical and Rheological Testing.

Analyte	Procedure
Particle size	T044-A-01712F
Thermogravimetric analysis	LA-560-112
Differential scanning calorimetry	LA-514-113
Specific gravity	LA-510-112
Percent water:	LA-564-101
Rheology	PNL-ALO-501 PNL-ALO-502
Physical properties	N/A

Scanning TGA and DSC are useful in determining the thermal stability or reactivity of a material. TGA measures the mass of a sample while the temperature of the sample is increased at a constant rate. In DSC analysis, the heat absorbed/evolved over and above the usual heat capacity of the substance is measured while the substance is exposed to a linear increase in temperature. The gravimetric weight percent water was determined by drying the sample for 12 to 24 hours in an oven at 103 to 105 °C and measuring the difference in the weight of the sample.

4.5.2 Chemical and Radionuclide Constituent Analysis

Most of the chemical and radionuclide analyses were performed at the 222-S Laboratory. The uranium and plutonium isotopic analyses, however, were performed at PNL. Duplicate analyses were performed on every tank sample. Table 4-4 lists the analytical methods used (Winters et al. 1990a, Winters et al. 1990b).

Sample Homogenization

The segment and core composite samples were homogenized using a mechanical mixer before analysis. This was done so that aliquots removed for analysis would be representative of the entire segment or core composite. Aliquots of the homogenized tank waste from core 33, segments 1, 3, 5, 7, and 9, were taken to determine the efficacy of the homogenization procedure. The samples were split into duplicates, acid digested, and assayed by ICP and GEA. This procedure is done to determine if the degree of mixing achieved by the as-planned homogenization procedure was sufficient to achieve sample homogeneity. Since the homogenization samples are evaluated concurrently or after the other core samples, the results provide only an estimate of subsampling error (or variation). They were not used in this case to ensure that homogenization was achieved before analysis. However, after review of the results, it appears that homogenization of the samples was satisfactory.

4.5.3 Organic Constituent Analyses

All organic analyses of the samples from tank 241-T-111 were performed at PNL. An EPA contract-laboratory-procedure-type organics-speciation analysis was performed on the core composites. No levels of organic compounds above the contract required quantitation limit were found in any of the samples, and they were not expected to contribute to the sample matrix. The organic analyses performed were volatile organic analyses, semi-volatile organic analyses, total organic halides, and extractable organic halides. Duplicates were performed for all of these analyses. Table 4-5 lists the analyses and procedure numbers.

At the 222-S Laboratory, the initial total organic carbon assays were done using a furnace oxidation procedure of a water digested sample. At PNL, the total organic carbon content for the solids was determined using the hot persulfate method. That method dissolves a sample in a 90 °C+ sulfuric acid solution to liberate inorganic carbon (carbonate). $K_2S_2O_8$ is then added, and organic carbon is converted to CO_2 , which is measured coulometrically. As stated in the "Executive Summary," these methods did not provide satisfactory results for the tank 241-T-111 waste samples. Later, furnace oxidation tests done at PNL gave results much higher than those from the 222-S Laboratory and more in line with the observed exothermic activity.

Table 4-4. Analytical Methods for Chemical and Radionuclide Analyses.

Analyte	Method	Procedure number
Hg	Cold vapor atomic absorption	LA-325-102
F-, Cl-, NO ₃ -, NO ₂ -, PO ₄ -, SO ₄ -	Ion chromatography	LA-533-105
		
CN ⁻	Distillation/spectrometric analysis	LA-695-101 LA-695-102
Ū	Laser fluorimetry	LA-925-106
Total Alpha	Proportional counting	LA-508-101
Total Beta	Troportional counting	LA-306-101
²³⁸ Pu, ^{239/240} Pu, ²⁴¹ Am	Alpha spectrometry	LA-503-156
²³⁷ Np	Alpha proportional counting	LA-933-141
Total Cations	Inductively coupled plasma	LA-505-151
⁹⁰ Sr	Beta proportional counting	LA-220-101
⁹⁹ Tc ⁷⁹ Se ¹⁴ C ——— ³ H	Liquid scintillation	LA-438-101 LA-365-132 LA-348-104 LA-218-114
129 <u>I</u> 59Ni	Low energy gamma analysis	LA-378-104 PNL-ALO-464
⁶³ Ni	Liquid scintillation	PNL-ALO-474
¹⁵⁴ Eu, ¹⁵⁵ Eu, ²⁴¹ Am, ¹³⁷ Cs, ⁶⁰ Co	Gamma energy analysis	LA-548-121
NO ₂ -	Spectrophotometry	LA-645-001
H+	рН	LA-212-103
As Se	Graphite furnace atomic absorption	PNL-ALO-214 PNL-ALO-215
Pu Isotopic	Fusion mass spectrometry	PNL-ALO-423 PNL-MA-597
U Isotopic	Mass spectrometry uranium laser	PNL-MA-597 PNL-ALO-445
TOC	Total organic carbon	LA-344-105 PNL-ALO-380 PNL-ALO-381
CO ₃ /C	Total inorganic carbon	LA-622-102
· · · · · · · · · · · · · · · · · · ·		

Table 4-5. Analytical Methods For Organic Analytes.

-Analysis	Method	Procedure Number PNL-ALO-335	
Volatile organic analysis	Gas chromatography/mass spectrometry		
Semi-volatile organic analysis	Gas chromatography/mass spectrometry	PNL-ALO-345	
Extractable organic halides	Microcoulometric titration	PNL-ALO-320.2	
Total organic halides	Microcoulometric titration	PNL-ALO-321	

4.5.4 Segment-Level Analyses

The objectives of segment-level analyses are to provide (1) information as a function of depth pertaining to the overall waste energetics (water content and chemical reactivity) and (2) the particle size distribution and other general rheological information. To accomplish these goals, the limited suite of analyses listed in Table 4-6 were performed on each homogenized segment. These analyses were conducted using the analytical procedures identified in Tables I5-1 and I5-2 of WHC-EP-0210, Rev. 3 (Hill et al. 1991). In addition, where appropriate, the information obtained from the segment-level homogenization tests will be used to enhance the interpretation of the data.

Table 4-6. Segment-Level Analysis.

Direct	Acid dissolution*
Thermogravimetric analysis differential scanning calorimetry Wt% H ₂ O particle size**	Inductively coupled plasma (metals) gamma energy analysis (137Cs) total alpha

^{*}Acid dissolution assays were performed on the homogenization test segments.

^{**}Particle size was done on non-homogenized segment material.

5.0 ANALYTICAL RESULTS: TANK 241-T-111

5.1 CHEMICAL ANALYSES

5.1.1 Analytical Results: Inductively Coupled Plasma-Atomic Emission Spectroscopy

Online inter-element corrections were performed for matrix interferences. The ICP has a built-in correction capability to adjust for moderate matrix interferences; however, there may be performance degradation on samples containing weight-percent quantities of iron, aluminum, or uranium. Corrections were made for moderate levels of aluminum, calcium, chromium, and magnesium in the samples. Corrections were made for high iron concentrations as well. Process or preparation blank values have not been subtracted from the results. In the water digestion and liquid grab sample assays, the single most prevalent element is sodium by at least an order of magnitude. Relative percent differences (RPDs) for water digestion results were high for some analytes (i.e. above the 20 percent acceptance criteria), but there was no consistent trend observed between cores-31 and 33.

In the fusion assays, some elements can appear to be at high concentrations because of the large dilution factors required for fusion samples. These high dilution factors propagate errors. Those analytes actually may be present only in concentrations marginally above the detection limit. For several analytes, higher quantitation was found in the acid digestion results. For purposes of determining inventories and making comparisons, the highest reliable average analytical result will be used between acid and fusion preparations.

In reviewing the data, a subtle bias or gradient was observed between the results for cores 31 and 33, with core 31 having slightly higher values overall than core 33. Although not readily discernable among the first tier analytes (sodium, bismuth, iron, and phosphorus), the difference between cores 31 and 33 is more evident in the second tier analytes (aluminum, lead, and magnesium). The RPDs between the individual core composite samples and their replicates were small, suggesting that the gradient is real and not an analytical artifact, however, this observation could be the result of the compositing procedure. There were nine segments used in the core 33 composites, but only eight segments in the core 31 composites (segment 6 was absent in core 31). Furthermore, sample variability may contribute or be wholly responsible for the observed difference.

RPDs for most-elements were within the 20-percent acceptance criteria for acid and fusion results, and generally were less than 10 percent for the major analytes. Potential sample contamination for boron and silicon exists because of the caustic nature of the samples and the glass vials used to store the samples in the laboratory; however, silicon routinely demonstrated a low bias. Calcium and magnesium for blank and spike recovery results may have been biased high by the powder used on the analysts' gloves when performing the assays. With the small sample sizes used in the assay, even trace amounts of powder have

the potential to impact the analyses. Low spike recoveries were noted for several analytes for differing reasons. Silver recoveries are commonly low because of the precipitation of silver chloride in the sample digestion. Poor spike recoveries of iron, magnesium, and calcium accompany high preparation blank results, but the overall correlation is poor. Spike failures frequently are noted for major elements when the spike concentration is insignificant compared to the analyte concentration in the waste matrix. Spike and standard results outside the acceptance criteria for these analytes do not necessarily invalidate the sample results for the ICP in general, or for those analytes in particular. Individual analyte failures need to be evaluated on a case by case basis. All of these behaviors could affect, and are considered in the interpretation of the results.

The detection limit for each analyte is provided for comparison with the results to aid in interpretation. All ICP analytes are reported in the data tables; however, those consistently contributing significant (i.e generally greater than approximately 0.2 weight percent) amounts to the composition of the waste matrix generally are relevant to bulk characterization. The fusion/acid ratio, which can be compared in the table, for most analytes indicates near total dissolution for the acid digestion assay. Average values for the analytes are reported to three significant figures. The full range of ICP analytical results can be found in the full data packages (McKinney et al. 1993). All reported concentration values are based on grams of wet sample, unless otherwise specified. Table 5-1 provides ICP analyte concentration information on the core composites as a function of the sample preparation. Table 5-2 provides ICP analyte concentrations as a function of depth for Core 33. Table 5-3 compares the water digestion core composite ICP values with the results from the grab sample.

Core 31

The most significant analytes measured by the water digestion of the core composite were sodium and phosphorous (probably as a soluble phosphate), and sulfur (probably as a soluble sulfate). Much smaller amounts of iron, chromium, and silicon also were measured. RPDs were elevated for these samples (between 10 and 20 percent), but were generally within the 20-percent acceptance criteria. For analytes with results outside the acceptance criteria, no clear trend between the two composites can be established. The water digestion result for core 31, composite 1 had several analytes with high RPDs, many of which are significant contributors to the waste: iron, lead, manganese, silver, bismuth, lanthanum, and strontium. Core 31, composite 2 had few analytes with results outside the acceptance criteria: barium, calcium, lead, and vanadium. All of the analytes noted are largely insoluble and that characteristic probably is contributing to the observed variability.

The results from the acid digestion preparation of the core composite samples had sodium, phosphorous, sulfur, calcium, chromium, iron, manganese, bismuth, and lanthanum as major analytes. RPDs are very good in general for most of the major analytes (generally between 5 and 10 percent), well within the 20 percent acceptance criteria. Analytes outside the acceptance criteria for core 31, composite 1 were antimony and boron (56 percent and 27 percent, respectively). However, neither of those analytes are substantial contributors to the waste matrix, and for analytes near the detection limit, reproducibility is not expected.

Table 5-1. ICP Cations. (4 pages)

Analyte	Acid-water digestion	Fusion digestion	Prep type	Core 31 comp. 1	Core 31 comp. 2	Core 33 comp. 1	Core 33 comp. 2
	Detection limit	Detection limit		Average concentration	Average concentration	Average concentration	Average concentration
	(μg/g)	(μg/g)		(μg/g)	(μg/g)	(μg/g)	(μg/g)
Al	2.4		Water	6.71	10.3	15.5	11.0
:			Acid	584	705	472	405
		12.0	Fusion	644	693	484	459
Sb	i7.7	· · · - · · · · · · · · · · · · · · · · · · ·	Water	17.7	17.7	17.7	17.7
ŧ			Acid	30.4	36.5	35.9	22.6
		85.5	Fusion	88.6	88.3	109	88.4
As	3.0	_	Water	3.0	3.0	2.99	3.0
		-	Acid	3.15	2.92	3.45	3.06
	3.3		GFAA ¹	3.3	3.3	3.3	3.2
a.a		15.0	Fusion	- 15.0	15.0	15.0	15.0
Ва	.3	-	Water	0.305	0.416	0.547	0.516
	··· <u>·</u> . (Acid -	57.0	64.9	66.8	87.3
-	_	1.5	Fusion	58.8	60.6	65.4	73.7
Be	0.1		Water	0.10	0.10	0.10	0.10
			Acid	0.105	0.097	0.117	0.104
		0.5	Fusion	0.501	0.499	0.499	0.499
Cd	.4		Water	0.40	0.40	0.4	0.4
			Acid	7.22	7.86	4.4	3.72
		2.0	Fusion	8.25	10.7	6.42	7.17
Ça	4.4 . [Water -	50.8	61.6	66.4	67.5
	. <u>.</u>	ļ	Acid	2,200	2,480	1,490	1,350
	-	22.0	Fusion	2,760	2,660	2,220	2,050
Ст	0.9		Water	209	229	224	211
			Acid	1,860	1,840 -	2,060	2,140
		4.5	Fusion	1,890	1,700	1,790	1,820

¹GFAA: Graphite furnace atomic absorption

Table 5-1. ICP Cations. (4 pages)

Analyte	Acid-water digestion	Fusion digestion	Prep type	Core 31 comp. 1	Core 31 comp. 2	Core 33 comp. 1	Core 33 comp. 2
	Detection limit	Detection limit		Average concentration	Average concentration	Average concentration	Average concentration
	(μg/g)	(μg/g)		(μg/g)	(μg/g)	(μg/g)	(μg/g)
Co	0.8		Water	0.8	0.825	0.821	0.825
			Acid	3.4	7.16	3.13	2.91
-		. 4.0	Fusion	10.1	10.8	13.3	11.8
Сu	.4		Water	0.4	0.4	0.4	0.4
			Acid	25.2	79.6	16.4	13.0
		2.0	Fusion	36.3	34.2	22.1	24.6
Fe	1.0		Water	79.6	140	132	159
·			Aeid -	19,200	20,000	17,500	17,300
		5.0	Fusion	20,500	19,600	15,900	16,100
Pb	6.2		Water	7.05	7.93	6.24	6.19
			Acid	475	543	201	168
		31.0	Fusion	440	484	267	269
Mg	0.3		Water	2.94	3.95	3.84	3.83
			Acid	435	479	305	290
		1.5	Fusion	438	443	268	272
Mn	0.2		Water	14.7	25.1	25.4	33.8
			Acid	6,190	6,140	6,710	6,280
	_	1.0	Fusion	6,380	5,940	6,220	6,590
Nι	1.7		Water	1.7	1.7	1.7	1.7
			Acid	151	157	110	108
		8.5	Fusion	. NR	- NR	^NR	NR
K	11.2		Water	734	783	712	648
			Acid	1,100	1,210	1,210	1,020
		56.0	Fusion	NR	NR	NR	NR
Se	7.6		Water	7.6	8.01	7.58	7.59
			Acid	7.98	7.4	10.3	7.79
	1.5		GFAA	1.5	1.5	1.5	1.5
		38.0	Fusion	38.0	37.9	38.0	38.0

Table 5-1. ICP Cations. (4 pages)

Analyte	Acid-water digestion	Fusion digestion	Prep type	Core 31 comp. 1	Core 31 comp. 2	Core 33 comp. 1	Core 33 comp. 2
	Detection limit	Detection limit		Average concentration	Average concentration	Average concentration	Average concentration
	(π ā /ā)	(μg/g)		(μg/g)	(μg/g)	(μg/g)	(μg/g)
Ag	0.5		Water	0.639	1.16	0.553	0.499
			Acid	203	227	44.3	30
		2.5	Fusion	214	221	39.5	37.1
Na	3.1		Water	34,000	35,000	30,800	32,000
			_ Acid	37,600-	38,700	35,000	36,300
		15.5	Fusion	39,800	39,000	33,900	35,200
V	0.5		Water	0.5	0.707	0.5	0.8
-	-	-	Acid	12.7	21.4	13.9	9.99
		2.5	Fusion	12.1	16.5	15.3	14.7
Zn	0.3		Water	0.3	0.3	0.3	0.3
	i	,	Acid	79.4	101	44.2	35
		1.5	Fusion	104	106	105	110
Bi	7.5		Water	115	191	231	270
			Acid	23,600	23,300	28,500	28,400
	-	37.5	Fusion	20,900	20,100	26,500	26,700
В	0.6		Water	3.31	3.19	5.54	4.25
		ı	Acid	27.1	23.4	29.4	32.2
	-	3.0 .	Fusion	3.0	2.99	4,84	4.84
Ce	10.1		Water	10.1	10.1	10.1	10.1
		·	Acid	32.6	28.6	37.8	35.8
		50.5	Fusion	50.6	50.4	50.4	_50.4
La	1.4		Water	6.02	8.52	13.8	15.8
		į	Acid	3,720	3,620	4,640	4,890
		7.0	Fusion	3,690	3,410	4,510	4,810
P	5.8		. Water	5,760	5,960	···· 5,300 ···	- 5,7 00
-			Acid	10,100	9,960	9,860	11,300
		29	Fusion	11,600	11,100	9,070	9,910
Si	1.3		Water	438	560	669	620
ĺ	į	į	Acid	482	471	528	394
		6.5	Fusion	5,960	5,840	5,460	5,410

Table 5-1. ICP Cations. (4 pages)

Analyte	Acid-water digestion	Fusion digestion	Prep type	Core 31 comp. 1	Core 31 comp. 2	Core 33 comp. 1	Core 33 comp. 2
	Detection limit	Detection limit		- Average- concentration	Average concentration	Average concentration	Áverage concentration
	(μg/g)	(μg/g)		(μg/g)	(μg/g)	(μg/g)	(μg/g)
Sr	0.3		Water	1.21	2.13	2.18	2.34
			Acid	282	280	305	334
		1.5	Fusion	3 <u>03</u>	280	291	317
S	- 2.7	-	Water	1,190	1,200	1,060	1,140
	•		Acid	1,230	1,260	1,140	1,220
		13.5	Fusion	1,350	1,310	1,080	1,160
Sn	1.6		Water	1.6	1.6	1.6	1.6
Ì		!	Acid	4.21	2.44	1.81	1.61
		8.0	Fusion	8.01	7.98	7.99	7.99
Ti	0.4		Water	0.4	0.4	0.4	0.4
		İ	Acid	29.4	33	8.9	6.46
		2.0	Fusion	72.9	72.4	22.3	24.1
Zr	NR		Water	0.8	0.799	0.798	0.798
		_	Acid	0.84	0.778	0.913	0.816
		NR	Fusion	4.00	3.99	4.00	4.00
Hg	0.125	NR	CVAA ²	1.59	1.83	1.20	1.08

NR = Not reported

Table 5-2. Tank 241-T-111 Core 33 ICP Selected Analyte Trending as a Function of Depth (Acid prep on segments).

Segment	Ca - (μg/g)	Cr (µg/g)	Fe (µg/g)	Mn (µg/g)	Ni (μg/g)	Na (μg/g)	Bi (μg/g)	La (µg/g)	P (μg/g)	Si (μg/g)	Տ (μg/g)
1 .	4,800	490	17,300	23,900	240	22,400	760	3	4,400	380	800
3	1,100	1,200	11,700	3.800	70	25,800	24,200	3,900	4,800	430	720
5	1,300	2,000	16,100	2,800	90	32,100	33,400	5,100	9,100	400	1,100
7 -	1,100	2,500	18,100	4,200	70	36,000	34,700	4,900	12,200	450	1,200
9	950	2,000	16,600	4,600	90	40,700	24,100	4,200	15,300	450	1,400

²CVAA: Cold vapor atomic absorption

Table 5-3. Grab Sample/Water Digestion Data--ICP Average Values.

Analyte	Grab sample avg. concentration (µg/g sample)	Core 31, comp. 1 (µg/g sample)	Core 31, comp. 2 (µg/g sample)	Core 33, comp. 1 (µg/g sample)	Core 33, comp. 2 (µg/g sample)
Ca	< DL	51	62	66	68
Cr	230	210	230	220	210
Fe	< .DL -	80	140	130	160
Ni	<_DL	15	25	25	34
Na	24,800	34,000	35,000	30,800	32,000
Bi	< DL	120	190	230	270
La	< DL	6	9	14	16
P	3,200	5,800	6,000	5,300	5,700
Si	60	440	560	670	620
S	750	1,200	1,200	1,100	1,100

< DL = below detection limit

This behavior is true in general for analytes with these characteristics. The silicon RPD is marginal at 19.26 percent, but acceptable. For core 31, composite 2, cobalt, copper, silicon, and tin were outside the acceptance criteria. Again, cobalt, copper, and tin were not significant contributors to the waste, and thus have little influence on the interpretation of the data. The only clear trend for this assay is for silicon. The variation observed for silicon was attributed to the solubility of the waste matrix (i.e. the waste is not completely solubilized by the acid), and therefore, the fusion results will be reviewed to quantitate silicon. Based on a ratio of the acid and water digestion results, most of the analytes are not water soluble, except as noted earlier, however, approximately 10 percent of the chromium is water soluble.

The results from the fusion preparation core composite had sodium, phosphorous, bismuth, manganese, lanthanum, silicon, iron, calcium, and chromium as major analytes. Nickel was detected in substantial quantities, but the results are considered unreliable because of sample contamination (the method uses a nickel crucible to perform the fusion), and nickel was not detected in significant quantity in the acid digestion assay. RPDs were quite low, generally less than eight percent for most analytes, demonstrating excellent agreement. The only analyte with an RPD outside the acceptance criteria for core 31, composite 1 was cadmium, which is not a large contributor to the waste and is near the detection limit. The only analytes with excessive RPDs for core 31, composite 2 were cadmium and nickel. Both of these analytes have problems associated with them that make these results suspect for this test method (proximity to the detection limit for cadmium and cross-contamination from the

crucible for nickel). Comparisons with the acid digestion results indicate that the samples were well dissolved by the acid preparation, but in some cases, and with silicon especially, the fusion dissolution was necessary to obtain reliable, quantitative results for the analyte.

Core 33

Sedium, phosphorus (probably as a soluble phosphate), and sulfur (probably as a soluble sulfate) were the most significant analytes measured by the water digestion of the core composite. Much smaller amounts of iron, chromium, and silicon were the other main analytes. RPDs were elevated for these samples (between 10 and 20 percent), but generally were within the 20 percent acceptance criteria. For analytes with results outside the acceptance criteria, no clear trend between the two composites or between cores can be established firmly, but there was some correspondence between the cores and their respective composites (i.e. there was some agreement between cores 31 and 33 composite 1, etc.), but the connection was quite tenuous. The water digestion result for core 33, composite 1 had several analytes with high RPDs, many of which are significant contributors to the waste. These analytes are aluminum, barium, iron, manganese, bismuth, and lanthanum. For core 33, composite 2, only calcium and vanadium were outside the acceptance criteria. The degree of variability observed for this particular sample preparation is not unexpected. Most of these analytes probably are in a form that is not readily water soluble and, depending on the sample matrix exposure to the solution media, substantial differences may be observed.

The results from the acid digestion preparation of the core composite samples are quite similar to core 31. They had sodium, phosphorous, sulfur, calcium, chromium, iron, manganese, bismuth, and lanthanum as major analytes, the RPDs decline significantly in comparison to the water digestion results, both in the number of analytes outside the acceptance criteria and the overall magnitude of the RPDs themselves. The RPDs are very good in general for most of the major analytes (between 5 percent and 10 percent), well within the 20 percent acceptance criteria. Analytes outside the acceptance criteria for core 33, composite 1 were antimony and selenium (88.5 percent and 33.0 percent, respectively). For core 33, composite 2, antimony and silicon were outside the acceptance criteria (31.2 percent and 48.7 percent, respectively). The results for antimony and selenium are not surprising. They are not far above their respective detection limits and neither is a substantial contributor to the waste matrix. The silicon RPD result is not unexpected. because the waste has solubility properties that make it resistant to acid digestion, making this assay marginal at best. Based on a ratio of the acid and water digestion results, most of the analytes are not water soluble, except as noted earlier; however, approximately 10 percent of the chromium is water soluble. In this case, for several analytes, some consistency is seen between composite replicates and cores. As noted earlier, core 33 sample results are, in several cases, about 25 percent or more lower than core 31. Significant changes in analyte concentration were observed in the ICP acid digestion results as a function of depth for core 33. The results are shown in Table 5-2. The changes observed for some analytes are notable, ranging from factors of 3 and 4 for calcium, chromium, and phosphorous to orders of magnitude for manganese, bismuth, and lanthanum. These swings

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in concentration suggest the waste is heterogenous on a tank-wide scale and that the waste in the tank is layered.

Properties and major analytes for the fusion digestion results are similar to core 31, however the differences observed between the analytical results of cores 31 and 33 become much less consistent with this group of assay results. RPDs were generally low, less than 12 percent, with most analyte results substantially less than that. Antimony, calcium, and boron had RPDs outside the acceptance criteria for core 33, composite 1. In core 33, composite 2, cobalt and nickel, which are not large contributors to the waste, had RPDs outside the acceptance level and antimony, boron, and cobalt are near their respective detection limits. Several of these analytes have problems associated with them that make these results suspect for this test method. However, the variation seen for calcium is not expected and there is no reason for its behavior. In this case, no consistency was observed between composites or cores, except for nickel in cores-31 and 33, composite-2, and the nickel results for this test method are suspect.

5.1.2 Analytical Results-Anion Assays

Core 31 Ion Chromatography Results:

Major anions detected are NO₃ and PO₄³, with smaller, but still substantial amounts of F and SO₄⁻². Nitrite and chloride are minor contributors to the waste. Indeed, the IC results for nitrite are considered an estimated result; however, the magnitude and range of concentration is confirmed by a spectrophotometric analysis. All RPDs for quantitated results are well within the 20 percent acceptance criteria, and generally are less than 10 percent. Comparisons of phosphorous and sulfur water-digestion ICP results with ion-chromatography results for PO₄³ and SO₄⁻² give good agreement (RPDs are less than 10 percent) and show that phosphorous is about 48 to 56 percent soluble (as phosphate) and sulfur is about 90 to 100 percent soluble (as sulfate). Table 5-4 illustrates the comparison between the water digestion ICP and IC results, and the relationship to the total amount of phosphorous and sulfur in the matrix. Table 5-5 presents the summary results for the IC analytes and other anions.

Core 33 Ion Chromatography Results:

Major analytes were the same as those found in core 31. RPDs for the minor analytes, F, Cl', and NO₂', were much higher in composite 1 than in composite 2, or in either of core 31's composites, exceeding 15 percent, but within the 20 percent acceptance criteria. Core 33, composite 2 had superior RPDs, all were less than 8 percent. Comparison of the phosphorous and sulfur water-digestion ICP results with IC for PO₄³ and SO₄⁻² gives good

Table 5-4. Comparison of Ion Chromatography and Water Digestion ICP Results for Selected Analytes.

Sample ID	Conce	O ₄ ³⁻ ntration g/g)	PO ₄ ³ RPD	Conc	SO ₄ -2 entration ig/g)	SO ₄ -2 RPD	Į.	O ₄ 3- ibility*		O ₄ -2 bility*
	· (IC)	(ICP.w)		(ĪC)	(ICP.w)	·	(ĪĆ)	(ICP.w)	(IC)	(ICP.w)
Core 31-C1	16,200	17,300	6.7	3,690	3,570	-3.3	50.8	54.2	100.4	97.1
Core 31-C2	17,400	17,900	2.8	3,740	3,600	-3.7	54.5	56.1	101.8	97.9
Core 33-C1	13,600	15,900	16.9	3,290	3,420	-3.3	42.6	49.8	89.5	93.1
Core 33-C2	15,100	17,100	13.2	3,470	3,660	-1.4	47.3	53.6	94.4	99.6

*Solubility is a ratio of the IC/ICP.w result to 31,900 μ g/g PO₄³⁻ and 3,675 μ g/g SO₄²⁻, the tank average of the converted fusion results ICP.w = Notation for ICP water digestion result

Table 5-5. Anion Results-Composite Data (water leach).

Analyte	Detection limit (µg/g)	Core 31 (μg/g)		Cor (μg	Grab sample (μg/g)	
		Composite 1	Composite 2	Composite 1	Composite 2	Average
NO ₃	100	44,300	43,800	36,900	40,100	30,300
PO ₄ ³⁻	100	16,200	17,400	13,600	15,100	8,400
SO ₂ 2-	100	3,690	3,740	3,290	3,470	2,900
Cl	10	470	497	401	432	490
F	10	3,090	3,130	1,370	1,630	2,100
NO ₂	50	952	525	878	817	1,320 (IC)
TOC:	500	3,490 -	3,990	2,000	3,000	420
TIC	500	650	824	823	950	670
Free OH	NM	NM	NM	NM	NM	3,000
pH*	NA	10.18	9.93	10.05	9.77	11.65
NH ₃ / NH ₄ ⁺ •	4,500	< DL	< DL	< DL	< DL	400

Table 5-5. Anior	ResultsComposite	Data	(water	leach).
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Analyte	Detection limit (µg/g)	Cor (μg	e 31 (/g)	Core 33 (μg/g)		Grab sample (µg/g)
		Composite 1	Composite 2	Composite 1	Composite 2	Average
Direct Cyanide ¹	5	< DL	< DL	< DL	< DL	< DL

NM = No measurement

NA = Not applicable

< DL = Below detection limit

Note: All IC results are obtained from a water leach preparation and are reported on a wet basis.

*Direct Cyanide, NH₃/NH₄⁺, pH, total organic carbon (TOC), and total inorganic carbon (TIC) are not IC analyses, but are grouped with the anions for convenience. NO₂ results reported are from spectrophotometric methods. At the 222-S Laboratory, TOC and TIC assays were done on water digested samples. When 325 performed segment level assays for these analytes, they did them on direct solid samples.

agreement, however, phosphorous shows a higher, but still acceptable variability, with RPDs of 13 to 17 percent, than sulfur as SO_4^{-2} , with RPDs of -1.4 to -3.3. Phosphorous is approximately 50 percent soluble as phosphate, whereas sulfur as sulfate is almost entirely soluble. Table 5-4 illustrates the comparison between the water digestion ICP and IC results, and the relationship to the total amount of phosphorous and sulfur in the bulk tank matrix. Note the ICP results are converted to phosphate and sulfate for comparison.

Core 31 and 33 Additional Anion Results:

- Nitrite, as determined from spectrophotometry, has good agreement between composites and acceptable RPDs. Concentrations between cores are observed to agree well also.
- The pH of the solids is measured using a water dilution of a solids aliquot. The pH of the mixture is measured and reported. The results for tank 241-T-111 waste from this method are consistent, ranging from 9.8 to 10.5, but are of limited utility because the sample preparation and assays only marginally represent the conditions in the tank. The pH as determined from a grab sample taken in 1994 (see Appendix C) is between 11.5 and 11.8, and is considered more reliable and more representative of the waste tank conditions.

- Ammonia/ammonium results for the solids were all below the detection limit for the method. This is not unexpected. Over time, ammonia probably was slowly released to the vapor space and removed by passive ventilation. A small amount of this analyte remains in the liquids, but the liquid is being removed.
- Total organic carbon (TOC) results indicate a substantial difference between cores 31 and 33 (core 31's TOC concentration is 33 percent higher than core 33's), but results for the individual core composites are very consistent. Core 31 composite 1 and composite 2 RPDs are negligible. Core 33, composite 1 and composite 2 RPDs are low, well within the acceptance criteria, but noticeable.
- Total inorganic carbon (TIC) results were a factor of three to four times lower than the TOC results and were near the detection limit. Therefore, they are not considered wholly reliable, however the results did provide acceptable reproducibility. TOC results are also considered questionable, especially when contrasted with the calorimetry results from segment-level data from both cores (segments 1-3 particularly), and a furnace oxidation value of 4.1 weight percent (dry), for core 33, segment 2 (Baldwin 1994).

The gradient observed in the ICP results remains noticeable for the anions, especially for TOC and F, but the overall effect is much less pronounced. This effect is expected and ascribed to the nature of the assay (i.e. water digestion assay results for soluble analytes in a mostly water matrix are expected to be reasonably homogeneous). TOC and F probably are present in insoluble forms, and thus are not as amenable to detection and are more likely to exhibit variations in concentration.

5.2 ANALYTICAL RESULTS--RADIOCHEMISTRY

5.2.1 Radiochemistry Assays--General Comments

Analyses appear to be consistent. Total beta measurements calculated using ⁶⁰Co detector efficiencies are largely in agreement with the sum of the major beta emitters, ⁹⁰SrY and ¹³⁷Cs [i.e. beta emission values must be adjusted by 1.42 and 1.51, respectively, the ratio of ⁹⁰SrY and ¹³⁷Cs to ⁶⁰Co detector efficiencies (Winters 1991)]. However, the total alpha values show only marginal agreement with the sum of the neptunium, plutonium, and americium/curium values. This behavior was attributed to the low activity of the samples and a recently discovered equipment problem. Technicians at the 222-S Laboratory identified an electronic problem with their alpha detection system involving the degree of discrimination between beta and alpha emissions. These two conditions are thought to contribute to the observed high bias of the total alpha measurements. Detection and quantitation of ¹³⁷Cs and most other radionuclides was based on the presumed ability of the fusion sample preparation to completely dissolve the waste. Water preparation samples were

done on selected analytes as directed in the test plan. GEA measurements show good agreement with alpha energy analysis for ²⁴¹Am. GEA analytical values are not corrected to account for decay, however most of the radionuclides present are long-lived enough and low enough in concentration so that any correction at this point would be modest.

5.2.2 Gamma Energy Analysis Results

The GEA data from the replicate samples of the core composites for cores 31 and 33 prepared by caustic fusion agree reasonably well with each other (RPDs generally were within 10 percent). However, there is a discernable gradient between cores 31 and 33 for 137Cs. This follows the general pattern observed for several other analytes, but in this case ²⁴¹Am-does-not appear to be affected. Review of the segment-level homogenization data for core 33 indicates that in general, the ¹³⁷Cs content is lowest in the bottom segments and highest in the upper segments, increasing in concentration from bottom to top. This behavior indicates that the ¹³⁷Cs is concentrated in the upper portion of the tank. However, the overall radionuclide content of tank 241-T-111 is extremely low. The 154Eu and 155Eu content is above the detection limits only in segments 1 and 3, suggesting that these isotopes are in the upper portions of cores 31 and 33. However because of the limited segment level data, too many conclusions should not be drawn from these observations. Grab sample results for ¹³⁷Cs is more than a factor of four lower than those obtained from fusion assays, indicating that the radionuclides, ¹³⁷Cs in particular, are very insoluble. Agreement between the same top and bottom aliquots in the homogenization check samples prepared by acid digestion were very good for analytes above the detection limit, with RPDs generally less than 10 percent and usually-much less than that, indicating sample homogeneity and/or relatively complete dissolution.

5.2.3 Total Alpha Analysis and Uranium Assay

Total alpha, plutonium, ²³⁷Np, ²⁴¹Am, and ²⁴⁴Cm analyses were performed on the fusion prepared samples of the core composites and selected assays were done on the liquid grab sample. Total alpha measurements were also performed on the homogenization check samples from segments 1, 3, 5, 7, and 9 from core 33. The total alpha activity was determined by drying a small aliquot of prepared sample on a counting plate and assayed with an alpha proportional counter. The plutonium and americium fractions were separated by solvent extraction or ion exchange and similarly counted.

The plutonium analyses are reported as total alpha ^{239/240}Pu. The process blank was two to three orders of magnitude lower that the samples, indicating little contamination occurred during sample preparation. The total alpha concentration frequently tends to be somewhat lower than the sum of the individual alpha emitters. The difference is probably caused by absorption by the salt residue on the counting mounts. However, in this case, substantial differences are observed between the total alpha measurements and the sum of the individual alpha emitters. The total alpha emissions are believed to be lower than the measurement

indicates because of a bias caused by beta emissions confounding the detector. The activity of the samples is so low that the offset used to discriminate between alpha and beta plateaus was not sufficient to provide accurate readings. Furthermore, analyses from PNL indicate a potential low bias for plutonium in the samples. The degree of discrepancy ranges from over a factor of two to nearly a factor of five. This degree of disagreement warrants further attention. Isotope content was determined by thermal-ionization mass spectroscopy. Little variation in the plutonium isotopic composition was observed between cores. Total alpha measurements vary widely as a function of depth, but exhibit a general decreasing trend the deeper the samples are in the waste. Segment 5 was an exception to this trend. The anomalous total alpha reading in segment 5 may be indicative of a process upset or change in waste management operations, such as the 5-6 waste that Anderson (1990) notes was co-mingled with the 2C waste in 1952.

Uranium measurements were obtained from laser fluorimetry of the fusion-prepared sample from the two core composites and their replicates. The assays show good agreement between duplicates for each individual core composite, but there is not good agreement between the core 31 and 33 replicates. Additionally, the gradient observed in the other assays is not evident here. Furthermore, analyses from PNL indicate a potential low bias for uranium in the samples. The degree of discrepancy is nearly a factor of two for corresponding samples, and thus warrants further study. Differences in concentration as a function of the water content of the samples is not deemed to be an issue because the additional characterization work done in resolving the energetics question indicates that little or no water is lost while the samples are stored. Time lag is also not considered relevant, since the difference in time between the two assays was small in comparison to the half-lifes involved. No general trend of the uranium concentration as a function of depth can be established because there is no segment-level data for this analyte.

5.2.4 Total Beta

Total beta, 90 Sr, and 99 Tc analyses were performed on the liquid grab sample and fusion prepared samples of the core composites. The total beta activity was determined by drying a small aliquot of prepared sample on a planchet and assaying it with a beta proportional counter. The 90 Sr fraction was separated by solvent extraction or ion exchange and counted. The 99 Tc fraction was separated similarly, but assayed using liquid scintillation. There generally is good agreement (RPDs were less than 10 percent) between duplicates, and preparation blank beta activities are orders of magnitude lower than the levels found in the samples, again indicating little contamination from preparation in the hot cell. Most of the beta activity in the tank samples is from 90 SrY and 137 Cs. There is also a trace of 99 Tc. The 90 SrY, 99 Tc, and 137 Cs data are consistent between the fusion core composites and their replicates, but in this case the gradient between cores 31 and 33 is observed for 90 Sr and 137 Cs. There is no data to determine if 90 Sr content varies as a function of depth.

Tables 5-6, 5-7, 5-8, and 5-9 show the average radionuclide concentrations found in the core composite samples. Table 5-10 shows average fission product concentration and total alpha concentrations as a function of depth.

Table 5-6. Consolidated Radionuclide Concentration Results for 241-T-111.

Analyte method	Detection limit	Core 31, composite 1	Core 31, composite 2	Core 33, composite 1	Core 33, composite 2
GEA analytes	(μCi/g)	(μCi/g)	(μCi/g)	(μCi/g)	(μCi/g)
⁶⁰ Co.f	(DL = 4.07E-04)	< DL	< DL	< DL	< DL
¹³⁷ Cs.f	$-(DL = 3.70E-04)^{-1}$	0.211	0.237	0.114	0.103
¹⁵⁴ Eu.f	(DL = 1.20E-03)	0.00108	0.00324	< DL	< DL
155Eu.f	(DL = 5.95E-04)	< DL	< DL	0.00307	< DL
²⁴¹ Am.f	(DL = 1.14E-04)	0.0459	_0.0409	0.0387	0.0443
Beta emitters					
Total beta.f	(DL = 9.35E-02)	20.6	21.5	9.59	8.83
Total beta	Calculated	20.7	21.5	10.5	9.9
³ H.w	(DL = 3.15E-04)	<. DL .	< DL	< DL	< DL
¹⁴ C.w	(DL = 2.25E-04)	< DL	< DL	< DL	< DL
⁵⁹ Ni.a	(DL = 4.00E-06)	8.3E-05	3.33E-05	4.44E-05	4.07E-05
⁶³ Ni.a	(DL = 4.00E-06)	0.0093	0.00358	0.00545	0.00459
⁷⁹ Se.f	(DL = 1.40E-04)	< DL	< DL	< DL	< DL
⁹⁰ Sr.f	(DL = 2.15E-03)	7.16	7.43	3.65	3.43
⁹⁹ Tc.f	(DL = 4.60E-03)	0.00514	0.00473	0.0114	0.0104
¹²⁹ I.f	(DL = 6.40E-03)	< DL	< DL	< DL	< DL

Table 5-6. Consolidated Radionuclide Concentration Results for 241-T-111.

Analyte method	Detection limit	Core 31, composite 1	Core 31, composite 2	Core 33, composite 1	Core 33, composite 2
Uranium and transuranics			_	_	
Total U.f (μg/g) (325)	(DL = 3.40E-02)	2,180 4,000	3,880 5,200	3,180 4,500	1,950 3,500
Total alpha.f	(DL = 7.01E-03)	0.358	0.359	0.377	0.379
Total alpha	Calculated (Range)	0.179-0.669	0.179-0.608	0.172-0.357	0.195-0.416
²³⁷ Np.f	(DL = 3.40E-02)	< DL	< DL	< DL	< DL
²³⁸ Pu <u>.</u> f	(DL = 1.00E-02)	< DL	< DL	< DL	< DL
^{239/240} Pu.f (325)	(DL = 3.50E-03)	0.138 <i>0.628</i>	0.136 <i>0.565</i>	0.134 <i>0.319</i>	0.147 <i>0.36</i> 8
²⁴¹ Am.f	(DL = 3.00E-03)	0.0414	0.0431	0.0382	0.0478
²⁴⁴ Cm.f	(DL = Not Reported)	< DL	< DL	< DL	< DL

< DL = below detection limit

Analyte. f = fusion digestion

Analyte.a = acid digestion

Analyte.w = water digestion

Total beta calculated determined by:

 $1.42 (2)(^{90}Sr) + 1.51 (^{137}Cs)$

Total alpha calculated determined by:

 $^{239/240}$ Pu + 241 Am

Table 5-7. Core Composite Uranium.

Core number	U _{FL} (222-S) (μg/g)	U _{FL} (325) (μg/g)	mass percent	mass percent
Core 31, composite 1	2,180	4,000	99.3074	0.6755
Core 31, composite 2	3,880	5,200	99.3098	0.6761
Core 33, composite 1	3,180	4,500	99.3125	0.6761
Core 33, composite 2	1,950	3,500	99.3161	0.6717

FL = Uranium measurement by laser fluorimetry.

222-S Assay date: 4/92 325 Assay date: 9/92 it to the technique

Table 5-8. Plutonium Concentration and Isotopic Distribution.

Core number	Total Pu α (222-S) (μCi/g)	Total Pu α (325) (μCi/g)	mass percent	mass percent	mass percent	mass percent	mass percent
Core 31, C1	0.138	0.628	0.005	96.7199	3.2109	0.0352	0.0151
Core 31, C2	0.136	0.565	0.0105	96.6351	3.2834	0.0496	0.0215
Core 33, C1	0.134	0.319	0.004	96.7540	3.1046	0.1071	0.0683
Core 33, C2	0.147	0.368	0.0105	96.5499	3.3436	0.0621	0.0337

222-S Assay date: 4/92

C1 = Composite 1

325 Assay date: 9/92

C2 = Composite 2

Table 5-9. Radiochemical Analyses of Grab Sample.

Analyte	Average concentration (µCi/mL)	Average concentration $(\mu \text{Ci/g})^1$
⁹⁰ Sт	0.001	0.001
¹³⁷ Cs	0.090	0.087
^{239/240} Pu	6.83E-05	6.58E-05
Total Alpha	0.0024	0.0023
Total Beta	0.233	0.224

Density of 1.036 g/mL used for conversion.

Table 5-10. Tank 241-T-111 Core 33 Radionuclide
Analyte Trending as a Function of Depth
(Acid prep on segments).

Segment	¹³⁷ Cs (μCi/g)	¹⁵⁴ Eu (μCi/g)	¹⁵⁵ Eu (μCi/g)	²⁴¹ Am (μCi/g)	⁶⁰ Cο (μCi/g)	Total alpha (μCi/g)
1	0.403	0.021	0.027	0.138	0.006	0.649
3	0.140	9.05E-04	0.002	0.014	0.0005	0.166
5	0.088	< DL	< DL	0.020	< DL	0.527
7	- 0.023	<- DL	∵< DĻ	-0.014	0.0005	0.350
9	0.013	< DL	< DL	0.050	< DĻ	0.262

5.3 TANK 241-T-111 CORE SAMPLE RHEOLOGICAL/ PHYSICAL MEASUREMENTS

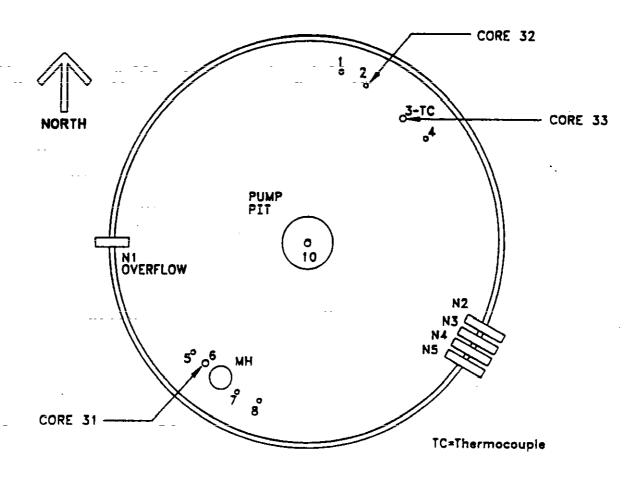
Physical and rheological assays consume substantial quantities of material. Tables 5-11 and 5-12 provide a breakdown of the total amount of sample available at the outset of the analylical effort. Figure 5-1 shows the location where the core samples were taken and the waste depth. Measurements of physical characteristics such as shear strength, viscosity, particle size, and settling properties were taken. These measurements are necessary for the design and fabrication of retrieval, pretreatment, and final waste disposal systems. Rheological assays were performed on samples from core 31, segments 2, 4, and 8. Particle size measurements were done on each segment of core 31. The data from segment 4 is not considered valid for these assays because it had dried before the measurements were taken. The drying process irreversibly changed the physical properties of the sample under investigation, and thus the sample is not considered representative. Therefore, the results from most of these assays will not be presented. However, in some cases it is useful to compare and contrast the results from the "representative" samples with the samples that dried.

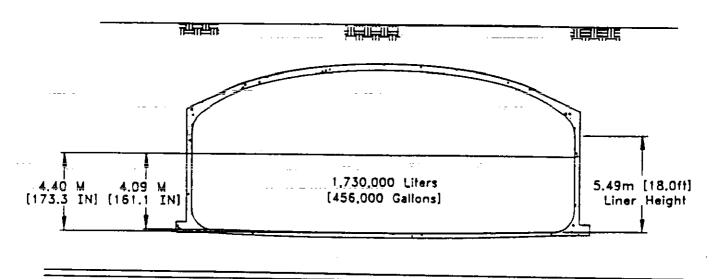
Table 5-11. Tank 241-T-111 Core Sample Description Summary.

Core number	Segment number	Solid sample mass-(g)	Liquid sample mass (g)	Approximate solid sample volume (mL)	Liquid sample volume (mL)	Approximate solid sample length
31	1	18.74	45.26	Not Recorded	~50	Not Resolved
31	2	178.68	4.12	150	NA	15 in. (38 cm)
31	3	162.2	NA	177	NA	18 in. (46 cm)
31	4	153.47	NA	148	NA	15 in. (38 cm)
31	5	190.94	NA	187	NA	19 in. (48 cm)
31	6	0	NA .	0	NA	No Sample
31	7	186.44	NA	177	NA	18 in. (46 cm)
31	8	186.44	NA	187	NA	19 in. (48 cm)
31	9	203.08	NA	187	NA	19 in. (48 cm)

2 for for \$2 median all

Figure 5-1. Current Condition of Tank 241-T-111.





				<u> </u>		
Core number	Segment number	Solid sample mass (g)	Liquid sample mass (g)	Approximate solid sample volume (mL)	Liquid sample volume (mL)	Approximate solid sample length
33	1	159.16	0	137	NA	14 in. (36 cm)
33	2	209.59	0	187	NA	19 in. (48 cm)
33	3	167.9	0	163	NA	16 in. (41 cm)
33	4	182.05	0	167	NA	16 in. (41 cm)
33	5	174.3	0	167	NA	17 in. (43 cm)
33	. 6	217.37	. 0	- 187	NA	19 in. (48 cm)
33	7	196.91	0	187	NA	19 in. (48 cm)
33	8	199.8	0	187	NA	19 in. (48 cm)
33	9	191.01	0	187	NA	19 in. (48 cm)

Table 5-12. Tank 241-T-111 Core Sample Description Summary.

5.3.1 Shear Strength

The shear strength of the waste from tank 241-T-111 was measured on the unhomogenized segment samples from core 31 (segments 2, 4, and 8). The shear strength measurements were made at ambient temperatures using a shear vane connected to a viscometer and rotated at 0.3 rpm. Shear strength (τ_*) is a semiquantitative measurement of the force required to move the sample. Because shear strength is dependent on sample handling, the measurement was taken without any sample homogenization. Small aliquots from the segments 2, 4, and 8 from core 31 were taken and assayed. The aliquots were transferred to a sample jar and allowed to settle for several weeks so that they could recover from the disturbance of sampling and extrusion. The extended delay between sample and analysis was permitted because it is believed that the longer the sample sits undisturbed, the closer it will resemble its original condition; therefore, the shear measurement is likely to be more representative. The shear stress (S_{τ}) of the sample was recorded as a function of time and the shear strength was calculated using Equation 1.

$$\tau_{s} = \frac{\left[\%\tau/100\right] * S_{\tau} * 4.9E + 05}{\frac{\pi * H_{v} * D_{v}^{2}}{2} + \frac{\pi * D_{v}^{3}}{6}}$$
(1)

			٠.
11/	п	T	٠.

%7/100 =	The ratio of the total torque to the maximum torque of the
	viscometer head, measured as a percentage of the full scale on the
	plot of the shear stress versus time diagram (dimensionless)

S_r = Instrument reading proportional to the torque (dimensionless)

 $4.9E+05 = maximum torque of the viscometer head (dyne<math>\bullet$ cm)

 $H_v = - \text{shear-vane height } (0.635 \text{ cm})$

No to family a

 D_v = shear vane diameter (0.635 cm)

The shear strength for segments 2 and 8 were found to range $5,000 \pm 2,300$ dynes/cm². Segment 4 is not reported because of sample drying. Although relatively low, the shear stress of the material exceeded the baseline value for the measurement system (200 dynes/cm²). Therefore, the values are considered to be valid and representative. Some additional drying of the sample may have occurred during the settling time, causing the shear stress to be higher than expected.

5.3.2 Shear Stress and Viscosity as Functions of Shear Rate

Shear stress measurements, as functions of shear rate, were performed on the as-received, 1 to 1 and 3 to 1 water to sample dilution of the sample at ambient hot-cell temperatures [ranging from 27 to 34 °C (81 to 93 °F)] and at 95 °C (203 °F). Drying the as-received sample at 95 °C posed difficulties in measurement; therefore no results of the rheograms for the samples run under those conditions are presented. In addition, the results from segment 4 are not presented because the drying that occurred compromised the sample properties, as discussed previously.

A rheogram for a material with a yield stress has two sections. The first section is a straight line beginning at the origin and climbing up the ordinate. This portion of the rheogram records the material as it acts like a solid or gel. When sufficient force is applied to the material to make it yield, the rheogram breaks sharply to the right; recording the material's behavior as a fluid. The point on the rheogram at which the sample's behavior transfers from a solid or gel to a fluid is the yield point or yield stress. This minimum shear stress must be exceeded to initiate fluid behavior. The tank wastes demonstrate both elastic and plastic behavior, depending on the amount of shear acting on them. The samples are elastic under low shear conditions (less than 50 s⁻¹), and plastic under high shear conditions (greater than 300 s⁻¹).

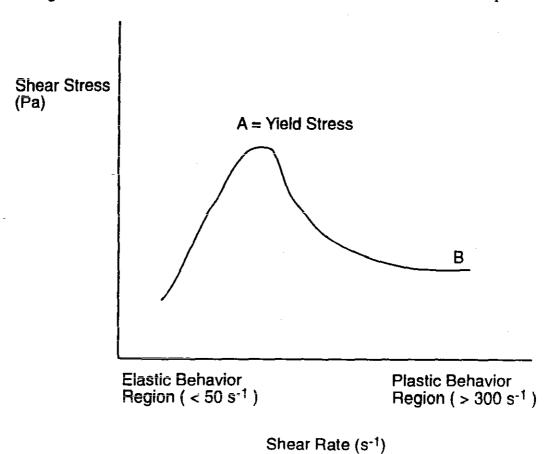
Viscosity measurements as a function of shear rate for the 1 to 1 diluted samples had viscosities near the limits of detection of the system (2 cP) for over the broad range of shear

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rates; however, some qualitative and quantitative information was obtained from the measurement trials. Even though the measurements were at the limits of detection, the qualitative and quantitative behavior was consistent and reproducible. Viscosity was observed to increase slightly, then decrease with increasing shear rates. The 1 to 1 dilution of the segment 2 and segment 8 samples exhibited tendencies toward yield-pseudoplastic behavior. In fact, the general behavior exhibited by the wastes is best described by a yield-pseudoplastic model, however the system was not modeled and empirical model parameters were not determined because the system was at the detection limits. No other measurements of the viscosity as a function of shear rate were made on the 1 to 1 dilution samples at 95 °C (203 °F) or the 3 to 1 dilution samples.

Figures 5-2 to 5-6 are general illustrations of the rheograms. They are not to scale and do not fully capture all of the nuances and detail that is contained in each measurement trial. However, when coupled with the description underneath each diagram, much insight can be obtained about the flow properties of the waste. If more detailed rheological information is required, the data package should be consulted. Note that in the figures, Point A is where the sample begins to register movement. Point B represents the behavior of the sample at the maximum shear rate of the viscometer.

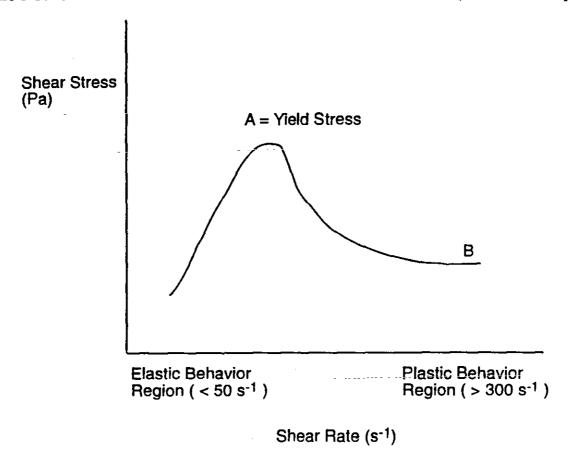
Figure 5-2. Shear Stress as a Function of Shear Rate: Direct Sample.



Shear Stress as a Function of Shear Rate: Direct Sample.

Sample: core 31	Sample number	Temperature (°C)	Point A shear stress range (Pa)	Qualitative behavior of rheogram	Point B shear stress (Pa)
Segment 2	80701	34	88 - 220	Wide variation at low shear, converging to a single value at high shear.	165
Segment 2	80703	33	200 - 680	Same	70
Segment 8	123201	33	36 - 108	Same	77
Segment 8	123202	- 33	0 - 108	Same	50

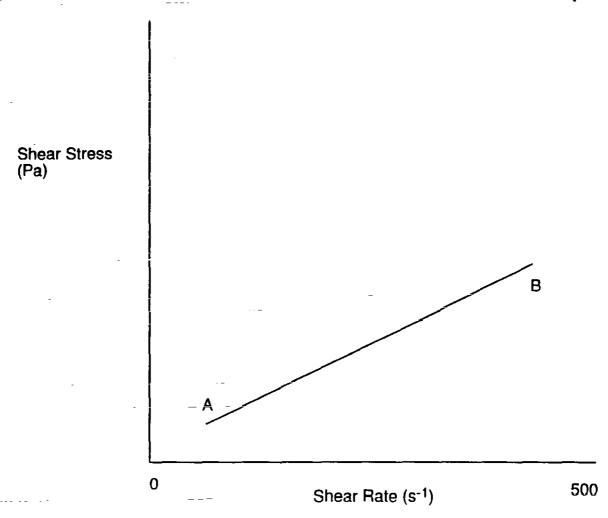
Figure 5-3. Shear Stress as a Function of Shear Rate: 1 to 1 Dilution, Water to Sample.



Shear Stress as a Function of Shear Rate: 1 to 1 Dilution, Water to Sample.

Sample: core 31	Sample number	Temperature (°C)	Point A shear stress range (Pa)	Qualitative behavior of rheogram	Point B shear stress (Pa)
Segment 2	1	27	0.6 - 4.2	Wide variation at low shear, converging to a single value at high shear.	2.4
Segment 2	_ 2.	27	- 1.2	Linear	2.8
Segment 2	3	95	1.0 - 7.0	Wide variation at low shear, converging to a single value at high shear.	2.0
Segment 2	4	95	0.7 - 1.1	Linear	1.2 - 1.4
Segment 2	5	95	1.0 - 2.4	Linear	2.2

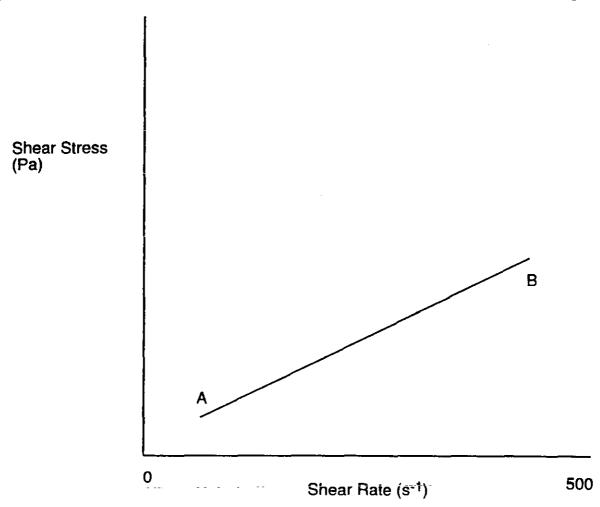
Figure 5-4. Shear Stress as a Function of Shear Rate: 1 to 1 Dilution, Water to Sample.



Shear Stress as a Function of Shear Rate: 1 to 1 Dilution, Water to Sample

Sample:	Sample number	Temperature (°C)	Point A shear - stress-range (Pa)	Qualitative behavior of rheogram	Point B shear stress (Pa)
Segment 8	1	27	0.4 - 0.6	Linear	- 2.8-
Segment 8	2	27	0.6	Linear	2.8
Segment 8	3	95	0.6	Linear	2.0
Segment 8	4	95	2.0 - 5.0	Erratic, non-linear	2.0
Segment 8	-5	95	0.2	Linear	0.7 - 0.9

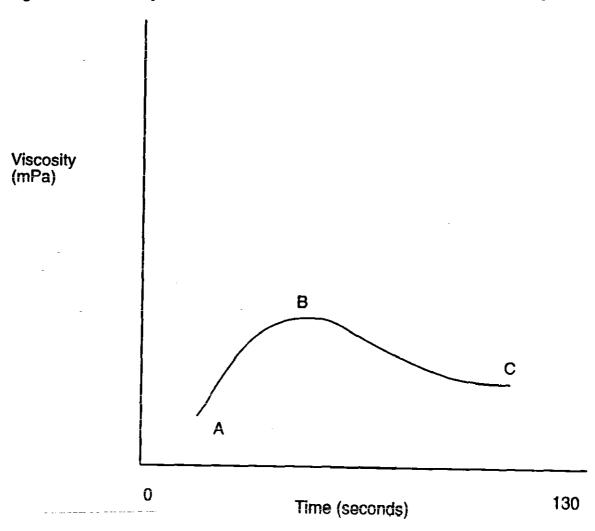
Figure 5-5. Shear Stress as a Function of Shear Rate: 3 to 1 Dilution, Water to Sample.



Shear Stress as a Function of Shear Rate: 3 to 1 Dilution, Water to Sample.

Sample: core 31	Sample number	Temperature (°C)	Point A shear stress range (Pa)	Qualitative behavior of rheogram	Point B shear stress (Pa)
Segment 2	1	27	0.05 - 0.25	Linear	0.75 - 1.1
Segment 2	2	27	0.2 - 0.35	Linear	1.2
Segment 2	3.	95	0.2	Linear	0.8
Segment 2	4	95	0.3	Linear	1.0
Segment 8	3	95	Not defined	Erratic, non-linear	0.4
Segment 8	4	95	0.3	Linear	1.0

Figure 5-6. Viscosity as a Function of Time: 1 to 1 Dilution, Water to Sample.



Viscosity as a Function of Shear Rate: 1 to 1 Dilution, Water to Sample

				•	•	
Sample: core 31	Sample	Temperature -(°C)	Point A viscosity (mPa)	Qualitative description of rheogram	Point B viscosity (mPa)	Point C viscosity (mPa)
Segment 2	1	30	0.65	Rises, levels off, then gradually declines	1.8 - 2.0	1.5 - 1.7
Segment 2	2	30	0.56	Slightly sinusoidal	0.6	0.6
Segment 2	3	30	0.80	Rises, levels off, then gradually declines	1.0	0.9
Segment 8	3	30	0.75	Flattened exponential growth and decay curve	1.0	0.75
Segment 8	4	30	0.85	Flattened exponential growth and decay curve	0.95	0.9 -

5.3.3 Particle Size Measurement

Particle size is analyzed by placing a small amount of sample in a dispersant, which is the liquid used to disperse and suspend the particles from the solid sample. Samples from each segment of core 31 were prepared and assayed. The prepared sample was placed in a particle size analyzer. The apparatus measures particle size by passing a thin beam of laser light through the dispersant. The diameter of a particle in the dispersant can be determined by the amount of light that it blocks as it passes through the beam. The dimension measured by this method is the value across the short diameter of the particle. This means that if a particle is oblong, the machine estimates the shortest length across the particle. The term "diameter" throughout this text will be used to describe any linear profile of any shape.

An important consideration involving the analysis of particle size is the dispersant used. The primary concern involved with the dispersant is that it may dissolve the particle. Any particles existing in the tank that are soluble in the dispersant will dissolve or decrease in size during the analysis. Depending on the dispersant, the particle size analysis may not represent the true particle size distribution in the tank. In the case of tank 241-T-111, water was used as the dispersing medium. If a true particle size distribution is required, the mother liquor, or drainable liquid of the tank should be used if possible because the tank particulates are already in equilibrium with the tank's mother liquor. The insolubility of the waste matrix suggests that the particle size data acquired should be acceptable.

The mean particle size in the number distribution runs in a narrow range from 0.93 to 1.23 microns in diameter for tank 241-T-111 waste samples. Table 5-13 presents the summary results of the measurements.

Table 5-13. Core 31 Particle Size Distribution by Number.

Segment	Mean (μm)	Standard deviation	Median (μm)
1	1.23	0.89	0.94
2	1.13	0.80	0.88
3	1.17	1.00	0.91
4	0.93	0.60	0.80
5	0.95	0.63	0.81
6	-	-	-
7	0.97	0.60	0.83
8	1.02	0.85	0.82
9	1.02	0.83	0.83

Table 5-14 presents the summary results of the volume distribution measurements. Assuming that the density of the solid material within the tank is constant, the volume distribution is also the best estimation of the mass particle size distribution of the tank.

Table 5-14. Core 31 Particle Size Distribution by Volume.

Segment	Mean (μm)	Standard deviation	Median (μm)
l	28.56	35.92	5.81
2	14.91	20.76	4.79
3	64.99	46.19	58.69
4	24.87	34.15	5.63
5	37.87	47.91	12.31
6	•	-	-
7	7.95	11.88	4.02
8	24.72	28.18	10.02
9	59.69	49.04	58.97

The number density graph is plotted over the acquisition range of the device (from 0.5 to 150 microns). The numbers of particles in each size range, shown as a percentage of the whole, are graphed against their respective size ranges to form a distribution curve. The figures show that the modes for particle size range between the origin and 2.0 microns. In

fact, over 80 percent of the measured particles fit within this narrow band. As with the number distribution, the volume distribution is represented by a probability volume density graph. The average particle size in the volume distribution is considerably larger than in the number distribution. The particle size in the volume distribution ranges over the full scale of the device, 0.0 microns to 150 microns in diameter, between the eight segments (core 31 segment 6 was empty, but the nomenclature for the other samples held). The analyzer calculates particle volume as the cube of the diameter.

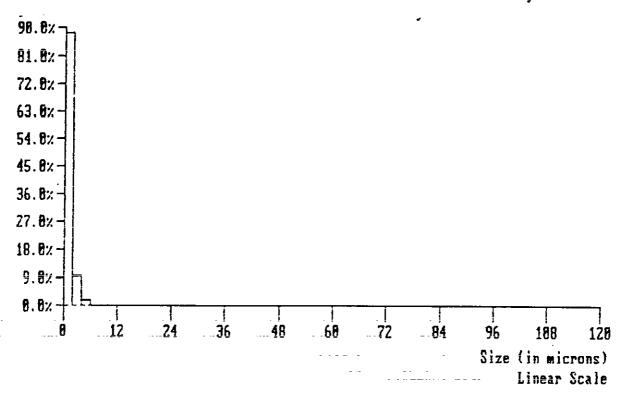
In core 31, approximately 70 percent of particle measurements for segments 1, 2, 4, 5, 7, and 8 fall into the range between 0.0 and 24.0 microns, and tend to be dispersed towards the smaller particle sizes. Segments 3 and 9 were exceptions to this rule. They had a distribution over the broad spectrum of particle sizes, with the particles generally much larger and more widely scattered over the 0.0 to 150.0 micron range and a slight tendency towards the smaller end of the scale. In segment 3, 23 percent of the particles were less than 24 microns. In segment 9, 40 percent of the particles were less than 24 microns.

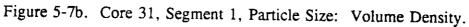
The disparity between the segment measurements possibly indicates a difference in waste type, or perhaps a transitional layer between two waste types that individually are physically similar, but when commingled, may precipitate larger particulate materials. In segments 1, 2, 4, 5, 7, and 8, over 60 percent of the particles in the sample have a diameter of less than 24.0 microns. In segments 3 and 9, over 50 percent of the particles have a diameter of greater than 24.0 microns. In the retrieval and subsequent treatment of the tank wastes, it may be desirable to design pumping or filtration systems for the tank particulate. Therefore, the volume distribution of the particles should not be neglected (i.e., particles with diameters of over 20 microns should be considered in these designs). In addition, variation in particle size distribution is believed to have an impact on analytical precision, especially with small sample sizes, and thus, should be considered when evaluating analytical results. Plots of the probability number and volume density for each core are presented as Figures 5-7a to 5-14b.

5.3.4 Settling Behavior of As-Received and Diluted Samples

This section analyzes the settling behavior and physical properties of the grab samples and the as-received 1 to 1 and 3 to 1 water to sample dilutions. The physical properties reported here include settling rates and volume percent for settled solids and weight percent and volume percent for centrifuged solids. The experimental procedures used to take these measurements were reported previously (McKinney et. al 1993). The physical properties of the grab samples are reported in Table 5-15. The properties for core 31 samples are summarized in Table 5-16.

Figure 5-7a. Core 31, Segment 1, Particle Size: Number Density.





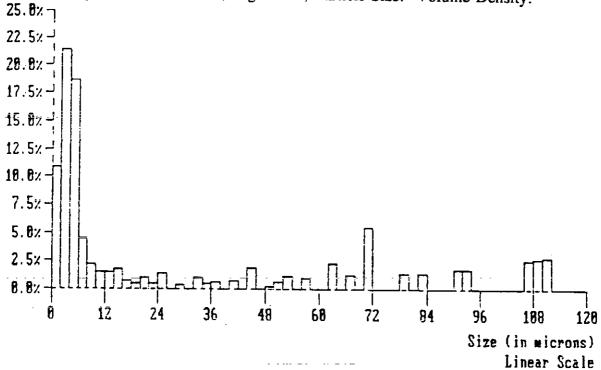


Figure 5-8a. Core 31, Segment 2, Particle Size: Number Density.

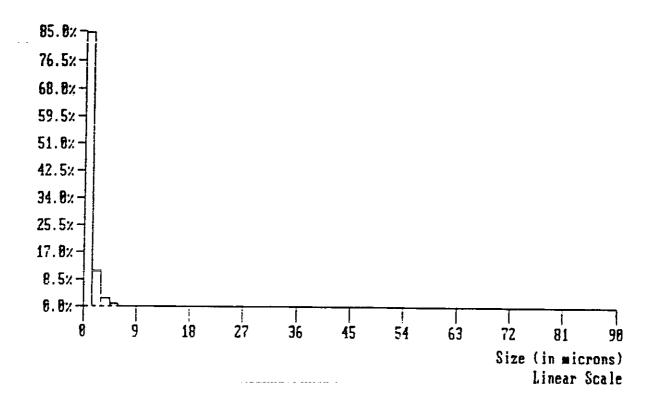


Figure 5-8b. Core 31, Segment 2, Particle Size: Volume Density.

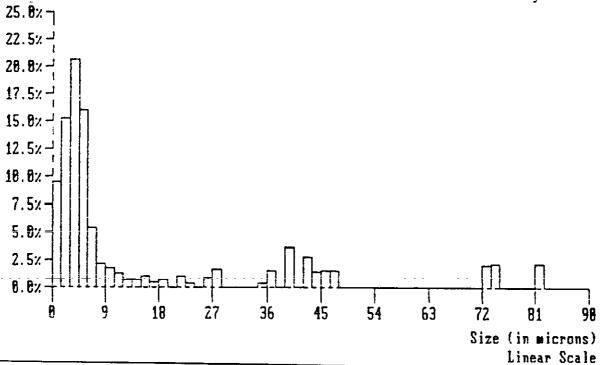


Figure 5-9a. Core 31, Segment 3, Particle Size: Number Density.

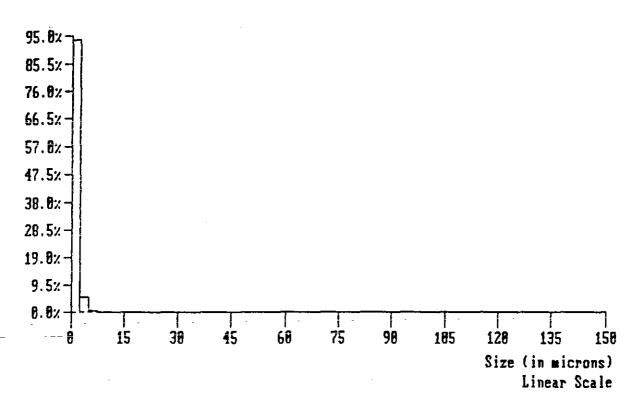


Figure 5-9b. Core 31, Segment 3, Particle Size: Volume Density.

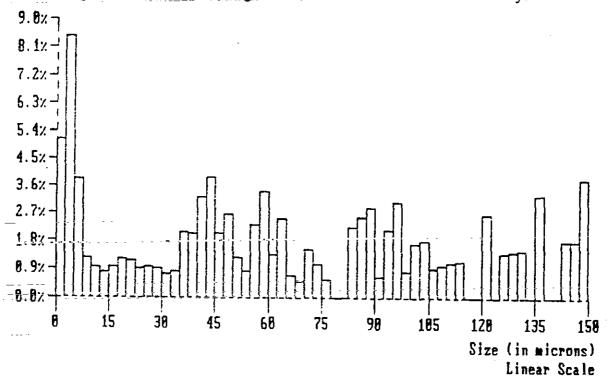


Figure 5-10a. Core 31, Segment 4, Particle Size: Number Density.

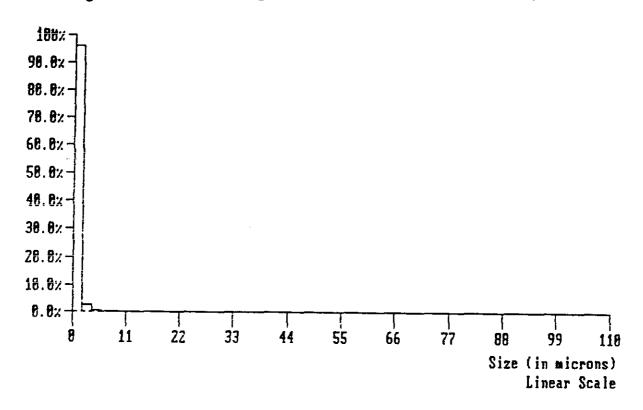


Figure 5-10b. Core 31, Segment 4, Particle Size: Volume Density.

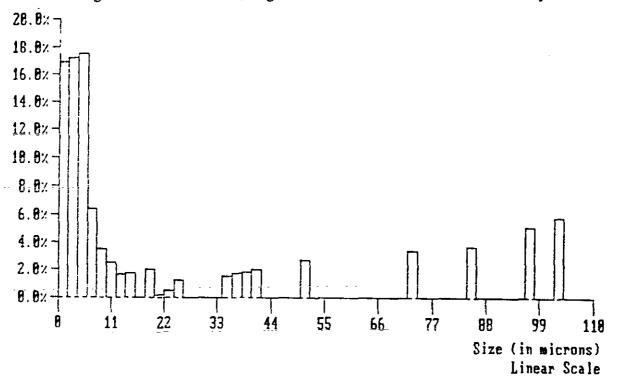
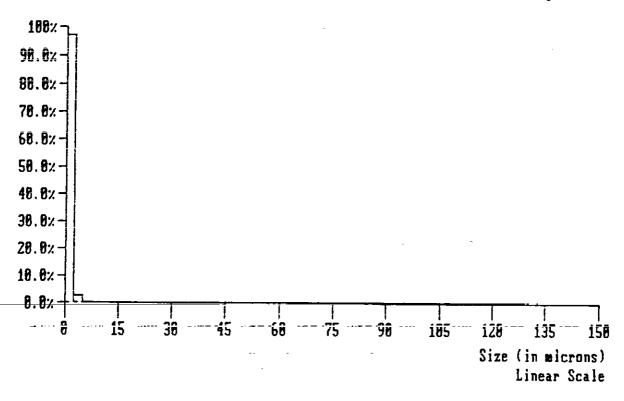
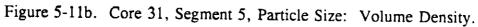


Figure 5-11a. Core 31, Segment 5, Particle Size: Number Density.





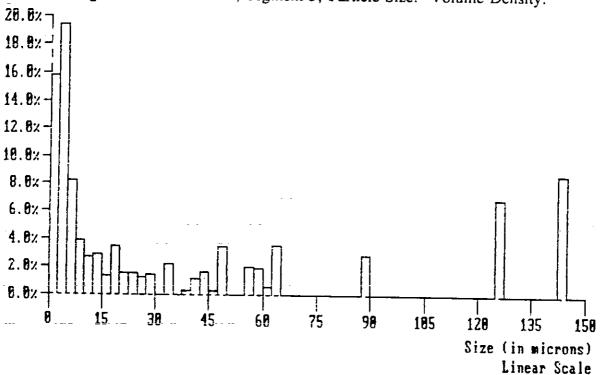


Figure 5-12a. Core 31, Segment 7, Particle Size: Number Density.

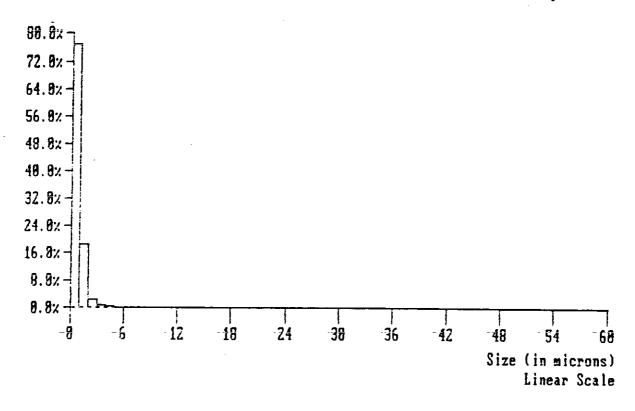


Figure 5-12b. Core 31, Segment 7, Particle Size: Volume Density.

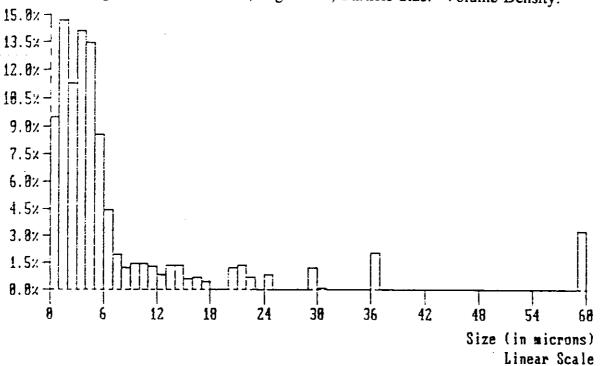


Figure 5-13a. Core 31, Segment 8, Particle Size: Number Density.

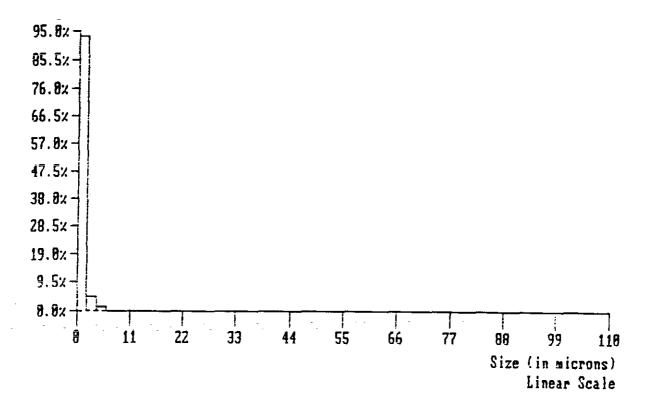


Figure 5-13b. Core 31, Segment 8, Particle Size: Volume Density.

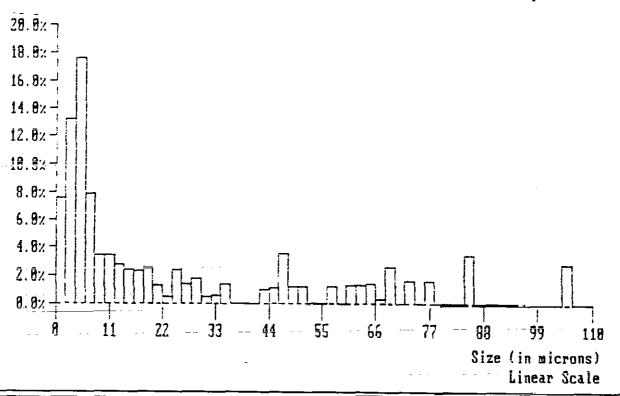
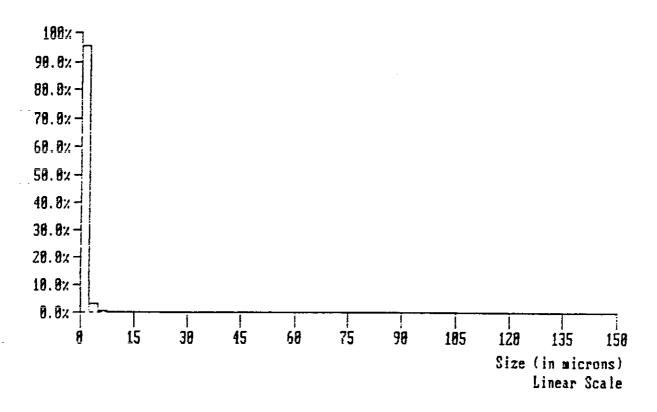


Figure 5-14a. Core 31, Segment 9, Particle Size: Number Density.



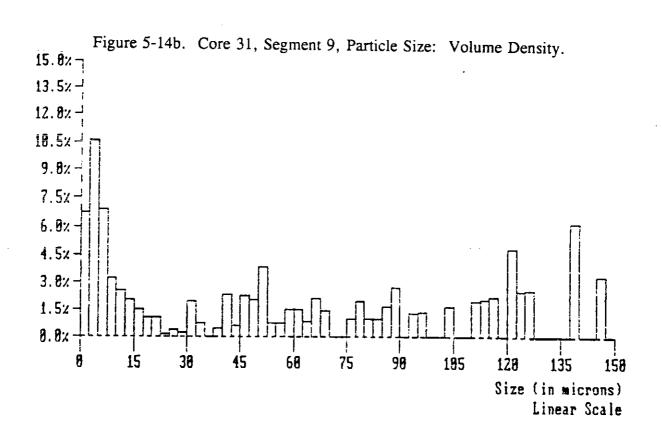


Table 5-15. Grab Sample Physical Properties Summary.

Property	Grab sample #1	Grab sample #2	Grab sample #3	Average value
Specific gravity	1.036	1.038	1.033	1.036
pH	11.57	11.59	11.78	11.65
Wt% H₂O	92.70	92.90	92.90	92.83
Wt% H ₂ O (TGA)	91.17	91.73	90.24	91.05
Settled solids (vol%)	< 1%	< 1%	< 1%	< 1%

The grab samples were clear yellow liquids with no particulate.

Table 5-16. Physical Properties Summary.

		_Sa	mple
Property		Core 31, segment 2	Core 31, segment 8
Settled solids (vol%)		100%	100%
Wt% Solids	As-Received	22.4	29.3
Wt% Undissolved solids	As-Received	19.0	25.4
Density (g/mL)]	1.19	1.28
Centrifuged samples			-
Vol%		65.8	71.9
Wt%	1 hour at	67.3	75.9
Centrifuged supernate density (g/mL)	1,000 gravities	1.07	1.10
Centrifuged solid density (g/mL)		1.22	1.34

No settling was observed in the as-received segment samples over a period of three days, and there was no standing liquid obtained from the samples. Two dilutions of 1 to 1 and 3 to 1 water to sample ratios, respectively were prepared, and the volume-percent settled solids for each of the dilutions are plotted as a function of settling time.

The 1 to 1 dilution for segment 2 reaches a final volume percent settled solids of 85 to 87 percent. Settling is observed throughout the three-day period, but the majority of the settling is observed in the first 10 hours. The 3 to 1 dilution reaches a final volume-percent settled solids of approximately 52 percent. Again, settling is observed over three days, and the first 10 hours is when the majority of the solids settle. Qualitatively, the settling behavior for both sample dilutions is a steep, nearly linear relationship between the initial

fluidization of the material and the first 10 hours of settling. After that, the final 10 percent of the suspended solids take up almost the rest of the time settling in a long, gradual decline, before coming to equilibrium.

The 1 to 1 and 3 to 1 dilutions for segment 4 were compromised by drying the sample before its assay. However, some observations and contrasts with the other samples are appropriate. Settling is mostly completed after 3 to 4 hours, and is complete after 10 hours. This is in sharp contrast to the other samples for which there is a long, asymptotic-like settling behavior observed for a substantial portion of the suspended solids (10 to 15 percent), after the initial settling phase. This behavior is suggestive of a colloid or gel for segments 2 and 8. In contrast, segment 4 appears to be a collection of discrete particles with no interaction between them.

The 1 to 1 dilution for segment 8 reaches a final volume-percent settled solids of about 80 percent (see Figure 5-17a). Settling is observed throughout the three-day period, but the majority of the settling is observed in the first 10 hours. The 3 to 1 dilution reaches a final volume-percent settled solids of approximately 40 percent. Again, settling is observed over three days and the first 10 hours is when the majority of the solids settle. Qualitatively, the settling behavior for both sample dilutions is a shallow, nearly linear relationship between the initial fluidization of the material and the first 10 hours of settling. The slope of this line is much more gradual than that of segment 2 for the corresponding dilutions. After the first 10 hours, the final 15 percent of the suspended solids take up almost the rest of the time settling in a long, gradual decline before coming to equilibrium. Table 5-17 summarizes the settling behavior for the samples investigated. Figures 5-15a through 5-17b illustrate the setting behavior over time.

Table 5-17. Settling Comparison for 1 to 1 and 3 to 1 dilutions for Core 31 Segments 2, 4, and 8.

Analyte	Segment 2 Segment		ent 4 Segme		ent 8	
Dilution: water to sample	1:1	3:1	1:1	3:1	1:1	3:1
Final volume % solids	87	52	22	22	80	40

5.4 ANALYTICAL RESULTS--ENERGETICS

TGA and DSC were performed on subsegment and core-composite material from tank 241-T-111. These two thermal analysis techniques are used to determine the thermal stability or reactivity of a material. In DSC analysis, heat flow over and above the usual heat capacity of the substance is measured while the substance is exposed to a linear increase in temperature, i.e., the change in temperature, divided by the time elapsed is constant (dT/dt = constant). While the substance is being heated, a cover gas (usually air or N_2) is passed over the waste material to remove any gases-being-released. The onset temperature

Figure 5-15a. Settling Rate Data for Tank 241-T-111 Core 31, Segment 2, 1 to 1 Dilution.

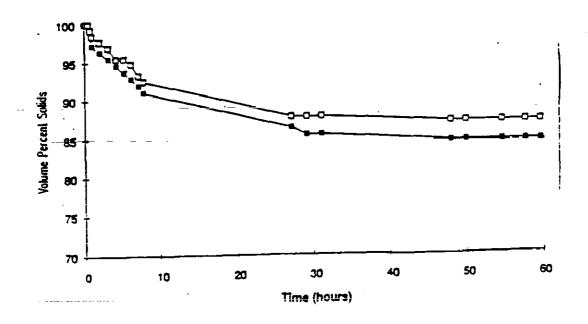


Figure 5-15b. Settling Rate Data for Tank 241-T-111 Core 31, Segment 2, 3 to 1 Dilution.

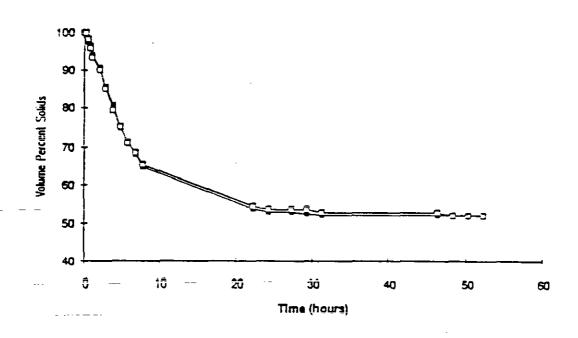


Figure 5-16a. Settling Rate Data for Tank 241-T-111 Core 31, Segment 4, 1 to 1 Dilution.

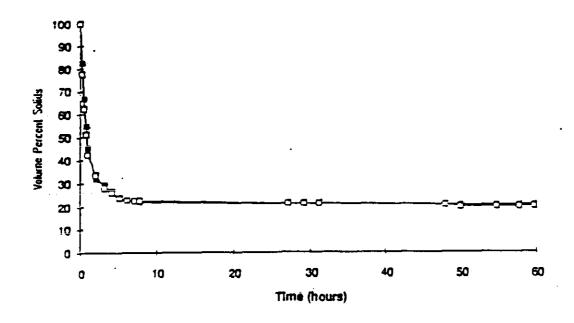


Figure 5-16b. Settling Rate Data for Tank 241-T-111 Core 31, Segment 4, 3 to 1 Dilution.

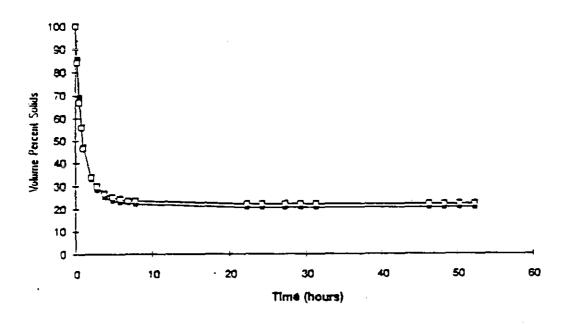


Figure 5-17a. Settling Rate Data for Tank 241-T-111 Core 31, Segment 8, 1 to 1 Dilution.

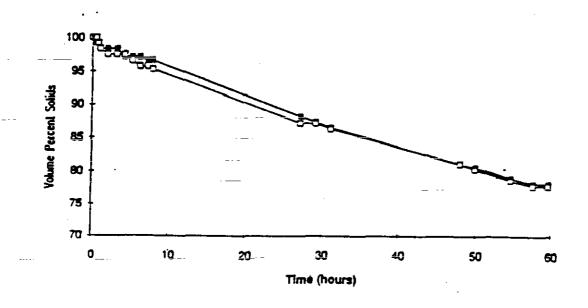
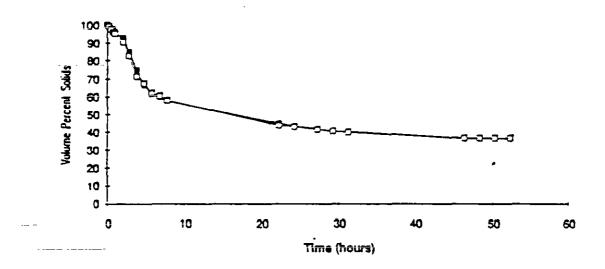


Figure 5-17b. Settling Rate Data for Tank-241-T-111 Core 31, Segment 8, 3 to 1 Dilution.



for an endothermic or exothermic event on a DSC is determined graphically. The endpoints of the event are determined and a line is drawn between them to establish a base. A line tangent to the initial side of the event is drawn until it intersects the base. From that point of intersection, a vertical line is constructed to the temperature scale at the bottom of the DSC curve. That temperature is the onset temperature of the event.

TGA measures the mass of a sample while the temperature of the sample is increased at a constant rate. The X-axis is representative of the running time of the analysis as well as the temperature increase of the sample during analysis. The Y-axis represents the weight percent of the sample and is effectively unitless. As with the DSC, a cover gas is passed over the sample during heating. Any decrease in the weight percent of the sample represents a loss of gaseous matter from the sample either through evaporation or a reaction that forms gas phase products.

DSC is often used to measure thermal decomposition temperatures, heats of reaction, reaction temperatures, melting points, and solid-solid transition temperatures. TGA is used to measure thermal decomposition temperatures, water content, and reaction temperatures. The two methods often provide complementary information.

5.4.1 Remarks on the Interpretation of Differential Scanning Calorimetry/Thermogravimetric Analysis Data

Tables 5-18, 5-19, and 5-20 summarize the results of the thermal analyses performed. Although DSC assays were performed on the grab samples, the results were uniformly negative (i.e. no exotherms were observed). Where exotherms were observed, there are two significant features seen on the DSC and one on the TGA plots. The endothermic DSC event overlaps the area where the substantial majority of the sample mass is lost as recorded by the TGA, suggesting that this endotherm is a result of the evaporation of water from the sample. The values presented in the tables may not exactly match the values derived from the DSC and TGA plots, especially exotherm magnitudes. This is because interpreting these semi-quantitative analyses requires considerable experience and judgement on the part of the analyst, and differences in perception and judgement between readers can be expected in a semi-quantitative analysis. Appendix A contains copies of all of the DSC and TGA traces and results.

Additional physical properties work was performed to expand the understanding of the exothermic behavior of some of the segments (WHC 1994, Delegard 1994). Table 5-21 presents additional energetics results for core 33, segments 1 and 2. These samples were dried under a vacuum at 60 °C before analysis, using either air or nitrogen as a cover gas. Even after drying, the samples retained 10 to 12 weight percent water. Table 5-22 presents a brief summary of the analytical results for the average sample properties as-received after centrifuging samples from core 31, segments 3 and 7, and core 33, segments 1 and 7, at 500 gravities for 113 hours.

5.4.2 General Comments on the Differential Scanning Calorimetry/ Thermogravimetric Analysis Behavior of the Samples

The first transition in each sample is endothermic, begins at the lower temperature limit of the analysis (30 °C), and essentially is complete between 140 and 180 °C. The most likely phenomenon occurring in this region is the release of the bulk and interstitial water in the core sample material. The endotherms exhibited in this region are substantial (typically in excess of 1,000 J/g). These values are per gram of wet sample. If divided by the mass fraction lost during analysis, they range from 1,600 to 1,900 J/g (dry) and correspond roughly with the heat of vaporization of water (2,260 J/g). The TGA water content corresponds reasonably well with the water loss observed in a gravimetric weight percent solids determination; however, the gravimetric weight percent water assay is consistently lower.

Table 5-18. Percent Water Analyses Results from Tank 241-T-111.

Sample I.D.	Core 31 average gravimetric Wt% loss	Core 31 average TGA transition 1 Wt% loss	Core 33 average gravimetric Wt% loss	Core 33 average TGA transition 1 Wt% loss
Segment 1	80.3	87.0	80.4	NM
Segment 2	82.4	87.0	85.7	80.6
Segment 3	86.0	85.0	81.8	88.5
Segment 4	77.3	82.8	79.9	89.5
Segment 5	80.9	88.0	78.2	88.8
Segment 6	No sample	No sample	75.8	84.4
Segment 7	76.8	84.8	71.7	85.8
Segment 8	76.6	85.6	75.4	84.8
Segment 9	75.8	71.0	76.0	85.2
Segment 9B	70.4	72.1	NA	NA
Composite 1	74.6	73.3	76.5	81.6
Composite 2	75.9	70.2	77.1	80.8

Table 5-19. Differential Scanning Calorimetry Energetics Results from Tank 241-T-111, Core 31 (wet basis).

		Transi	tion 1	Transition 2		
Core sample	Range (°C)	Avg. onset (°C)	ΔH range (J/g)	Range (°C)	Avg. onset (°C)	ΔH range (J/g)
Segment 1	43 - 141	43	1,088 to 1,406	200 - 387	200	-259 to -273
Segment 2	46 - 149	46	1,108 to 1,643	200 - 398	225	-256 to -264
Segment 3	45 - 160	50	1,210 to 1,233	195 - 405	198	-263 to -448
Segment 4	44 - 165	44	1,235	200 - 390	200	-55.7
Segment 5	NR		Endotherm NR	NA	NA	No Exotherm
Segment 6	NA					NA
Segment 7	50 - 1 6 4	- 50	1,488	165 - 400	ŇÁ	No Exotherm
Segment 8	50 - 153	50	1,534	154 - 400	NA	No Exotherm
Segment 9	61 - 158	61	1,437	159 - 400	NA	No Exotherm
Composite 1	NR		Endotherm NR	256 - 339	257	-23.6 to -37.0
Composite 2	NR		Endotherm NR	260 - 334	262	-18.5 to -22.9

NOTE: To convert from J to cal, divide by 4.18. NOTE: Negative ΔH indicates an exotherm.

NA = Not applicable.

NR = Not resolved.

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Table 5-20. Differential Scanning Calorimetry Energetics Results from Tank 241-T-111, Core 33 (wet basis).

-	Т	ransitio	on 1	Transition 2		
Core Sample	Range (°C)	Avg. onset (°C)	ΔH range (J/g)	Range (°C)	Avg. onset (°C)	ΔH range (J/g)
Segment 1	49 - 168	49	958 to 1,604	168 - 374	184	-218 to -293
Segment 2	45 - 179	50	1,346 to 1,496	168 - 438	174	-454 to -645
Segment 3	NA		Endotherm NR	237 - 400	237	-49.3
Segment 4	NA		Endotherm NR	NA		No Exotherm
Segment 5	NA		Endotherm NR	NA		No Exotherm
Segment 6_	. NA		Endotherm NR	NA	- -	No Exotherm
Segment 7	NA ·		Endotherm NR	NA		No Exotherm
Segment 8	NA		- Endotherm NR	NA	-· <u></u> · ·	No Exotherm
Segment 9	NA NA		Endotherm NR	NA	- .	No Exotherm
Composite 1	NA		Endotherm NR	NA		No Exotherm
Composite 2	NA		Endotherm NR	NA		No Exotherm

NOTE: To convert from J to cal, divide by 4.18.

NOTE: Negative ΔH indicates an exotherm.

NA = Not applicable. NR = Not resolved.

Table 5-21. Differential Scanning Calorimetry Energetics Results from Tank 241-T-111, Core 33 (dry basis).

-		Fransiti	ion 1		on 2	
Laboratory-core sample-air/N ₂	Range (°C)	Avg. onset (°C)	ΔH range (J/g)	Range (°C)	Avg. onset (°C)	ΔH range (J/g)
222-Score 33, seg. 1, Air			NRDried	158 - 405	NR	-1,857 to -1,882
222-S core 33, seg. 2, Air	****	-7777	NRDried	130 - 425	NR	-251 to -269
222-S core, 33 seg. 2, N ₂			NRDried	130 - 430	NR	-288 to -309
222-S core 33, seg. 2, N ₂			NRDried	128 - 418	NR	-180 to -187
222-S core 33, seg. 2, N ₂			NRDried	123 - 421	NR	-163 to -175
222-S core 33, seg. 2, N ₂			NRDried	121 - 438	NR	-336
325 core 33, seg. 2, N ₂			NRDried	107 - 394	199	-836 to -898

NOTE: To convert from J to cal, divide by 4.18.

NOTE: Negative ΔH indicates an exotherm.

NA = Not applicable. NR = Not resolved.

Table 5-22. Additional Segment-level Physical Properties Measurements (1994).

Analyte	Core 31, segment 3		Core 31, segment 7		Core 33, segment 1		Core 33, segment 7	
	As- received	Centrifuged	As- received	Centrifuged	As- received	Centrifuged	As- received	Centrifuged
Gravimetric water (%)	79.53	64.96	74.72	62.06	79.56	65.49	74.07	59.95
TGA (%)	76.72	55.36	74.06	55.83	78.08	51.37	78.1	45.15
Density (g/mL)	1.24	1.09	1.19	1.20	1.16	1.19	1.20	1.29
ΔH exotherm range (J/g)	-112 to -191	-465.3 to -546.9	-10.2 to -33.1	0	-249 to -254	-822.4 to -838.1	-37.5 to -41.4	Ō

When there is a second transition it is usually substantial and the energetic behavior is readily quantifiable in all of the samples analyzed where exotherms are observed. Any weight loss in the second transition region (generally temperatures above 200° C) was not readily quantifiable, whether exotherms were observed or not. The results for the samples from segments 1, 2, and 3, which are from the upper portion of the tank, indicate significant differences in thermal behavior compared to other samples from deeper in the tank, further suggesting a difference in waste type. In addition, because of the observed exothermic behavior for the top-3-segments from cores 31 and 33 and the results from Baldwin (1994), the present TOC assay is not considered capable of measuring the TOC in the waste. Resolution of the actual magnitude of the exotherm, its reaction mechanism, and speciation of the-fuel is still continuing.

5.5 OVERALL ANALYTICAL DATA SUMMARY

Several characterization and safety issues are defined by certain bulk amounts or weight percent of a given analyte. Table 5-23 presents the nominal concentration and calculated bulk amounts of the analytes in the waste matrix. The gross waste inventory in the tank is estimated to be 2,171,000 kg wet solid. It is assumed that by the publication of this report, no drainable liquid will remain in the tank. Appendix C presents the data, assumptions, and calculations used to determine the following values.

Table 5-23. Overall Data Summary and Inventory Estimates. (3 pages)

GROUP	Process history ¹ (Agnew 1994)	TRAC (Jungfleisch 1984)	Analytical data range (most quantitative assay)	1992 cores 31 and 33 average	Tank inventory based on 1992 core data
CATIONS	μg/g	. µg/g	µg/g .	μg/g	kgs
Be.f			< DL	. < <u>DL</u>	< 0.22
B.a			23.4 - 32.2	28	60.8
Na.f	74,200	0	33,900 - 39,800	37,000	80,300
Mg.a			290 - 479	377	820
Al.f	0	0	459 - 693	570	1,240
Si.f	2,800	0	5,410 - 5,960	5,670	12,300
P.f	(as PO ₄ ³⁻)	(as PO ₄ 3-)	9,070 - 11,600	10,400	22,600
S.f	(as SO ₄ 2-)	(as SO ₄ ² ·)	1,080 1,350	1,230	2,660
K.a	3,700	0	1,020 - 1,210	1,140	2,460
Ca.f	0	0	2,050 - 2,760	2,420	5,260
Ti.f			22.3 - 72.9	48	104
V.f	454		12.1 - 16.5	14.5	31.5
Cr.a	33	719	1,840 - 2,140	1,980	4,290
Mn.a	90	507	6,140 - 6,710	6,330	13,700
Fe.f	9,400	10,300	15,900 - 20,500	18,500	40,200
Ni.a	0	0	108 - 151	132	285
Co.f			10.1 - 13.3	4.5	9
Cu.f			22.1 - 36.3	29.3	63
Zn.f			- 104 - 110	106	231
As.a			< DL	< 3.3	< 7.2
Se.a-			< DL	< 1.5	< 3.3
Sr.f	0		280 - 334	300	651
Zr.f	0	0	4	NR	
Ag.f.		0 -	37.1 - 221	130	278
Cd.f			6.42 -10.7	5.8	12.6
Sn.a			1.61 - 4.21	2.5	5.4
Sb.a			22.6 - 36.5	31.4	70
Ba.a		0	57.0 - 87.3	69	150
La.a	12,700	512	3,620 - 4,890	4,220	9,160

Table 5-23. Overall Data Summary and Inventory Estimates. (3 pages)

GROUP	Precess-history (Agnew 1994)	TRAC (Jungfleisch 1984)	Analytical data range (most quantitative assay)	1992 cores 31 and 33 average	Tank inventory based on 1992 core data
CATIONS	μ <u>ε</u> / <u>ε</u>	μg/g	µg/g	μg/g	kgs
Ce.a	0	0	28.6 - 37.8	33.7	73.2
Hg.CVAA		_	1.08 - 1.83	1.43	3.1
Pb.f	0	. 0	267 - 484	365	792
Bi.a	23,200	963,000	23,300 - 28,500	26,000	56,300
U.LF	140		1,950 - 5,200	3,550	7,700
RADIONUC	LIDES	·			· · · · · · · · · · · · · · · · · · ·
Analyte	μCi/g	μCi/g	μCi/g	μCi/g	Ci
Total Alpha			0.166 - 0.649	0.368	NA
Total Beta			8.83 - 21.5	15.1	NA
²⁴¹ Am		0.0092	0.0382 - 0.0478	0.0425	92.4
^{239/240} Pu	0.009	0.055	0.134 - 0.628	0.304	660
¹³⁷ Cs	0.086	0	0.103 - 0.237	0.166	360
[™] Tc			0.00473 - 0.0114	0.0079	17.2
⁵⁹ Ni			3.35E-05 - 9.4E-05	5.04E-05	0.11
⁶³ Ni			0.0036 - 0.011	0.0057	12.4
⁹⁰ Sr	0.176	1.84	3.43 - 7.43	5.41	11,800
¹⁴ C		0	< DL	< DL	
³Н -			< DL	< DL	
ANIONS				-	
Analyte	μg/g	μg/g	μg/g	μ <u>g</u> /g	kgs
OH.	8,900	15,700	3,300 - 6,000 ²		
NH,			< DL	< DL	
F	24,600	0	1,370 - 3,130	2,300	4,990
CI ⁻	0	0	400 - 500	450	977
NO ₂	0	0	525 - 952	793	1,720
NO,	51,600	0	36,900 - 44,300	41,300	89,660
CO ₃	0	0	as TIC	812	1,760
Total PO, 3	35,400	438,000	27,800 - 35,500	31,900	70,100

Table 5-23. Overall Data Summary and Inventory Estimates. (3 pages)

GROUP	Process history ¹ (Agnew 1994)	TRAC (Jungfleisch 1984)	Analytical data range (most quantitative assay)	1992 cores 31 and 33 average	Tank inventory based on 1992 core data
CATIONS	μĝ/ĝ	μg/g	μg/g	μg/g	kgs
ANIONS					<u> </u>
SO ₄ -2	1,450	0	3,290 - 3,690	3,680	7,990
CARBON-BI	ARING SPECIES		-		·
Analyte	μg/g	μg/g	μ <u>8</u> /8μ	μg/g	Kgs
TOC	2,000		2,000 - 3,990	3,120	6,770
TIC		0	650 - 950	812	1,760
C ₂ H ₃ O ₂	0	0	NM		
C2O42.	7,200		NM	***	
C ₆ H ₅ O ₇	0	0	NM		
Fe(CN) ₆ ⁴⁻	0	0	NM		
EDTA	0	0	NM	***	
HEDTA	0	0	NM	***	
PHYSICAL 1	PROPERTIES				
Wt% Water (Grav.)	72.5%		74.6 - 77.1	76.0%	1.65E+06 kg
™t% Water (TGA)	72.5%		70.2 - 81.6	76.5%	1.66E+06 kg
Bulk Density (g/cm³)	1.21	1.8	1.19 - 1.28	1.24	***
Supernatant Density (g/cm³)	1.058		1.033 - 1.038	1.036	

Process history estimates are determined using a simple linear combination based on the proportions of 2C and 224 waste contributed.

²OH is estimated from mass and charge imbalance.

6.0 INTERPRETATION OF ANALYTICAL RESULTS

Tank 241-T-111 had a relatively straightforward process history as documented in the transfer records. It received very few major types of waste that were likely to deposit solids during its operating history. The waste types, in chronological order, were as follows:

- 2C waste
- 224 Waste
- Decontamination streams from T Plant (221-T).

The purpose of this section is to attempt to identify and reconcile the location of the tank waste solids, thereby estimate the tank inventory for various analytes of importance.

The waste profile was identified by examining the available segment level assays for analytes or characteristics distinct to the waste types that were disposed in the tank, and then combining that information with what is known regarding the tank's process history. The first waste placed in the tank through the cascade inlet from tank 241-T-110 was 2C waste. Study of the process stream compositions indicates that this waste would be comparatively high in bismuth and phosphate in addition to the ubiquitous sodium, nitrate, and iron found in nearly all waste types. Elevated levels of fluoride, chromium, and sulfate are also expected. Qualitatively, the waste has been observed to be a gelatinous material. Anecdotal reports during this time also indicate that the cascade lines may have clogged in tank 241-T-111 and in other tank cascades receiving this type of waste. The 2C solids volume was measured to be between 723,000 L (191,000 gal) and 931,000 (246,000 gal) in 1953 (Anderson 1990).

The tank then received 224 waste. The solids from this waste are high in manganese, lanthanum, and fluoride, however, the 224 waste may have been combined with the 2C waste before being discharged to the tank. The estimated solids volume contribution for this waste type in tank 241-T-111 at the end of bismuth phosphate production in T Plant was between 594,000 and 802,000 L (157,000 and 212,000 gal).

The last major waste type disposed in the tank was T-Plant decontamination waste. Much of the physical and chemical composition of this waste is unknown because it was a catch-all, consisting of various process residues, unused stock solutions, and aqueous decontamination solutions that contained surfactants (such as Turco). Few predictions can be made from

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studying the historical process flowsheets in this case. Lack of analytical data and/or transfer records with regard to T-Plant effluents later in tank 241-T-111's service life are a great source of uncertainty regarding the waste near the surface. One observation is that slightly higher TOC values may be anticipated near the surface because of the use of detergents in the decontamination waste stream and its chronological discharge sequence with respect to the other wastes. This waste stream is estimated to occupy the top segment of waste, or 198,000 L (53,000 gal). The sum of the wastes would range between 1.52 million L and 1.93 million L (401,000 gal and 511,000 gallons), well within the reported range described in Anderson (1990). This waste volume would measure between 389.4 to 491.0 cm (153.3 to 193.3 in), measured from the centerline and distributed evenly across the tank. The present surveillance level status is almost exactly in the middle of this range 1.73 million L (456,000 gal) and 440.2 cm \pm 1.3 cm (173.3 \pm 0.5 in) (Rios 1994).

Two common characteristics of all these waste types are high water content and relatively low activity. None of the waste streams disposed in this tank were concentrated through the evaporator; therefore, the waste tank would not have any salt cake. The uranium, plutonium, and fission product content of these wastes are uniformly low.

6.1 Review of the Analyte Profiles

The following conclusions are drawn from review of the available composite and segment analyses presented in Section 5, and the historical information presented in Section 2.

Core 31

The chemical analyses of core 31 indicate there are at least two primary types of material in distinct layers in the tank. The DSC traces for segments 1, 2, and 3 show exotherms far out of proportion to the measured organic content in the wastes. The temperature range where the wastes begin to show reactions are from 170 to 400° C, but do not appear to be self-sustaining. Instead, the overall energy profile is highly endothermic, probably caused by the large amount of bound water (70 to 80 weight percent) that is evaporated from the sample before a reaction is initiated. The overall physical and chemical properties of the waste in the tank roughly correspond to the expected behavior and composition of 2C waste, with large quantities of bismuth, iron, and phosphorous, and no exothermic behavior. But, there is a substantial contribution of manganese and lanthanum in the composites. However, these analytes can be found in 224 waste, and historical data indicates that 224 waste was added to tank 241-T-111 later in its service life. Therefore, the analytical and historical data correspond reasonably well, except for the anomalous energetic results. At this time, there is no adequate explanation for the observed exotherms in the upper segments of tank 241-T-111 waste.

Core 32

Every segment of core 32 was compromised during the sampling process in some fashion. When the sampler operated properly, liquids were the only material recovered from this core. However, the sampler valve failed repeatedly. Photographs taken before and after the sampling event reveal that there was a plastic bag in the vicinity of the sampling area before coring operations. After core sampling, the bag can not be seen. It is surmised that the sampling drill string was obstructed by the bag during core sampling operations, causing the corruption of the samples. All samples from core 32 were rendered unusable or categorized as non-representative. Therefore, no assays were performed and no analytical results are reported.

Core 33

The chemical analyses of core 33 indicate there probably are three primary types of material in distinct layers in the tank. The extremely high manganese values in the top first segment of the tank are attributed to a combination of T-Plant decontamination waste (i.e. the result of a final process flush during the decontamination of T Plant) and 224-waste solids deposited late in the tank's service life. Proceeding deeper into the tank, beneath the first segment, the distribution of bismuth, manganese, lanthanum, and chromium in the composites and the analyte profiles from the homogenization results through segment 3 support the conclusion that this material is still 224 waste. Segment 3 itself is suspected to be a transition layer, containing the boundary between the 224 waste and the 2C waste, based on the exothermic behavior of the waste as a function of depth. The TOC analysis indicates moderate amounts of residual organics in the waste.

Cesium - 137 concentrations between the core composites vary within a factor of two, and the variation in the ²⁴¹Am is less than 13 percent. But the change in concentration as a function of depth for each of these analytes is much more significant. Both analyte profiles show a decreasing trend as a function of depth in core 33. The ¹³⁷Cs concentration decreases by more than a factor of 30 over the depth of the tank, and ²⁴¹Am decreases by a factor of 10 through segment 7 before rebounding somewhat in segment 9. The ¹³⁷Cs and ²⁴¹Am concentrations as a function of depth in core 33 show profiles consistent with the wastes believed to be associated with the segments: low overall ¹³⁷Cs and ²⁴¹Am values. But the relative radionuclide concentrations for the suspected 224 and T-Plant decontamination wastes are higher than the 2C wastes. The ⁹⁰Sr concentration is also low, and ⁹⁰Sr is similar to ¹³⁷Cs in the magnitude of the change in concentration and as a function of location for both cores. However, there are no high radionuclide values anywhere in the tank, and the tank temperature further confirms the tank's low radionuclide content.

Lanthanum and phosphate/phosphorus demonstrate an increasing concentration profile as a function of depth. Between segments 1 and 3 and 5 there is an abrupt change in the concentrations of calcium, chromium, manganese, bismuth, lanthanum, and phosphorous, although after segment 5, the analyte concentrations tend to plateau, mirroring the behavior

of the radionuclides. In addition, the DSC traces for segments 1, 2, and 3 show exotherms far out of proportion to the measured organic content in the waste. The temperatures where the wastes begin to show reactions range from 170 to 400 °C, but these reactions do not appear to be self-sustaining. Instead, the overall energy release is highly endothermic, probably from the large amount of bound water that is evaporated from the sample before a reaction is initiated.

In conclusion, the physical and chemical properties of the waste in the bottom half of the tank correspond to the expected behavior and composition of 2C waste, and the upper half of the tank is suspected to be 224 waste with a high manganese layer located in segment 1. Therefore, the analytical and historical data correspond well with the historical fill pattern, except for the anomalous energetic results. Presently there is no adequate explanation for the observed exotherms in the upper segments of tank 241-T-111 waste.

6.1.1 Entrance, Exit, and Mixing Effects on Analyte Distribution

Figure 5-1 shows an elevation and plan view of where the core samples were taken. Important items to note are the arrangement and location of the risers and cascade lines (inlet and outlet). Their configuration can have a substantial impact on the distribution of waste in the tank, and observations with regard to sampling. However, the waste entrance and exit points for the tank over its service life are not well documented, therefore the spatial relationship and proximity to the sample risers is not known. The decant "float and flex" pump contained a 6.1-m (20-ft) section of flexible hose that could traverse a relatively wide area under and around the pumpout riser. The cascade fill line where 2C waste entered the tank is closer to the core 33 sample point than to the core 31 sample point. There was very limited discharge/transfer traffic from the other separations plants within T farm or with the tank farms as a whole, so no highly enriched layer of radioactive material is expected to lie on top of the waste as has been observed in other tanks sampled. However, the relative concentration of radionuclides is observed to be higher in the upper portions of the tank.

As new wastes entered the tank and were distributed across it, the material under and around the tank pumpout could have been disturbed, and occasionally solids could have been transferred, in behavior similar to the last in, first out principle. It is believed that the material beneath the waste inlets, the cascades and perhaps a riser, would have been disturbed initially, but over time, large stratified layers resistant to mixing eventually would have built up. No deliberate mixing of the wastes was performed; therefore, where segment-level data is available, distinctions between waste types can be made. Some of the larger particulate materials discharged to the tank initially may have settled out near the inlet because they were not as flocculent or as easily suspended as some of the other solids. This settling behavior may have provided a slight degree of separation. Therefore, the influence of the waste inlet and outlet locations can provide insight to the analyte distribution and waste profiles between cores 31 and 33. These factors may account for the nominally observed lateral heterogeneity between cores 31 and 33; however, it must be noted that this lateral

heterogeneity is very slight and there may be several other factors contributing to this observation.

6.1.2 Waste Profile

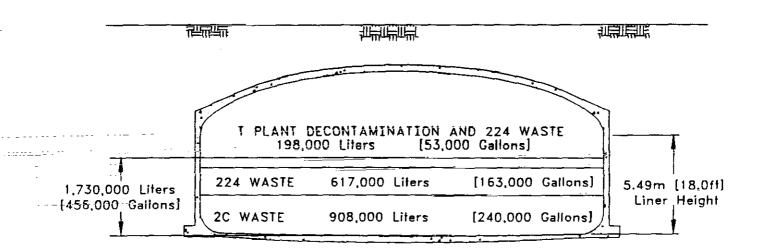
Given the historical and analytical data, it is possible to develop a rough representation of the wastes as they presently are configured in tank 241-T-111. Figure 6-1 shows a representation of the overall waste profile of tank 241-T-111 and the assumed volumes, boundaries, and positions of the various individual layers as they are believed to exist.

6.2 WASTE SUMMARY AND CONDITIONS

Historically, 241-T-111 was a non-watch list single-shell tank with no previous indication of a potential safety issues. All waste receipts were 2C waste, 224 waste, or residues from the T-Plant cleanout. Flowsheet records indicate that small amounts of oxalate are present in 224 waste (at 0.028 M) (Schneider 1951). As a result of DSC measurements, tank 241-T-111 was added to the Organic Watch List on March 4, 1994 and further concerns have been raised about the tank's integrity.

When tank 241-T-111 was sampled, normal paraffin hydrocarbon was used as a hydrostatic fluid, presenting a potential contamination source and bias for the DSC. However, when taking the first segment, no normal paraffin hydrocarbon was used, and sample recovery for the other segments in the cores was excellent. No voids or separable liquid layer were observed on extrusion of the samples, and results from the gas chromatograph/mass spectrometer for normal paraffin hydrocarbons components showed only trace amounts. precluding significant sample contamination from the hydrostatic head fluid. Furthermore, experiments have demonstrated that normal paraffin hydrocarbons steam distill away before reaching the reaction temperatures observed in DSC assays of the samples. In the deeper segments and in the composites, core 31 demonstrates some activity, but not to the extent shown in the upper segments. In core 33, no significant exotherms are evident in the composite. The photographs of the extruded waste material demonstrated that the all of the core samples were viscous, gel-like materials with very little free liquid, and that they held their shape relatively well. During the physical testing of the core 31, segment 2 solid sample, when the sample was centrifuged, there was some separation of free water from the gel. Later physical testing did show some separation of liquid from the waste matrix for segment 7, but again, no separable phase was observed in the liquid (Delegard 1994). In each case, the liquid did not appear to have any distinct layers, therefore no liquid organics are considered to contribute to this waste matrix. In addition, although historically there have been problems with both volatile and semi-volatile organics analysis methods and

Figure 6-1. Waste Profile of Tank 241-T-111.



- Dished bottom and tank layer 1: 2C, 908,000 L (240,000 gal)
- Tank layer 2: 224 waste, 617,000 L (163,000 gal)
- Tank layer 3: T-Plant decontamination and 224 wastes, 198,000 L (53,000 gal)

holding times on these samples, all of the results from these assays show levels of these compounds below EPA contract laboratory procedure quantitation limits.

This information, coupled with data from the organic vapor monitor taken during the recent liquid sample effort (Appendix C), supports the contention that there are no substantial liquid organics in the tank. The organic vapor monitor results from the vapor space of tank 241-T-111 show a reading of 9.2 parts per million over a three minute monitoring interval: a detectable concentration, but far below the established safety criterion of 20 percent of the lower flammability limit. This result from the organic vapor monitor is not unexpected because small quantities of ammonia were observed in the grab samples and historically ammonia was used as a process chemical in the tank wastes. Over time, the ammonia is believed to have dissipated slowly to the levels observed today. It is also believed that the vapor space of this tank is relatively homogenous, and therefore, large concentration gradients for organics and other materials in the vapor space are not plausible. Differences of 2 to 10 times are within the realm of possibility (10 times being an extreme). Factors higher than that are not considered credible. For comparison, the vapor space of tank 241-C-103, which has a known organic liquid layer, was found to be 200 ppm. Furthermore, a simplified modeling scenario of tank 241-T-111 shows that if normal-paraffin-hydrocarbon-type liquid organics were present, they would be in much greater quantities than detected in the organic vapor monitor (see Appendix C).

To conclude, the waste in tank 241-T-111 is a complex material, primarily made up of water and organic and inorganic salts in a gel-like matrix. The insoluble solids are a mixture of phosphates, silicates, hydrated oxides, and hydroxides in combination with calcium, chromium, lanthanum, iron, bismuth, manganese, and uranium. The soluble analytes are primarily sodium, nitrate, sulfate, and fluoride. Phosphorous is nearly evenly divided between its soluble and insoluble forms. Substantial exotherms were detected in segments 1, 2, and 3, with the reactivity tailing off in segment 4 of core 31 and in segments 1, and 2, with the reactivity tailing off in segment 3 of core 33. The exotherms were reported to be similar in size and temperature range. It is important to note that no exotherms are observed until the sample has been dried and heated to approximately 180 °C. The exotherms themselves do not appear to be kinetically fast, thus the reaction may not be able to sustain itself without being thermally driven, as they are in the DSC apparatus. The organics present appear to be in the form of slightly soluble salts, contributing to a gel or sol-like structure. This behavior would be consistent with the very high moisture content observed in these tanks and with the types of organics historically indicated to be in the waste: aqueous decontamination solutions using surfactants, not normal paraffin hydrocarbons or TBP-type organics. This would help explain their low volatility and reactivity, their presence in the solid phase, and their relative absence in the liquids. These materials are carbon-bearing and will, if given enough impetus in the form of a thermal driver, react; however, the reaction does not appear rapid or self-sustaining, and will not occur without first removing the water from the gel. No other safety issue was found after critically reviewing the analytical, historical, and surveillance data.

Hypotheses Regarding Recent Behavior of 241-T-111

Recently, surveillance data indicated that there had been a relatively abrupt level change in tank 241-T-111 over the past 16 to 18 months. Approximately 2.5 cm (1 in) of gradual sludge growth was observed over the last eight-year period, followed by an approximately 4.1-cm (1.6-in) decrease in surface level over 16 months. This situation is cause for concern. Tank 241-T-111 is an assumed leaker, therefore there is the possibility that the tank has leaked again and waste is being lost to the environment. This is not an acceptable condition. Tank Farm Operating Procedures (Boyles 1992) specify that when a level drop in a tank is observed, the observation must be accompanied by certain actions, and that a leaking tank must be pumped to eliminate any remaining drainable liquid to prevent further environmental contamination. However, the observed level decrease was not large enough to trigger action past investigation under present guidelines (Boyles 1992), and it was not resolved that a leak was the only explanation for the observed behavior. In fact, there were observers who believed there was another safety issue requiring consideration involved. The final outcome of the review, however, was the decision to pump the liquid in the saltwell of tank 241-T-111 to tank 241-SY-102 (Jenkins and Engelman 1994). This section recaps some of the alternative explanations presented for the surveillance data and provides some of the strongest and weakest points of the arguments put forth.

• Tank is a "Re-leaker"

Evidence supporting this point of view is strong and it is one of the most favored explanations for the observed level drop. Tank integrity was questionable for a long period of time, and the tank was declared a leaker in 1984. Corrosion of tank liner is also evident from in-tank photographs. However, the change in surface level could be a localized phenomenon and in-tank photographs and observations of the extruded core material indicate that the waste is viscous and cohesive. Substantial damage to the tank-liner and shell would be necessary for the evacuation of over 11,000 L (3,000 gal) of material and this condition is not indicated.

Long Cycle Gas-Release Tank

This proposition is more speculative, but warrants consideration. Cyclic, gas generating behavior previously has been observed in Hanford-Site waste tanks. Specifically, tank 241-T-110, the cascade source for tank 241-T-111, is on the Gas-Generation Watch List. Viscous waste material, such as that observed in tank 241-T-111, also appears to be able to retain generated gases. However, overall tank information, specifically the analytical chemistry results, indicate tank conditions necessary for gas generation as currently understood (i.e. high radionuclide and complexant levels), do not exist.

• Intrinsic Waste Matrix Changes

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This explanation is even more tenuous than the previous one, but again, the present understanding of the high-level waste tank matrices is very limited, so it is within the realm of possibility. Composition and structure of the waste (small particles, high water content, high ionic strength solution, and possible organic surfactants), may be similar to an emulsion. Over time, the low-level radiolytic action and thermal cycling of the waste may break down the colloidal characteristics of the matrix. The breakdown may be uneven, and there may even be a slurry-growth phase before the separation of the emulsion and the loss of free water through pinhole corrosion. However, that same degree of uncertainty regarding the content and structure of the waste matrix makes this hypothesis highly speculative without much more characterization information than currently exists.

• Structural Subsidence of the Waste

This particular condition may be a contributing factor to the observed behavior, since it is acknowledged that the overall configuration of the waste generally is not very well known (and that the surfaces of the waste beds can be highly irregular), however it is not considered the main cause. After disposal into the tank, irregularities in the waste bed may have formed. The shifting of the waste bed over time as a result of gravitational compression is potentially responsible for the sudden drop in waste level. However, this rationale does not adequately explain the observed increase in tank level without involving another agent (i.e. a slow tank intrusion). Also, further slumping of the waste was not observed in the recent in-tank photographs, and the waste surface was observed to be reasonably uniform.

Other Possibilities or Combinations of the Preceding Agents

There may be more than one mechanism involved in the observed behavior, such as a long-term slow intrusion coupled with a subsequent upset and relatively rapid loss of material, especially of liquid, because the matrix appears to be quite viscous. There are a limited number of measurements taken and their location is fixed; therefore, local irregularities may be exacerbated. In addition, measurement error and bias also are significant when considering the magnitude of the drop in comparison with the error band.

6.3 TWRS PROGRAM ELEMENT CHARACTERIZATION SYNOPSIS

This section provides selected results obtained from core sampling for some of the most pertinent analytes for the various TWRS program elements, including vitrification, Retrieval, Pretreatment, and Waste Tank Safety. Analytes of interest will be reported on a level of resolution commensurate with the available data and program direction. Watch-list tanks will have segment or subsegment level analyses reported, while non-watch-list tanks are analyzed on a core composite basis. Analytes of interest to multiple programs generally will be reported only in one section. Further detail can be found in the body of the report or in the data packages.

6.3.1 Retrieval Program Data Summary: Physical Properties

A major objective of the characterization program is to measure the physical properties of the waste to support waste retrieval technology development. The analytical methods to determine the physical properties of the waste as it actually exists in the tank require 50 to 100 g of unhomogenized sample. In some cases, the limited amount of sample recovered constrains the number of analyses that can be performed. At the time of the sampling and analysis of tank 241-T-111 waste, no data quality objective existed to define the scope of the analyses. However, several analytes that specifically relate to physical properties were determined to be of interest to the program and are summarized here. The physical characteristics of tank waste are required to develop design criteria for waste retrieval equipment, provide a basis for simulated waste development, and to provide a basis for validation of equipment testing using design criteria and simulated waste. Selected rheological and physical properties are presented in Table 6-1. Further information regarding these analytes can be found in Section 5.3.

Analyte Data range Specific gravity (g/mL) --solids 1.19 - 1.28 --liquids (grab sample) 1.036 Shear strength $5,000 \pm 2,300 \text{ dynes/cm}^2$ Viscosity (mPa●s) --1:1 dilution @ 29 °C Less than 2 mPa s (cP) Settled solids (Vol %) 100% Weight % solids 22.4 to 29.3 Weight % undissolved 19.0 to 25.4 solids Particle size (µm) --number distribution $85\% < 2 \mu m$ --volume distribution $70\% < 24 \mu m$

Table 6-1. Retrieval Program.

6.3.2 Final Disposal Program Data Summary

Bulk Constituent Concentrations for Pretreatment

Programmatic decisions pertaining to the design of pretreatment and final disposal systems shall be based upon the average characteristics of the tank waste. Therefore, the majority of the laboratory analyses shall be conducted on representative core composites. However, as noted in other documentation (Bell 1993), segment, subsegment, and additional analyses will be performed when directed. The constituent concentrations and inventories shall be calculated by either treating the core samples as random samples and averaging the results, or by using a spatial model. The calculated values will include an estimated total quantity of each selected analyte and its corresponding confidence interval (CI) based upon analytical and sampling variability. Again, no data quality objective existed to define the scope of the analyses at the time tank 241-T-111 was sampled and analyzed. However, several analytes relating specifically to the most significant chemical and radiological contributors and their solubility properties were determined to be of interest to the program and are summarized here. Chemical analytes of interest are presented in Table 6-2. Trace analytes and more comprehensive chemical and radiological characterization information can be found in Section 5.

Low-level and High-Level Vitrification Program

The final disposal option for Hanford-Site wastes has been determined to be vitrification after partitioning into low-level and high-level fractions. This program has characterization needs in addition to those described for core sampling. The vitrification process will be performed after the solids have been pretreated. Therefore, the core sample information will provide preliminary bounding design conditions for the vitrification plant. Further characterization for technology development and regulatory compliance will be necessary on the pretreated waste that will be fed to the vitrification plant. Although the data requirements for this option are not formally defined, the analytical requirements for the previous Hanford Waste Vitrification program generally are applicable and are identified in the Hanford Waste Vitrification Plant Feed Characterization Requirements, Revision 4 (Wagner 1992). These requirements are quite similar to the pretreatment program requirements, and therefore are presented together in this section (see Table 6-2). For more specific information on a particular analyte not given in this table, consult the data package (McKinney, et al. 1993) or the appropriate table in Section 5.

The analytical program for vitrification not only entails determining if a waste type is suitable for disposal as glass, but also includes determining the physical and chemical characteristics of the glass for process-control purposes and to ensure regulatory compliance. Sampling and analysis plans will be developed on an individual basis for each tank or process batch. The characterization needs for these efforts include analyses for metals, water-soluble anions, radionuclides, semi-volatile organics, and rheological and physical testing for both the feed and vitrified product.

Table 6-2. Concentrations and Solubility of Principal Waste Components.

Analyte	Tank average concentration water prep: (μg/g)	Tank average concentration fusion prep: (µg/g)	% Water soluble
Calcium	61.8	2,420	2.55
Chromium	218.3	1,980	11.03
Iron	127.7	18,500	0.69
Aluminum	10.9	570	1.91
Sodium	33,000	37,000	89.19
Bismuth	201.8	26,000	0.78
Lanthanum	11.0	4,220	0.26
Silicon	571.8	5,670	10.08
Uranium	No measurement	3,550	===
Zirconium	0.8	4.0	20.00
Phosphate	15,600 (IC)	32,300	48.29
Sulfate	3,550 (IC)	3,680	96.47
Nitrate	41,300	Not Applicable	
Fluoride	2,300	Not Applicable	
TOC	3,120	Not Applicable	
Radionuclides	(μCi/g)	(μCi/g)	
⁹⁰ Sr	0.00097 (Grab)	5.41	0.018
¹³⁷ Cs	0.087 (Grab)	0.166	52.33

Tank 241-T-111 presently is not scheduled as an early feed for pretreatment and vitrification. The following characterization objectives need to be addressed in a data quality objective supporting the design of retrieval, pretreatment, and final disposal systems for early feed tanks.

• Provide extensive characterization of the chemical and radiological contents of the waste (solids and supernate) as it currently exists in the tanks to evaluate how it can be processed and to verify if the composition variability study envelope coverage for key analytes is adequate.

- Estimate the waste fraction that will remain after sludge wash pretreatment and estimate the feeds for the low-level and high-level streams for vitrification.
- Simulate sludge washing pretreatment on the waste material. This will provide a detailed understanding of the sludge wash process and obtain empirical data on soluble species removal.
- Determine the physical and rheological properties of the waste before and after simulated sludge washing to support the design of a waste retrieval system.
- Provide a supply of sludge washed material to be used as feed material for a laboratory scale vitrification.
- Satisfy the general characterization requirements for physical, chemical, and radiological analytes.

6.3.3 Waste Tank Safety Program Characterization Data Summary

Safety Screening

The tank safety screening data quality objective will be used to classify 149 SSTs and 28 double-shell tanks that contain high-level radioactive waste into specific safety categories for issues dealing with the presence of ferrocyanide, organics, flammable gases, and criticality (Babad and Redus 1994). The analytes used to make this classification are fuel energy value, total alpha concentration, weight percent moisture, and gas composition. The following table presents the analytes of concern, the criterion for classification, and the analytical result from the tank, where available. Further information on the tank contents are presented in subsequent sections. Because of the exothermic response of the waste material from the upper portions of tank 241-T-111, it has been placed on the Organic Watch List, and further studies trying to resolve the nature of the observed reaction are continuing. Table 6-3 provides a comparison of the tank values with the safety screening criteria.

Table 6-3.	Tank 241-T-111	Comparison to	Safety	Screening	Criteria.
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Analyte	Safety issue/criteria	Tank result
Fuel energy value (cal/dry g)	Organics, ferrocyanide, flammable gas; -125 cal/dry g	Greater than -215 cal/dry g
Total alpha concentration	Criticality; 1 g ²³⁹ Pu/L	0.0053 g/L
Percent moisture	Organics, ferrocyanide, flammable gas; 17 wt%	76 wt%
Gas composition	25% lower flammability limit	NA

NA = Not available

Criticality Safety

The criticality safety program has indicated that plutonium and uranium isotopic analyses on each core composite and the bottom six inches of each core is required to alleviate the concern for the potential of tank criticality. Therefore, upon extruding the last segment in a core, the waste to be tested shall be homogenized before a small aliquot is taken and analyzed for plutonium and uranium isotopic analyses by mass spectroscopy. The analyses will indicate whether the fissile species have settled in a concentrated layer at the bottom of a tank. However, this analytical criterion was established after the sampling and analysis of tank 241-T-111 waste. Tables 6-4 and 6-5 present a summary of the core composite data for uranium and plutonium concentration. After reviewing the data, the isotopic analyses from both cores 31 and 33 were found to be very consistent with regard to composition, and the total alpha content indicated from the core 33, segment 9 homogenization data is extremely low (0.262 μ Ci/g). Therefore further re-analysis and isotopic resolution of the core sample material from the lower portion of cores 31 and 33, segment 9 is not warranted. For this tank to exceed established operating limits for fissile material in the tank farms, a concentration of 1.58 μ Ci/g ^{239/240}Pu was calculated as a threshold limit value (see Appendix C).

Table 6-4. Core Composite Uranium.

Core No.	U _{PL} (222-S) (μg/g)	U _{FL} (325) (μg/g)	mass percent	mass percent
Core 31, composite 1	2,180	4,000	99.3074	0.6755
Core 31, composite 2	3,880	5,200	99.3098	0.6761
Core 33, composite 1	3,180	4,500	99.3125	0.6761
Core 33, composite 2		3,500	99.3161	0.6717

FL = Uranium measurement by laser fluorimetry.

Table 6-5. Plutonium Concentration and Isotopic Distribution

Core number	Total Pu α (222-S) (μCi/g)	Total Pu α (325) (μCi/g)	²³⁸ Pu mass percent	²³⁹ Pu mass percent	²⁴⁰ Pu mass percent	²⁴¹ Pu mass percent	²⁴² Pu mass percent
Core 31 C1	0.138	0.628	0.005	96.7199	3.2109	0.0352	0.0151
Core 31 C2	0.136	0.565	0.0105	96.6351	3.2834	0.0496	0.0215
Core 33 C1	0.134	0.319	0.004	96.7540	3.1046	0.1071	0.0683
Core 33 C2	0.147	0.368	0.0105	96.5499	3.3436	0.0621	0.0337

Organic Tanks

The following characterization objectives support resolution of this unreviewed safety question and safety issue and support retrieval, pretreatment, and final disposal systems design. Table 6-6 provides a comparison of the tank core composite values with the Organic Data Quality Objective Criteria.

- Determine the overall waste energetics and properties governing waste reactivity behavior in the tanks.
- Determine the spatial distribution of ¹³⁷Cs and ⁹⁰Sr.
- Determine the concentration of TOC and the speciation of organics present in the waste.
- Satisfy the general characterization requirements for physical, chemical, and radiological analytes.

Table 6-6. Data Quality Objective Decision Limits for Organic Tanks. (Babad 1994)

Analyte	Decision threshold	Tank result
TOC (Dry wt%)	> 5 wt% (dry basis)	1.3 wt% (tank) 4.1 wt% (Core 33, Seg. 2)
Moisture content (wt%)	< 17 wt%	76 wt%
Presence of organic layer	Yes/No	No
Tank temperature (°C)	90 °C	16 °C
Total fuel content	-125 cal/dry g	In excess of -215 cal/dry g

Further analysis and secondary analytes for measurement were found to be unwarranted for the tank composites; however, the first two segments of core 31 and three segments of core 33 will undergo additional study. The results from this series of assays will be incorporated into the revision of this characterization report as they become available.

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7.0 QUANTITATIVE/STATISTICAL INTERPRETATION OF THE DATA

This section contains the results of the statistical analysis of data from two core samples obtained from tank 241-T-111. Section 7.1 contains a description of the core-sample data used in the statistical analyses and some general observations regarding the data. Section 7.2 contains mean concentration estimates, and the associated 95 percent CIs, for each of the analytes in tank 241-T-111 waste. Section 7.3 contains estimates of the spatial variability (variability between cores), and estimates of the analytical error from the core composite data in tank 241-T-111. Further information can be found in Jensen (1994).

Two types of analytical error were estimated from the core composite data: variability between composite samples within the same core and variability between the primary and duplicate analyses within each core composite sample. Estimates of the analytical measurement error were used to test the significance of the spatial and compositing variability. Spatial variability was significant (i.e. substantially greater than zero at the 0.05 significance level) in 40 out of 79 analytes in the tank. The compositing variance was significant for 38 out of the 79 analytes.

7.1 APPROACH

Cores 31-and 33 were the two valid core samples taken from tank 241-T-111. The segment recoveries for each core were given previously in Section 4. Two core composite samples were made for each core from the homogenized solid segment waste. Primary and duplicate results were obtained from each core composite.

The ICP acid digestion, ICP fusion dissolution, ICP water leach, IC water leach analyses, selected radiochemistry, and other GEA were performed on all composite core samples. These were the analytical results used in the statistical treatment of the data. In the tables in Appendix B, the data are identified by the analysis method, the type of dissolution, and analyte; e.g., the notation ICP.a.Al refers to aluminum, acid digestion, and an ICP analysis. The core composite sample results used also are contained in Appendix B, together with ratios of the mean of each sample and duplicate divided by the detection limit for that pair. The data package for tank 241-T-111 (McKinney et al. 1993) contains a complete report of the sample results along with the laboratory quality control data. The core composite data for each analyte are illustrated in Appendix B.

Statistics were calculated for analytes with concentrations greater than 10 times their detection limits (DL). Personnel within the TWRS Information Management Systems have identified a list of analytes that have exception to this rule. Table 7-1 lists the analytes specified. Statistics were calculated for the analytes from this list if the concentrations were greater than three times their DL. For a number of analytes, the concentrations of some samples were greater than a particular limit (3 or 10 times the DL), while the other samples were less than that limit. In these cases the statistics were calculated using all of the data

whether it was above or below the particular limit (3 or 10 DL). The above rules do not apply to alpha or beta/gamma counting methods. The ratios (mean/DL) reported in Appendix B are provided to show how large the analyte concentrations are relative to the DL.

Table 7-1. Special Analyte List.

Aluminum	Nitrate
Bismuth	Nitrite
Calcium	Phosphate
Chromium	Carbonate
Iron	Fluoride
Silicon	Chloride
Sodium	TOC
Zirconium	Cyanide
	-

A close examination of the figures in Appendix B reveals several outlier data points in the core composite data:

ICP.a.Co: The primary result for core 31, composite 2 of 11.7 ug/g is over three times the duplicate result. The other results for ICP.a.Co all fall in the range 2.7 ug/g to 3.8 ug/g. The detection limit for ICP.a.Co is 0.8 ug/g.

ICP.a.Cu: The duplicate result for core 31, composite 2 of 127 ug/g is about four times the primary result of 31.7 ug/g. The detection limit for ICP.a.Cu is 0.4 ug/g.

NO₂: Both the primary and duplicate results from composite 2 of core 31 are about half the results for core 31, composite 1. The average of the results for core 31, composite one is 952 ug/g; the average for composite two is 525 ug/g. The detection limit for NO₂ by water digestion spectrophometric analysis is 50 ug/g.

7.2 MEAN CONCENTRATION ESTIMATES

One of the tasks outlined as part of the waste characterization effort (Bell 1993, Winters et al. 1990a, Winters et al. 1990b), is to estimate the constituent inventories in the waste. The inventories are estimated by computing mean concentrations and 95 percent CIs on the mean concentrations for each constituent. The estimate of the inventory and CI on the inventory of an analyte in the tank are equal to the corresponding mean concentration estimates and CI multiplied by the volume of waste in the tank.

7.2.1 Statistical Methods

The concentration estimates are given in the form of 95 percent CIs on the mean concentration. It is assumed that each sample and its duplicate are analyzed independently of one another. The two analytical results are used to estimate the analytical measurement error. Because of the hierarchical structure of the data, the analytical measurement error alone is not the appropriate error term to use in computing the CIs. A linear combination of the analytical measurement variance and the spatial variance is the appropriate variance of the mean for the CIs. Appendix B contains a description of the statistical model and formulas used to calculate estimates of the mean, variance of the mean, and the CI on the mean.

7.2.2 Statistical Results

Table 7-2 contains the summary statistics by analyte for ICP acid digestion, ICP water leach, ICP fusion dissolution, IC, and selected radiochemical and physical analyses. The summary statistics are as follows:

- y mean of the concentration data
- $\hat{\sigma}^2(\bar{y})$ estimated variance of \bar{y}
- df degrees of freedom
- 95% LL lower limit to the 95 percent CI on the mean
- 95% UL upper limit to the 95 percent CI on the mean.

For some analytes the lower confidence limit (95 percent LL) was negative. Because concentrations are greater than or equal to zero, any negative 95 percent LL values were set equal to zero.

The CIs in Table 7-2 are wide relative to the range of the data. The CIs are wide because only two cores were used to estimate the spatial variability. A minimum of two core samples is needed to estimate a tank's spatial variability.

Table 7-2. Concentration Estimate Statistics. (4 pages) (Units μ g/g Except Radionuclides μ Ci/g)

Analyte	ÿ	$\hat{\sigma}^2(\bar{y})$	df	95% LL	95% UL
ICP.a.Ag	1.26E+02	7.86E+03	1	0.00	1.25E+03
ICP.a.Al	5.41E+02	1.06E+04	1	0.00	1.85E+03
ICP.a.B	2.80E+01	7.56E+00	1	0.00	6.30E+01
ICP.a.Ba	6.90E+01	6.46E+01	1	0.00	1.71E+02
ICP.a.Bi	2.59E+04	6.38E+06	1	0.00	5.80E+04
ICP.a.Ca	1.88E+03	2.12E+05	1	0.00	7.72E+03
ICP.a.Cd∗	5.80E+00	3.03E+00	1	0.00	2.79E+01
ICP.a.Co●	4.30E+00	1.63E+00	1	0.00	2.05E+01
ICP.a.Cr*	1.98E+03	1.63E+04	1	3.57E+02	3.60E+03
ICP.a.Cu	3.35E+01	3.53E+02	1	0.00	2.72E+02
ICP.a.Fe	1.85E+04	1.21E+06	1	4.55E+03	3.25E+04
ICP.a.K	1.14E+03	2.24E+03	1	5.34E+02	1.74E+03
ICP.a.La	4.22E+03	3.00E+05	1	0.00	1.12E+04
ICP.a.Mg	3.77E+02	6.36E+03	1	0.00	1.39E+03
ICP.a.Mn	6.33E+03	2.68E+04	1	4.25E+03	8.41E+03
ICP.a.Na	3.69E+04	1.56E+06	1	2.10E+04	5.27E+04
ICP.a.Ni	1.32E+02	5.12E+02	1	0.00	4.19E+02
ICP.a.P	1.03E+04	1.21E+05	1	5.90E+03	1.47E+04
ICP.a.Pb	3.47E+02	2.64E+04	1	0.00	2.41E+03
ICP.a.S	1.21E+03	1.06E+03	1	8.00E+02	1.63E+03
ICP.a.Si	4.69E+02	9.17E+02	1	8.40E+01	8.54E+02
ICP.a.Sr	3.00E+02	3.75E+02	1	5.39E+01	5.46E+02
-ICP.a.Ti	1.95E+01	1.39E+02	1	0.00	1.69E+02
ICP.a.V	1.45E+01	6.58E+00	1	0.00	4.71E+01
ICP.a.Zn	6.50E+01	6.46E+02	1	0.00	3.88E+02

Table 7-2. Concentration Estimate Statistics. (4 pages) (Units μ g/g Except Radionuclides μ Ci/g)

Analyte	ÿ	$\hat{\sigma}^2(\bar{y})$	df	95% LL	95% UL
ICP.f.Ag	1.28E+02	8.05E+03	1	0.00	1.27E+03
ICP.f.Al	5.70E+02	9.70E+03	1	0.00	1.82E+03
ICP.f.Ba	6.46E+01	2.45E+01	1	1.73E+00	1.28E+02
ICP.f.Bi	2.36E+04	9.08E+06	1	. 0.00	6.18E+04
ICP.f.Ca*	2.42E+03	8.27E+04	1	0.00	6.07E+03
ICP.f.Cd*	8.12E+00	1.76E+00	1	0.00	2.50E+01
ICP.f.Cr	1.80E+03	-1.56E+03	1	1.30E+03	2.30E+03
ICP.f.Cu	2.93E+01	3.56E+01	1	0.00	1.05E+02
ICP.f.Fe	1.80E+04	4.05E+06	1	0.00	4.36E+04
ICP.f.La	4.11E+03	3.08E+05	1	0.00	1.12E+04
ICP.f.Mg	3.55E+02	7.31E+03	1	0.00	1.44E+03
ICP.f.Mn	6.28E+03	1.88E+04	1	4.54E+03	8.02E+03
ICP.f.Na	3.70E+04	6.00E+06	1	5.82E+03	6.81E+04
ICP.f.Ni	8.14E+03	6.41E+06	1	0.00	4.03E+04
ICP.f.P	1.04E+04	8.42E+05	1	0.00	2.21E+04
TCP.f.Pb*	3.65E+02	9.38E+03	1	0.00	1.60E+03
ICP.f.S	1.23E+03	1.13E+04	1	0.00	2.58E+03
ICP.f.Si	5.67E+03-	5.41E÷04	1	2.71E+03	8.62E+03
ICP.f.Sr	2.98E+02	6.24E+01	1	1.97E+02	3.98E+02
ICP.f.Ti	4.79E+01	6.09E+02	1	0.00	3.62E+02
ICP.f.Zn∗	1.06E+02	7.17E±00	1.	-7.22E+01	1.40E+02
ICP.w.Al●	1.09E+01	5.75E+00	1	0.00	4.14E+01
ICP.w.Bi	2.02E+02	2.40E+03	1 -	0.00	8.24E+02
ICP.w.Ca∗	6.16E+01	3.32E+01	1	0.00	1.35E+02
ICP.w.Cr	2.18E+02	2.45E+01	1	1.55E+02	2.81E+02
ICP.w.Fe	1.28E+02	3.11E+02	1	0.00	3.52E+02

Table 7-2. Concentration Estimate Statistics. (4 pages) (Units μ g/g Except Radionuclides μ Ci/g)

Analyte	ÿ	$\hat{\sigma}^2(\bar{y})$	df	95% LL	95% UL
ICP.w.K	7.19E+02	1.54E+03	1	2.21E+02	1.22E+03
ICP.w.La*	1.10E+01	1.42E+01	1	0.00	5.89E+01
ICP.w.Mg★	3.64E+00	5.45E-02	1	6.75E-01	6.61E+00
ICP.w.Mn	2.47E+01	2.36E+01	1	0.00	8.65E+01
ICP.w.Na	3.30E+04	2.44E+06	1	1.31E+04	5.28E+04
ICP.w.P	5.68E+03	3.24E+04	1	3.39E+03	7.97E+03
ICP.w.S	1.15E+03	2.38E+03	1	5.29E+02	1.77E+03
ICP.w.Si	5.72E+02	5.35E+03	1	0.00	1.50E+03
IC.w.Cl	4.50E+02	1.11E+03	1_	2.56E+01	8.74E+02
IC.w.F	2.30E+03	6.46E+05	1	0.00	1.25E+04
IC.w.NO₂ ★	8.97E+02	2.10E+04	1	0.00	2.74E+03
IC.w.NO ₃	4.12E+04	7.77E+06	1	5.82E+03	7.67E+04
-IC.w.PO ₄ 3-	1.55E+04	1.53E+06	1	0.00	3.13E+04
IC.w.SO ₄ ² -	3.54E+03	2.85E+04	1	1.40E+03	5.69E+03
GEA.Am-241★	4.24E-02	2.61E-06	1	2.19E-02	6.29E-02
GEA.Co-60	3.64E-04	2.68E-10	1	1.56E-04	5.72E-04
GEA.Cs-137	1.66E-01	3.35E-03	1	0.00	9.02E-01
Gross.alpha	3.73E-01	1.96E-04	1	1.95E-01	5.51E-01
Gross.beta	1.51E+01	3.48E+01	_ 1	0.00	9.00E+01
TGA.Percent.H ₂ 0	7.65E+01	2.23E+01	1	1.64E+01	1.37E+02
NO ₂	7.93E+02	8.76E+03	1	0.00	1.98E+03
Percent.H ₂ O	7.60E+01	5.81E-01	1	6.63E+01	8.57E+01
Pu-239/240	1.39E-01	9.19E-06	1	1.00E-01	1.77E-01
Sr-90	5.41E+00	3.53E+00	1	0.00	2.93E+01
TOC*	3.12E+03	3.83E+05	_1 _	0.00 -	1.10E+04
Tc-99●	7.92E-03	8.90E-06	1 .	0.00	4.58E-02

Table 7-2. Concentration Estimate Statistics. (4 pages) (Units $\mu g/g$ Except Radionuclides $\mu Ci/g$)

Analyte	ÿ	$\hat{\sigma}^2(\bar{y})$	df	95% LL	95% UL
U∗	2.79E+03	2.01E+05	1	0.00	8.50E+03
рН	9.98E+00	7.79E-03	_	8.86E+00	1.11E+01

- Analytes with a portion of the data below 3 times the DL.
- *: Analytes with a portion of the data below 10 times the DL.

7.3 COMPARISON OF THE VARIANCE COMPONENT ESTIMATES

Using the hierarchical structure of the core composite data, estimates of the between-core spatial variability, the compositing variability, and the analytical-measurement variability can be obtained. The spatial variance is a measure of the variability between cores. The compositing variance measures the variability between composite samples within the same core. The analytical-measurement variance is a measure of the difference between the analytical results from the sample and duplicate samples. This variance includes, among other things, the sample handling error and the chemical analysis error.

The estimate of the variance of the mean is a linear function of the spatial, compositing, and analytical-measurement-variances. To help evaluate the magnitude of these three variance components, estimates of each variance component are given.

7.3.1 Statistical Methods

Estimates of the spatial variance $(\hat{\sigma}^2(S))$, compositing variance $(\hat{\sigma}^2(C))$, and analytical-measurement variance $(\hat{\sigma}^2(A))$, were obtained for each analyte using restricted maximum likelihood estimation methods. Restricted maximum likelihood estimation is discussed by Harville (1977).

To test the significance of the variance components, an analysis of variance (ANOVA) was calculated using the hierarchical statistical model described in Appendix B. The mean square error terms in the ANOVA table were used to perform an F-test on the spatial variability and the composite variability. The p-values given in Table 7-3 were derived from the results of these tests.

7.3.2 Statistical Results

The restricted maximum likelihood estimates of each component of variability along with the p-values (significance level) from the F-tests also are given in Table 7-3. P-values less than 0.05 indicate that $\sigma^2(S)$ or $\sigma^2(C)$ is significantly different from zero at the 0.05 significance level.

The p-values from the tests on $\sigma^2(S)$ were less than 0.05 for 40 out of the 79 analytes in tank 241-T-111 waste. Thus, for these 40 cases, differences between the results in the two cores were statistically significant. The p-values from the tests on $\sigma^2(C)$ were less than 0.05 for 38 out of the 79 analytes in tank 241-T-111 waste. This indicates that, relative to the analytical error, differences between composite samples were significantly greater than zero in 38 cases. Conversely, for 41 out of 79 cases, differences between composite samples were not statistically significant. The number of analytes (and the amount that they contribute to the waste) for which $\sigma^2(S)$ and $\sigma^2(C)$ were statistically significant further suggests that the waste is heterogeneous.

7.4 MASS BALANCES

A method to help ensure the data are consistent and reasonable is to perform a mass and charge balance on the core composite sample data. This activity is a rough quality-control check and provides insight to some of the properties of the matrix. To do this, the assumption in performing the mass balance is that the anions, cations, and water are all associated in some manner, but the exact chemistry of the association is not considered. Analytes contributing less than 0.2 weight percent, generally trace ICP analytes, AA analytes, and radionuclides, are considered negligible in this assessment. The assays that will contribute analytes to the mass balance are ICP acid or fusion (whichever gives higher quantitation), IC, TOC, and the gravimetric weight-percent water measurement.

Without considering the physical and chemical properties of the waste matrix and the context of the process history, the mass balances produced from these assays may be biased low. However, this bias is expected because it is known that there are analytes present that were not measured in the analysis of the samples. The IC anions only measure the water-soluble components; there is a substantial insoluble residue that must contain additional anions. Bias may be impacted substantially by chemical form, accountability, and variability in oxide or hydroxide content. Assumptions regarding the chemical combination of some of the analytes will be made and inserted into the mass/charge balance, presented in Table 7-4. Generally, this consists of assuming that some analytes are precipitated as an oxide or hydroxide, and that the shortfall indicated from the charge balance in microequivalent [μ equ.] is present as one of those two analytes.

[&]quot;Statistically significant for the purposes of this analysis means substantially greater than zero at the 0.05 level of significance.

Table 7-3. Variance Components Estimates. (3 pages)

Analyte -	∂²(S)	Test: $\sigma^{2}(S) = 0$ p-value	∂²(C)	Test: $\sigma^2(C) = 0$ p-value	∂²(A)
ICP.a.Ag	1.56E+04	0.001	1.94E+02	0.000	3.16E+00
ICP.a.Al	1.89E+04	0.055	4.79E+03	0.000	1.06E+01
ICP.a.B	1.25E+01	0.093	9.28E-01	0.387	8.59E+00
ICP.a.Ba	6.91E+01	0.263	1.20E+02	0.000	3.54E-01
ICP.a.Bi	1.27E+07	0.000	6.25E+03	0.360	3.75E+04
ICP.a.Ca	4.10E+05	0.010	2.17E+04	0.074	9.95E+03
ICP.a.Cd★	.5.96E+00	0.004	1.90E-01	0.039	5.34E-02
ICP.a.Co∗	8.92E-01	0.376	8.08E-01	0.389	7.84E+00
ICP.a.Cr	3.17E+04	0.007	1.36E+03	0.081	6.75E+02
ICP.a.Cu	3.39E+02	0.290	1.67E+02	0.369	1.14E+03
ICP.a.Fe	2.33E+06	0.013	1.55E+05	0.063	6.25E+04
ICP.a.K	2.45E-24	0.911	8.86E+03	0.000	1.75E+02
ICP.a.La	5.91E+05	0.003	1.63E+04	0.019	2.88E+03
ICP.a.Mg	1.25E+04	0.005	5.18E+02	0.001	1.58E+01
ICP.a.Mn	2.97E+04	0.254	4.29E+04	0.030	1.00E+04
ICP.a.Na	2.76E+06	0.057	6.73E+05	0.016	1.05E+05
ICP.a.Ni	1.02E+03	0.000	6.13E+00	0.091	3.38E+00
ICP.a.P	2.37E-22	0.608	4.77E+05	0.000	1.25E+04
ICP.a.Pb	5.20E+04	0.003	1.41E+03	0.000	1.91E+01
ICP.a.S-	1.11E+03	0.268	1.94E+03	0.003	1.25E+02
ICP.a.Si	2.57E-31	0.946	7.61E-14	0.418	7.34E+03
ICP.a.Sr	6.38E+02	0.079	2.18E+02	0.004	1.54E+01
ICP.a.Ti	2.75E+02_	0.001	4.69E+00	0.000	7.83E-02
ICP.a.V	1.83E+00	0.423	2.25E+01	0.000	2.30E-01
ICP.a.Zn	1.22E+03	0.021	1.35E+02	0.008	1,42E+01
ICP.f.Ag	1.61E+04	0.000	6.23E+00	0.295	1.83E+01
ICP.f.Al	1.90E+04	0.005	6.78E+02	0.030	1.57E+02

Table 7-3. Variance Components Estimates. (3 pages)

Analyte	∂²(S)	Test: $\sigma^2(S) = 0$ p-value	∂²(C)	Test: $\sigma^2(C) = 0$ p-value	ô²(A).
ICP.f.Ba	3.99E+01	0.102	1.71E+01	0.012	2.19E+00
ICP.f.Bi	1.81E+07	0.001	2.88E+04	0.392	2.94E+05
ICP.f.Ca±	1.51E+05	0.008	2.07E-19	0.804	5.86E+04
ICP.f.Cd∗	2.17E+00	0.126	1.21E-20	0.649	5.38E+00
ICP.f.Cr	3.72E-15	0.994	5.63E+03	0.013	1.21E+03
ICP.f.Cu∗	6.97E+01	0.005	2.00E+00	0.119	1.43E+00
ICP.f.Fe	8.00E+06	0.003	1.68E+05	0.079	8.13E+04
ICP.f.La	5.96E+05	0.010	3.82E+04	0.012	4.98E+03
ICP.f.Mg	1.46E+04	0.000	5.15E-44	0.913	1.60E+02
ICP.f.Mn	1.66E-11	0.553	7.02E+04	0.011	9.70E+03
ICP.f.Na	1.17E+07	0.007	5.43E+05	0.044	1.65E+05
ICP.f.Ni	1.17E+07	0.033	4.18E-12	0.498	4.39E+06
ICP.f.P	1.57E+06	0.028	1.95E+05	0.049	6.45E+04
ICP.f.Pb★	1.85E+04	0.003	4.35E+02	0.030	1.02E+02
ICP.f.S	2.15E+04	0.017	2.00E+03	0.008	2.13E+02
ICP.f.Si	1.06E+05	0.005	1.00E+03	0.369	6.83E+03
ICP.f.Sr	6.12E-16	0.661	1.15E+02	0.222	2.69E+02
ICP.f.Ti	1.22E+03	0.000	2.66E-01	0.331	1.12E+00
ICP.f.Zn∗	5.69E-23	0.440	1.10E-22	0.839	5.74E+01
ICP.w.Al●	7.35E+00	0.208	6.93E+00	0.061	2.72E+00
ICP.w.Bi	3.90E+03	0.102	1.30E+03	0.123	9.66E+02
ICP.w.Ca*	1.86E-11	0.144	1.69E-22	0.862	2:66E+02
ICP.w.Cr	5.66E-20	0.979	9.52E+01	0.001	5.62E+00
ICP.w.Fe	6.16E+01	0.436	9.05E+02	0.077	4.29E+02
ICP.w.K	2.28E+03	0.148	1.57E+03	0.001	4.73E+01
ICP.w.La∗	2.71E+01	0.015	1.34E+00	0.226	2.23E+00
ICP.w.Mg★	3.04E-21	0.597	1. <u>61E-01</u>	_0.098	1.14E-01

Table 7-3. Variance Components Estimates. (3 pages)

Analyte	∂²(S)	Test: $\sigma^2(S) = 0$ p-value	∂²(C)	Test: $\sigma^2(C) = 0$ p-value	ô²(A)
ICP.w.Mn	2.51E+01	- 0.266	3.19E+01	0.130	2 .51E+ 01
ICP.w.Na	4.53E+06	0.030	6.84E+05	0.002	2.88E+04
ICP.w.P	3.98E+04	0.222	3.94E+04	0.089_	_2.13E+04
ICP.w.S	3.95E+03	0.091	1.50E+03	0.014	2.13E+02
ICP.w.Si	8.53E+03	0.113	1.98E+03	0.272	4.72E+03
-IC.w.Cl	-1.99E+03	0:043	1.15E-10	0.509	9.66E+02
IC.w.F	1.28E+06	0.001	1.37E+04	0.102	8.44E+03
IC.w.NO₂:★	3,99E+04	0.015	5.22E-16 -	0.510	8.64E+03
IC.w.NO ₃	-1.42E+07	0.038	2.46E+06	0.013	3.44E+05
IC.w.PO ₄ ³⁻	2.59E+06	0.082	8.54E+05	0.030	1.99E+05
IC.w.SQ ₄ ² -	5.27E+04	0.032	7.04E+03	0.071	3.14E+03
GEA.Am-241★	2.99E-32	0.786	8.13E-06	0.062	4.59E-06
GEA.Co-60	3.58E-10	0.046	5.14E-36	0.813	7.12E-10
GEA.Cs-137	6.61E-03	0.003	1.87E-04	0.000	1.88E-06
Gross.alpha	3.42E-04	0.065	7.74E-05	0.098	4.58E-05
Gross.beta	6.93E+01	0.000	3.14E-01	0.026	6.62E-02
TGA.%.H ₂ 0	3.17E+01	0.008	3.78E-33	0.935	5.18E+01
NO ₂	1.92E-14	0.789	3.46E+04	0.000	9.50E+02
Percent H ₂ O	8.99E-01	0.128	2.55E-01	0.259	5.46E-01
Pu-239/240	2.49E-28	0.736	2.01E-05	0.171	3.33E-05
Sr-90	7.04E+00	0.000	1.79E-02	0.209	2.61E-02
TOC⋆	6.10E+05	0.113	2.98E+05	0.006	2.72E+04
_Tc-99●	1.76E-05	0.001	2.63E-07	0.027	5.75E-08
U⋆	6.24E-06	0.831	7.91E+05	0.001	2.99E+04
pН	7.11E-28	0.601	3.06E-02	0.001	1.19E-03

^{•:} Analytes with a portion of the data below 3 times the DL.

^{*:} Analytes with a portion of the data below 10 times the DL.

Table 7-4. Core 31 and Core 33 Mass and Charge Balance. (2 pages)

Analyte	Core 31 average concentration (µg/g)	Core 31 charge (µequ/g)	Core 33 average Concentration (µg/g)	Core 33 charge (µequ/g)	Core 31/33 RPD
Ca ⁺²	2,710	135.50	2,140	107.00	23.51
Cr ⁺³	1,850	106.73	2,100	121.15	-12.66
Fe ⁺²	20,100	717.86	16,000	571.43	22.71
Mn ⁺⁴	6,160	448.00	6,400	465.45	-3.82
Na ⁺	39,400	1,713.04	34,600	1,504.35	12.97
Bi ⁺³	23,500	337.32	28,500	409.09	-19.23
La ⁺³	3,670	79.21	4,770	102.95	-26.07
Si ⁺⁴	5,900	842.86	5,440	777.14	8.11
U-+6	3,820	96.30	3,280	82.69	15.21
PO ₄ 3-	30,100	-950.53	31,700	-1001.05	-5.18
SO ₄ ²⁻	3,650	-76.04	3,460	-72.08	5.34
NO ₃	44,100	-711.29	38,500	-620.97	13.56
F	3,110	-163.68	1,500	-78.95	69.85
TOC	3,740	-85.00	2,500	-56.82	39.74
Anion recon	ciliation				<u></u>
O ₂ 4-	17,116	-2,139.50	-16,926	-2,115.75	·
θН	5,965	-350.88	3,325	-195.59	
Water conter	nt				
H ₂ O	735,000	0	790,000	0	-7.21
H ₂ O Calc.	785,109	0	798,859	0	-1.74

Analyte	Core 31 average concentration (µg/g)	Core 31 charge (µequ/g)	Core 33 average Concentration (µg/g)	Core 33 charge (µequ/g)	Core 31/33 RPD
Total	949,891	-0.10	991,141	0.05	
Percent difference*	-5.01		-0.89		

Table 7-4. Core 31 and Core 33 Mass and Charge Balance. (2 pages)

 O_2^4 : Represents the overall mass and charge of oxygen added to manganese, uranium, silicon, TOC, and chromium as part of the assumptions given in Section 7.4. This notation is not meant to imply actual chemical form.

OH: Represents hydroxide amount calculated to reconcile charge balance NOTE: Neither of these analytes (oxygen or OH) are analytically determined.

--Relative Percent Difference (RPD) =
$$\frac{\text{Core 31 value - Core 33 value}}{\left(\frac{\text{Core 33 value + Core 33 value}}{2}\right)} \times 100$$

*Percent difference is determined from a Total of 1.0E+06 $\frac{\mu g}{g}$ i.e.,

A significant source of error can be reduced by assuming all phosphorous is present as PO₄³. The water digestion ICP values for phosphorous (converted to PO₄³) and PO₄³ values from the IC agree well. The ratio of soluble to insoluble phosphorous (taken as phosphate) indicates that it is only about 50 percent soluble. The process history of the tank also indicates that large amounts of phosphate were used to encourage precipitate formation. Therefore, an assumption that the phosphorus determined by ICP in the fusion acid/assay (and converted to phosphate) represents total PO₄³ is not unwarranted. The phosphorus in the ICP fusion assay is converted to PO₄³ and added to the other anions in the charge balance. This step will avoid double counting in the mass and charge balance calculations.

The following other assumptions will be made for the purposes of simplifying the calculations: manganese is assumed present as MnO_2 , uranium is present as $UO_2(OH)_2$, silicon is present as SiO_3 , TOC as $C_2O_4^{2-}$, and chromium as Cr_2O_3 . These forms are not the

only likely speciation of the analytes; however, the waste matrices are too complicated to represent every possible, or even probable compound present.

In the case of these waste materials, the disparity between the gravimetric water measurement and the TGA water content suggests (1) drying of the sample before the gravimetric assay; (2) incomplete drying during the gravimetric test, which biases the results low; or (3) one or more endothermic events occurring in the same temperature range (chemical reactions or phase transitions resulting in the loss of mass). One or more of these factors may be responsible for the observed trend. However, in this case, the analytical results and chemical assumptions that were made with regard to the waste matrix reconcile well.

7.5 SUGGESTED COMPONENTS OF WASTE MATRIX

The actual composition of the waste matrix is quite complex and trace amounts of various compounds probably exist in the tank. However, with some simple assumptions regarding how the anions and cations will combine, a list of the most probable compounds that exist in the waste matrix and contribute significantly to its overall makeup can be developed.

Table 7-5 is a condensed version of a more general chart found on page D-147 in the *Handbook of Chemistry and Physics 64th Ed.* (Weast 1984). It provides solubility data on some of the most common anions and cations. The oxidation state shown in the table for the cations is the most stable. However, precipitates may form for multivalent cations under varying conditions, and so precipitates are reported as likely, if conditions and anions in the assessment of the analyst warrant it.

From the chromatographic data, suspected solubility behavior, and process information, chloride, nitrite, and carbonate will not be significant mass contributors to the waste matrix. Sodium, SO_4^{2} , and NO_3 are highly soluble, and thus probably do not contribute much to the insoluble solids. However, they contribute significantly to the overall solids content of the waste (dissolved + insoluble solids). Phosphorous is one of the most prevalent analytes, is approximately 50 percent soluble, and contributes substantially to both the soluble and insoluble solids. No analytical measurement of hydroxide was made for the solids (although there was an OH assay of the grab sample), but it is known that in the process history of tank 241-T-111, basic solutions were added routinely to the tank. The following are likely candidates for the insoluble solids:

- Bismuth phosphate, BiPO₄
- Bismuth hydroxide, Bi(OH)₃
- Calcium fluoride, CaF₂

- Bismuth fluoride, BiF₃
- Bismuth trioxide, Bi₂O₃
- Calcium phosphate, Ca₃(PO₄),

2 to No. 4 to the face the

Table 7-5. Probable Solids in the Waste Matrix.

	NO;	PO ₄ 3-	-SO ₄ ²	F	OH-	Si (as SiO, ²)	Oxide 2	CO ₃ ² ·
Bi ⁺³		PPT		PPT	PPT		PPT	-
Ca ⁺²		PPT	PPT	PPT		PPT	PPT	PPT
Cr ⁺³	<u>i</u> ĝ	PPT	PPT	PPT	PPT	4	PPT	
Fe ⁺³		PPT		PPT	PPT		PPT	PPT
Na ⁺	127							
La ⁺³			PPT .	- PPT -	PPT	-	PPT	PPT
Mn ⁺⁴		- PPT	,,	- PPT	PPT	PPT	PPT	PPT
U+6	NL	PPT	PPT	NL	PPT		PPT	

PPT = Precipitate forms.

NL = Precipitate formation not likely under tank conditions.

- Calcium carbonate, CaCO₃
 Calcium oxide, CaO
- Calcium sulfate, CaSO₄
- ← Chromium(II) fluoride, CrF₂
- Chromium phosphate, CrPO₄●2H₂O
- Chromium dioxide, CrO₂
- Chromium oxide, Cr₂O₃
- Iron(III) fluoride, FeF₃
- Iron(III) hydroxide, Fe(OH)₃
- Iron(III) phosphate, FePO₄
- Lanthanum oxide, La₂O₃
- Manganese diflouride, MnF₂
- Manganese phosphate, $MnPO_4 \bullet H_2O$
- Manganese dioxide, MnO₂
- Manganese(III) hydroxide, MnO(OH)
- Uranyl hydroxide, UO₂(OH)₂

- Calcium chromite, CaCr₂O₄
- Calcium silicate, CaSiO₃
- Calcium hydroxide, Ca(OH)₂
- Chromium(III) fluoride, CrF₁
- Chromium hydroxide, Cr(OH)₂
- Chromium monoxide, CrO
- Iron(II) fluoride, FeF₂
- Iron(II) hydroxide, Fe(OH)₂
- Iron(II) phosphate, Fe₃(PO₄)₂
- Lanthanum hydroxide, La(OH)₃
- Lanthanum fluoride, LaF₃
 - Manganese triflouride, MnF₃
- Manganese hydroxide, Mn(OH)₂
- Manganese oxide, Mn₃O₄
- Uranyl phosphate, UO₂HPO₄•4H₂O
 - Uranyl sulfate, $2(UO_2SO_4) \bullet 7H_2O$.

Insoluble aluminosilicates are suspected of binding the ¹³⁷Cs. ⁹⁰Sr may be held by several possible insoluble ionic compounds. There are many more possible and complex compounds that conceivably could exist in the waste matrix. This list is not meant as authoritative or exhaustive, and the alkaline nature of the media may substantially alter the phase equilibria for some of these materials. However, it does provide a reasonable starting point for any further speciation work.

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8.0 CONCLUSIONS

Analyses of the waste show a very small number of analytes comprising a disproportionate majority of the waste. Water is the single largest analyte, making up over 75 percent of the solids mass. Less than one percent of the total mass of the tank waste is drainable liquid. Calcium, chromium, iron, manganese, sodium, lanthanum, bismuth, and silicon constitute approximately 10.7 percent of the solids mass. PO₄³⁻ and NO₃⁻ constitute approximately 7.4 percent of the total (i.e. soluble and insoluble) solids mass. The fraction of the total anions that nitrate and phosphate represent cannot be determined adequately because the analytical method measured only soluble anions and it is known that there are insoluble oxides and hydroxides that are not assayed at this time. The TOC was measured and found to be less than 1.3 weight percent (dry basis) in each core and for the tank as a whole. However, the TOC assay method is believed to biased low for this waste matrix and individual segment results have been observed to be higher than the bulk value for the tank.

The only significant gamma emitter found in the waste was ¹³⁷Cs, and it was found at very low levels. No meaningful regional concentrations (hot spots) of radioisotopes or fuel were detected along the vertical axis in either core. The ¹³⁷Cs concentration was relatively constant between individual core composites and their replicates; however, the concentrations between core 31 and core 33 differed by a factor of 2. In addition, the ¹³⁷Cs concentration decreased by a factor of nearly 3 as a function of depth between segment 1 and segment 9 of core 33. The major source of radiological activity was ⁹⁰Sr, which was also at a very low level. The bulk waste temperature in the tank, obtained from a thermocouple tree, ranges between 16 to 20 °C (61 to 68 °F). The radiological activity of tank 241-T-111 waste material was quite low, ranging from 0.3 to 10 mR/hr, measured through the drill string. No significant radiological activity was found in the drainable liquid in the tank or in the water digestion of the samples, but radionuclides were liberated readily in the acid digestion sample preparation, as indicated from the homogenization data. This suggests that ⁹⁰Sr and ¹³⁷Cs are insoluble.

Cores 31 and 33 appear to have a T-Plant process flush disposed on top of 224 waste, overlying a 2C waste heel. ²⁴¹Am and ¹³⁷Cs decrease substantially as a function of depth, thus their profiles nominally agree with fill histories, although individual batches and process upsets can show characteristics contrary to the general trend. These observations are consistent with the historical information regarding the transfer history, and the ICP element distribution through the segments. Discernable exothermic behavior was detected in the upper segments of both cores 31 and 33, even though the magnitude of the exotherms did not agree with the TOC present. This lack of an identifiable fuel source was attributed to the inability of the persulfate oxidation method to adequately quantitate TOC in this waste matrix. Another hypothesis under investigation is the potential contribution of an exothermic inorganic reaction that has not been identified.

Historical data indicated that 2C and 224 wastes were not expected to give any exothermic response at all, and could not create a propagating hazard. This prediction has been upheld

for 2C wastes as no exotherms were observed. In the 224 wastes, even though some small amounts of oxalate were indicated in the flowsheets, no reaction was expected. Calculations of the bulk waste inventory and inventories for several analytes of interest to the various safety issues [organics (as TOC), NO₃, ¹³⁷Cs, ⁹⁰Sr, plutonium, and water] were made. The calculated TOC by weight percent was smaller than the watch list criterion on a bulk basis. but the energetics results do not reconcile well with that interpretation, and indicate that the organic content in the tank may be disproportionately partitioned between the upper 100 cm of waste and the remainder of the tank. However, it is important to note that the organics concentration, even in this hypothesized enriched layer, may be too low to support a self-sustaining reaction in its present state. Reactions were observed only after all water had been removed from the waste matrix, and water makes up over 75 weight percent of the waste, providing an enormous heat sink to be overcome before reactions can be initiated. Both the historical and analytical data from tank 241-T-111 strongly indicate that the waste lacks the fuel concentration needed to sustain any propagating exothermic behavior or a heat source intense enough to trigger a reaction. None of the other calculated bulk inventory values exceeded any level of concern (see Table 8-1).

Experimental and analytical evidence from tank 241-T-111 waste suggests the risk from organic compounds in this particular Hanford-Site high-level waste tank is acceptable and that a propagating exothermic reaction under current and near-term tank operating conditions is not credible.

Table 8-1. Comparison of Tank 241-T-111 Analyte Values to Safety Issue Criteria.

Analyte	Safety issue criteria ¹	Calculated/ measured value	
ΔH (dry basis)	-75 cal/g	In excess of -215 cal/g	
^{239/240} Pu	50 kg	9.2 kg	
Temperature	300 °F (149 °C)	16 °C (60.5 °F)	
Heat load	11.72 kw	0.08 kw	
Organic content (TOC, Dry basis) (10% sodium acetate equivalent)	3.0 wt% TOC	1.3 wt% TOC (This result is likely biased low)	

¹(Lindsey 1986, RHO 1988, Boyles 1992, Reep 1992)

WHC-EP-0806

a No. 150 to mention are a 2 d

8.1 RECOMMENDATIONS

The following recommendations are made based on the data and analyses presented in this report and the goals of the characterization effort.

- Investigate the potential existence of alternate transfer paths from T-Plant.
- Examine more closely the chemical behavior, reactivity, and composition of Turco decontamination agent.
- Continue to characterize tank 241-T-111 sample matrices at PNL.
- Investigate the possible kinship between tank 241-T-111 and other tanks.
- Research and develop improved assay methods for TOC.
- Investigate observed discrepancy for alpha-emitting radionuclides (especially ^{239/240}Pu and U).

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Appendix A: Analytical Data

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WHC-SD-WM-DP-024 ADDENDUM-2, REV O

P.O. Box 1970 Richland, WA 99352

Tank 241-T-111 Core 31 and Core 33 Introduction and Narrative

Introduction

The analyses in this data package were performed by the Westinghouse Hanford 222-5 Laboratory or the Battelle Pacific Northwest Laboratory (PNL) under the guidance provided in the "Waste Characterization Plan for the Hanford Site Single Shell Tanks" (WHC-EP-0210) and the "Sampling and Analysis of Ten Single Shell Tanks" (WHC-SOW-91-0006). The quality control for single shell tanks is described in appendix D of WHC-EP-0210. Additional support data can be found in appendix I of WHC-EP-0210. Laboratory operations at 222-S are performed according to the "Quality Assurance Project Plan for the Analysis of Highly Radioactive Samples in Support of Environmental Activities on the Hanford Site" (WHC-SD-CP-QAPP-002) unless superseded by the waste characterization plan, appendix D, the associated SOW, or the Technical Project Plan (TPP).

Tank 241-T-111 (referred to as T-111 in the remainder of this package) is a single shell tank built in 1944 with an operating capacity of 500,000 gallons. T-111 received 488,000 gallons of "2C" and "224" type wastes through February 1976, and is classified as EHW (extremely hazardous waste). During early remediation efforts, ending in 1978, nearly 25,100 gallons of liquid was pumped out of T-111. The "2C" waste was from the second decontamination cycle of the bismuth phosphate process at the B and T plants. This waste consists of the liquid remaining after precipitation of the plutonium. The "224" waste was from the final decontamination and concentration stage of the bismuth phosphate process. In this stage, first the by-products and then the plutonium are precipitated with lanthanum fluoride. Historical data indicates the major components and estimated concentrations of these waste streams are as follows.

<u>Component</u>	224 (mol/L)	2C (mol/L)
81	0.0062	0_0092
·· ĈF - ·····	Ô.0009	0.0025
F	0.31	0.22
Fe		0.023
oxalate	0.028	
_ K	0.26	==
La	0.0014	
Mn	0.0046	
Na .	1.75	2.04
nitrate	1.06	1.27
hydroxide	0.59	
- phosphate	0.049	••
<u>Ş</u> i		0.37
sulfate		0.62

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Based on prior knowledge, cores from T-111 were expected to contain 8.5 segments. The first segment is normally the partial segment. Three cores were obtained from T-111 - cores 31, 32, and 33. Core 32 was not analyzed as the recovered material, all liquid, was considered not representative of the tank. No drainable liquids were recovered from cores 31 and 33. Drainable liquids are defined as quantities greater than 25 mL. Less than 25 mL quantities are blended back into the sample. Homogenization tests were performed on segments 1, 3, 5, 7 and 9 from core 33. Two composite samples were prepared for core 31 and core 33. Sample identification is obtained from the "sample point" section on the analytical card. A sample from core 31. segment 1 is designated "Till-C31-Si". Similarly, a sample from composite 1 core 33 is designated "T111-C33-C1", and a homogenization sample from core 33. segment 5 is designated "TIII-C33-S5H".

All of the segments, with the exception of segment 9 of each core, were a homogeneous brown-black mud-like material. The bottom half of segment 9 had a white layer. The homogenization test for segment 9 used the entire segment (both phases) from core 33. Additional analysis (acid digest, water digest, ICP, GEA, TA and IC) were requested for the white layer on material from core 31. This sample is designated "TIII-C31-S98". All samples from T-III exceeded the six month holding time limit.

Fifteen samples were sent to PNL. Volatile Organic Analysis (VOA) were requested on odd numbered segment samples for both cores and rheology on even numbered segment samples for core 31. The VOA requested for segment 9 is on the white phase only. Samples from both composites from each core were submitted for semi-VOA, Extractable Organic Halides (EOX), As. Se, Ni-63, Pu and U isotopic analyses. These samples and the requested analyses are as fallows.

VOA: core 31, segments 1, 3, 5, 7, 9.

core 33, segments 1, 3, 5, 7, 9.

Semi-VOA, EOX, As, Se: core 31, composites 1, 2.

core 33, composites 1, 2.

Ni-63, Pu and U isotopic: core 31, composites 1, 2.

core 33, composites 1, 2.

VOA, semi-VOA: hot cell blank.

field blank.

Rheology & Physical: core 31, segments 2, 4, 8.

The required QC is listed in table 0-6 of appendix 0, WHC-EP-0210. summary, the requirements are:

- one laboratory control standard per analytical batch,
- one blank per analytical batch,
- one matrix spike per core-or per matrix change,
- 100 % duplicates on all homogenization test samples and core composite samples.
- one duplicate per analytical batch for direct segment samples.

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----a-duplicate to-verify each detected exotherm for DSC analysis.

The exceptions to the above guidelines are:

- % water is always ran in duplicate,
- --Sr-90; -Se-79; -Tc-99; I-129; Pu and Am have a spike or carrier added to each sample, so no additional matrix spikes are required,
- GEA and pH do not require a spike,
- a matrix spike for Np is requested on each sample,
- ICP, HYAA, CVAA, and IC require additional, method specific QC. Instrument calibration and check standards are run according to specific procedure protocols.

Sample analysis are repeated at least once if the spike recovery is outside of 100 % ± 25 % (provided the spike concentration is at least 25 % of the sample concentration), or the duplicates have a relative % difference greater than 20 % (provided the analytical results are greater than 5 times the detection limit). If the original problem remains after one re-run, and the chemist has no explanation or "fix", additional re-runs are not required. These situations should be described in the "Analytical Batch Summary Sheet" described below. The entire batch does not have to be re-run unless the standard or blank for the batch failed.

-A-"Characterization Change Notice" was implemented during the T-111 analysis. The intent of the form is to document minor changes in analyses and deviations from the SOW or waste characterization plan. These are attached at the end of the summary section.

A "Chemist's Batch Summary Sheet" form was introduced for T-111 sample analysis and appears at the end of the analytical batch. Since this form was introduced after analysis started, one may not accompany each sample batch. The intent of this form is to document the chemist's initial data evaluation, and to provide a means of obtaining descriptive results from the chemist and chemical technologist on each sample batch. While not required as part of the final package, these forms will aid in preparing this narrative and provide insight into some of the problems encountered during sample analysis.

All analytical data for the package appears in a summary section. The summaries are lotus spread sheets intended to aid in review of analytical data, and may not include all of the associated laboratory control standards, dates, or laboratory identification numbers (J numbers). The spread sheet summaries display analytical results, blank data, detection limits, LMCS recoveries and spike recoveries, and using this data, calculates averages for duplicates, the relative percent difference between the duplicates, a mass and charge balance (described below), a ratio of the result to the detection limit, and various inter-comparisons throughout the summary. Secondary spread sheets are supplied that compare the core composite and homogenization duplicate and average results. Since a spike is not required for each sample, spike data is copied with each associated sample. "Less than" values are not used in any of the spread sheet calculations.

WHC-SD-WM-DP-024 ADDENDUM-2, REV O

All analytical results are "wet weight" values, based on the actual extrusion weights. To convert to dry weight, divide the "weight wet value" by the weight fraction of solids (100 - %water / 100). All the preparation blanks and detection limits are reported in the same units as the sample results by assuming a typical sample size. This is done to provide a direct comparison of the background and degree of contamination present in the method to the sample results. Since the sample-size is estimated, a "less than" value provided for a sample could be slightly lower than the given detection limit or blank value.

A charge balance is calculated for each core composite sample in the analytical summary. The charge balance is calculated by ratioing the equivalent charge of the cations to the anions. As an example, the equivalent charge of sulfate is calculated as follows:

average ug/q sulfate milliequivalent weight of sulfate * 1000

Only the major components are used for these calculations, those with a concentration greater than 1000 ug/g. The cations used are Ca, Cr, Fe, Mn, K, Na, Bi, and La from the acid digested ICP results. The anions used are fluoride, nitrate, phosphate (calculated from the P on the acid digested ICP result), sulfate (calculated from the Σ on the acid digested ICP result), and acetate (calculated from the TOC results). The ratio ranges from 1.3 - 1.5. Proper incorporation of hydroxide would probably improve these ratios, since there is a relatively large uncertainty in the assumed molecular speciation model.

A mass balance is calculated for each core composite sample in the analytical summary, using an "oxide model" as the basis for the calculations for metals and the anions were present in their normal forms. The mass balance is the sum of the weight fractions of the major constituents (those with a concentration greater than 1000 ug/g). As an example, the weight-fraction calculation for CaO-is-as follows:

(ug/g of Ca * molecular weight of CaO) (atomic wt of Ca * 1000000)

The components used in calculating the mass balance are Bi, Ca, Cr, Fe, Mn, Na, U, F, nitrate, phosphate, sulfate, water, K, and La. It was assumed that all but K, Na, and F were present as oxides. The ratio ranged between 0.93 and 0.96.

SECTION I: EXTRUSION AND SAMPLE DATA

Visual

A summary of the extrusion data, entitled "Physical Properties Summary", is provided with the segment data. This is a compilation of all the recoveries for each segment of each core, with an associated percent recovery, penetrometer reading, and density. Photographs of the extruded segment are included in the data package. All volumes given that are less than 180 mL (100 % recovery) are estimates provided by the hot cell chemist. Segments marked with an asterisk on the summary sheet were recovered from sample

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_______containers with the valves left in the "open" position (the waste was thick enough to remain in the container, even though the valve was not closed).

The extrusion data for core 32 will be included in the core 31 package.

Core 31

- Segment 1: The sampler was almost empty. It contained about 50 mL of homogeneous, black-brown, low viscous solids. The entire sample was used in VOA and DSC/TGA sampling. There was a tiny amount of liquid, not enough to observe characteristics.
- Segment 2: The sampler was completely full of homogeneous, dark brown to black solids with the consistency of swamp mud. There were no drainable liquids.
 - Segment 3: The sampler was completely full of homogeneous, dark brown solids that had the consistency of swamp mud. There were no drainable liquids.
 - Segment 4: Most of the sampler was full of the same dark brown homogeneous solids. The top 1/8 of the extruder was filled with a more "liquidy" solids than the rest. There were no drainable liquids.
 - Segment 5: The sampler was completely full of the same dark brown homogeneous solids and contained no drainable liquids.
 - Segment 6: This sampler was empty, the valve was left in the open position.
 - ---Segment-7:--The-sampler was completely full of the same dark brown homogeneous solids and contained no drainable liquids.
 - Segment 8: The sampler was completely full of the same dark brown homogeneous solids and contained no drainable liquids.
 - Segment 9: The sampler was completely full of solids, the same dark brown solid material at the top, but gradually whitened down the segment. While the color varied, the consistency of the material was the same as all the other segments in this core.

Core 32

- Segment 1: Contained about 50 mL (58.70 g) of "dark dishwater grey aqueous" liquid with about a 2 mm layer on the surface. There was a layer of bubbles covering the thin top layer, and adhering to the sides of the jar. No solids were collected.
- Segment 2: Contained about 170 mL (88.35 g) of dark grey liquid with dark grey particles suspended in the liquid. There were only minimal solids (enough for a VOA sample) that had the consistency of loose mud or pudding and was dark brown/black in color.
- Segment 3: Contained 180.90 grams of dark grey aqueous material. No solids were present. No volume was recorded.

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- Segment 4: Contained 182.48 grams of dark grey aqueous material. No solids were present. No volume was recorded.
- Segment 5: Contained 191.05 grams of dark grey aqueous material. No solids were present. No volume was recorded.
- Segment 6: Contained 185.52 grams of dark grey aqueous material. No solids were present. No volume was recorded.
- Segment 7: The sampler was empty, the valve was left in the open position.
- Segment 8: The sampler was empty, the valve was left in the open position.
- Segment 9: Contained 158.24 grams of liquid, with about 150 mL of aqueous phase and 8 mL of NPH.

No samples were taken from core 32 as the material was judged to be non-representative of the material in the tank when compared to cores 31 and 33.

No photographs were taken for segments 1, 3, 4, 5, 6, 7, 8, 9.

Core 33

- Segment I: The sampler was completely full of nearly black homogeneous solids, and contained no drainable liquids.
- Segment 2: The sampler was completely full of the same nearly black homogeneous solids, and contained no drainable liquids.
- Segment 3: The sampler was completely full of the same nearly black homogeneous solids, and contained no drainable liquids.
- Segment 4: The valve was left open on this core assembly. The extruder was about 75% full dark brown/nearly black homogeneous solids, and contained no drainable liquids. Color was slightly lighter than previous segment.
- Segment 5: The valve was left open on this core assembly. The extruder was about 7/8 full of the same dark brown homogeneous solids, and contained no drainable liquids.
- Segment 6: The sampler was nearly full of the same dark brown homogeneous solids, and contained no drainable liquids.
- Segment 7: The sampler was nearly full of the same dark brown homogeneous solids, and contained no drainable liquids.
- Segment 8: The sampler was nearly full of the same dark brown homogeneous solids, and contained no drainable liquids.
- Segment 9: The sampler was completely full of solids, the same dark brown solid material at the top, but the bottom 5-6 inches was light colored. While the color varied, the consistency of the material was the same as all the other segments in this core. The segment

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was divided into two parts, one was the bottom 5 inches of the segment which contained most of the white material. A VOA sample was taken from the top, and a DSC/TGA taken from the bottom. These samples were later combined for the homogenization test.

A field blank and a hot cell blank were also taken for T-111.

Sub-samples for T-III

The following numbers are the "J" numbers assigned as tracking numbers by the 222-S laboratory. The analytical batches may contain information from other sample numbers because of the way the samples were grouped for analysis. Results from other samples should be ignored, since they will be included in the data package for that specific segment.

	CORE 31	CORE 33	CORE 33 (Homog. Tests)
-Segment 1	J308, J309	<u> </u>	J396, J397, J398, J399
Segment 2	J310	J321, J323	NA
-Segment 3	J311	J324	J400, J401, J402, J403
Segment 4	J312	J325	NA
Segment 5	J313, J315	J326, J327	J404, J405
Segment 6	none (empty)	J328, J329	NA
Segment 7	J315, J316	J330	J406, J407, J408, J409
Segment 8	J317	J331	NA
Segment 9	<u> </u>	J332, J333	J410, J411
Segment 9B	Ĵ412, J413, J414	NA	NA
-	J439, J440,	J447, J448, J449,	J426, J432, J433, J438, J455, J456, J457, J464, J480, J493, J606, J607, J631, J632, J641, J642,
Core 31 composite	2J418, J419, J485	J427, J434, J441,	J450, J459, J467, J475,
Core_33_composite	J445, J451,	J428, J429, J430, J452, J453, J460, J487, J488, J489	-J435, J436, J442, J443, J461, J462, J468, J469,
Core 33 composite	2 J42 2, J42 3,	J431, J437, J446,	J454, J463, J471, J477

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Hot cell blank

J494, J495, J496, J498, J499, J500, J502, J503, J504, J506, J508, J513, J517

SECTION II: SEGMENT ANALYSIS

TGA

Thermogravimetric Analyses and Differential Scanning Calorimetry (DSC) are performed on every visible phase of the waste, prior to homogenization, to evaluate the thermal properties of the waste. The TGA results give a measure of the weight loss with increasing temperature. The result obtained from TGA is a weight % water. The TGA % water (percent weight loss at 100 degrees C) may vary considerably, due to a combination of the small sample size and sample heterogeneity. The procedure number for DSC, TO42 A-01712F, has been changed to LA-514-113, Rev. A-0. The procedure number for TGA, TO45 A-00712F, has been changed to LA-500-112, Rev. A-0. These procedures are routine Analytical Operations now, rather than Process Chemistry procedures.

The TGA results were 5-7% higher than the gravimetric wt % water, and ranged from 70-87%. Where duplicates were ran, there was good agreement, and results were consistent over all segment samples

% Water

More representative % water measurements are obtained with the gravimetric measurements, also reported in the summaries. Duplicate % water analysis were requested for all segment samples. However, the duplicates for segments 2, 3, 4, 5, 8, 9 from core 31 and segments 1, 3, 4, 5, 7, and 9 from core 33 were not requested until one month after the original samples were run. These latter values are added to the summary sheet, but due to the summary sheet format, the LMCS standard results are not included. The duplicate values for segments 2 and 4 from core 31 (32.60% and 59.60%) are probably low due to the sample drying out between analysis; or from a non-homogeneous "soupy sample" that would cause uneven sampling. There is no more sample for a re-run. However, the original sample results and the TGA agree quite well. There was no sample for a duplicate analysis of segment 8 from core 33.

DSC

DSC is used to identify the potential of exothermic reactions from the waste upon heating. In all cases, the identified exotherms were 3 to 4 times smaller than the detected endotherms. The calibration frequency of the instrument is determined by the chemist and is based on the instrument performance on laboratory standards.

Core 31: Exotherms were detected on the sample and duplicate for segments 1, 2, and 3. A very small exotherm, nearly an order of magnitude smaller, was detected for segment 4, and was not verified with a duplicate. No exotherms were detected for segments 5, 7, 8, 9, and 98 (the white layer). The temperature ranges for the exotherms were between 200 and 405 °C, with the energy around 300 J/g. The small exotherm had an energy of 55.7 J/q.

Core 33: Exotherms were detected on the sample and duplicate for segments 1 and 2, and only the sample for segment 3. The duplicate did not show an exotherm. The temperature ranges for the exotherms were from 179 to 438 °C,

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with the energy around 300 J/g. The exotherm for segment 3 was small, about 50 J/g. No exotherms were detected for segments 4-9.

Particle Size

Particle size analyses are performed on every segment from every core to support the evaluation of potential waste retrieval methods. Small aliquots of the wastes were suspended in a water matrix using sonification before measuring the particle size by laser technology. The data are summarized in a table and four plots of probability number and probability volume density and distribution graphs. The procedure number for particle size analysis listed in the Technical Project Plan (WHC-SD-CP-TP-070, rev 0) is incorrect, and should be T044-A-01712F.

Penetrometer

<u>Penetrometer readings are taken for each segment at extrusion.</u> All of the segments were soft, with readings <100 psi.

Segment 9B (white layer)

In addition to the above analyses requested for each segment, ICP, IC, GEA, and TA were requested on the white portion of segment 9 of core 31. The sample was taken after the white layer had been homogenized. The ICP and TA resemble the homogenized segment data, and the IC resembles the core composite data. The spike recoveries for the ICP were biased low. There were no exotherms detected, and the % water was 70 %, about 5-10 % lower than the other segments.

General Comments

Because of the qualitative nature of TGA, DSC, particle size, and penetrometer analyses, no detection limits or precision and accuracy statements are required.

SECTION III: HOMOGENIZATION TEST ANALYSES

Introduction

Waste tank samples have a wide range of physical characteristics. Methods for homogenizing waste samples are limited by their applicability to hot cell operations and how well they lend themselves to remote operation and decontamination for reuse. Homogenization tests have been incorporated into the single shell tank characterization program to obtain estimates of the errors resulting from homogenization techniques employed during analysis. The evaluation of this data will support decisions made using the data and identify areas where homogenization procedures may be improved.

Homogenization Tests

Homogenization tests are performed on segments. After the segment has gone through the homogenization procedure, two sub-samples are taken from two different (top/bottom, left/right) locations in the homogenized segment. Each of the sub-samples are analyzed in duplicate to obtain estimates of the

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analytical error. The differences in the results obtained on the two subsamples will provide homogenization error. A statistical analysis (beyond the scope of this narrative) of the variance must be performed to differentiate between the analytical and homogenization errors. Each homogenization sample was analyzed by GEA, ICP, and TA on an acid digested sample. The acid digestion is the same as that used for normal ICP analysis.

Sub-samples

Homogenization tests were performed on segments 1, 3, 5, 7, and 9 from core 33. See the listing above for the J numbers assigned to the homogenization samples. The results are compiled in the summaries entitled "T-lll Homogenization Analyses" and "Homogenization Test Summary". One spike was considered sufficient for the homogenized sample data. A comparison of the RPD's of the duplicate results of each sample to the RPD of the averages of the two samples (see summary) shows virtually no difference for those elements well above their detection limits. This tank probably is not a good test of homogenization procedures, as the segment material was very uniform at extrusion. Even the white material in segment 9 was chemically similar to the rest of the segments, with the exception of segment 1. Segment 1 has much higher concentrations of Mn, Pb, Ag, and In when compared to the other segments, while 81 and La are much lower. The spike for the gross alpha on segment 5 was-consistently high even after several re-runs.

SECTION IV: CORE COMPOSITE ANALYSES

Sample Preparation

Samples are prepared for core composite analyses using an acid and water digestion, and a KOH fusion. ICP and HYAA are run on the acid preparation, ICP, IC, nitrite by spectrophotometry, pH, ammonia, TOC, TIC, carbon-14, and tritium on the water digestion, and ICP and the other radiochemical analyses on the fusion preparation. The KOH fusion is contaminated with Ni from the crucibles used. LMCS standards for the acid digestion go through the sample preparation procedure. Standards for the water digestion and fusion are not run through the sample preparation procedures.

ICP

The ICP data is reported using a CLP software package from WARD Scientific Ltd (WARDS), and is used to generate the spread sheet summaries instead of the analytical cards. The calculations on the analytical cards are preliminary calculations performed by the instrument technician to access instrument operation and are not used in the final report.

WARDS utilizes the concept of a sample delivery group (SDG), and batches spike, duplicate, instrument control standards, blanks, serial dilutions, interference checks, and narrative information for each sample group. If the result is below the detection limit, the CLP report format will return a 200% RPD for duplicates and a 100% for serial dilutions. On all pages except the "duplicate" page (form VI-IN), any values less than the detection limit are replaced with the instrument detection limit, and should be interpreted as a "less than" value. The duplicate page (form VI-IN) reports what the instrument measures, and may include negative numbers. Since it is this page

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that is used to generate the "result 1" and "result 2" columns in the summary sheets, negative values are included.

On the page in the WARDS package designated "COVER PAGE - IN", the answer to the third question, "...were raw data generated before application of background corrections?" has been changed to "no" based on the chemist's interpretation. This change does not affect the data or evaluation of the data in any of the WARDS packages generated in support of the SST program. The WARDS package Sample Delivery Group (SDG) number has a "/" suffix, where the "is "N" for no sample preparation, "W" for water digestion, "F" for fusion, and "A" for acid digestion. This was not implemented until after analysis began on T-111, and so most of the WARDS packages will not have this identifier. The case narrative provided with each batch flags areas that would not meet CLP criteria. The WARDS package is configured to report all elements requested in the SOW except T1, Th, and U, and report the additional elements B, Ce, La, P, Sr, S, and Ti. Form I-IN will not show negative results for individual determinations, while Form VI-IN does. The WARDS results used in compiling the master summaries are taken from the VI-IN form.

ICP analyses are ran on acid and water digests, and a KOH fusion. In the data summaries for the T-111 package, the preparation blank and the detection limit values for ICP have been converted from ug/L to ug/g by estimating the typical sample size and dilution involved in the preparation procedure. The acid and water blank and detection limit results (ug/L) were divided by a factor of 10 because the acid and water preparation procedures dissolve approximately 1 g of original sample in 100 mL of liquid. The fusion blank and detection limit results were divided by 2, as this preparation procedure dissolves approximately 0.5 g of sample in 250 mL of liquid.

Failure criteria for an entire WARDS package are under development. Since each ICP run generates results for 32 elements, there will always be some failures on each run. These failures are identified in the narrative associated with each WARDS package. Failure criteria should be based on the absolute failure rate of either spikes, duplicates, LMCS standards, and calibration standards, but tempered by the relative importance of the individual element that failed. For example, LMCS and spike recoveries may be generally high for Na, Ca, and Fe from contamination of the blank and sample during preparation, and Si and B from the use of glass containers during sample preparation. Spike or LMCS failures noted for these elements should not invalidate the entire WARDS package.

Several factors noted in the WARDS packages for T-111 create low spike recoveries and were noted by the ICP chemist. Silver recoveries are commonly low due to the precipitation of Silver Chloride. Poor recoveries of Iron, Magnesium, and Calcium accompany high Preparation Blank values, but the correlation is poor. Spike failures for major elements are frequently caused by a high element concentration in the sample. When the added spike concentration is insignificant compared to the concentration of the element present in the sample, a failure generally occurs.

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The major elements detected by ICP analysis of the acid digestion were Ca. Cr. Fe, Mn, K, Na, Bi, La, P, and S (all greater than 1000 ug/g). Cr and Pb concentrations (1.86E+3 and 4.75E+2 ug/g respectively) would both exceed the 100 ug/g TCLP limit, assuming 100% leachability. The WARDS package performs inter-element corrections for Al, Ca, Fe, Mg, and Cr. Calcium and magnesium contamination often occurs as a result of the powder on the gloves used by laboratory personnel. The ICP analysis of the fused sample is used to evaluate the completeness of the acid digestion procedure, and to identify any acid insoluble compounds. The K results in the WARDS package often over ranges on the fusion, giving the impression that no K was in the sample. Ratios of the ICP results for major elements appear in the data summaries, and are used to estimate consistency and speciation. The fusion/acid ratio shows good agreement for Al, Fe, Na, P and Sr. Si and Ti are an order of magnitude higher in the fusion, while Bi is slightly higher in the acid digestions. Based on the ratio of the acid and water ICP data, about 10 % of the Cr is water soluble (CFVI). Most of the Na is soluble, but only 50 % of the phosphate, and essentially none of the Mn is soluble. There was no La reported for the water digested composite LMCS standard for core 33, and no Ir detection limits reported for the core composite samples. The Si results for ICP on all T-III analyses are known to be biased low due to an undetermined interference.

CVAA and HYAA

Cold vapor atomic absorption (CVAA) was used for Hg analyses. The Hg results for T-111 ranged from 1.02-1.88 ug/g. The spike for core 31 was high (122%) and low for core 33 (82%). However, both were within the plus or minus 25% limits. All RPD's were less than 13%. Due to instrument and sample matrix problems, the As-and-Se elemental analyses were performed at PNL by graphite furnace AA.

Anion and Wet Chemical Analysis

Anions were determined on a water digestion of the sample. Ion chromatography (IC) was used to determine F, Cl, nitrate, nitrite, phosphate, and sulfate. _All-duplicates were within limits (<20 % RPD) and the results were consistent over all the composite samples. Post digestion spikes are used because of the high concentrations of the anions. Nitrate, phosphate, and F concentrations were high, consistent with the historical data available on T-111 waste. Spike recoveries were biased low for 3 of the 4 composite samples, but within the 75-125% limits. The S determined on the acid ICP, converted to sulfate, agrees very well with the sulfate determined by IC, while the same comparison of P with the phosphate shows the converted P a factor of 2 higher than the phosphate. This indicates that only 50 % of the phosphate is soluble. The pH is very consistent at about 10, while only "less than" values were obtained for ammonia. All the TIC results were near the detection limit, with good reproducibility and spike-recoveries. TOC results were nearly an order of magnitude higher (2000-3000 ug/g). All analytical units were changed to ug/g for the data summaries, and may not agree with what is on the analytical card. All cyanide results obtained were "less than" values. Both the % water and 16A results were between 75 and 80 % with good precision over all the core _composite samples.

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Radiochemical Analysis

Most of the radionuclides are determined on the fusion due to the acid insolubility of actinide and Sr compounds. Tritium and C-14 are determined on the water digestion, and Ni-63 (analyzed at PNL laboratories) is determined on the acid digestion. There are indications that I-129 is lost during the acidification of the fused sample, which will require the development of an alternate digestion procedure for future work. In the cases where two spike recoveries are reported in the analytical summaries, one under the other, the spike recovery is used to correct for chemical yield. The second spike (bottom) corresponds to the duplicate result. Pu-238 is reported as part of the Pu-239 analysis, using the spike data from the Pu-239 procedure. Cm-244 would be reported only if seen on the Am-241 mount. The relative counting error reported is based on gross counts, and will under-estimate the counting error in those samples close to the detection limit (where the gross counts and the background counts are nearly the same). Calibration data for GEA labeled as detector "14" is data for detector "4".

Ouring calibration updates, the new calibration file name for individual detectors have a "1" added (4 to 14) to avoid erasing the old calibration data prior to validation of the new calibration data. For T-111, the new calibration file name was not changed back to "4". Calibration of the alpha proportional counters (APC) is performed as follows. A standard source is prepared by the standards laboratory and used to calibrate one of the three detectors. Since the calibration standard is not durable enough for daily use, a secondary source is prepared and the d/m value of the secondary source is assigned using the calibrated detector. The secondary source is counted daily on all three APC's and tracked using the LMCS system. The detectors pass the daily check if the calculated efficiency for each detector is within 50% ± 0.25% (100% "recovery"). The LMCS data for the APC's and EDP codes are listed in the "Precision and Accuracy Statement" section.

Poor spike recoveries were obtained for uranium on core 33, but the data across all the core composite samples was consistent (between 2000-3000 ug/g).

Total alpha and total beta RPD's were normally under 5% with good recovery on the spikes. The total alpha results were twice the sum of the Pu and Am results, indicating 1) the potential presence of undetermined alpha emitters, 2) beta cross-talk (unlikely due to the beta activity present), or 3) incorrect Pu or Am recoveries (incomplete exchange with the spike). This last scenario is unlikely due to the consistency of the ratios, and the excellent agreement between the Am-241 results obtained by gamma spectroscopy and alpha spectroscopy. The ratio of the total beta to the sum of the Sr/Y and Cs-137 results show good agreement (.991 to 1.00) for the activity balance of the beta emitters. When sample "J-470" was rerun for poor spike recovery, a blank was not included in the analytical batch.

Sr-90 spike-recoveries were consistently 90-95%. Results for core 31 were around 7 uCi/g while core 33 was around 3.5 uCi/g. These results vary more than the elemental Sr results from the fusion ICP, where the results are consistent across both cores (280-310 ug/g). Tc-99 results were low, within a factor of 5 of the detection limits but spike recoveries were adequate (67-74%). The results for Se-79, H-3, and C-14 were at or below detection limits, while spike recoveries ranged from 85-95%. No standard material is available

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for Se-79. The tritium procedure LA-218-113 was changed to LA-218-114 for T-111 analysis. This procedure uses microdistillation columns in place of the normal glass distillation equipment.

The Pu spike recoveries ranged from 50% to 90%, while Am recoveries were poor, 10-20%. No Pu-238, Np-237 or Cm-244 were present above detection levels. Soike recoveries for Np-237 were consistently low at around 70%.

Am-241 and Cs-137 were the only isotopes that were consistently found at quantifiable concentrations by gamma spectroscopy. No spikes were requested for gamma spectroscopy. As stated above, the beta activity based on the Cs-137 results from gamma spectroscopy agreed with the gross beta determination, and the Am-241 results from gamma spectroscopy agreed with the alpha—spectroscopy results for Am. I-129 results were all "less than" values, therefore no counting error could be determined. The preparation blank for I-129 for core 33 was invalid based on spike recovery, but was not re-run as all results were "less than" values. The spike recoveries ranged from 39% to 57%.

Physical Measurements

DSC, TGA, and weight % water was performed on each core composite. Exotherms were observed (in duplicate) for both composite samples from core 31, but no exotherms were observed for core 33. This is not surprising, as the exotherms were small on the segment analysis and were done prior to homogenization. TGA and % water results are in the same range as found on the segment samples, 75-85%, with the TGA results being consistently higher than the % water.

TCLP Results

A TCLP extraction was performed on the waste using a TCLP procedure scaled down a factor of 10 for radioactive samples. The matrix spike was added before the TCLP extract was preserved and acidified (per SW-846 procedure). recognizing that possible precipitation of the spike material could result in low spike recoveries. In order to meet the duplicate, spike, and blank requirements of appendix D, the following sampling and digestion scheme was used on T-111. Composite 1 and composite 2 from core 31, composite 1 from core 33, and a blank were run through the TCLP digestion procedure. The composite 1 TCLP digests from core 31 and core 33 were split into two sub-_samples, and one sub-sample from each digestion was spiked with the TCLP metals. All sub-samples were acidified. Each sub-sample was acid digested in preparation for analysis, with the unspiked sub-samples from both cores again prepared in duplicate (split into two samples prior to acid digestion) to determine analytical error. The composite 2 and blank TCLP digests from above were acidified, acid digested and submitted for analysis (no spike or duplicate measurements).

The RPD's were high for As, Ba, and Pb, but these analytes were present in concentrations less than 6 times the detection limit. Cd was the only element with spike recoveries within acceptable limits (greater than 75%). A comparison of spiking procedures is being performed on the next single shell tank (C-110). Cr and Se both exceeded the TCLP limit. Good agreement was obtained for those elements well above the detection limits.

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Field Blank and Hot Cell Blank

Field blanks are prepared by filling a sample container in the field, shipping it to the labs with the samples, extruding in the hot cell and collecting as drainable liquids. The hot cell blank is a rinse of the extruder tray. The required analysis for these samples are ICP, GEA, HYAA (As, Se), CVAA (Hg), TA, TB, TOC, IC, and VOA and semi-VOA. The raw data for the field blank and the hot cell blank is included at the end of the core 31 data package. No duplicate samples or spikes were required on the blanks. The ICP was ran on direct sample, so no corrections were made on the preparation blanks predetection limits. The ICP LMCS standard did not include Sr. There was insufficient sample for duplicate ICP analysis for the hot cell blank.

Hot cell blank: The major constituents in the ICP analysis were Na at 5 ug/mL, B at 1.5 ug/mL and Si at 4.75 ug/mL. All the rest were less than 1 ug/mL. The balance of the analytes determined (F, Cl, NO2, NO3, SO4, PO4, As, Se, Hg, TOC, TA, TB, and GEA) were "less than" values.

Field blank: The major constituents in the ICP analysis for the field blank—were similar to the hot-cell blank. The Na was at 8.07 ug/mL, B at 1.76 ug/mL, and Si at 4.47 ug/mL. The rest of the constituents were less than 1 ug/mL. Results greater than the detection limits were obtained for F, NO3, and PO4 on the IC, TB, and Cs-137 on the GEA. These levels of contamination are insignificant when compared to the concentrations of these analytes in the T-111 samples.

Accuracy and Precision Statements

Accuracy and precision estimates for the procedures used in analyzing T-111 core samples were compiled from the 222-S Laboratory Measurement Control System (LMCS) data base over the period of April 1, 1992 through July 22, 1992 (the time period that the samples were ran in the laboratory). The average percent recovery and standard deviation for each analyte are determined from multiple analysis of standards containing those constituents. Each analyte and analytical system has a specific EDP code. A different EDP code is used for each parameter, instrument, method, or standard used in the system. When more than one EDP code is provided, then more than one analytical system was monitored. If no data was available for an EDP code for the time frame of interest, the historical values based on earlier data has been provided. In these cases, the number of measurements noted as "n" is recorded as "0". Historical data typically is based on 50 measurements.

No standard Se-79 material is available from commercial suppliers of isotopes for the preparation of a standard. The LMCS standard data for DSC only indicates that an exotherm is detected, and 100 % recovery is noted. The results do not reflect errors caused by sample matrices, or special sample preparation procedures used prior to applying some of the methods.

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The accuracy and precision estimates for T-111 are as follows.

Method ICP	Analyte	EDP	<u>_n</u>	Average % Recovery	Relative Std. Dev.	Comments
LA-505-151	AT .	\$102	46	96.7	4.9	
. , . , ., ., ., ., ., ., ., ., ., ., ., ., .,	Sb	\$104	50	100.2	3.0	Historical
. == : :	Ās	\$106	5	99.6	2.5	111310111041
	Ba	5108	35	96.9	2.6	
	8e	-5110		102.3	3.2	- (
	Cd	S116	21	98.5	3.5	
	Ca	5118	42	96.5	3.5	
	Cr	S122	56	100.3	3.0	
	Co	Š124	6	98.5	2.0	
	Cu	\$126	18	96.5	1.8	
=	Fe	5130	39	98.5	2.7	
	Pb	S134	35	98.3	4.7	
	Hg	S138	30	98.1	3.7	
	Mn	S140	33	95.0	2.8	
	Ni	S148	31	98.9	2.7	
	K	S152	37	99.3	2.7	
	Se	S156	5	100.1	4.9	
	Ag	\$160	17	99.1		
	Na	S162	49	96.1	3.5 2.3	
	Ÿ	5182	12	98.9	1.4	
	Žn	- 5184	36	99.0		
	Bi	S112	23	100.1	2.4	•
	8	5114	9	95.8	4.7	
	La	S132	14	103.2	3.0	
	P	\$150	31	96.9	3.0	
	_\${	S158	23	94.4	4.2	
	<u>\$r</u>	- S1-54	24	96.3	5.1	
-	\$	51 66	32		2.2	
	Sn	S174		100.1	3.2	
	Ti	S174	9 13	97.1	4.6	
	Źr			98.1	2.2	
	Ce	S186	9	101.4	3.8	
	~~	S120	•	103.9	1.3	

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Method	Analyte	EDP	<u>n</u>	Average % Recovery	Relative Std. Dev.	Comments
CVAA LA-325-102	Нд	R716	18	102.8	6.0	-
IC LA-533-105 SPEC	F C1 NO3 NO2 PO4 SO4	R978 R968 R976 R970	20 22 22 18 21	101.4 99.0 96.8 101.2 97.5 96.9	5.9 4.4 3.9 4.9 3.7 3.9	ι
TOC/TIC LA-622-102	TOC	S215 S223		99.6	3.6	
LA-344-105 Dist/SPEC -LA-695-101 LA-695-102	CD3	S346 S242 S244		99.4 99.5	3.5 2.0	·
Dist/Titrat LA-634-102	e NH3	S235 S236		99.6 99.6 -	6.4 6.4	
Fluorimeter LA-925-106	U '	S267	111	100.8	8.2	
Alpha LA-508-101	Tot Alpha	S510	62	103.8	8.8	
LA-508-104 LA-508-114	APC 1 APC 2 APC 3 Det 14 Det 16	C011 C012 C013 C142 C162		99.9 100.0 99.5 99.7 103.5	0.11 0.09 0.10 4.8 5.3	
Beta LA-508-101 LA-508-114		S515 C143 C163	176	98.3 99.0 99.7	4.4 3.1 7.4	
Gamma LA-548-121	GEA Cs-137 Co-60	R901 R905	Ž8 28	101.9 100.6	2.4	
Alpha Spec LA-503-156	Pu-23 9 - Am-241	R211 - R201		99.0 94.2	5.3 8.2	

WHC-EP-0806 WHC-SD-WM-DP-024 ADDENDUM-2, REV 0

_	-					
	Analyte-	EDP	u	Average % Recovery		Comment
Alpha LA-933-141	Np-237	\$380	20	72.8	13.1	•
8eta LA-220-101	Sr-9Ò	S376	30	93.7	6.4	i , ,,
LSC LA-438-101	Tc-99	5363	67	108.0	7.8	
LSC LA-365-132	Se-79	no st	andard	material av	ailable	
.Gamma LA-378-104	I-129	S928	16	107.8	9.1	
LSC LA-348-104	C-14	R90 9	26	91.6	7.3	
LSC LA-218-113	H-3 .	R907	10	84.7	4.4	
Gravimetric LA-564-101		\$360	63	100.3	1.3	
Thermal LA-561-112	TGA	\$362 R826	6	98.8 98.5	1.4 1.9	
Thermal LA-S14-II3	osc	S230	16	100	0	
pH LA-212-103	H+	S348	4	100.6	0.15	

Stew M5/Cinny 9/30/92

Steve McKinney Senior Scientist

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WHC-SD-WM-DP-024
ADDENDUM 5, REV 0-A

Project Number _____

Internal Distribution

Own September 18, 1992

se McKinley

from MR Orie MW Th_

No. Sec. James James 1880.

Subject GTRA (As. Se) Analytical Results for Till (Core II # II)

<u>Graphite Furnace Atomic Absorption analyses were performed September 17 and 18 on the SST samples from Tank Till Core 31 & 13 core composites following procedures FML-ALO-214 (As) and FML-ALO-215 (Se). The digestions were performed according to FML-ALO-101 in the Shielded Analytical Laboratory and the analytical work was conducted in the Laboratory 303 using a F-E 5100 AA (WB76757). Results and all supporting data are archived in GFAA File: GF091792.</u>

The ICT/CCV analytical control standards for As and Se analyted between 99% and 108% recovery. The two pre-digestion blank spikes (or "process control standards") recoveries were 138% \$ 104% for As, and 57% \$ 58% for Se; the Se results are flagged with an "N" to indicate that the recoveries are less than 75%. The low control recovery on Se is indicative of a loss of Se during preparation; reanalysis was not performed due to lack of sufficient material. The results reported in the Summary Report may be bias low as much as 50% (i.e., if an adjustment is made for the control recovery, the Sa results are 3.0 µg/g "U" instead of 1.5 µg/g "U"). The pre-digestion spiked samples for As recovered well (i.e., 95% & 33% at 5% dilutions), whereas the spiked samples for Se showed no recovery (i.e., at 10% dilutions) indicating a severe matrix interference.

The results for the samples, duplicates, spikes and controls is presented in the attached GFAA Summary Report.

concur: Justed Bernen 7/23/82

WHC-SD-WM-DP-024 ADDENDUM 5, REV 0-A

GRAPHITE FURNACE AS ANALYSIS SUPPLARY, REPORT -- TARK BILL COME 33/33

rel	Contro Nees,	Spike	Spike Rec	Saspia Sipika US/8		Post Splke Lec			1	Post			Patt	Sample 119/6	L Lug#	PHI	Saspia 10#	Analyte
1/38X 2 57X	3636 1632	3494 3494	95% 8/A	4.47	U	61X 101X	0.64 6.15	N/A N/A	"-W]	100% 82%	3.30 1.50	4 4	102% 77%	3.30 1.50			Eli-conpl (11)	A+ 6+
					÷	I		M/A M/A	w	101X 74X	3.30 1.50	, W	100x	3 - 30 1 - 50			EST-comps	A4 6+
1 1041 1 581	2601 1441	2494 2494	OJK N/A	4,15	u u	1111 74%	0.46 0.15	W/A W/A	"-UJ	92X 81X	3.30 1.50		971 801	3.39 1.50			C31-coupl	A 6 6 0
							KT) N/A N/A L	J- 3 3	avx 45x	3.10 1.46		94X	3.30 1.50	9594 99594	85 - Q	C33-corbs	
				4.15	3	79	KT 3.)	N/A N/A L	- % J	45X /	3.10 1.46		96X 64X	1.30 1.50	995 9 (995 9 (92-6	CSS-combg	A4 5•

------AA5100 PE (9/1/92): IDL >> As = 3.30 ug/L, Se = 0.75 ug/L; CBDL >> As = 10 ug/L, Se = 5 ug/L sulle 8200A pllution; IDL >> As = 8.66 ug/g, Se = 8.15 ug/g; CRPL >> As = 2 ug/g, Se = 1 ug/g. The analytical spike is 20 ug/L for As and 10 ug/L for Se.

ICV/CLV used daring analyses: As, Se -- ICF (CV2(1290))
The pre-digestion spike is 25 ug/L for both As and Se.

Procedure: PHI-ALO-216 (As), PHI-ALO-215 (Se); MAIE: GF/A PE 5100 ug/6757

U . [Analyte] . IDL .. Note the value reported for the nample is the IDL.

^{8 .} IDL . (Analyte) . ERPL

M . Fre-disection Spike Recovery not within control limits 475-125%).

W - Post digestion splits recovery not within control timits (50-150%).

- 870 not within control limits.

WHC-SO-WM-DP-024 ADDENDUM 5 REV OA

Susan McKinley October 9, 1992 Page 2

or the second common way

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The relative percent difference (RPD) is calculated with sample 92-08280-HZ and its duplicate to indicate the overall analytical precision (3% for "Ni and 25 for "Ni).

A blank spike and matrix spike were prepared using 53 Ni; no NIST traceable 53 Ni was (or is) available for spiking purposes. The blank spike recovery is 95%. The matrix spike recovery is 91%. Thus the average batch bias is -7%.

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WHC-SO-WM-DP-024 ADDENDUM 5 REV DA

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RADIOCHEMISTRY STANDARDS AND CONTROLS Nickel-63 and Nickel-59 Supporting Documentation

This page indicates the <u>location for supporting documentation</u> for radiochemical analyses performed in support of ⁵⁹⁸⁶³N1 determinations.

Detector Controls

- LEPS detector background, control counts, and efficiency determination, see control and background files for M&TE # Diode A, Diode B, and Diode D.
 - LSC detector background and control counts, see instrument log, Packard 2250XL, Lab 54, Bldg 325.

Solution_standards:

- SINi standard #54124-16-0, LRB 54124, page 16, Bldg. 329, room 10.
- Nickel (stable) carrier solution for yield determination, see LRB 53293, page 30.

Flourescence on the SNi X-ray peak

· ALO office, 101SY 59863Ni report

Performance checks of pipets and balances, see LRB 53647 pg. 83, 85, & 36 and -- - LRB 54371, pg. 41.

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average of two enelyses

TABLE 13a. SST Core 31 & 33, Radiophemietry Daka Pu-230, Pu-239/240, Pu MS Impropio.

ADDENDOM 5 REV DAWHC-SD-WM-DP-DZ.C

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WHC-SD-WM-DP-024 ADDENDUM 5 REV DA

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	Pacific Northwest Labora	1000

Project Number

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L. R. Greenwood File/LS

October 9, 1992

To Susan McKinley

From J. H. Kaye J. H. Hagen

Analytical Results for Mi-59 and Mi-63 in Tank
Till. Cores 31 and 33

Nine core composite samples were received from the hot cell for ⁵⁹⁴³Ni analysis (92-08278-H1, -H2, -H3; 92-08280-H1, -H2; 92-08282-H1, -H2; and 92-08284-H1 and -H2). These represent fused composite samples and duplicates, and a fused hot cell blank from Tank Till, Cores 31 and 33. These samples were analyzed with their corresponding QC samples (two reagent blanks, a blank spike, a matrix spike, and a sample duplicate) according to procedures PNL-ALO-473, Revision-I, Nickel Separation from Radioactive Solutions, PNL-ALO-464, Procedure for Gamma Counting and Data Reduction in the Low-Level Counting Room, and PNL-ALO-474, Liquid Scintillation Counting.

Revision 1 of PNL-ALO-473 has been submitted to the ALO office. In this revised procedure, the option of 3 dimethylglyoxime precipitations was used. This was followed by anion exchange and the remainder of the procedure (steps 5.1 through 5.5, then steps 5.13 to end).

A summary of the results obtained is given in the attached table.

Chemical yields are determined by ICP measurement of stable Ni on an aliquot of the final solution prepared for LSC. The yield based on gravimetric means is determined but is not used for calculation of results. It is performed to determine if electroplating was successful. The error in the ICP measurements is quoted at ±15% relative; however ±5% is a better estimate of the error based on typical performance criteria with ICP analysis on ideal matrices such as presented with the samples submitted for analysis (2 volume percent nitric acid and no interfering ions). The data are therefore calculated with the error of ±5% applied to the ICP determination.

The fusion blank is used to monitor contamination due to hot cell operations; the reagent blanks are used to monitor contamination due to the laboratory analysis. The fusion blank and reagent blanks are calculated similarly to the composite fusion samples in order to provide a value which can be compared to the samples. The fusion blank gave a slightly positive result for ²³Ni, but this is insignificant compared to the sample results. A mean sample size of 0.3006 grams was used for the reagent blanks. A value of 60 ml was used for the sample volume analyzed for the reagent blanks, as this amount was used for seven out of the eleven sample volumes.

Detection limit estimates for the reagent blanks were approximately 4E-06 for both $^{53}\mathrm{Ni}$ and $^{59}\mathrm{Ni}$.

WHC-EP-0806 WHC-SD-WM-DP-024 ADDENDUM 5 REV 0 A



Project Number

Internal Clamburgon

File/LB

- September 30, 1992

To SG McKinley

Ann CO Harvey 60 Home

Subject Radiochemical Results for Till (Core 31 & 33)

The radiochemical analyses were performed on four core composites, two each from Core 31 and 33. The solid core composite was prepared by potassium hydroxide (KOH) fusion, following procedure PNL-ALO-103. Duplicate samples and a single process blank were analyzed. The samples were analyzed for Total Alpha Plutonium, Uranium and for isotopic composition of the Pu and U. There was generally good agreement between duplicates for Uranium except for samples 92-08280-H1 and H2. However, the results were verified by re-analysis. The agreement of duplicates for Pu was poor, differing by a factor of approximately two. The duplicates for samples 92-08278 and 92-08280 were reanalyzed with good agreement with the original measurement, indicating acceptable analytical performance. The results of the individual analysis are included with the data. The average of the analyses is reported in the summary tables. The reason for the disagreement is unknown.

______ The Nickel-63 analysis are not complete and will be reported at a later date.

Cancur: 3:3-5-9-93

54-1900-001 (10/00)

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u-lauer, U HS leotoplo u-lauery Date table 13bt 857 Core 31 & 33, Rediochemiatry Date

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49.3219	4500'0	1979 0	9500 B	\$	3.10	des/	92-08284-B-II
99.3198	1100.0	\$019.0	£ 500 0	ÿ	91.5	47/1	12-0-20200-26
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89.3099	1100.0	\$414.0	5900.0	•	2 300	94/1	92-08280-M-R
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Nickel-63 and Nickel-69	Results	Tank T-111.	Cores 31 and 33
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41.010	Cardin Train	NI-63	t/- one	APD	NI-59	+f- one	RPD	1
ALO Sample ID	Sample Type	(inicroCVQ)	elgma*		(microCVg)	*Amgla		•
92-08278-111	Composite t fusion, Core 31	7.34E-03	4.0E-04		7.16E-05	1.9E-05		
92-08278-112	Composite & fusion, Core 31	1.13E-02	8.2E-04		9.43E-05	2.7E~05		
92-08278-113	Fusion Blank	6.17E-08	1.4E-08	•	<2E-06		•	> #5
92-08280-111	Composite 2 fusion, Core 31	7.32E-03	4.0E-04		6.53E-05	1.8E05		6 3
92-06280-112	Composite 2 fusion, Core 31	3.62E-03	2.0E-04		3.30E-05	1.0E05		m +
92-08280-H2	Composite 2 fusion, Core 31	3.53E-03	1.9E-04	3	3.36E-05	9.4E-08	2	* (/
	(duplicate)		1			I		
92-08282-111	Composite II fusion, Core 33	6.36E-03	3.5E-04		4.74E-05	1.5E05		33 ± € (1)
92-08282-112	Composite il fusion, Core 33	4.54E-03	2.6E-04		4.14E-05	1.2E-05		•
82-08284-111	Composite 2 fusion, Core 33	4.26E-03	2.3E-04		3.76E-05	1.0E05		æ (⊏
92-08264-112	Composite 2 fusion, Core 33	4.92E-03	2.7E-04		4.37E-05	1.2E05		₩.
	Aougont Blank 1 * *	<5E-06	•		<2E-08	I		
	Rosgent Blank 2**	<6E-06	1		<5E-06			0.
ı			. '		•			
	' i		1					
	i i	bebbA	+f−,pna		Becovered	1/- On a	Recovery	
	' 1	(dpin)	• il Qana •		(dpm)	*Acngla	-(berceut)	Þ
1	Blank Spike (NI-63)	776	12		735	. 40	95	
	Matrix Spike (NI-63)	776	12		704	77	91	
	mentur often fragon)	,,,	1		,,,	•••	• • • • • • • • • • • • • • • • • • • •	

^{*} One sigma uncertainties are leased on propagation of mass, volume, counting, and yield determination uncertainties.

Entered By Date 10/9/92
Hoviewed By Letter Julien Date 10-9-12

^{**} Two reagent blanks were used for this batch of analyses. The results have been normalized relative to the sample volumes utilized.

WHC-EP-0806

WHC-SD-WM-DP-024 ADDENDUM-5, REV-0-A

ORGANIC COMPOUND ANALYSIS REPORT

SEMIVOLATILE COMPOUNDS

SAMPLE ANALYSIS REPORTED

Analysis of SST samples from Cores 31 and 33 Composites, Building 2228 Hot Cell Blanks, and Field Blanks for semivolatile organic compounds by gas chromatography/mass spectrometry (GC/MS) is the subject of this report.

SAMPLE DESCRIPTION AND PREPARATION

Sample ID	ACL Lab Number
Core 31 Composite 1	92-08277
Care 31 Composite 2 Care 33 Composite 1	92-0827 9 92-08281
Core 33 Composite 2 222S Hot Call Blank	92-08283 92-08285
Field Slank	92-08286

The samples were received on 07/30/92. Extractions of both samples and spiked samples were performed according to the following procedures.

 Extraction 		PNL-ALO-120 for the solids
		PML-ALO-122 for the water sample
	partitio	ning
• Extraction	location	Analytical Chemistry Hot Cells and labs
_		306/308, 325 building
• Extraction	i type	Sonication, medium level soil for core samples
		Separatory funnel, medium level water for
		2225 Hot Call Blank and Field Blank
		an ies

• Sample/Extract storage 4°C(+/-2°) temperature

The procedure used to extract the matrix spiked core sample resulted in a two phase extract concentrate. The second phase was eliminated by adding approximately 20 mL methylene chloride to the extract concentrates, and mixing each of these solutions with 20 grams sodium sulfate for apout 30 minutes.

ANALYSIS METHOD

• GC/MS procedure:

PNL-ALO-345.

GC/MS instrumentation:

HP-5890/5970 GC/MS (WB38473)

• GC/MS location:

Lab 325, 325 building.

	Analyte.	\$14.4L	PREP BLK	Result 811	Result 17	Average	RPD
		B Rec	(44/4)	(ug/g)	(ug/g)	(ug/g)	3 6
	14.309					·	
	USC	EXOTHER	'NA	EXOTHERM	EXOTHERM	НА	MA
·		l					
	10 309						1
	164	100 7	НА	87.0	84.0	47.0	0.1/3
		<u> </u>					
	14.309						
A WATER		99.2	HA	(NO. N	79.6	80.3	t.13

			d	-,		· · · · · · · · · · · · · · · · · · ·	
	Amilytti	814 81	PREP BLK	Result #1	Result 82	Avenge	AIID
		A Nec	(ug/g)	(ug/g)	(vg/g)	(vg/g)	Æ
•	10.340	<u> </u>					- ",
	DSC	EXOTHER	NA	MASHTOKS	EXOTHERM	HA	MA
	10-310						
	TOA	100.7	NA	87.0	NA	HA	MA
						_	
	17:310						
A WATER		99.1	'NA	J 112.4	32.6	57.	94.4

							
	Apolyte	\$14.41	PREP BIK	Result #1	Result 12	Avenge	APD G1A
		R Rec	(mg/g)	(ug/g)	(ug/g)	(ug/g)	S.
	10.341						
	DSC	EXOTHER	NA	MASHTOXS	EXOTHERM	NA	HA
	· ;						
	14-111		<u> </u>				
	TOA	100.7	NA.	85.0	NA.	NA	NA
			l				
	14.311			_			
S WATER		99.2	NA	62.2	97.3	87.6	\$6.0

Segment 4

	Analyta	51001	PAEP BLK	Result #1	Result #2	Average	APD
		A Rec	(mg/g)	(ug/jj)	(ug/g)	(44/8)	
	20 313						
	DIC	EXOTHER	NA	Смантоха		NA	NA .
	10.313						1
	TGA	100 7	NA	62.6	. NA	HA	NA
						}	1
	14.312						
S WATER		. 19 2	NA	Jns	39.6	40.5	25.84

camen 5

	Analyta	84.01	PREP BLE	Reads #1	Rends 173	Avenge	APD
		# Rec	(ug/g)	(jug/g)	(vg/g)	(4g/g)	
	10.313	1					
·····	DSC	EXOTHER	NA .	NO EXO	NO EXO	НА	NA
	14.313		<u> </u>			 	}
	IGA	99 2	NA	84 0	[HA	HA	NA
	10.313	<u> </u>					
WATER	1	99 2	NA	74.4	41.4	85.4	11.9

Stamus 1

	Analyte	514 81	PREP DI K	Result #1	Result #3	Average	APD 018
Ì		S Acc	(ug/g)	(ug/g)	(4g/g)	(vg/g)	*
	10.316						1
	DSC	EXOTHER	HA	NO EXG	HO EXO	НА	НА
	10.316	<u> </u>					
	IOA	99)	NA	85.1	81.4	64.1	0 (1)
	10.316			_]			
S WATER		100 7	НА	76.4	77.2	76.0	104

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	Assins	514 61	PREP BLK	Result #1	Result #2	Average	MD
· · · · · · · · · · · · · · · · · · ·		II Rec	(44/4)	(vg/g)	(ug/g)	(ug/g)	
	14 307						
	DSC	EXOTHER	NA.	NO EXOTHER	НА	HA	NA
			<u> </u>				
<u> </u>	10.307	<u> </u>			1	i .	
	104	99.3	. NA	45.6	HA	NA	NA
		<u> </u>					
	10.347				-		
& WATER		100 7	NA	76.7	76.4	76.6	0.31

Segment f

	Analyta	1848	PREP BLK	Repub #1	Result 82	Avenge	AI D
		S Ruc	(vg/g)	(ug/g)	(ug/g)	(µg/g)	5
	10 310					1	
	υsc	FROTHEN	NA	но ехо	HA	NA	NA
	10.316						
	104	99.3	HA	71.0	NA	NA	NA
11	18.336						
S WATER		101.0	HA	76 0	74.7	75.0	2 90

Scament Sh

	Analyis	514 61	PREP BLK	Result 81	Result #2	Ачельде	APD
		1 Acc	tug/g3	(ug/g)	(ug/g)	(Hg/g)	, s
	18 414	_					
	DSC	EXOTHER	: NA	NO EXO	но ехо	НА	НА
	14.414						
	104	97.4	- NA	72.1	73 0	72 (0.14
	18.414						
& WATER		99 9	HA	69.3	71.2	10.4	142

	· 	- r		Special An	alysis of Segn	nent 9 (white	layer)	······································		,		
	Asslyte	INCIN		Pinep bl K	Result 01	Brank /2	Average	ALD .	५५ .स	Spile	DETLIM	Ratio
.414 .		144 (m/L)	* Au	(1-9/8)	(ng!/g)	(ug/g)	(ug/g)		(vg/g)	S Rec	(ug/g)	MDL
P-Acid .	AI	JUL JULO	114 40	1 34E-01	J 1.48E-01	1.51E-01	1.50E-02	7.29	99.86	45.40	2 40E+00	62
	<u>\$\$</u>	3000	91 PU	3 11E-01	UJ -1 016.00	-6.54B-00	-4.21E+00	-11.17	20.44	80 60	1.77E-01	-0
	A4	, xwo	60 60	1 00E-00	M -4 17 E-00	-7.55E+00	-5.69E+Q0	-54.40	99.84	71.60	3 006-00	-1
	Bo	10000	64.50	3 QUE -01	1 71E-01	5.652-02	\$ 44E-02	0.94	199.74	80.30	1 00E-01	1097
	No.	10000	91 40	1 00E-01	UJ -1 27E-01	-7.44E-01	-1 01E-01	-51.74	199.76	88.80	1.006-01	-1
	C4	10000	91 00	4 QUE-01	J 1 14E-00	2.47E-00	2.52E-00	4.24	199.76	17.40	4 00E-01	
	Ca	10000	128 00	9 40E-01	丁 • 95E+01	9.29E+02	9.12E-02	3.69	100.76	33.60	4.40E-40	201
	Or .	Souo	94 7 U	1 00E-01	丁 I NE-01	1.99E-03	1.91E-0)	0.74	99.41	34.10	10-300 t	3701
	C ₀	Jour	93 10	1.03E+00	UT 1 11E-00	3.43E+00	3 47E-00	12.14	199.76	\$0.10	I COE-OI	
	Chi	jouo	34 FO	\$ 04E-01	J	9 03E-00	P.15E-00	3.43	22 10	89.50	4 602 01	2.1
	F.	SUL)	151 20	1 ME-00	1 32E-04	1.54E+04	1.53E+04	1.47	59.88	-203.0	1.00E.00	1321
	Ph	souo	90 30	4 2UE+0U	J 1 46.01	9 43E-01	9 01E-01	0.10	** **	84.70	4 20E-00	
	Ma	XXQ	101 40	1.07E-01	J 25E-02	1 11E-01	2.30E+01	4.26	99.41	61.30) (ME-OI	761
	Ma	xw	11 20	3 00E-01	J 4 191E-01	3.04E-03	3 DIE-01	1.00	99 68	-49	2 QUE-01	23
→	NI	3000	11 20	1.70E-00	T #228-01	\$.04E+01	1.99E+01	1.51	99.11	87.50	1.70E-00	
3	K	KUO	94 60	1.13E-01	J 101E-01	4 07E-02	3.93E-02	4.71	99.41	99.10	1.128-01	
J	34	3000	98 20	7.40E-00	J -1 10E-01	-3.74E+01	-7.17E-01	-30.01	99.11	13.90	7.40E-60	31
	Ag	3000	48 00	1 WE-01	UJ 1.14E-00	9 286-01	J 01E-00	20 15	99.41	31.10	1 00E 01	
	Na Na	1000	157 00	4 20E-01	J 4018-04	4.04E-04	4 025-04	0.74	199.76	-346.1	3.10E-QU	
	v	10000	94 90	3 OUE-91	LT 131E-00	1 526-00	1.45E+00	944	199.76	172.30	3.00E-01	12971
41	Za	10000	94 8Q	2 01E-00	J 2 10E-01	1.04E-01	1 07E-01	3.09	199.76	90.70	3 00E-01	
	Di	NO.	66 20	7.50E-00	J 2 39E-04	2 40E-04	1 60E+04	0 24	99 10	- \$40 \$		49
		3000	113 40	3 44 E +00	J 166E-01	3 20E-01	3 07E-01	13 45	99 11	47 00	1.50E+00	3161
	C ₄	3000	104 00	1 01 E-01	J 4.19E-01	4 U9E-01	\$ 24E-01	4.76	99.44	97.40	6 (WE 0)	31
		3000	98 00	1.40E-00	4.14E-01	4 41E-01	4.58E-03				1 01E-01	
	 	10000	PI 10	3.80E-0U	J 147E-04			1.17	99.68	-13.90	1.40E-00	3273
·	 	┤─── -}-		·	1	1 47E-04	1.68E-04	117	199.76	-26 90	1 NE-10	2898
·	<u>si</u>	10000	274 00	2 17E-01	- -	4 186-03	4.17E+07	3.31	199.74	394.20	1.308+00	316
	<u>\$</u>	10000	H 20	1.22E+00	J 417E-01	4.19E-02	4 18E-03	0.42	199.74	81.40	1008-01	1393
		3000	113 70	1.436-01	J 1.47E-01	3 49E-03	1.48E-01	1.34	99.11	34 40	2.7UE+00	344
	Se) XCO	91.20	2 30E-00	U.T 1 03 E . 00)	\$ OUE-CO	4 02E-00	4141	911	11 10	1 608-00	
	11	, sow	94 00	4 00E-01	J 4.43E-60	4 21E-09	4.35E+00	4.49	99 88	89 SU	4 COE-01	10
	u	10000	88 30	1 00E-01	U.T. I ILE OI	-1.11E-01	-1.13E+01	-4.08	99.81	131.20	8 UOE-01	-14

Addendin D. Rev. O

Special Analysis of Segment 9 (white layer)

	T	r———	r—	ı ~ *~~~ ı	ແນ້ອາຊ ດາ ວຸລຸຕິເ		_ 		, -	·	
 	Analyte	84.81	PREP BILE	Beauli 21	Result 62	Average	RPD	Spike	DET LIM	Retio	
<u> </u>	10,414	S Bec	(va/a)	(48/6)	(ug/g)	(ug/g)		S Rec	(ug/g)	WDL.	
IC-Wolce	P	107 00	<1 @0E-01	J 4.77E-01	4.24E+03	4.51Æ+03	11.76	119.00	1.00E-01	450.50	
	q	f# 10	<1 (IUE-01	J 143E-03	7.33E-01	4 48 E+03	22.41	N.44	1.00E-01	M 80	
	HOS	03.90	<1 (00E-03	J 6 12E-04	3.48E-04	\$ 00E-04	11 93	111.00	4.00E-02	300 00	
	NO2	101 00	<1 00E+03	ĭ 8 42E-01	6.54E+02	7.47E-01	24 83	99.20	1.006-02	7.49	
	PØ4	114.50	<1.00E-01	J 1116-04	2 42E+04	2.73E+04	9.45	99.00	1.00E+03	273.00	
	504	114 00	<1 00E-03	J 3 31E-03	4.72E+03	1.11E-01	15.44	109.00	1.00E+02	31.83	<u>-</u>
	P/PO4					1.67Æ-00					
	8/804					8 66E-01					
The state of the state of		×		860 SX 1 60	भारतसम्बद्धाः	2788 P. O.					· • • • • • • • • • • • • • • • • • • •
	i	Stå (FL	PREP BLE	Brook #1	Result #2	Average	RPD	Spile	DET LDA	Ratio	S OIT ENR
3	18.414	& Acc	(vCV _{II})	(vCi/s)	(uCifg)	(ucu)		N Ace	(uCi/g)	MDL	
RAD-Pus.	14	104 40	≪6 71E-4	1.1JE-01	1.43E-01	1.40Æ-01	17.86	114.40	6.70E-04	250.75	2 30
the section of the section	िकार ह	\$ E #		1.0		11 YE 17 1				* 1	
	14,414			_							
OEA	C4-137	103 00	<# 2E-3	1 03E-03	9.72E-03	1.00/6-01	5.79	NA	7.10E-01	133 47	2 40
	Eu-134	NA	<2 23E-4	<2 48E-4	@.11E-4	NA	NA	NA	1.40E-04	MA	
	Eu-155	HA	<1.32E-4	□ ##E-4	<1.89E-4	NA	NA	HA	1.19E-04	MA	
	Am-241	HA	Q 61E-4	2 \$0E-03	3 ME-02	3 ME-03	1.40	NA	2 21E-01	125.44	3 100
	Ce-60	101 00	<1 41E-1	<4.172E-5	<7.4E-3	NA	NA	NA	\$ 15E-05	NA	

WHC-EP-0806
WHC-SD-WM-DP-024 ADDENDUM_2, REV 0

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1 00 - 01	10-300 t TJ	actus.	HA	HA	NA :	NA	\$ COE-01	1 00	
) 00E-00	11 1 00E-00	pulse "	NA	NA	NA	NA	1 00E-00	1 00	İ
1 14E-00	M 100E-00	64794	HA	HA	HA :	HA	1 (10) - (10)	1 00	8
4.74E-00	W-3001 LU	8084 ·	HA	MA	NA :	МÀ	4 (W.L000	1 40	1
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8 00E-00	WT 1 00E-00	pope	NA	NA	HA	NA	1 005-00	90	ļ
4 00 E -00	UJ 1008-00	ecos.	HA	HA	NA .	NA	4 008-00	5 00	} :
1005-01	J 1148-02	ecost	HA	HA	NA	NA	1 00 E-01	GR 40]
4 20E-01	W + 208-01	sout.	NA	NA	NA ·	NA	4 XVE-01) s	
3 00 £ -00	J 1011-01	\$400 B	HA	HA	HA	HA	QG-3QG (M to	
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1 405-01	UJ 7 001-01	8498	MA	NA	NA -	HA	7 408-01	54∞]
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1 toE-01	J 4.91E-01	boths.	NA	NA	HA	HA	\$ 102-01	तम ०३	١i
1 00E-00	UJ 1 ODE-00	3404	HA	NA .	MA	HA	00-300 t	EZ 8	Į
) OUE-00	J 1678-01	9494	HA	HA	MA	HA	00-300 €	d II] :
1.10E-0L	UJ 1 20E-01	3.004	HA	NA	HA	NA	7 10E-01	1.00	l
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WHC-EP-0806

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ļ		Assiyis	Bi4 #1	PREP BLK		Remit #1	_ _	Berok 12	Average	RFD		Rec	DET LIM	Reliës	
	· · · · · · · · · · · · · · · · · · ·	14.517	A Rec	(ug/ml)		(ug/ml)	.	(ug/ml)	(ug/sel.)	5 .	·	5	(ug/mL)	MDL	1
Į	HYAA	As	94 40	CJ 02-4	្រា	T 4 10E	ا_	Q 50E-3	NA :	NA	HA		Ø 50E-3	NA	
l		Se .	101 00	<1 9B-4	u	J 4 10E-1	,	<2 10E-1	MA	MA	NA	-	<0.50E-3	HA	
- {	CYAA	Нв	95 00	<3.02-4	LL:	J 4 10E-	<u>.</u>	<0.10E-1	HA	HA	HA		<0.50E-1	HA	
- 1															管制智力
1		18.504					_ _								
ļ	IC-Walks	r	101.00	<.1	ιί	J		<.1	NA	HA		100	<.1	HA:	
ŀ		a	101 00	< 1	\mathbf{L}	J <.	<u>.</u>	<1	NA .	NA		120	<.1	HA:	
ł		ноз	94 40	<10	и.	J 416	١	<1.0	NA .	NA		85 9	<1.0	NA.	
		HO3	101 00	<1.0	L	J <10	٠	<10	HA .	NA		99.7	<1.0	NA.	
		ю	95 70	<1.0	u	J 41	•	<1.0	NA -	NA		97.3	<1.0	AA	
- {		504	92.90	<1.0	LL	J 41	٠	<1.0	NA .	NA	,	95.7	<1.0	NA	
- {		কিনিক কৰি কৰি		************							7			3 24 14 2	
		10.513												i.	
	BOC-Waller	TOC	90.70	3 vo c	U	JU	3	4.1	HA .	НА	HA	:	. X.0	NA.	
4	· >							1 1 1 1	7 125					्राच्या वर्षे राजावीय	
နှင့်	-35	Assigns	5.4.41	PREP BLK	1	Remk #1		Frenk 12	Average	AtD.		h Ree	DET LIM	Batio	Relativa
ָרָט ו	<u> </u>	18.496	S Acc	(vCVal)		(vCi/sel.)		(isCifeL)	(uCl/int.)				(4/al)	NDL	Count Err
- [IAD-Woler	TA	106.90	<4.31E-4		<3 11E-	٦	<4.39E-4	HA	HA	NA		4.39E-04	NA.	1 00E-01
							7			F. West	, i	** *			1 4 1 (d. 1)
·		18.494										,			
	•	TB	100 10	9.30E-3		<0.33E-		<7 94E-1	NA	NA	HA		\$.38E-01	NA	6 40E-00
ĺ		91-49	* * * * * * * * * * * * * * * * * * *	19 No. 1	11	•		" \		1 1 1 1 1 1		1171	1 1 (N) 1 1 1		*
- [18.496			1									1	
Ì	OEA	CS-137	101 00	<1.17E-6		<1 19E-	•	<1.20E-4	NA	NA	NA		1.17E-01	NA	HA
ľ		Eu-134	N/A	<2 37E-4	<u> </u>	<) 07E-4	•	Q ME-4	NA	NA	NA		2.47E-00	NA	НА
		Eu-133	N/A	<1 018-4		<1 13E-	•	<1.77E-4	NA	NA	НА	·	1.02E-04	·	HA
j		Am-241	NIA	<3 ME-4	1	<0.11E-	- ا،	Q 41E-6	HA	NA	МА		3.34E-06	1	NA .
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	1 141134	1 .6.4	1	au		Field Blaid Rewall	10 Amia	A10 3141	l	PHENT]		
oud Javi	10-0	१ क्षेत्र	17. 17.5		1 they	(TVP)	12:20	(L Pn)	9916 B		•MissA	<u> </u>	tor e
17 +1	3 voE-01	VN	41,44) AH	10 ((1/b/)	\$6-30f f	20-320-1	10-30- 2	00 (14)	(1 h) pp+			1: Y - 10
60 I	to-111'1	YH	VH.	00 0	to-31f.1	to-344.1	to-141 [M	No Jul 6-	or He	troot		<u> </u>	
001	10-300 (VN	VN	60 6	10-300 (10-300 (10-200 1 17	10-300 (or lel	00%	••		
11 1	60-300 E	VH.	VN	terie	00-311 (00-3017	00-1119 9	00-300 €	or idi	0000#	**		
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u i	60-300 P	YH	. AH	40 E1	m-31f +	4 115-00	00-300 + TT	00-31(1)	06 501	eccet	13		
H-fi	10-30+	MA	MA	50 C	50-314 F	10-111'L	T 1 mil at	10-30+ 1	or \$0	00001	ر و		
aff	00-3014	YH	VM	00 €	00-300-0	00-700 0	111 0 m 1 -00	00-300 (01 101	(Basel)	15		
on h	00-300 f	YH	VNI	00 P	00-300 0	00-300 1	00-1001_[1]	00-300 t	OF POS	annot	•5		
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Ţ	10-300 [VH	HV.	00 ff	10-300 1	10-300 1	10-1700 I_T7	10-300 I	00 (01	OCO\$	• 1		
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3	00-300 f	VN	VM	0F II	10-369 £	10-314.F	10-1111 C	0r)- 300 €	101 10	0770	711		
4	00-300 t	VN	YN	et i	10-314)	10-300 P	10-310 1 C	300 €	OK COO!	coot	911		
A T	10-3001	VN	YH	00 9	10-30t-1	To-Hot 1	10-302 1 IYI	10-30(1)	er for	timoli	IH		- ∢
录	20-3211	VH	HY	to 1	10-7111	10-321-1	20-321 1_[7]	£0-371 f .	ti (c)	0004	-		- -36
है	10-300 L	VH	VH	∞ •	10-201 L	10-309 L	10-3001_[7]	10-300 L	ot 101	0007			
ON V	00-300 f	-VH	VN	∞ •	00-200 f	00- Joo f	00-301 (CT)	00-300 f	00 00	one			
र ठूलर	10-301 (VH		nt.	10-1401	10-3141	10-30t C	10-301 (Ol: (14	topo!	N.		
DEDE	00-300 f	-VH	V#	∞ •	00-200 f	00-300 f	00-3001 111	00-300 f	00:101	00001			
nyı leti	00-300 f	VN	VH	14 9	10-312)	10-341+	10-3117	00-300 t	00 101	cocot			
ત્ય	10-30f £	VH	VN	00 0	10-30F £	10-301 L	10-305 L_ETT		ot sof	enos			
60 142	00-300 9	VH	N.N	nı	10-39(1	10-316 1	10-311 1	00-300 9	00 99	0005			
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и і	10-30+1	- VN	VN	W11	10-345 £	10-311 €	10-311 1 TM	10-30+1	or rol.	DUDL	*1		 -
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H 1H	10-30t I	- VH		91 (10-311)	10-311-1	10-319 - [10-142 €	CE EL	10000		<u></u>	
91 21	on- 100 f	-VH		001	10-319 6	10-311 1	10-311 1	00-300 (-		! <u></u>	
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WHC-EP-0806

			· · · · · · · · · · · · · · · · · · ·		Field Black		!				
	Analyse	314 01	PREP DI K	Reside #1	Result #3	Average	APD	Spk Res	DET LIM	Ballo	
· · · · · · · · · · · · · · · · · · ·	18:516	A Acc	(ug/ast.)	(ug#m1)	(ug/mL)	(ug/mlL)	5	. 5	(ug/m1.)	M/DL	
ID AA	A•	98 40	<3 0E-4	UJ 42 10E-3	<2.50€-3	NA	NA :	MA	≪3.50E-3	MA	
'	Se	104 00	<3 0E-4	UJ @ 108-1		NA ·	NA	NA	<2.50E-3	HA	
CYAA	Ha	95 00	<3.06-4	UI die-i	<2.50E-3	NA	NA	NA	<2.50E-3	НА	
	The Control of										7.推翻音
	18.505				-		1				
IC-Wales	,	191 00	<.1	UJ 1.502-61	3.30E-01	J.40E-41	3.40	1.005-02	<1	HA	
	a	103 00	<1	UJ <	<.1	MA .	MA.	1.302-03	<1	MA	
	ноз	H 40	<1.0	以丁 (1.30至-00	4.200+00	MA	MA	1.398-01	<1.8	NA	
	ноз	101 00	<10	UJ <1.0	≪1.6	NA	NA	9.97E-01	<1.0	NA	
	104	95.70	<10	W BHE-00	2 41E+00	2.50£-00	3.61	9.732-01	<1.0	HA	
	504	92 90	<1.0	W <10	≪1.0	NA	HA	9.37E-01	<1.0	NA	
			1. ş								: \$ 10 m
· · · · · · · · · · · · · · · · · · ·	10.512	1	1						1	1	
TOC-Water		94.70	Sua C	W 43	€3.3	NA	NA	HA	500	NA	
→		1 2 3 3							***		V 418/5
-37	Assiyte	514 81	PREP BLK	Result #1	Result #3	Average	MD	Sph Rec	DET LIM	Ratio	Relative
7	10.497	E 844	(«CVm1.)	· (uCl/al)	(vCV=1.)	(vCV=IL)	I		(uC:VmL)	NOL	Count Est
RAD-Water	TA	101 90	<4 3PE-6	<4 19E-6	<5.15E-6	NA	NA	HA	4.39E-04	HA	70 70
	18 7 18		1985 A. F.	क्षक हैं। पर इस्के	19 11 12	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			100 100 98 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5		-0.00
	18.497	1	1]]			1		
	T.D.	100 10	<9.30E-3	1 16E-04	1.306-04	1.18E-04	3.39	HA	9.34E-01	1.26	3 50
·		11 11 11 11		RATE OF THE		Y 6 8	****				
	14.497						1,			1	
OEA	Co-137	101 00	<1.17E-6	1.32E-01	1.64E-01	5.39E-05	2.50	NA	1.17E-06	47.74	3 40
	Eu-154	HA	4.11E-1	Q HE-4	4.09E-6	NA	NA	НА	1.87E-06	HA	HÁ
	Eu-133	NA	<1.41E-4	<1.93E-4	<1.648-6	NA	NA	НА	1.01E-06	HA	MA
	Am-241	NA	<3.545-6	Q.71E-6	Q.71E-6	NA.	NA	HA	3.54E-04	NA	HA
• •	Co-60	101.00	<1.048-6	<1 07E-6	<1.118-6	NA.	HA	MA	#.04E-04	NA	NA

T-111 Core 31 Composite 1

, <u></u>		- .		<u></u>	·				·	<u> </u>				
 	Analyte	IMCS NA		PREP DLK	Result #1	Rend #3	Average	N/D	5 pt +44	Epika	DET LOS	Relig	2019	Mail
F4[.44]		·M (19/1)	3 Au	(hq/p)	(46/8)	(ug/g)	(mg/g)	8	(lug/e)	1 14	(ug/g)	I/DL	Joel	64
ICP-Acid	AI	1000	121.40	1 23E-0	J 5.81E-02	\$.406+02	\$.44E-02	141	11 11 11	41 40	2 40E-00	245.31		
	56 111	1000	P6 80	1.77E-01	LLT 3.19E-01	1.10E-01	3.04E-01	34 03	53 94	106 10	1.77E-01	1.72	1	
	Α.	1000	87.90	3 00E+00	11 T 3 17E-00	1.11E-00	3.15E-00	147	33.94	47.00	3 00E-40	19.05		
	Ba	2000	49 90	3 99E 01	J 3.73E-01	1.64E-01	\$.70E-01	1.58	107 67	91.70	3 00E-01	190.14		
	Be	2000	H 40	1.00E 01.	WT 1.00E-01	1.042-01	1.032-01	148	107.07	\$4.40	1 00E-01	1.05		F
	CI	2000	57.80	-4.44E-01.	UT 7.11E-00	7.195-00	1.12E-00	0.74	100.01	88 89	4.90E-01	14 01		
	C ₄	2000	111 00	6.122-01.	丁 1 26E-01	. 2118-01	3.20E-01	100	101 01	-40 40	4 40E-00	171 %	0.11	
	C ₄	1000	93.20	9 00 \$-01	J 1.002-03	1.01-01	1.M2-01	325	HE	0.10	10-300 €	2064 94	b	•
_:	C4	2000	11.10	10-200	LC 3.36E-00	3.428-06	3.40E-00	1.31	107.07	89.60	4 00E-01	4.25	مَا	
	<u> </u>	1900	14 10	4 00E-B1	J 1.17E-01	1.40 E-01	3.528-01	14)	13 H	8F.30	4 00E-01	61 12	P	Ι_
- 	Pe .	1000	91.60	4.37E-00	J 1.95E-04	1.002-04	1.07E-04	3.10	11 14	-610 6	1 (46.00	19316.3	5,	+ FC-
	<u></u>	Jau -	11 10	4 30E-00	J 4 81E-03	4.40 E-02	4.738-03	3.32	33.84	70.10	4 305-00	76.34		I
<u></u> -	Mg	1000	97.30	3 145-00	J 4 31E-07	4.37E-02	4.33E-01	1.37	33 H	41 30	3 OUE-01	1430.74	-6-	EP-0806
· 	Ma	1000	# 40	2 00E-01	J 6 31E-03	6.07E-03	6.19E-03	3.76	33.H	-234.3	3 00E-01	30937.1	100	∏ &
P.	พเ	1000	17.70	1.70E+00	J 1.HE-01	1.49E-02	1.11E-02	3.61	11 14	H 10	1.702-40	10 11	-8-] 8
₩	K	1000	105 80	1 67E-Q1	1.11E-03	1.012+01	1.106-01	2 24	11 14	31.70	1.11E-01	97.07		1
	<u> </u>	1000	80 80	7.608-00	UT ONE-OD	7.92E-00	7.94E-00	1.47	33 84	8.00	7.60E-00	1 01	- B	<u> </u>
		1000	95 90	3 00E-01	7 2 01E-03	3 01E-03	3 015-03	• 11	51.H	87.90	3 00E-01	403 24	<u> </u>	
	H•	2000	111 00	1.102-03	J 1 80E-04	3.71E+04	3.74E+04	2 42	107.67	-331.3	3.10E-00	12111.0	443	00
	V	XOO	1) 60	-3.31E-01	J 131E-01	[2)E-01	1.37E-01	3.62	103 \$1	\$0.40	3.00E-01	25.39		<u> </u>
		3000	96 20	9.19E-01	J 1 20E-01	7.41E-01	7.ME+01	6.57	107.87	88.30	10-300 E	264.80	2	
, —	Pi	1000	14 40	7.50E+00	T 117E-04	3 HE-04	3.HE-04	1.57	53 94	-319 0	7.30E-00	3161.34	644	. 00
	·	1000	113.10	2.74E-01	10-310 E-01	1.48-01	2.71E-01	71.17	33 M	25.00	6.008-01	43 21		
·	<u></u>	1000	107.00	1 01E-01	M 1 11E-01	3.39E-01	3.24E-01	7.02	13.M	97.70	1.91E-01	3.33		
	<u> </u>	1000	93.50	1 40E-00	丁 171E-01	3.44E-0J	3.72E-03	1.77	5) 14	-11.40	1.40E-00	2457.05	0.04	00
	P	2000	97 60	9 11E-00	J 1 00E-04	1.02E-04	1.01E-04	1.10	107.67	473.40	\$.80E-00	1748 84	0.21	01
	H	2000	201 00	\$ 01E-01	J 4 31E-07	J.201-02	4.07E-07	19 24	107.87	1.40	1 30E-00	170 PH		<u> </u>
	<u> </u>	2000	10.40	3 00E-01	丁 3 65E-03	3.718-07	2 928-02	3.42	107.87	\$1.00	3 00E-01	936 P6		
		1000	100 40	9.59E+00	J 1.24E-01	1.312-03	1.336-41	2.23	J).H	\$1.00	3 70E-00	454.66	0.11	0.1
	Ba .	1000	94.50	1 44 E-00	117 4.112.00	4.212-00	4.212-00	3.13	11.14	89.40	1 608-00	341		
	71	1000	89.80	4 00E-01	J 1 116-01	3.ME-01	2.ME-01	1.04	M	\$3.50	1-006-01	73.50		
	21	2000	61 10) OCE -01	U T 47E-01	10-3K t	8 40E-01	1.48	107.57	0 60		MA		Γ
		<u> </u>				L]			

T-111 Core 31 Composite 1

Direct/Unhom	Analyte	518 81	PREP DLK	Result #1	Bleault #3	Average	MD	Sph Rec	DET LD4	Batla	imeq	Mai Bal
		3 Res	(mg/g)	(145/2)	(ug/g)	(lug/g)	8	18	(ug/g)	IVPL	Jone	4/61
	10.433											
	DSC	EXOTHER	NA	EXOTHERM	EXOTHERM	HA	NA	HA	NA	NA		
理學 计图象经验	T. W. Carlot	\$ 11.5		型型: 計畫(E)		8 to 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	* J. 18		4.7	** 7.1	13.5	
	1413	Ï		(5)	(6)	(A)						
	TOA	91.00	NA	T 85.30	41.30	73.30	32.74	NA	HA	NA		
	17124	erre 👫			1998年本共200	基準日第18		有数字 数				*
	18 416, 417			(*)	(£)	(5)						
S WATER	AWTR	pt).70	HA	74.40	74 00	74.60	0.54	NA	MA	NA		0.746
	No. of the Control of	14. A		5.1% V 3	8	1 1 1 1 1 V	28 7 F	ा स्व				
	1426											
НУЛА	A6	PHL										
	84	PHL										
	Hg	101.2	49 12			1.50E-00	23.04	123 90		12.77		************
18 18 18 P. S. C. P. S. C. P. S. C.	11012		*	11100000000000000000000000000000000000	表现 新江		X X	11 1	10.43	1.0		
	18 440											
CN-Dir	CH CH	99.00	ರ 0	<1.0 UJ	<4.5	HA	HA	40.10	3.0	NA	0 000	
	Culif=Q14					NA		I				

WHC-EP-0806
WHC-SD-WM-DP-024 ADDENDUM 2, REV 0

·	Analyte	IMCS	•	PREP BLK	Bicault #1	Beaute 57	Average	APD	514 .44	Spike	DET LD4	Raile
J 437		a disted	S Rec	(46/6)	(4/4)	(44/0)	(ug/g)		(4/4)	A Rec	(ug/a)	WDL
P-Weist	Al	Socio	95 60	1 40E-CO	(LJ 6 4)E+00	6.99E+00	4.71E-00	8.33	1250	99.50	2 406 -00	2 80
	54	Jouo	90 40	1 7/E-01	LLT 1.77E-01	1.77E-01	1.77E-01	9.10	1750	74.10	1.77E-01	1 40
	14.	\$000	98 40	3 OUE-00	(1) WE-40)) 00E-00	J.00E-00	0.10	2300	44.20	3 toE-00	1.00
	De .	10000	90 30) COE-01	W 3 OUE-01	1.09E-01) DIE-01	3.13	2500	88.80	1 008-01	1.01
	De .	10000	N N	1.00E-01	W-346.0 TH	1.008-61	1.00E-01	0.10	2500	\$8.60	1.008-01	1.00
	લ	lanco	97.90	4 00E-01	115 4 005 41	4.008-41	4.008-01	0.10	2300	91.20	4 00E-01	1.00
	C ₄	10000	00,70	2.10E-01	W 1.128-01	. 3.05B-01	J.012-01	1.33	2300	91.00	4.40E-00	31.53
	Cı	Jaug	100 40	10-300 (J 2012-02	2.115-02	2.008-02	2 25	1250	94.80	9 COE-01	3)3.31
	C.	10000	P4 40	\$ 00E -\$1	W 7.002-01	10-300 4	1 00E-01	0.10	2500	97.60	6 COE -01	1.00
	<u> </u>	3000	95 80	4 00E-01	MJ 4 WE-01	4 00E-01	4 (GE-0)	8.10	1250	\$4.90	4 (OE-9)	1.00
	F.	3400	97 00	00-300 (J 1028-02	\$ 70E-01	1.96E-01	34.97	1250	P4 70) toE -00	79 63
	n	JOLO	PH 00	4 20E-00	UJ . 198-00	7.91£-00	7.05E-00	24.34	1250	91,20	6 20E •00	1 14
	Ma	Seco	11 70	4 40E-01	(II) 216-00	2 47E-(00	J.ME-IIO	10.77	3250	99,70	3 (06-01	9.01
	Ма	X00	93 10) COE-01	J 1 93E-01	1 01E-01	1.47E-01	42.70	1250	93.7u	3 (0-30)	73 45
	<u></u>	KW	91 50	1.76E-60	W 1 70E-00	1.70E-00	1.70E-00	0.10	3150	98.10	1.701-00	1 00
	K	\$400	103.10	1 136-01	J 124E-02	7.40E+02	7.14E-03	1.65	1230	63.10	1 125-01	65.51
	Se	\$000	95 60	7.40E-00	UJ 1.11E-00	7.40E-(10	7.40E-00	0.10	110	191.40	7.40E-00	3.00
	As	3000	101.00	10-300 (W 4 ME-01	7.70E-01	6.39E-01	43.78	#130	102.50	10-300 t	1.24
	Na :		94 40	31 88E-01	J 198-04	3.41E-04	3.40E-04	0.47	31500	-27.70	3 10E-00	10975.7
	V	10000	95 10	19-300 lt	WJ 4 ME-01	3.00E-04	1 00E-01	0.10	1:500	77.40) mt-01	1.00
	Za	10000	97.90	-1:49E-00	LW 3 00E-01	3 00E -01	3.00E-01	8 10	2300	97.00	3 (voE -01	1 00
	ы	1000	99 60	1 36E-00	J 1.47E-02	8 34E+01	1.13E+03	33.50	1250	39.00	7.50E-00	11.11
· 	3	SOUO	87 80	4 WOE-61	J 3 11E-00	3.50E-00	1.11E-00	11.44	1330	64 10	6 WE-01	3 31
	C ₄	101.0	99.00	1 015-01	LU 1 01E-01	1.01E-01	1.01E-01	0 10	1250	29.30	1 DIE-01	1.00
	La	3000	99.10	1.40E+00	J 7.018-00	\$.00E-00	6.92E+00	33.93	1250	100.50	1.40E-40	4.30
·· ··············	,	16000	93.40	\$.\$0E-00	J 3.01E-01	1.612-01	\$.74E-01	4.43	2500	\$2 00	\$ 80E-00	993.36
• •	N	10000	80 00	1.30E-00	J 3.305-07	3.452-02	4.55E-02	42.44	2100	99.10	1.308-00	334.55
· 	3,	10000	\$7.10).00E-01	M2 148-00	1.148-01	1.31£-00	45.14	2100	99.00	1 00E-01	4.03
		1000	97.10	1 70E-00	J 1.188-01	1.312-01	1.102-01	3.13	\$2.50	11.70	3 708-00	41 49
	5.0	3000	92.70	1.80E-00	W 1 108-00	1.40E-00	1.608-00	0.10	1130	9) 00	1 406-00	1.00
	11	1000	91.60	4 WE-01	UJ 4.00E-01	4 00E-01	4 00E-01	0.10	1350)) 40	4 00E-01	1.00
	24	1000	98 30	1 COE 401	11 T' 99E-01	\$ 00E-01	1 00E 01	0 10	1		1	7.40

WHC-EP-0806

	Analyse	514 41	THEP BLK	Besule Mi	Kroult #3	Avamae	MD	Spl Rec	DET LD4	Ratio	9 m4	Mail Bel
		1 Res	(ig/g)	(ug/g)i	(ug/g)	(ug/g)		5	(lug/g)	NDL	lons	(4)
	F#1437	!	11									
IC-Wates	ip.	P4 50	< OUE-1	J 101E-01	3 14E-03	3 09 E -0 3	3.57	117.00	1.00E-01	301 50	0 16	9 00
	(C)	101 0	<1 00E-1	J 4 40 E-01	4 738-07	4 70E-03	1.40	111 00	1 00E-01	46 95	- 1	• 00
	ING3	93 20	<1 00E-3	J 441E-04	4 456-04	4 43 E-04	9 90	101.00	1.00E-03	443 (40	9 71	004
	INO/S	\$0 10	<1 00E+2	UJ (1 ME)	<1 10E1	MA	HA	104.30	1 00E-01	NA		8 00
·	PO4	94 50	<1 00E+2	T 107E-01	1 14E-01	1 616-04	6 81	· \$4 00	1 COE-01	161.50		0 0)
·	303	90 70	<1 00E+2	J 1 4DE+01	3 69E-01	3.69E-01	9,00	81.70	1.00E-02	34.90		0.01
<u>i</u>	P ACID ICTO	ı IC				1.91E-00						
	A A CID ICT/SO					9 97E-01			10			
THE REPORT OF THE	Y 15 Y 1		3 3 3 5 5 7	110-111)	() () () () () () () ()	***	X1111		1484		14.7 × 1.
	18:457]									
Pot-witer	HO3	100 0	<\$ 00E-1	J 0.13E-03	0 492-01	9.5116-413	0.63	971.34	3 00E-01	17.04		• 40
	HOD OCHOD E	pee.	<u> </u>			44.113						
>	HON OCHONO		<u> </u>	l		9 430					1	
4 4	केंद्री हुई	14 14 6.		ांगर क्षात् ह		表的意识	語が声が				7.0	₹ **
— .	18 457				, · , <u> </u>							
pH	pH .	100 7	НА	丁 10 17	10 10	10.11	0.20	HA	NA	HA	7	
the search of the				推入民工	alpar ser	1971 7	\$P\$ (1)	1, 77				Heije
	18.457											
NH)-Will	ина	87 6 0	≪4300	<4500 UJ	<4500	NA	NA	93.40	4.50E-03	HA		***************************************
		T 1								3.5	W. 4 1 1 1 1	
	18.457								:			
TOC-Wake	10C	98 30	≪650	J 164E-01	3.30E-01	1.49E+D3	10.49	93 80	\$ 00E-01	4.98		
'	Acetau. Eq		1.72E-04						1		0.291	9 000
		terior (g. t.)						4 7 7 7	114		-	
	18.457								1		 	
TIC- vales	003	101.1	<500	U.T 6.10E-02	<3 00€+3	NA	MA	103.50	\$ 00E-03	NA	0 00.0	0 000

	Assiyie	INC NO		PREP' DE K	ksoult #1	Remb #3	Avenge	MD	ध्रु । स	Spike	DET LEA	Ratio	- 44	И
1.444		1646	# #44	(vg/s)	(Hg/g)	(ug/g)	(41/6)		(4/1)	# Au	(146/8)	MDL	lone	Ì
CP-Fusion		3000	93 30	1 20E-01	J 6 56E-01	6.32E-02	4 44 E-03	3.74	1250	P4:30	1.20E+01	33.70		1
		5000	10.70	8 83E-01	11. J & 81E-0	8.87E+01	\$ ME-01	0.20	1150	101.10	\$ 33E-01	1.04		j
	^^.	3000	91.30	1.50E-01	10-30E-01	1.30E-01) 10E+01	0.20	2300	44.00	1.30E-01	1 1.00		1
	<u>Be</u>	10000	97.00	1.505-00	J 6 04E-01	3.77E-01	1.ME+01	3.34	2300	11.60	1.10E-00	39.31		Ī
	B	10000	H 10	1.15E-00	UJ 1.00E-01	3.01E-01	J.01E-01	0.20	2300	99.76	\$ 00E-01	1.00		1
	C4	10000	99.40	2 00E-00	WJ 9.146-00	7.ME-00	8.21E-00	33.11	2500	100.40	3 00E+00	4.13		1
	C ₄	10000	92 40	1 315-63	丁 1 955-01	. 3.502-63	1.741-01	12.70	2500	M.8 0	1.20E-01	125.39		Ī
	<u> \ </u>	1000	101 30	4.50E-00	J 1.925-01	1.648-01	1.002-61	3.91	1250	104.00	4.50E-00	420 03	출	Ī
	C•	10000	99.10	4 00E-00	UJ 1.03E-01	P.70E+00	1.01E-01	7.44	2500	29.40	4.60E+00	3.33	1	Ī
		Kuo	93 60	3.31E-00	J 3 39 E-01) 44E-91	3.43E-91	244	1370	94.34	3 COE -GO	16.17	Ρ̈́	ī
	F.	3000	97.80	7.13E-01	J 201E-04	2 02£-04	1 01E-04	2 00	1210	123.30	3.00E-00	4103.72	Ę	-
·	n	, tuo	91.70) 10E-01	J 4 31 E-02	4.37E-01	4.40E-03	\$.76	1250	01.99	3 106-01	14.19	B	-
`	Мв	1000	99 50	1 34E-01	T 4.51E-01	4.242-02	4.34E-01	6.37	1210	100.00	1.50E-00	393 00	٦	
, 	Ма	3000	94.60	\$ 04E-00	丁 6.47E-03	6.39E-03	4.31E-03	2.43	1250	101.20	1.00E+00	6361.59	22	-
	М	tous	94 80	B 91E-01	J \$ 41E-01	1.94E-03	3.77E-0)	144	1250	107.00	0 SOE-00	679.30	•	•
	K	suco	100 10	3.40E-01	UJ 5.40E-01	\$ 41E-01	J 4HE-01	0.20	1230	1344.3	3.60E-01	1.00	8	-
	Se	soco	\$3.20	3.0UE-01	UT 3.80E-01	3 81E-01	1 WE-01	9.20	1230	N.40	3.80E-01	1.00	ADDEND	,
	AB	1000	100 80	8 SUE-00	J 2 10E-01	3 10E-03	1 14E+03	3.59	1250	103 10	2.50E-00	85.74	E	•
	114	10000	93 60	4 49E-0)	J 401E-04	3.ME-01	3.91 E-04	3.28	2500	109.80	1.336-01	2569.99	2	1
	٧	10000	P4.30	1.50E-00	J 1.37E-01	1.116-01	1.31E-01	17.23	2500	89.90	2.50E+00	4.85		1
	Za	lando	\$9.40	-1.346-01	J 1.11E-02	9.71E-01	1.04E+01	13.00	1500	99.80	1.50E-00	69.22	<u> </u>	1
	D)	3000	PH 40	3.75E-01	J 114E-04	3 01E-04	1.09E+04	4.47	250	79.40	3.75E-01	330 23	0	1
	D	arra (\$1.10). UUE -00	UJ 1.00E-00	1.01E-00) (O)E-00	9.20	1:250	83.10	3 00E -00	1.00		1
	C.	Kuq	101.30	1 01E-01	LT 1.01E-01	. 3.06E+01	J.04E+01	9.20	1250	97.80	\$ 01E-01	1.00		†
	La	1000	99.30	7.00E-00	T 1.71E-01	3.416-01	3.44B+03	3.24	1210	101.40	1.00E+00	527.61		t
-	,	10000	101.10	3.90E-01	J 1.198-04	1.125-04	1.16E-04	3.67	2500	101.70	2 90E-01	29 4 74		┪
•	54	10000	73.10	6.798-01	J 6 04E-01	1.012-03	J.MI-43	175	2500	79.10	4.308.00	917.29	0.4)	1
	84	10000	97.30	1.308-00	J 1941-01	3 048-67	1.6)4-62	3.30	2500	94.90	1.10E-00	202.19	9.49	+
_ 	-1.	5000	101.00	2.77E-01	J 1.348-03	1.318-63	1.332-03	0 07	1210	102.40	1.33E-01	100 21		ł
 , 		\$600	93 00	9.146-00	W 100E-00	8 02E+00	8 91E-00	0.20	1230	93.50	1 00E-00			╁
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T-111 Core 31 Composite 2

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ŵ	13434	00-301 (@*************************************	11.101	22 0	10-311	10-318 €	10 3M I	1 101 -03	00 (K1	000t	PH.	
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	i	i	00 9	10 101	41 £	10-30L'L	I	10-30K ([7]			.]		l

T-111 Core 31 Composite 2

Direct/Valvan.	Anilyte	314 #1	PREP BLK	Beside #1	Result #2	Avengo	(ILD	Spt. But	DET LD4	Balle	meq.	Mu Bul
		A Dec	(19/8)	(ug/g)	(ug/g)	(44/6)	5		(ve/a)	INDL	Jons	(e/e)
	I# 414											
	DIC	EXO	HA	EXOTHERM	EXOTHERM	NA	. NA	NA.	NA:	NA .		
					4 1 1			* (*)	41.44	7/10	****	
	18 134			(#)	(5)	(A)						
	IGA	97 jū	HA	71.30	49.30	70.20	2 05	NA	NA.	NA		
	3.指标: 10° 2		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1					2				
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	Sai	PHL	<u> </u>]			<u> </u>					
CVAA	Hg	101 3	< 123		1 19E-00	1 PTE-00	4.10	122 00		EAR		
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	10.441		<u> </u>			<u> </u>		i]		
H-Dir	CH	99 90	40	on UJ	941	НА	NA	44 30	3 00E-00	NA	<u> </u>	
	CINIF4CH4					HA						

WHC-SD-WM-DP-024 ADDENDUM_2, REV 0

C-EP-0806

A-47 3

								Sjå ndd	Spile		Ratio
	04464	A Acc	(46/8)	(ug/g)	(ug/g)	(44/4)	1	(m/L)	S Rec	(ug/g)	N/DL
Al	y co	93 60	3 40E-00	UJ 1 01E-01	1 04E-01	1 01E-01	1.70	1250	\$9.50	3 40E-00	4 31
56	30.0	98 40	1.77E+01	UJ 1 71E-01	1.77E-01	1.77E-01	9 (4)	1250	74.10	1.77E-01) at
As	3000	91 40	\$ 00E-00	[1] 1 00E-00	3 00E -00) 00E-00	9 60	2500	44 20	3 008-00	· · · · · ·
<u> #•</u>	10000	91.30	3 CUE-01	[4] 5 32E-01	10-300 t	4 14E-01	\$5.84	2500	98.80	3 00E-01	1 31
Be	10,000	94 90	1 WE-01	UT 99E-02	9 99E-02	9 9/E-01	\$ 00	2500	98 60	10-300 t	1 0
ca	16000	97.90	4 00E-01	UJ + OVE +DI	4 00E-01	4 00E-01	9 00	2500	97.20	4 00E-01	1 00
(Ca	Iduua	90 70	3 10E-01	UT 6 09E-01	3 43E-01	6 16E-01	21.73	2500	91.00	4.40E-00	14 0
Cı	3000	100 40	00E-01	J 2 30E-01	3 24E-03	1 29E-01	0 94	1250	91.80	9 COE-01	254 51
C•	ISILO	96 40	\$ 00E 01	UJ 6 SIE-OI	3 ME-01	8 23E-01	4 27	2500	97.60	0 COE-01	10
cs	3 IZU	95 80	4 (WE-01	11 T 4 WE-01	4 COE-01	4 WE-01	0 00	1250	96 FG	4 GOE-01	1 (4
Fo) sun	V1 00	1 00F-00	J 1 31E-01	1 30E-03	1 40E-01	13.57	1250	\$4.70	1 00E-00	140 4
Ph.	3440	94 00	4 30E-00	[1] 09-1E-00	4 93E-00	7.01E-00	23.10	1250	93.30	\$ 20E-00	1.2
Me	sow)	11 70	4 40E-01	J 4 21 E-00	5 44E-00	3 91E-00	14.41	1230	\$9.70	3 COE-01	13.1
Ma	34-00	93 60	2 00E-01	J 2 70E-01	2 31E-01	2.51E-01	13 27	1230	93.70	3 to E - 01	137 30
NI	scou	37.30	1.70E-00	11 T 1 70E-00	1.70E-40	1.70E+00	0 .00	1250	91.10	1.70E-00	100
K	34.00	101 10	1 136-01	J 1.11E-01	7 636-02	7.43E-02	0.05	1230	45.10	1.12E-01	49.9
Se:	\$000	93 90	7 40E-00	UT 0 44E-00	7.19E-00	\$ 01E-00	10.53	1230	101.10	7.40E-00	1.01
	x.00	101 00	3 COE -01	UJ 1.07E-00	1.24E-00	1.16E-00	14 47	1250	102.50	\$ 00E-01	7.37
No.	10000	P6 60	3 04E-01	J 1 30E-04	3.51E-04	3.50E-04	0 14	2500	-27.70	3 10E-00	11301.4
v	Iuco	93 50	3 00E-01	[1] 8 20E-01	5.94E-01	7.07E-01	31.49	2500	97.40	\$ (IDE-01	1 44
Za	laco	97.90	-3 49E-00	UJ 3 00E-01	3 00E-01	10-3u0 t	9 00	2500	97.00	J 00E-01	1.00
Bi .	50.00	99 60	7 30E-00	J 205E-03	1.768-02	1.91E-03	13-42	1250	39.90	7.50E+00	25 47
	5000	#1 MG	\$ WE-01	J 3 17E-60	3 27E-00	J 19E·W	4.72	1250	16.10	4 (OE-01	3 12
C4	\$600	P9 00	1 01E-01	W 1 01E-01	10-310 1	1.016-01	0 00	1750	99.30	1.01E-01	I OU
La	SOUD	99 10	1 40E-00	7 9 15E-00	7.49E-00	\$ 17E-00	10.55	1730	100.50		4 69
	cont	P3.40	3.80E-00	J 6 11E-01	3.81E-03	J.94E-01	5 01	2500			1021 04
si	1000)	80 00	1.10E-00								430 39
S.	14000	97 80	1 00E-01		1.97E-00						7.10
	\$000	P7 10	2 70£ -60		1 10E-01						444 44
Sa	3000	12.70	1 40E · (co		1 60E-00						
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REV

WHC-SD-WM-DP-024 ADDENDUM_2,

100 0 (120 0 1 43 \$ 00E-05 OF 101 10 20 1 31E-03 (LT : 19E-01 1315 4-3(1 10-3411 101.1 (300 too 451 81 .**š**... , ់ រូ ខ្លួន 1134 : 3 lee o A Malion HO-394 1 1 31 00 10 10-310 (te-300 f 11.1 1 895.403 10-361) 059> 06 84 100 10C-M-1111 65+ FI . . , 11 yr 13 19.99 . . _CY1 00(1) YH 10-30f P 09 10 VN YN 97.40 <4500 1917.4 - 81114 457 21 4.3 . . !! % 1.5 141 121 4: 1 4 VH VH VH 06 0 16.9 16.6 YH LON Hq Hq 457 #1 1.32 1000 . 15. 15 رائد. ٠ VH HOS (IC) WOS (FREC) HON DEJUNOS Elec 1) (1 00 0 69:01 1 00E-01 93.30 14 0 1 316-03 1 338 -03 10-311 5 1+300 (> 0 001 1914 4- 29/9 451 01 00-310-1 S ACTO ICP/SOH IC 1.318-00 A ACID ICP/POR IC 100 EE.FE 10-300 1 11.30 10-311 C 00.00 1.741-03 1.778-01 (1 to[+3 06 96 101 100 10-31L1 _C 134 00 1 006 03 01°0L 59°E 10-3161 10-3161 1 300 (+ 1 00 t4 101 00 0 (1340 1> _ET) 1 005-03 93°00 YN. VH 10-316 t · 300 f> 09 96 HOS 10 g PL . 411.50 1 00E-03 06 00 69 B HO-365 P ____ 1.316.04 10-316 1 t+300 t> 04 64 KON 00 6 10-35L + ____ 23.45 99 B 10-300 F 00 14 TO-BLE-01 1 196-03 CI COE+1 00 14 00 0 91 0 313.30 1 00E-01 00.65 Pt t 10.360 € _C 3.13E-03 10-341 6 1+300 (> 1.401 1211.W-DI 65 P F ((1/1) (H/Pn) 74/4 ((8/3m) (1/m) (2/20) (8,11-) 100 . . -3. Mat 144 1414 DEI 104 514 Rec Q1A Average Beauth da Retolt #1 Milian ALE BLK IF PIS

T-111 Core 31 Composite 2

WHC-EP-0806

Fusion ICP/Acid ICP

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			V		CI	0.13	Al		. 9 915	l	At 1	• • •
Healts field	Tet Cal	'	3 76		Ma	9 00	Cı		0 115		Fe	• •
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100		,			1			1				
	Analyte	5471	PREP DLK	Result #1	Result #2	Avenge	MD	Sid Rec	DET LIM	Raile		Hat But
		B \$44	(1-4/4)	(ug/g).	(ug/g)	(148/4)			(44/8)	MDL	luus	(4/4)
	10 447			: :								
liver	U	104.6	Q HE-2	丁 3 75 6-03	4 (X)E+01	J.84E-01	6.45	104.4	2.50E+02	13.50	9 013	0 004
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that had the	77 7 7 7 7 213 8 7	100		10 .	\$ 1 p.	3 1 3 7	1 1 1	10 30		4 En.	31.7	
	Assiyis	518 81	Prop BU	Acoust 88	Result 12	Average	APD	Spl Rec	DET LD4	Ratio	Rel.	
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NAD-Fus.	TA	101 9	4 14E-3	J 1 69 E-01	3.50E-01	3.59E-Q1	1.20	101.1	7.10E-01	30 43	3.10	
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	T.	M 10	<3 11E-3	3 138-01	2 14E-01	2.15E-01	1.40	76 20	0.33E-02	229.41	1 20	·
	IB/Co-St	1				1 00E-00		:				
The second second					l 					38.		
	14.447					· · · · · ·				<u> </u>		
)EA	Ca-137	101 0	7 BDE-04	3 34E-01	2 34E-01	2.37E-01	9 94	NA	3.70E-04	\$40.54	1.160	
	Eu-154	NA.	Ø.1E-4) 24E-01	<1.06E-3	NA	NA NA	NA	1.20E-01	NA NA	11 000	
	Eu-133	NA	<4 11E-4	(-31) E-3	Q.13E-3	MA	HA	NA	3.93E-04	NA NA	NA AM	
	Am-241	NA	<1.17E-)	4 16E-02	4 03E-03	4.09E-01	3 42	NA	1.14E-01	35.44		
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# 1.43			3 // 1		3 872 4		The second second	17/1	4 07E-63	NA	MA .	
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·	Ln-310	· NA	C) 40E-)	<1 O(E-3	<1 01E-3	NA .	NA	49 PG		NA	J	

White ICP/Acid-ICP

T-111 Core 31 Composite 2

	Amiria	844 #1	Prop Mile	Acc	iuli #1	Result #3	Average	APD	Spå Rina	DET LOA	Balje	Ref.	·
		5 844	(wCifg)		Ci/g)	(uCi/g)	(uCi/g)	\$	5	(uCi/g)	MDL	Q Es	
	18 463	l											
	Am- 341	1601	<1 70E+1	1	1 ME-01	4 66E-01	4 31E-01	16 24	9.00	\$.70E-03	7.56	30.40	
									11.90			110	
	Cus-244			avec de	uc ted		NA	NA	HA	NA	HA		
		Am-341 (GE	A)/Am-241 (ALP	HA)			9.49E-01						
	Marsy San Side S												
	14.467								1				
	MP-237	70 30	<6 49E-3	us	@ 24E-1	← 24E-1	NA	NA	78 70	3.24E-02	MA	3.20	
11 2.114 5		3		1		**	*	4. ()	2 2 1		1 11 11		
	18.467												
	TC-99	316.3	<141E-4		4 16E-01	4.90E-03	4.73E-01	7.19	78.10	1 50E-04	3.56	4.40	
									771.30				
	\$5 C. 45	P. F				()				47.7	1		
3	18.44T			[[
	-129	119.5	<1 40E-3	M	<1 93E-3		NA	NA	\$0.20	1.72E-02	NA		1
	I	I							\$1.10]		1

WHC-EP-0806
WHC-SD-WM-DP-024 ADDENDUM 2, REV 0

WHC-SD-WM-DP-024 ADDENDUM-2, REY O

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T-111 Core 31 Composite 2

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	C 31C	cı 🗀			C31CC3			CDCCI				CDCC2		
Assiyte	Besch #1	Remit #2	Average	Real #	Berndt #2	Average	ľ	Bench #1	Brail (1)	Avenge	Γ	Result #1	Brack #2	Avenge
ICP-acid	(48/4)	(4/4)	(ug/g)	(44/4)	(46/6)	(44/4)	\	(44/6)	(44/a)	(44/4)	-	(44/4)	(4/4)	(4g/g)
Al	5 84E-02	\$ 102-03	\$ \$4E-01	1 01E-01	7.03E-01	7.01E-01	1	4.71E-02	4.71#-01	4.77E-01	77	4 04E-413	4.01 E-02	4 01E-02
56	3.19E-01	1 40E-01	3 04 E -01	1 49E-01	3.41E-01	1 41E-01	-	5 14E-01	2 00 E-01	3.39E-01	-	2412-01	1.01E-01	1 262-01
A	3 1715-00	3.43E-00	3 11E+00	1 61E-00	3 07E-00	1.91E-00		3.50E-00	3 39E-00	3.41E+00		1 10E-G0	1.23 E-00) 06E-00
Ba	3.75E-01	1 44E-01	\$.70E-01	4 44E-01	6 11E-01	4 49E-01		4.70E-01	4.478-01	4.64E-01	-	\$ 79E-01	9 41E-01	\$ 73E-01
Be .	1 041-01	I OLE-OI	I OJE-OI	9 34E 01	1.018-01	171E-01	-	1.20E-61	1.116-01	1.17E-01	-	1.008-01	1.04.6-01	1 01E-01
CA	7.338-00	7.19E-GO	7.33E-00	1 ME-00	7.74E-00	7.64E-00	I -	4.70E-00	4.09E00	4 40E-00	Ī	3.00E+00	3.44E-00	3.71E-00
C4	1 14E-01	3.11E-01	1 30E-01	1.41E-01	3 36E-01	1 44E-01		1.50E-01	1.49E-01	1.49E-03	Γ	1.HE-01	1.31E-03	1 316-01
Ce	1.19E-01	1.43E-03	1.64E-01	1 ME-01	1.84E-03	1.84E-03		1 01E-01	2 05E-01	1.04E-01	7	3 135-03	1.14E-01	1.14E-03
C•	1 10E-00	3 47E-00) 40E-00	1 17E-01	3 79E-00	1.74E-00]_	J 10E-00	1.11E-00	J.11E+00		1 70E-00	J.13E-00	3.91E-00
Ci.	2 17E-01	241E-01	3 51E-01	3 17E-01	1 17E-01	1.94E-01		1 45E-01	1.44E-01	1.44E-01	ŢŢ.	1.30E-01	1.20E-41	1.30E-01
Fo	1.95E-04	1 19E-04	1 97E-04	3 00E-04	101E-04	3 008+04		1.76E+04	1.74E+01	1.75E-04	-	1.72E-04	1.71E-04	1.73E+04
th	4 11E-02	4 49E-02	4 75E-01	1 44E-01	\$ 42E-02	1 43E-02		2 01E-02	1 00E-01	3 01E-02	-	1.492-01	1.47E-01	1 60E-02
Mg	4 14E-01	4 11E-02	4 11E-01	4 41E-01	4 71E-01	4.79E-03		1 01E-01	1 01E-01	1 01E-03	<u> </u>	3 97E-03	1.84E-01	1 POE-03
bla	4 11E-01	4 07E-01	6 19E-01	4 14E-01	4 14E-03	4 148-01		6.77E-03	4 41E-01	4.71E+03		6 31E-01	4 32E-03	6 21E-0)
М	1.MEKR	1.49E-01	1.31E+01	1 37E-01	1 17E-01	1 378-07		1.10E-07	1.01E+01	1.10E-01		1 01E-03	1.09E+02	1 01E-03
K	1118-01	1 00E-03	1.108-01	1 30E-01	1 31E-01	1 315-03	<u> </u> _	1 31E-01	1.312+01	1.31E-01	Ĺ	1016-01	1 02E-01	1 01E-01
34	8 04E-00	7. 92E-00	7.91E-00	1 11E-00	7 ME+00	7.40E-00		1.30E-01	8 40E+00	1015-01		7.40E-00	0 10E-00	1.79E+00
<u> </u>	2 03E-02	1 01E-01	2 01E-01	1 21E-01	1 25E-02	1 27E-02		4.39E-01	4.48E+01	4.43E+01	L	3 19E-01	2.01E-01	3 006-01
Ha	3 00E-04	3 71 E-04	3 74E-04	3 ME-04	1 17E-01	3 87E-04		3 50E-04	J. 49E-04	3.50E-04		3 67E-04	J.4JE-04	3 63E-04
<u>v</u>	1.31E+01	1 27 E-01	1 27E-01	10-311 E	3 13E-01	3.14E-01		1.47E-01	1.118-01	1.39E-01		1 04E-01	9.18E-00	9 99E-00
Za	1.70E-01	7 41 E-01	7 94E-01	1 04E-03	9 48E-01	1016-01		4.47E-61	4 34E-01	4.47E+01	.	1.54E-01	3.45E-01	3.50E-01
81	3.37E-04	1 HE-04	1 14E-04	1 31E-04	3 11E-04	1.11E-04		3 HE-04	1.44E-04	3.41E-04	L	3 81E-04	1.448-04	3 ME-04
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14	3.74E-03	1 44E-01	3 72E-01	1 19E-01	3 44E-03	3 41E-01	L	4 67E-03	4.41E-01	4.64E-03	<u> </u>	4 86E-01	4.91E-01	4.69E-03
P	1 00E-04	1 02E-04	1 01E-04	# #4E-01	0 ME-01	9.94E-01		9.75E-01	9.97E+0)	9 84E-03		1 14E-01	1.13E-04	1.13E+04
Si	4 16E-01	\$.29E-01	4 41E-01	3 11E-01	4.14E-02	4 71E-03		4 80E-01	3.73E-02	\$ 14E-01		3 14E-02	4.90E-01	1 ME-03
5,	2 45E-02	1.71E-01	1 41E-01	1 (06-0)	3 40E-03	1 30E-01		1016-01	3 01 E-172	1 01E-03		3 31E-03	3 37E-01	3 ME-03
3	1 31E-03	1.21E-01	1 1)E-0)	1 27E-01	1 26E-01	1.24E-01		1.14E-01	1.44E-03	1.14E-01		1 22E-01	1 33E-01	1.225-03
Sa	4.13E-00	4 34E-00	4 31E-00	3 13E-00	1 74E-00	1 44E-00		1 NOE-00	1.41E-00	1 01E-00		1 JOE-00	1.72E-00	1 412-00
Ti	3 93E-01	3 ME-01	10-314.S	10-311 C	3 34E-01	3 30E-01		8 80E-00	\$ 00E-CIO	# POE-00	-	4 10E-00	\$ 42E-00	₫.44E+00
Z,	8.47E-01	1 34E-01	1 40E-01	1 30E-01	8 04 E-01	7.74E-01		9 20E-01	9 03E-Q1	9 11E-01		7.70E-01	1 41E-01	() 16E-01

A-54 55

	CHC	CI			CHCCL		i	CHECI				C33CC3	
Analyte	Reads #1	Brank #2	Average	Beach #1	Renk #2	Avarage	Remain #1	Pearl 12	Average		Read (1	Read of	Avenge
	(ug/g)	(4/4)	(ug/g)	(19/8)	(ug/g)	(4g/g)	(ug/g)	(40/4)	(ug/g)		(ug/g)	(44/4)	4
DIC	EXOTHER	EXOTHER	НА	ENOTHER	EXOTHER	NA.	но ехо	но ехо	NA	Γ	NO EXO	HO EXO	HA
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TOA	83 30	61.30	73 10	1 13E-01	4 91E-01	7 02E-01	0 21E-01	0.10E-01	8 14E-01	Γ	1 70E-01	7.ME-01	4 04E-01
741.3	15.8 971	\$ \$ 10 h	3 "t-1"	31, 35					40. A. O.	L	阿斯斯斯	C	\$ \$1.0 M
	(#)	(5)	(8)	(4)	(6)	(5)	(#)	(16)	(5)	Ľ	(#)	(#)	(#)
E WIR	74.40	74 60	74 40	1.19E-01	7.39E-01	7.10E-01	7.57E-01	7.786-01	7.65E-01		7.64E-01	7.712-01	7.71E-01
State of the	Japan Ja	्रम् श्रीका		: ¥*:		7 - 1 - 1		· 注:"	2 1 % 13 5		1:41 (1)	24.13.45	
H	1.48E-00	1.70E-00	1 50	1 44E-00	1.79E-00	1 43E-00	1.22E-00	1.102-00	1.30E-00		1.15E-00	1.02E-00	1.016-00
4 (7)	111		S F 18 18 18 18 18 18 18 18 18 18 18 18 18				1 5 2		14 10 1	Γ			
CH	<40	<11	<4.5	911	911	NA	<4.81	<4.90	NA]	<4.61	<4.76	NA

Analytic Brail #1 Brail #2 Arrigg Brail #2 Arrigg Brail #2 Arrigg Graph Gr		CHE	CI (CHCC2	1 -			CHCCI			.	C33CC3	
A 18E-00 6 99E-00 17E-01 17E-0	Analysa	Bendi #1	Binds (2	Average	Result #1	Reads #7	Aveilage	[Beach #1	Read 12	Average		Brank #1	Beach #2	Avenge
A 18 C A 18 C A 18 C C A 18 C C A 18 C C C C C C C C C	ICP-Water	(44/1)	(44/4)	(44/4)	(44/6)	(19/4)	(44/4)	_	(44/4)	(46/8)	(44/4)		(44/4)	(4/2)	(44/4)
As 3 00E-00 3 00E-00 3 0E-00 3	AI	4 41E-00	4 90 E-00	4 71E-00	1 016-01	I DIE-OI	1 03E-01		1.77E-01	1.HE-01	1 11E-01	-	1 02E-01		1.10E-01
Ba	53.	1 TIE-01	1.77E-01	1 77E-01	1.77E 01	1.77E-01	1.77E-01		1.77E-01	1.77E-01	1.77E-01		1.77E-01	1.77E-01	1.77E-01
	As .	1 00E-00	\$ 00E-00	3 00E-00	3 00E-00	1 00E-00	1 00E-00		2 99E-00	3.94E-00	2 ME-00		1 00E-00	3 89 E-00	1.00E-00
C4	Ba	3 GOE-01	3 OF E-01	3 016-01	\$ 32E-01	1 00E-01	4.14E-01		7.11E-01	3.788-01	3.47E-01		3 81 E-01	10-31E E	3:14E-01
C4 3 12E-01 3 09E-01 1 01E-01 1 01E-01 1 01E-01 3 01E-01 3 01E-01 3 10E-01	Be	9 19E-01	1008-01	1 00E 01	P PVE OI	9 99 E-03	99E-02	I_	1 10E-03	9 91E-03	9.99E-03		1 10E-01	0.94E-02	1 99E-02
C1 20TE-01 21TE-02 20E-02 3 10E-02 3 10	C4	4 00E-01	4 00 E -01	1 00E-01	4 COE 01	1 00E-01	4 (OE 0)		1 99E-01	1 00E-01	3 ME-01		4 00E-01	3 ME-01	1 ME-01
Co 7 99E 01 8 00E 01 8 00E 01 8 18E 01 7 99E 01 8 18E 01 7 99E 01 8 18E 01 7 99E 01 8 18E 01 8 18E 01 8 18E 01 6 18E 01 7 99E 01 4 00E 01 4 00E 01 4 00E 01 4 00E 01 7 99E 01 8 19E 01	C4	\$ 1:2E-01	3 03 E-01	1 04E-01	4 49E OI	1 43E-01	4 14E-01	I_{-}	# 13E-01	7.16E-01	# 44E-01		4.13E-01	9.35E-01	4.75E-01
C. 4 OPE 01 4 OPE 01 4 OPE 01 4 OPE 01 4 OPE 01 4 OPE 01 4 OPE 01 4 OPE 01 3 PPE 01	Cı	3 01E-03	2 11E-02	1 09E-01	1 30E-01	3 39E-03	3 29 E -01		1 24E-01	1 17E-01	3 24E-02		3 09E-02	\$ 13E-07	1 11E-02
Fe 101E-07 3 NG-01 7 NGE-01 1 31E-01 1 31E-01 1 30E-07 1 40E-01 1 41E-01 1 41E-01 1 31E-01 1 30E-07 1 40E-01 1 41E-01 1 41E-01 1 31E-01 1 30E-07 1 40E-01 1 41E-01 1 31E-01 3 31E-01 1 31E-01 1 31E-01 1 31E-01 1 31E-01 1 31E-01 1 31E-01 3 31E-01 1 31E-01 3 31E-01 1 31E-01 3	Co	7 99E-01	1 00 E 01	10-300	1 11E-01	1 99E-01	8 215 0		8 43E-01	7 98E-01	0 21E-01	_	1 99E-01	8 10E-01	1 11E-01
Ph	Cu	4 00E-01	1 00/E-01	10-300 t	4 COE 01	1 00E-01	4 00E-01		3 99E-01	3.99E-01) 99E-01		4 00E-01	3 ME-01	3 99E-01
1	F.	1036-03	1 70%-01	7 ME-01	1 51E-01	1 305-03	1 40E-01		1 47E-01	1 14E-03	1 115-01	_	1 38E-03	1.40E+01	1.49E-02
1	ln	6 IBE-00	1 11E-00	7 65E-00	191E-00	6 81E-00	7 018-00		4 34E-00	6.17E-00	6 34E-00		4 19E-00	4 19E-00	6. HPE-00
1	Me	9 33E-00	1 47E-00	1 ME-00	4 23E-00	3 44E-00	3 95E-00	<u> </u> _	4 04E-00	J 60E-00	1 HE-00		171E-00	3.95E-00	3 43E-00
E 7 21E-01 7 0E-02 7 19E-01 7 0E-02 7 19E-01 7 0E-02 7	bla	1.052-01	101E-01	1 47E-01	1 XVE-01	3 31E-01	7 11E-01	L	1048-01	10-E-01	10-346.5		1.42E-01	J. ME-01	J. 10E -01
\$\frac{1}{3}\$\$\$\$\frac{1}{3}\$\$\$\$\frac{1}{3}\$	M	1.70E+00	1.10E-00	1.702-00	1.70E-00	1. NE-00	1 702 00		1.7UE-00	1.70E-00	1.70E-00		1.70E-00	1.70E-00	1.70E+00
A4	(K	7.21E-01	1 40E -01	1 HE-01	1 0)E-01	7.61E-02	7 AJE-07		7.10E-01	7.04E-01	7.11E-01		6 30E · 07	6 47E-02	4.40E-02
No	Sa	7.59.E-00	7 60 E-00	7 40E-00	8 44E-400	7.19E-00	9 91E-00		7.58E-00	7.14E-00	7,31E-00		7.19E-00	7.38E-00	7.19E-00
V 4.99E-01 3 00E-01 3	14	4 99 2-01	7.70E-01	6 39E-01	1 07E-CIO	1 14E-00	1 14E-00		4 08E-01	4 ME-01	1.51E-01	Ľ	4.99E-01	4.99E-01	4.07E-01
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Bi	V	4.99E-01	J 00E-01	J 00E-01	1 30E-01	1.ME-01	7.07E-01		4 99E-01	4.09E-01	4.99E-01		6.76E-01	9.21E-01	1.90E-01
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Co 1.01E-01 101E-01 10	Bi .	1:47E-07	1 34E 01	1 15E-02	1 01E-03	1.76E-01	1 916-02		1.50E-01	1 01E-01	3 31E-02		1 67E-01	1.71E-01	2. XI E-02
La 7 01E-00 3 00E-00 4 02E-00 9 31E-00 7 69E-00 8 37E-00 1 33E-01 1 38E-01		3.118-00) 10E 00	3 31E-00	1 11E-02	3 77E-00	1 19E-00		3 51 E-00	1.54E-00	\$.\$4E-00		1 04E-00	4.44E-00	4.21E-00
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So 1.40E-00	Sr	1.46E-00	9 34E-01	1 31 E-00	1 79 E-00	1.07E-00	3 11E-00		3 14E-00	1.97E-00	1.14E-00		3 29E-00	1 JPE-00	1.HE-00
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Asslyte	Beault #1	Bond of	Average	. Rende #1	Binds 17	Average		Neral 1	Remb 17	Avenue		Beach #1	Beach #2	Average
ICP Milos	(vg/g)	(44/4)	(+g/g)	(44/4)	(48/6)	(+g/g)	l	(4g/g)	(49/8)	(ug/g)	L	(ug/g)	(4/4)	(44/4)
AJ.	4 54E-03	4 33E-03	44E-01	1 04E-0	1 10E-01	6 91E-01	╽ᆜ	4 NE-01	4.63E-02	4. ME-01		4 39 E-07	4.59E-03	4.59E-02
SA	8 ME-01	6 87E-01	4 ME-01	6 43E-0	4 4)E-01	8 83E-01		B AJE-01	1.29E-03	1.09E-02		4 NE-01	I ASE-OL	6.84E-01
At	1 10E-01	1.10E-01	1 50E-01	1 10E-0	1 30E-01	1 70E -01	\mathbf{l}_{-}	1 30E-01	1.50E-01	1.50E-01		1.50E-01	1 30E-01	1.50E-01
B4	4 04E-01	5.73E-01	\$ 88E-01	4 13E-0	J ME-01	4 04E -01		6 48E-01	6 40E-01	6 34E-01		7.305-01	7.41E-01	1.17E-01
84	\$ 00E-01	3 01E-01	\$ 01E-01	4 99E-0	4 99E-01	4 998 01	I	3 00E-01	4 99E-01	4.99E-01		4.99E-01	1 00E-01	4.99E-01
C4	9 14E-00	7 34E-00	4 23E-00	1 41E-0	7 14E-00	1 01E -01	[4 04E-00	4.74E-00	4.47E-00		7.44E-00	4 ME-00	7.17E-00
C.	1 91E-01	3 36E-03	2 76E-Q3	3 4)E-0	2 49E-03	2 44E-01		L #3E-01	3 30E-03	3.12E-05	Ľ	1.97E-03	1 10E-03	2.05E-05
Cı	1 93E-03	1 ME-09	1 89E-01	1.738-0	1 47E-01	1 70£ -01		1 74E-01	1.81E-01	1.79E+03	Ľ	1 425-03	1.82E-01	1.62E-01
C4	1 01E-01	9 70E-00	101E-01	1 01E-0	1.11E-01	1 04E -01	_	1 30E-01	1.37E-01	1.338+01		1.48E+01	00-200	1.16E-01
<u>~</u>	3 19E-01	3 61E-01	1 416-01	\$ 42E-0	1 41E-01	3 42E-01	i_	2 24E-01	3 14E-01	3.31E-01		2 41E-01	3 30E-01	2 44E-01
F.	101E-04	3 03 E-04	3 01E-04	1 97E-0	1 15E-01	1 101 -04		1.57E-04	1.62E+04	1.59E+04	Γ	1 41E-04	1 618-04	1.41E+04
P	4 55E-02	4 17 E-01	4 40E-03	4 ME-0	4 81E-01	4 44E -07		1.77E-02	2 61E-02	1 47E-01		2 67E-02	1.72E-01	2.49E-02
Ma	4 37E-01	4 14E-01	4.31E-01	4 54E-0	4 318-01	4 4)E -01		2 14E-01	3 74E-03	2.40E-07		2 49E-01	3.74E-02	2.72E-01
Ma	4 47E-01	10-341 6	4 14E-03	4 D2E-0	1 HE-03	5 ME-03		6.13E-0)	4 29E-01	6.22E-03		4 59E-01	4 19E-01	6.39E-03
М	J 41E-01	\$ 14 E-01	3.77E-03	3 40E-0	T.41E-01	\$ 43E-03		P 09E-01	9.49E-05	9.29E-01	Γ	1 43E-04	9.81E-03	1.21E-04
K	1 40E-01	5 41E-01	1 412-01	1 196-0	3 39E-01	10-31L		\$ 40E-01	3.39E-01	3.30E-01	Γ	\$.30E-01	1 40E-01	3.39E-01
54	3 80E-01	3 (1) E-01	3 80E-01	3 79E-0	3 70 E-01	3 79E-01		3 00E-01	J.79E-01	3.00E-01	Γ	3.79E-01	10-308-61	1.00E-01
A	2 14E-07	3 10E-01	3.14E-0)	1.36E-0	3 17E-01	3 31E-01		4.00E-01	3 49E-01	3.952-01		3.74E-01	1 40E-01	3.71E-01
He .	4 01E-04	3 14E-04	3 84E-01	S ME-O	1 41E-04	3 POE-04		3 ME-04	1.41E-04	3.39E-04	Γ	3 32E-04	1) SIE-04	3.32E-04
V	8.32E-01	1.11E-01	1.31E-0	1.74E-0	1.11E-01	1 61E-01		1 69E-01	1.342-01	1.512-01	[1.51E-01	1.41E-01	1.472-01
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B)	3 14E-04	2.05/E-04	3 09E-04	2 018-0	2 02E-04	3 01E-04	7	3 4JE-04	3 44E-04	3 63E-04	Γ	2 61E-04	3 73E-04	2.47E-04
	1 00E-00	3 01E-00	3 00E-00	3 09E-0	2 998-00	1 196-00		4.32E-00	J.36E-00	4.ME-00		1.10E-00	4 50E-00	4.842-00
C	\$ 01E-01	1.04E-01	5 04E-01	S DIE-0	104E-01	\$ 04E-01		3 05E-01	3 04E-01	3 04E-01		5 04E-01	11.01E-01	3.04E-01
l e	3.73E-03	3 43/E-01	3 49E-01	3 158-0	1 14E-01	3 418-03	1	4 43E-03	4.14E-01	4.51E-03		4.74E-01	4.04E-03	4.01E-01
P	\$ 19E-04	1.125-04	8.16E-04	1.11E-0	1.11E-04	1 11E:04		#. 11E-01	\$ 99E-03	1.07E-03		9 91E-01	# DIE-01	8.91E-01
H	\$ 04E-01	3 48E-01	\$ ME-03	3 31E-0	1.71E-03	\$ \$4E:01	-	\$.39E-01	1.336-01	3.44E-03		1.41E-01	\$ 41E-01	\$ 41E-01
Se	1.91E-01	3 04E-01	1 03E-02	3 01E-0	1.70E-01	1 40E-01	-	# 64E-02	1.97E-01	1.91E-02	-	3.34E-01	1 DOE-07	3.17E-01
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Ti	7.23E-01	7.32E-01	1.29E-01	1.1HE-0	7.13E-01	7 24E-01		2: 278-01	1 25E-01	3 31E-01	1	2 31 E-01	1.81E-01	3 41E-01
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Segment 1

	Assilyte	\$14.81	LUEL BTR	Result #1	Republ #2	Avenge	APD
<u> </u>		A Rec	(4e/a)	(46/4)	[(ug/g) !	(44/3)	
	14.3)0						
	DSC	EXOTHERM	HA	EXIOTHERM	EXOTHERM	HA	NA
	10.319						
	IQA	99.3	НĂ	77.6	NA	HA	, NA
	10010	_				 	
S WATER		100 7	НА	61.1	79.6	80,4	8.07

Segment 2

	Assiyte	516 81	PREP BLK	Result #1	Besuk #2	Average	HPD.
		S Rec	0-6/6)	(ug/g)	2/ve/a)	(ug/g)	. 5
	10.525			1	No.		
	D\$C	EXOTHERM	NA	EXOTHERNO	EKOTHERM	NA	NA
	14.323		<u> </u>				
	TOA	99 3	NA ·	90.5	80.4	824	0.12
	10.323						
& WATER		90.1	NA	85 6	85.8	45.7	0.23

Lamest

	Andjis	S16 #1	PREP DLK	Result #1	Roonk #2	Average	BPD
		A Rec	(ug/g)	(48/8)	(ug/g)	(ug/g)	3
	18 324						
	psc	EXOTHERM	NA .	Смязито ка	HO EXO	HA	NA
	10.524						
	IOA	99 5	HA	01.5	НА	HA	MA
	10 324						
S WATER	· •	98.1	HA	11.7	93 0	31.9	8 33

	Analyte	Sid #1	PREP BLK	Result #1	Result #2	Avenge	Arb
		S Rec	(wg/g)	(ug/g)	(ug/g)	(46/8)	16
	10.323						
	usc	EXOTHERM	NA	NO EXO	HA	NA	HA
	10 325					,	
	IGA	99 3	HA	00.5	HA	NA	NA
	1#325				···		
N WATER		94.1	NA	80.4	79.3	79.0	1.1

Segment 5

	Analyte	\$1d #1	PREP BLK	Acreh #1	Result #2	Arenge	APD
		ik Rec	(mg/g)	(vg/g)	(ug/g)	(up/a)	5
	10.336						
	DSC:	EXOTHERM	NA	NO EXO	MA	NA.	HA
				1			
	14.336						
	104	99 5	HA	14 ()	NA	HA	NA
	10.324						
WATER	·	98.1	HA	79. H	77.0	78.2	2.0

Scouted 4

	Anine	514 #1	PREP DLK	Result #1	Actub 42	Average	APD.
		E Hec	(ug/g)	(ug/g)	(ug/g)	(ug/g)	
	14 329						
	DSC	EXOTHERM	NA	NO EXO	NO EXO	NA	NA
	14 329		 				
	IGA	91.6	NA	H.7	84.0	84.4	0 03
	10)29	_					
MATER		101 0	.	78.3	78.4	78.5	0 18

Kgmin 7

	Analyse	516 01	PAEP BILK	Republ 474	Result #2	Average	ND
		A Acc	(lug/g)	(ig/g)	(vg/g)	(m/g)	
	14.330						
	DSC	EXOTHERIA	NA	NO ENO	HA	NA .	HA
	14.330]
	104	94 4	NA	45.6	NA	NA	NA
							[
	12.330						
A WATER	-	101 0	HA	74.7	48.4	71.7	0.3

Segment B

	Asulyte	2001	PREP BLK	Result 81	licosh #2	Average	ND
		1 Acc	(mg/g)	(vg/s)	(vg/a)	(vg/a)	
	10,514						
	DHC	EKOTHEM	NA	NO EXO	HA	NA	NA
	14.331						
	TOA	P4 4	NA		NA	NA	NA
	18.331						
WATER		101 0	NA	J 734	as sample	NA	HA

Segment

	Analyte	514 #1	LYES DER	Result 81	Result #2	Avenue	ND
		B Rec	(44/8)	(ug /g)	(ug/g)	(46.4)	
	16:332		ļ	_			
	DSC	EXOTHEM	НА	HO EXO	NO EXO	NA:	NA
	ian						
	IOA	<u> </u>	HA	85.2	HA	NA .	NA
	14.333						
WATER	<u> </u>	101 0	NA .	71 0	74.9	74 0	7

		•	PREP BIK	A	Dient C	Avende	2	Spt 143	77.05	DET LEA	Perle
_	(1/h) Pr	# Dec		3	i	; ;	J	3		() ()	70
,	0001	1.01	1 016 01	1 945-03	_	4 14E-03	1.11	2	=	3 40E -CO	17.5
נים	000	Ī	10.311	1486-01	1 76E-01	1 125-01	=======================================	7 = 1	11.7	1.776-61	=
1	000	8	1 006 -00	4 F-E-60	1 776-00	4 146-00	5.2	117.76	2) CDE -CD	5
Τ,	000	3	10-311-6	10-311	10-31+1	(14E-0)	I =	11111	3	1008-61	2
111	000	-	1 00 5 01	1 4 6 01	1 17E-01	1.596-01	=	81.11	3	1 00 E-01	= -
<u> </u>	0000	1.13	10-300+	> XOE OI	10-3111	1.016-41	5	817 SI	1	1005-01	47.10
5	007	131.0	10-351 +	1 106-03	1116-01	1 146-03	3 2	111 111	143.1	4 40E-00	3 8 1
١-	900	91.1	9 00 E-01	10-309 9	1.11E-01	1 146-03	2	110.76	13.7	10 300	3
Ь	000	1 69	1 COE -01	4 126.00	4.748-00	4 746 -00	-	31.00	3	1008-01	3
٦	88	2	4 00E -01	10-371	6 P9E-01	1 PZE-OI	1.4	11 11	3	4 00 5-01	8 11
i F	900	701	4 496-00	2000	P 30	1 TE &	2	200	1111	1 00E-00	1144
h	000	=	8 30	1.17E-41	9 746 02	148-07	=	12.5	=	20.00	13 61
1 -	8	916	8 391	7 116-01	1 415-02	60-300 t	2	2011	5	1000	343.9
	8	5	1001	7	1 516-04	2017	3	1	24.1	108E-01	
¥-	000	=	8	106-41	146-01	1 156-02	=======================================	10 70	2	892	100
1	000	8	9	OF	1126-01	1 0715-01	9.78	210.76	1	1.126-01	3
, -	100	=	1 405-00	1 236-01	10 361	1 215-01	111	110.76	=	7 60E-00	3
j.	8	=	0 300	4 216-41	933	1116-07	==	310 76	=	10-300	2
i	OCC	111.5	17.6-01	2	2 SE-2	306-04	3	27.50	=	20-301	3103 11
1	000	2	10-397 9-	1 (TE 4)	1676-01	136-01	L II	11.5	2	100.00	=======================================
	000	Z	8.35	1 116-01	1 115-01	1716-02	=	15	=	10000	3
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Ь	3000		10-306	4 31E-03	4 SIE-03	1 XE-03	r,	11.11	=	1 10E-00	
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, m.	¥	2 15E-4	10-396	17.6-01	1 34E-01	8	ī	3	2 2		
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WHC-SD-WM-DP-024 ADDENDUM_2, REV 0

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		10 117	10-31E E	YH	EL 0	10-30-1	10-346 1	10-301	4-351 E	YH	112-07	
		336 05	10-3411	VH	u l	10-369 E	10-3191	10-31L E	1-391 1>	YN	\$\$1-03	
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EK 0211	10-300 f	5 60	91 111	41 17	10-319	3 616 -03	1 ME-03	10-300 6	L 16	0001	手说	
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15'04	1 13E-01	1 451	91.911	01 0	10-3111	10-3111	10-351.1	10-30t 1	1 401	0001	D_{x}	
99.961	00-30L I	0 (6	76.616	11.1	10.311.01	3 30E-03	to-396 E	1 30E-00	£ 16	0001	工前	
LIFIEL	10-300 t	1300	96 911	äi	10-3()	3 44E-04	1 41E-04	10-300 E	0 (0	0001	五司	
3654.69	10-300 (101	96 911	<u>;; ; </u>	10-354 F	1.00 3 00 L	TO-310 1	00-395 8	0 14	0001	支前	 -
96 101	00-30E 9	1 16	91.111	i) L	to-Bot 4	10-368 B	10-310 t				[*눈길]	
9 16091	00-300 I	1 135-	35, 511	iii	I TIE-OI	10.3(1)		00-306 9	£ 90	0001	T w	
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		1 99	66.54£	<u> </u>	00-361.0	00-315 P	00-3(9)	10-300 1	1) 60	3000		
10:140	10-300 6	1111	96'811	#1	(17E-07	4 90E-93	CO-310 S	10-300 6	6.04	0001	<u>E</u> 5	
10 1401	4 40E-00	8 411	23.715	100	10-301 1	E0-300 P	10-366 +	10-311 1	0 192	2000	ट च	
19 18	10-300)	D 10	EE TEE	14 0	10-35) 1	10-3110-1	10-3() 1	10 300 t	1.40	000£	드	
1911	10-300 I	6 11	62.545	16.6	10-3111	10-3E1 6	10-311 1	10-3001	8 (4)	000E	<u>m</u> =	
sa cot	10-300 (£ 11	et res	E5 E	10-360 9	60-3f0 à	10-371 4	10-311 [0 10	COOK	モゴ	
Lit !	00-300 f	8 66	SF.F11	80 81	00-300 (3 44 E-00	00-3111	3 00 [-00	\$ 04	0001	They	
101	10-311	£ 10	11.111	196	10-300 E	3 OFE-OF	10-310.1	10-317 I	9 M	0001	TI 42	. 1
18 141	1 40E-00	LUI	91 911	111	1148-03	to-att	111.03	1 01E-01	1011	0001	L īv	P12Y-4:31
NDF	(1/34)		(1/3/1)	B	(1/10)	(4,40)	(8,80)	(1,14)	27 M M	(VIn) pp		46E #1
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This is to a sequence winester Coin 18

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WHC-EP-0806 WHC-SD-WM-DP-024 ADDENDUM 2, REV O

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l	Assirte	I MC3 1		PAEP BI K	Beach #1	Bench 52	Avenge	ND	Spl +44	Spike	DETIM	Ratio
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ICT-Acid	<u> 4 4 7</u>	1000	- 814 4	7 01E-01	1 44 E -02	1.35E-02	1.47E-01	7.99	114.76	137.7	1 40E-00	67.40
<u> </u>	LA MI	1000	94.9	1 77E-01	1 40E-08	3.13E-01	1.41E-01	nn	114.74	91.7	1.77E-01	1.34
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	O I	1000	<u></u>	4 ODE-01	7.24E-00	7.05E-00	7.15E-00	2.93	110.74	0.1	4 00E-01	17.00
	F. J	1000	99 7	\$ 73E-00	1.01E-04	1.74E-04	1.79E-04	1 83	110.76	-141.1	1.00E-00	17947.0
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ļ	MI	1000	<u></u>	3 (OE-00	1 04E-01		1.04E-01	8 33	118.74	107.3	1 00E-01	643.60
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Analyte	Rende #1	Result #1	avenge	RPD S		Repub #1	Result #2		IUP S		RPD B
]	(1/1)	(44.1)	(4/1)			(44/4)	(44/4)	(4/4)		Γ	evenges
<u> </u>	177 450	41 31	633 329	3 117	I_	607 807	314 070	595.939	3.983	7	6.394
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<u> </u>	1 071	3 623	1 111	10 410		3 964	4 425	4 107	10.874		9.334
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4	1.409	1.122	1.366	22 670		1.316	0.780	1.052	50.076	1	18.391
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C ₀	44 129	49.487	47 405	7 011		49 237	41.725	44.981	9.603	Н	1.730
La	3133 255	5355 620	5254 447	1 151		3134 365	4911.341	5017.054	4.245	·	4 606
P	\$764.313	8900 917	18772 900	2 435	-	9550 631	9039 846	9299.331	5.501	H	4.491
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Sa	13.716	13 924	13 321	1 317		15 091	13 166	15 329	3.045	-	10 316
Ti	1.004	2.474	1 240	21 075		1 259	3 082	3.171	44 007	-	3.107
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WHC-EP-0806
WHC-SD-WM-DP-024 ADDENDUM 2, REV 0

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WHC-EP-0806 WHC-SD-WM-DP-024 ADDENDUM 2, REV 0

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					Water ICP/Acid ICI	P	Walne ICP/Pleases ICP	<u> </u>	Punios ICP/Acid IC	7
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·	Analyte	M4 81	PREP BLE	Re away #1	. Acout 12	Avenge	MD	Spt Roc	DET LOA	Ratio	94	Ma Bal
· ·		1 Acc	(vg/a)	(ug/g)	(vg/g)	(ug/g)	#	5	(ug/g)	A/DL	lone	(4'4)
DATE 4-14-92	12.470							l				
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·	Anilyta	24 11	Prop Bill	Decule #1	Beach 12	Avenge	APD	Spt Res	DET LD4	Ratio	B ol	
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	Eu-135	NA	<4 DE-4	3 10E-01	1 97E-01	3.07E-01	4 20	HA	3.93E-04	3.15	41.0	
	Am-241	HA	<1.37E-1	3:78E-07	1.11E-02	3.67E-02	4.13	HA	1.14E-03	33 93	a a	
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	Anilyta	B-6 (1)	Prop Bil	Bende #1	Rende 12	Average	MPD	Spå Res	DET LO4	Ratio	Bal.
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Date:\$/19/93	18.470]				}				
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· · · · · · · · · · · · · · · · · · ·	Analyte	 	PAEP BLE	Rent 01	Reach 172	Atende	APD	Spt. Rec	DETLO	Bulle	ICLY Lies	Rendelia
	 	3 Acc	(m/1)	(mg/L)	(mg/1)	(mg/L)	<u> </u>		(mg/L)	NOL	(a ₄ /L)	
Dale:	10.611	{	ļ						ļ	 _		<u> </u>
ICIA-ICP	<u>^4</u>	107 00	1 00801	J 1.80E-01	3 47E-01	3.99E-03		23 80	J.00E-03	3.07	\$.00	0.01
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	<u>a</u>	95 10	4 (10€-0)	J 100E-01	3. XVE-01	3.75E-07	 	101.50	4.00E-0)	6.40	1.00	0.01
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Date:	18.608	 		[<u> </u>	 	 	 	
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Ω Ω Ω	3 •	11 (4	10-3111	01.08	10.58	tt o	10-310-1	10-310 1	10-310 [10-311 1	04 (0)	0006		
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Ŀ	¥ 5	will	10-3004	06 961	10 18	112.1	(0-2+1·g	10-391 E	10-211 F_E	10-300 \$	00 14	0001		
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	Analyta	34 81	PREP BLK	Result 61	Remb #3	Average	MD	Syd Reci	DELTM	Batle	. 1044	Mai Bal
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or six applica	े जिस्	38 M A - 1		16 3 (8° 35		21127	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	9 X.S.			1911	1 2 3
	18.437			(5)	(5)	(8)	1					
	IGA	94 50		8 30E+01	7 84E-01	6 04E+01	5.45		NA	NA		
STREET SECTION		25 11	रामा जनसङ्ख्या	7-10 (C)	f far by			13.3 gr. 200		137.34		
	14.423]		(#)	(5)	(8)						
A WATER		\$9.70	HA	J 7.41E-01	7.71E-01	7.71E-D1	1 62	NA	NA	NA .		0.77
						1 4 4	1 N 4 1 1 1		, 541 X 1 X 1	11:11	1 2000	
	14 431											
НУАА	٨٠	na										
	Se	PHL	l							<u> </u>		
CYAA	118	101 40	<1.13E-1	J 1.15E-00	1 03E-00	1 01E-140	11.09	62 710	1	6 64		
	5: 5 :13:13			4.1341, 115	\$110 mm.		The Land		建筑工程	Proprior		
	J#.446											
CH-Dir	애	99 30	<3 0060	UJ 4111	<4.74	NA	NA	100 70	\$.00E-00	HA	9 900	
	C+HIF+CH4		1		}	NA :			1		6 000	•

WHC-SD-WM-DP-024 ADDENDUM_Z, REV 0

DATE	Audyte	LAICS A4	<u> </u>	PREP BLE	Acade 81	Brock 18	Average	MD	Syd add (Syike	DET LD4	Ratio
14.441		1444	1 km	(ve/a)	(vg/a)	(14/4)	(vg/g)		(y/L)	II Rec	(40/4)	MAL
CP-Water		1000	78 70	2.40E+00	UJ 01E-01	1.19E-01	1.10E+01	15.54	1230	107.30	1.40E-00	4 31
		1000	100 10	1.T/E-01	4.J. 1.71E-01	1.77E-01	1.77E-01	8.09	1250	107.50	1.77E-01	100
		3000	ox 16	3 00E-00	LLT 1 OVE-00	2 99E-00	\$ 00E+00	0.10	2500	47.50	3 00E-00	1,00
···	Ba	10000	99 00	1.00E-01	UT 5 01 E-01	\$.32Æ-01	J.14E-01	5.97	1500	19.30	1 00E-01	1,00
		10000	101.70	1.018-01	UJ 8.99E-01	9.91E-03	1 19E-01	0.10	2300	100.20	1 00E-01	1 00
		10000	101.00	4.00E-04	(LJT 4 00E-01	1 99E-01	3.092-01	0.10	1500	04 001	4 (DE -0)	1 44
		10000	91 10	1.ME-01	<u> </u>	0.HE-01	4.71E-01	77.04	1500	, 91.00	4 40E-00	15 3
	Cı	, suo	101 70	10-300	J 1098-01	3 136-03	2.11E-01	1.21	1330	103 00	9.00E-01	114 (
 		10000	U 66	1 00 8 -01		8 50E-01	\$ 15E-O1	6.17	2.500	99 20	8 COE-DI	1 01
·	<u> </u>	, two	24 40	4 00E 01	UJ 1 OUE-OI	3 99E-01	3 198-01	0.10	1330	, 97.50	4 CDE-01	Įα
	Fe	xw	99 50	1 00E -00	J 1.318-01	1.60E -01	1.198-02	1.36	1330	98 50	I COE-CO	159 31
	r	xuo	11 60	4.30E-00	UT 4.196-00	4 49E+00	- 4 19E-00	0.10	133	91.70	6.30E-00	100
	<u>}\</u>	1000	101 10	1.71E-00	UT 1 71E-00	3 B7E-00	3.47E-00	4.03	1250	101.40) 00E-01	12.70
	bla	: XULO	94 00	1 00E-01	J 1 12E-01	10-314.6	J.34E-01	3.26	1230	. 95.10	3 00E-01	164.81
	NI	1000	100 80	1.306-00	LJ 1 10E-00	1.306-00	1.70E-00	0.10	1230	100.40	1.70E-00	1 0
		1000	01 10	1.126-01	J 4 30E-03	₿ 47E-02	6 442-02	6.50	1214	99.10	1.17E-01	37 pc
	\$6	3000	11 70	7.40E-00	LT 7.198-00	7.11E-00	7.19E-00	0.10	1330	91.40	1 60E-08	100
· · · · · · · · · · · · · · · · · · ·	^^_) XXXX	103 60	1 to E 01	WT 4 99E-01	4.998-01	4 PF-01	0.10	1230	103 00	1 00E-0)	100
	Ne Ne	10000	M 90	4.50E-01	T 131E-04	JI.19E-KH	3.20E-04	0.66	2500	90.40	1. (DE -QQ	10334.1
	Y	10000	100 50	1 00E-01	UJ 6 16E-01	1 31E-01	7.998-01	30.49	2500	08.10	\$ CUE-01	1.40
	<u> </u>	1000	101.10	-3.31E-CO	10-300 TU	1 D-300 S	3 AOE-01	0.10	2500	(16.20)	3 OUE-01	100
	al .	1000	101.70	7.50E-00	J 1.41E-01	\$.13E-02	1.70E-01	1.96	1250	14.40	7.10E-00	34 07
		3000	11 10	1 04E-01	J 1045-00	4.44E-00	4.35E-00	8.19	1250	67.60	6 WE-01	1.00
	C ₀	5000	101 40	1.01E-01	W 1018-01	1.012-01	1.01E-01	0.10	1230	107.10	1.01E-01	100
	<u>ta</u>	1=1		1.40E-00	J 1.31E-01	1.578-01	1.31E-01	0.10	1230	103 00	1.40E-00	11 2
		10000	97.20	\$.00E-00	J 174E-01	3.44E-03	1.70E-01	1.34	2500	127 (0	1 AUE-OU	941 21
·	Si	10000	63.30	1 49E-01	J 1118-01	6.33E-03	\$ 30E+03	0.58	2500	107.40	1.30E-00	476 P
	31	10000	99 00	1 00E-01	J 129E-00	2 J9E-00	3.34E-00	4.24	2500	90 30	3 WE-01	7.71
	3	J000	100.30	1.70E+00	J 1.14E-01	1.14E+01	1.14E-01	0.41	1250	1011 00	3.70E-00	422 \$1
	Sa	3000	94 NO	1.40£+00	11 1 10E-00	1 60E-00	1.40E-00	0.10	1350	91.80	1 608-00	1 00
1	TI .	\$000	107 10	4 00E-01	LAT 4 00E-01	3 29E-01	1.098-01	0.10	1210	· •	4 00E 01	1 (4)
	Li .	10000	101 40	1 002-01	LLT I ME OI	7.942-01	7.098-01	0.10	1250		-	

WHC-SD-WM-DP-024 ADDENDUM 2, REV 0

19114-5	103	01 101	(100	10-366 L_CY)	(0-301	to-30f ¢	OZ 18	OC 101	pot	OB I	\$10 0	too O
14-15-4-31-13	199 81		·		 							
	1.1			Y 3	Capacital Line	1. 1. A.	110 (4.5)			1,3 A g	1,21,5	
	bg sinisy		1 47E-04				<u> </u>				ORS @	TO 0
MIN'S	201	OC M	0(1>	10-300 (L)7	10-300 f	1 001-00	00 0	00 16	004	00 9		
/IE + 13-13	10.43		·	<u> </u>		·						
at 130 5 1 12	Links (NIE)		20.5	116,252		J = 2 7 3.	35.5	, (- 2).	A. L. Conne	ing.	34.5	
131-W-E1	tilit	or oot	006+>	(1) <1300	00(1>	VN	W	01.19	00(1)	VH		
£4-91-9 31V	18-163		-]								
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·	Hd	0 001	VH	J + 01E-00	\$ 13E-00	00-31T.Q	24 0	YH	AN	VN		
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24/82/6191	10.41	1							· · · · · · · · · · · · · · · · · · ·			
A Light of Little		10 10	9 14		te i i r	Y-1-15-7&1	J_1 , h_{2g} , $\{x_i\}$	49-59-14		\$ 100		
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	P ACID ICP/IPO					00-316.6			***	34 63		10 0
<u> </u>	101		<1 00E+1	(0-311 (T	# 41E-01	(0-3() (21 6	or til	£0-300 f	00 001		10 0
	101		£+300°1>	10-30f I _D	10-316-06	10-316.1	tt't	00 (4	to-300 1	26.6		00 6
	HOS		on: 3</td <td>1 04E-01</td> <td>20-345 (</td> <td>10-310 P</td> <td>n i</td> <td>00 (01</td> <td>to-300 l</td> <td>06 00+</td> <td>f# 0</td> <td>60</td>	1 04E-01	20-345 (10-310 P	n i	00 (01	to-300 l	06 00+	f# 0	60
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	b		<1 00E+1	(0-365 1 -	10-319 1	1,615.01	10)	OF 18	10-3001	00 (91	60 0	00 6
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19)	in/inity	# Wee	X 18 43A4	10 Haman	P# street still	192/24 V	034		PG 1 L3 0	oh all	100	ואיו אין
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DATE:	Assigna	13403 84		PREP BLE	Republication	Book #2	Average	DIA	554 044	Spile	DET LIM	B) tie	
474		1-41 (14/1)	# Rec	(vg/g)	(ug/g)	(vg/g)	(4/2)	<u> </u>	(4/1)	S Ros	(ug/g)	1/DL	hes
P Fuelca		XXX	P1 10	1 205-01	J 4 39E-03	4.3PE-02	4.508-01	0 07	1330	\$1.00	1.20E-01	38 25	
	56	3000	PH 90	8 87E-01	10-310 F TA	# 85E-01	8 ME-01	a .30	1350	83 60	8 83E-01	1.00	
		\$400	93 10	1.50E-01	10-30E-01	1.50E-01	1.306-01	0.20	2500	43 60	1.50E-01	1 00	
	Ba	1 arias	93 70	1 50E-00	7 X0E-01	7.41E-01	7.37E-01	3 01	2500	92.00	1.50E-60	49.14	
	Ba	10000	6) 10	1 048-00	UJ 4.99E-01	3 OUE-OL	4.09 E-01	0.20	1100	94 2D	1 04E-00	0 47	<u> </u>
	C4	10000	#1 00	3 (UE+00	(1 7.41E-00	4.84E+00	1.17E-00	8 63	2500	<i>9</i> 0 00	3 COE-00	3.59	
	Ca .	10000	PO 60	2 40E-01	J 122E-01	2.14E-01	7 01 6-01	12.7	2300	6 71.20	2 40E-03	8.34	
<u> </u>	C)	300.00	95 70	4 30E-00	丁 1 92E-03	1 #2E-01	1 878-0)	0.14	1250	93.30	4.50E-00	404 16	王
	<u></u>	10000	92 40	4 00E-00	11J 1 41E-01	\$ POE -00	1.34E-01	49.3	2300	90.10	4 60 E -60	2.96	7
		XXX	91 20	2 216-40	J 261E-01	1.30E-01	2 44E-01	12 6	1230	91.40	3 25E-00	10.94	Ö
	P.	KUO	P2 50	3 398-01	J 1 61E-04	1 41E-04	1.6HE-04	0 13	1330	124.50	3 30E-01	621.74	P F
	n	3000	95 40	1 10E-01	丁 2675-02	1.72E-02	3 60 E-03	8.76	1250	84 20	3.10E-01	1 49	
	210	Jaco	F3 40	7.735-00	丁 2 49 E-02	2 74E-03	2.7AE-02	8.00	1230	91.30	7.75E-00	35 06	P
	Ma	tuan .	9U 20	1 33E-01	J 4.39E-03	6.39E+03	4.30E-03	0.01	1210	\$2.50	1.55E-01	423.44	22
	M	\$CO0	1) 70	1 07 2-02	J 1.41E-04	0.01E-01	1.31 E-04	37.1	1330	84.10	1.07E-02	112 41	
	K	xus	91 40	\$ 60E-0)	10-30E-01	3.40E-01	1.11 E-01	0.20	1250	-1097	1 40E-01	1.00	営
	Se	KOO	93 60	\$ BUE-08	UJ 1.79E-01	3 80E-01	3.80E-01	0.20	1230	86 60	10-3UE (1.00	ADDENIOUS
	14	\$000	84 90	1 PIE-00	J 3 74E-01	1 00E-01	3.71E-01	1.47	1210	97 90	3 #1E-00	12 77	12
	Ha	10000	92 30	4 ME-01	J > > > > > > > > > > > > > > > > > > >	3.51E-04	3.57E-04	0.33	2300	79 00		35.47	
	٧	ious	92 40	2.50E-00	J 1.31E-01	1.42E-01	1.47E-01	5.M	2500	91.70		1.07	, 1/27
	Z ₄	10000	PH 40	1.378-01	J 1 19E-01	1.02E-02	1.102-03	15.4	2300	92 40	1	101	R
	e.i	soco	97.30	3 73E-01	J 261E-04	2.73E-04	2 67E-04	4.52	1330	33.30	·	711 40	•
		\$000	81 20	\$ 85E-00	UJ 1 10E-00	4.51E-00	4 84E-00	10.9	1230	78 80	· · · · · · · · · · · · · · · · · · ·	0.63	9
·	C ₄	1000	101 101	\$ 05E-01	LT SOLE-OF	1 01E-01	J 04E-01	9.20	1250	103.20		1.00	
	La	xuo	101 10	7 00E-00	☐ 4.78E-03	4 64E-0)	4.016-01	1.36	1230	PH.50		47.17	
· · · · · · · · · · · · · · · · · · ·	- _F	10000	93 80	\$ 75E-01	T PPIE-03	1.11E-01	1.016-01	001	2500	95 20	1		{ ·—
	si	Iaco	49.60	1 19E-01	J 11E-01	1.41E-01	3.41E-01	9 04	2300			113.30	
	S ₁	10000	93 90	4.37E-00	J 1 ME-01			I	1 	64 50	·	11 69	013
		1000		- '	J 116E-01	2 \$1E-01	3 17E-03	110	2500	91 50	·	77 (5	
	\ <u>`</u>	╌╏╼╼╼╾╂╼	P6 40	1.35E-01		1-16E+03	1.14E+03	0.52	1250	P6 70	-	83.79	-
	- Sa	X00	93.50	\$ COE-CO	ILT I ME-00	\$ 00E+00	7.99E-00	0 20	1250	92.10	 	100	
		1000	<u> </u>	3 (0.5-00	J 131E-01	3 31E-01	2 41E-01	1	1530	91.10	3 006 -00	13 03	l
1	<u> </u>	<u> </u>	-	4 COE-00	M7 3 80 E-00	4 005-00	4.00E-00	0.20	<u> - </u>	_	4 00E -00	1 -	l

		Water ICP/Acid ICP	Water ICE/Fusion ICP	1		Parios (CP/Acid (CP	
	· ·	Cs 0.10	M.		0 024	AI I	1.13
Charge Bel Tot Cas	3.55	Ma 0.01	Ci		0.116	Po ! G	0.93
For An	2.54	41 (.57	Pe i		0.010	Ma	0.97
CaUAA	1.39	0.14	Ma		0.005	34	94
	1	He 0.55	Maria Cara		9 D10	3) g:	1.74,
bisterial Belance: onlide mordel	0. H j	AI 0.01		100	0.374		1.m
		E 044	4		0.115		0.97
		P 9.30	₿¢ :		4.001		9.95

Deu:4/27/92	Analyte	341 (1	PILEP OL E	Reads #1	Acoult #2	Average	NO	Sed Bu	DET LD4	Rasia	844	Mai Bal
No. 471		\$ Rec	(ug/g)	(vg/g)	(44/8)	(49/4)			(ing/g)	D/DL	lea	(4/6)
Fluct	u	101 10	Q 40E-1	J 1.83E-013	10 E-01	1.932-03	12.63	44.10) 40E-03	3.77	0 014	0.007
	Ussucva				<u> </u>	6.47E-01	- 4		<u> </u>			
10. 在於學科關係		1 12 1				THE RESERVE	19.3	1000		\$ 100		
	Analysa	24 /1	Prop Bill	Brouk et	Brouk 177	Average	nio .	Spt Bac	PET LIM	Ratio	As).	
	<u> </u>	1100	(LCV4)	(uciva)	(uCVa)	(uCVe)			(NCAN)	N/DL	QEA	
DATE 4-24-02	14.474											
JAD-Fm.	TA	H.70	C 30E-1	J 3.97E-01	3 97E-01	3.97E-01	0.00	89.80	1:30E-04	749.04	1.400	
·	TA/Po-Am					3.01E-00						
1987年	2. 电影·多数		108 -913 SE		75.85		Yelde			16,1876		1 - 1 - 3
DATE-4-26-92	14.471									[- 		
	19	M 30	<4 132-1	8.95E-00	8 81E-00	8.89E+00	1.33	91.40	4.11E-02	144.33	1 300	
	TIVC+St					9.14E-01		· · · · · · · · · · · · · · · · · · ·				
1974年1974			14. 14. 2					3: 4:2	319	1.48 h.		
DATE-4-11-11	14:474											
OEA	Cr-137	N 00	4 64E-02	1.01E-01	1 01E-01	1.0)E-01	0.97	NA	3.70E-04	279.73	1 400	
	Eu-154	NA	<1 01E-1	C) HE-4	<1 01E-1	HA	на	NA	1.20E-01	NA	NA	
	Eu-158	NA	<4. NE-4	<1.49E-3	<1.40E-3	MA	HA	NA	1.01E-04	MA	NA	
	Am-213	NA	<1.39E-1	4 61E-02	4.24E-02	4.43E-02	1:34	MA	1.14E-01	314.16	7.840	
~ ~~~	C4-144	NA	NA	NA *	NA	NA	HA	HA	на	НА	HA	
	Co-40	94 00	a ne-	Q.118-4	4 39E	NA	NA	MA	4 Ø7E-01	NA	NA	-
1000年									MARKET			
DATE S-692	18.471											
	PU239/40	74.30	C) HE-1	J 147E-01	1 336-01	1.47E-01	7.46	84 80	3 30E-01	41.14	0 300	 -
_	PU-330	74.30	C HE-1	<1.04E-3	<1.13E-3	NA	NA	76 60	1 COE-01	HA	0 900	

	Analyta	5.4 (1	PREP BLE	Rend #1	Acoult #2	Average	APD	Sjå Res	DET LM	Ratio	(Re lotte
Di stc. 5/19/92	14 471	3 Acc	(48/0)	(vg/g)	(vg/g)	(vg 'g)	16	5	(ug/g)	₽/DL	1 Q Er
	Am-241	\$1.90	6 32E-03	J \$ 116-02	4.44E-02	4.78E-02	14.03	19.40	6 CUE-01	7.1%	1.70
				.				18.70			
	Cn-364	l	<u> </u>	nome detected		NA	MA			MA	
		A-241 (GEA)/A-241 (ALPHA)			9 8/12-01				· · · · · · · · · · · · · · · · · · ·		
1941 3446	10000000000000000000000000000000000000			THE CO.	ongest av 1			47.2.1			
DATE 4-14-92	10.471				·						
	HP-333	44 20	8 47E-01	WZ 4 318-1	Q 14E-8	NA	NA	69 20	3.408-67	NA	1 40
3 to 18. H	4 1 3 3 3 3		1444 A		98 J. 75	* **			1981 · /	1 35	
Ditt;	10.				· · · · · · · · · · · · · · · · · · ·						
	TC-90	104 40	1 00E-03	J 101E-01	1 01E-02	1.04E-03	1.92	47.00	4.402-0)	2.26	1.70
	7 7 7			To the second		了 最高数据	7.57	68 40		3 (3)	महापुर है
DATE 4-13-93	18.471					 		1			
	1-129	111 30	LOST	US 4178-1	<1.73E-3	MA	NA	30.70	6.40E-01	HA	
								34.80			
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	* 3 * * * * *	13,39				To Part of	V = 134,92%	* 1		M	36.76
DATE 4-21-93	10.471.										1
	\$r-90	92 20	Q 136-)	J 3 41E-00	1 17E-00	J.43E-00	3:31	95.40	3.15E-03	1393 O	0 10
								94.10			
	32 4 30	1111	31 4444	3 10 10 81					44 19 19 19 19 19 19 19 19 19 19 19 19 19	1837	
DATE:4-24-92	10.443										
	H-3	81 40	Q 112-4	US OHE-	<0.13E-4	NA	NA	H.30	3 15E-Q4	NA	6 30
		Transfer of	18.6	187-6		1111	3 . C . S . S . S	11	1900 P	वृत्ते हुन्। प	* 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
DATE.4-24-92	18.463								<u> </u>		<u> </u>
	C-14	88 90	₽3UE-4	UT QUE-	Q NE-4	НА	NA	67.80	3 25E-04	NA .	4 24
	\$ 14K 100	计数数数	भूत भू भव	No. of the last of			1000	3 1 1			1
D4TE4-29-91	18.471									l	
	Se-79	H/A	«1 40E-4	U <1.36E-1	<1.39E-4	NA	NA	89 30	1.408-04	NA	4.50
	1							87.50	\		

Then were so TCLP enclysis required for this core composite sample.

WHC-SD-WM-DP-058 REV. O-B

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WHC-EP-0806

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188.5 J/g calculated dry (wt. basis) value compared to average May exotherm values of 280.9 J/g on a calculated dry (wt. basis) value. The exotherm trace obtained in May is less well resolved than the January scan. The exotherms are very broad and not well separated from the water endotherms, so the baselines have been conservatively estimated.

Comparison of the current 222-S data (May 1994), and the original January 1992 data obtained for Core 33 segments 1 and 2 is quite interesting. The original data (Jan.1992) was obtained on a wet sample under air. The January 1994 and current data are on a dried sample as previously described. The wet sample (Jan. 1992), Core 33 segment 1 yielded an average value of 1151 J/g dry (wt. basis), and 2848.5 J/g dry (wt. basis) for segment 2. Both of these values are considerably different from the values obtained on the dried sample either in January or May 1994.

Samples from Core 33 segment 2 were also run by PNL in January 1994, reference 6. Results for Core 2 samples run under nitrogen purge were 942 J/g dry (wt. basis) and 1011 J/g dry (wt. basis) for an average value of 976.5 J/g dry. Sample preparation for the January-1994 samples run by both labs were similar. The PNL sample was vacuum dried at 60 degrees C for 24 hours. At 222-S the sample was vacuum dried (35 torr) at 60 degrees C to a constant weight. The two laboratories used different heating rates for the DSC. The 222-S lab used 10 C per minute, while PNL ran their sample at 5 C per minute. This difference should not effect the magnitude of the exotherm, but may cause a shift in the curve. The instruments used by the laboratories were also different. The instrument used at 222-S is a Mettler M3 balance DSC Model 20 TG-50, and at PNL a Perkin Elmer Model 2 was used.

: -

##C-SD-WM-DP-058

Tank 241-T-111 Limited Analysis of Core 33 Segment 2

Introduction

The analyses in this package were performed by the Westinghouse Hanford Company (WHC), under the guidance provided by the Analytical Integration Characterization Program (AICP), refer to references 1, 2, and 3. The analyses presented in this report provides additional information on the energetics previously observed for Tank 241-T-111, reference 4. There were no exotherms that exceeded the safety criteria.

Sample preparation and analyses were performed by the Analytical Services of WHC. Core 33 segment 2 was placed in an oven on 1/10/94 at 1500 with an initial weight of 28.64 grams (gms) and removed on 1/12/94 at 0735 and was stabilized in a hood for 1 hour, the final weight of the sample was 6.82 gms. The dried sample was stored in archive and was retrieved in May to perform this body of work. The specific analyses requested included Differential Scanning Calorimetry (DSC), which was performed under a nitrogen atmosphere on the sample previously dried under vacuum (35mm of Hg) at 60 degrees C. In addition, Thermographic Analysis (TGA) were performed on the dried archive sample. The analyses were performed to the quality assurance protocols reference 7.

Data Evaluation of Tank 241-T-111

Thermogravimetric analysis on the dried samples yielded an average value of 11.2% weight loss interpreted to be water loss. Differential Scanning Calorimetry of 5-10 mg. and 10-20 mg. sample sizes under nitrogen purge were performed. This analysis (May 1994) of Core 33 segment 2 showed consistent low broad exotherms, with no significant differences caused by sample sizes. These results are consistent with previous work performed on Core 33 segment 2 in January 1994, reference 5. The average BSC exotherm value from January was

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References:

- Internal Memo, D. R. Bratzel to J. G. Kristofzski, "Letter of Instruction on Centrifugation of Tank 241-T-111 Samples," dated April 22, 1994.
 7E720-94-119 Westinghouse Hanford Company, Richland Wa.
- 2. Internal Memo, D. R. Bratzel to J. G. Kristofzski, "Letter of Instruction for Tank 241-T-111 Archive Samples, "dated April 29, 1994.

 7E720-94-120 Westinghouse Hanford Company, Richland Wa.
- 3. Internal Memo, D. R. Bratzel to J. G. Kristofzski, "Revised Letter of Instruction for Tank 241-T-111 Archive Samples." dated May 2, 1994.

 Internal Memo, D. R. Bratzel to J. G. Kristofzski, "Letter of Instruction for Tank 241-T-111 Archive Samples," dated April 29, 1994.
- 7E720-94-122 Westinghouse Hanford Company, Richland Wa.
- 4. WHC-SD-WM-DP-024, Rev O. Tank 241-T-111 Core 31 and Core 33.
- 5. WHC-SD-WM-DP-058, Rev O. Tank 241-T-111 Reanalyses of T-111 limited Analysis.
- 6. Pacific Northwest Laboratories, Tank Waste Characterization Project. Tank 241-T-111 Core 33 Data Report Rev. O. January 10, 1994. Pacific Northwest Laboratories, Richland Wa.
- 7. WHC-SD-WM-PLN-047, Appendix A, Core Sample Characterization Quality Assurance Project Plan.

			4 1				
Sample Prep.	Analyte	Safety Criteria	Std. #1	Result #1	Result #2	Average	RPD
		J/g dry	%	J/g	J/g	J/g	%
			J1667-5511	J1668-5711	J1668-5711		
Direct	DSC	>522 J/g dry	96.0	251.2	269.0	260.1	6.84
	· · · · · · · · · · · · · · · · · · ·		J1667-5511	J1669-5711	J1669-5811		
Direct	DSC	>522 J/g dry	96.0	309.2	287.5	298.4	7.27
			J1667 - 5511	J1670-5711	1670-5811		
Direct	DSC	>522 J/g dry	96.0	180.2	187.1	183.7	3.76
	· · · · · · · · · · · · · · · · · · ·						
	!		J1667 - 5511	J1671-5711	J1671-5811		
Direct	DSC	>522 J/g dry	96.0	162.7	175.3	168.8	7.46
	!		J1667 – 5511	J1672-5711	1672-5811		
Direct	DSC	>522 J/g dry	96.0	336.2	335.9	336.0	0.09
1							i
			%	%	%	Average %	
1			J1667-5512	J1668-5712	J1668-5812		
Direct	TGA		99.1	11.68	10.06	10.9	14.86
	1						
			J1667-5512	J1669-5712	J1669-5812		
Direct	TGA	<u> </u>	99.1	10.66	9.90	10.3	7.38
:							
			J1667-5512	J1670-5712	1670-5812		
Direct	TGA	<u> </u>	99.1	11.35	11.99	11.7	5.47
·			J1667 - 5512	J1671-5712	J1671-5812		
Direct	TGA		99.1	12.07	12.08	12.075	0.08
-						[<u>-</u>	
			J1667-5512	J1672-5712	J1672-5812		
Direct	TGA	 	99.1	11.21	11.10	11.15	0.99
		<u> </u>	}]	· · · · · · · · · · · · · · · · · · ·]	

WHC-EP-0806
WHC-SD-WM-DP-058

Westinghouse 1 Hanford Company

WHC-SD-WM-DP-058 REV. 0-XB 110 5/24/94

Internal Memo

From:

Analytical Integration

7E720-94-119

Phone:

373-3579 R2-18

April 29, 1994 Date:

LETTER OF INSTRUCTION FOR TANK 241-T-111 ARCHIVE SAMPLES Subject:

To:	J. G. Kristofzski	T6-06		
•	cc: H. Babad	R2-78	K. L. Kocher	T6-06
	G. S. Barney	T5-12	R. P. Marshall	T6-14
	D. B. Bechtold	T6-0 9	L. M. Sasaki	R2-12
	C. DeFigh-Price	R2-31	P. Segall	H4-19
	D. B. Engelman	R1-49	B. C. Simpson -	R2-12
	J. M. Grigsby	H4-62	D. A. Turner	R2-78
	C. S. Haller	R2-12	for T. E. Whelan HKA	S1-57
	J M Kier	T3-01	LMS File/LB	

Reference:

Internal Memo, D. R. Bratzel to J. G. Kristofzski, "Letter of Instruction on Centrifugation of Tank 241-T-111 Samples," dated April 22, 1994.

This letter of instruction provides direction to the 222-S Laboratory for the performance of differential scanning calorimetry (DSC) and thermal oravimetric analysis (TGA) tests on archived material from single-shell tank 241-T-111. This letter also cancels the centrifugation and moisture measurements requested in the previous letter of instruction (Reference). The previously requested tests were determined not to be an effective means of addressing the U.S. Department of Energy, Headquarters' concerns regarding the safety of pumping the tank. Alternative testing to estimate the moisture that will remain in the waste after pumping is being considered and may be requested in the future. However, it has been determined that _additional information on the waste energetics is necessary to evaluate the safety of pumping the tank.

Purpose: Perform additional DSC analysis and TGA on tank 241-T-111 archive sample to more accurately measure the exotherm observed in the sample.

Safety Considerations: Radioactive materials will be used in these tests, therefore all laboratory procedures applicable to radiological control will be adhered to in conjunction with as low as reasonable achievable (ALARA) practices dealing with hazardous materials in the preparation and handling of these samples.

Work_Scope:

The sample to be analyzed in the archived sample from tank 241-T-111, core 33, segment 2 which had been dried, but never used, for adiabatic calorimetry analysis. Homogenize the sample to ensure that it is well mixed.

Walter France (MI) Sept 96

J. G. Kristofzski Page 2 April 29, 1994

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7E720-94-119

- 2. Obtain three aliquots from the sample and perform DSC and TGA analyses in duplicate on each aliquot. The sample size for the DSC analysis should be 5 to 10 milligrams. Perform the DSC under a nitrogen atmosphere.
- 3. Obtain two aliquots from the sample and perform DSC and TGA analyses in duplicate on each aliquot. The sample size for the DSC analysis should be twice the amount used in step 2 above (i.e., 10 to 20 milligrams). Perform the DSC under a nitrogen atmosphere.

Quality Assurance: If the sample and-duplicate for a given aliquot do not agree within 25 calories/gram, rerun the DSC using a sample from the same aliquot. Approval designator Q has been established for this work; samples and the indicated number of duplicates shall be run. The DSC shall be performed in accordance with procedure number LA-514-113 and the TGA shall be performed in accordance with procedure number LA-560-112. The requirements of WHC-SD-CP-QAPP (Quality Assurance Project Plan for the Analysis of Highly Radioactive Mixed Waste Samples in Support of Environmental Activities on the Hanford Site) shall be implemented as applicable. The data shall be reviewed to ensure that all quality assurance/quality control requirements were met.

Data Evaluation: The 222-S Laboratory shall evaluate the results of the analyses against the DSC analyses performed previously on the tank T-III samples that show exotherms in excess of 200 calories/gram (both cores 31 and 33) at both the 222-S and 325 laboratories. The evaluation should include a best estimate of the sample energetics along with uncertainties and the rationale behind the estimate.

Reporting Requirements: A letter presenting the results of the measurements will be the final deliverable of this task. The letter shall include a summary table of the results and a narrative. Advance notification of results via cc:mail is requested for representatives from the Characterization Program (D. R. Bratzel), Characterization Support (B. C. Simpson), Waste Tank Safety (H. Babad), Waste Tank Stabilization (D. B. Engelman), and TWRS Safety Analysis and Engineering (J. M. Grigsby). All analytical results shall be reported by May 3, 1994, the data evaluation shall be reported by May 5, 1994, and the letter report shall be issued by May 9, 1994.

WHC-SD-WM-DP-058

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J. G. Kristofzski Page 3 April 29, 1994

Funding for this task is provided under task package control number N54D2. A cost estimate should be provided to the Characterization Program by May 3, 1994.

If there are any questions or comments regarding this letter, please contact L. M. Sasaki at 373-1027 or B. C. Simpson at 373-5915.

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D. R. Bratzel, Manager Analytical Integration Characterization Program

pjm

Westinghouse Hanford Company WHC-SD-WM-DP-058

REV. 0-X B

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Internal Memo

From:

Analytical Integration

7E720-94-120

Phone:

373-3579 R2-18

Date: May 2, 1994

Subject: REVISED LETTER OF INSTRUCTION FOR TANK 241-T-111 ARCHIVE SAMPLES

To:	J. G. Kristofzski	T6-06		•
	cc: H. Babad G. S. Barney D. B. Bechtold C. DeFigh-Price D. B. Engelman	R2-78 T5-12 T6-09 R2-31	K. L. Kocher R. P. Marshall L. M. Sasaki P. Segall B. C. Simpson	T5-06 T6-14 R2-12 H4-19 R2-12
**** * * * * * * * * * * * * * * * * *	J. M. Grigsby C. S. Haller J. M. Kier	H4-62 R2-12 T3-01	D. A. Turner T. E. Whelan LMS File/LB	R2-78 S1-57

Reference:

Internal Memo, D. R. Bratzel to J. G. Kristofzski, "Letter of Instruction for Tank 241-T-111 Archive Samples," dated April 29, 1994.

This letter of instruction (LOI) provides a revision to the referenced LOI. This revision expands the work scope to include additional differential scanning calorimetry (DSC) and thermal gravimetric analysis (TGA) tests on wet (as received) archive sample from tank 241-T-111, core 33, segment 2 and revises the due dates for the reporting of results. All other requirements of the referenced LOI remain the same.

<u>Additional Work Scope:</u>

- The additional sample to be analyzed is the 20 gram of wet sample from the tank 241-T-111, core 33, segment 2 archived sample. This sample is being shipped from the Pacific Northwest Laboratory (PNL) 325 Building to the 222-S Laboratory for analysis. PNL is not able to analyze the sample at this time due to the stand down at the laboratory. Homogenize the sample to ensure that it is well mixed.
- 2. Obtain three aliquots from the sample and perform DSC and TGA analyses in duplicate on each aliquot. The sample size for the DSC analysis should be 5 to 10 milligrams. Perform the DSC under a nitrogen atmosphere.
- 3. Obtain two aliquots from the sample and perform DSC and TGA analyses in duplicate on each aliquot. The sample size for the DSC analysis on the wet sample should be five times the amount used in step 2 above (i.e., 25 to 50 milligrams). Perform the DSC under a nitrogen atmosphere.

Change to Reporting Due Dates:

Analytical results and preliminary data evaluation on the dried archived sample shall be reported by May 6, 1994. Final evaluation of both dry and wet sample results, in the form of a letter report, shall be made within four full working days after receip he sample from PNL.

A-103

J. G. Kristofzski Page 2 May 2, 1994

WHC-SD-WM-DP-058

REV. 0-KB LUW 424/4 7E720-94-120

If there are any questions or comments regarding this letter, please contact L. M. Sasaki at 373-1027 or B. C. Simpson at 373-5915.

J.M. Salai for D. R. Bratzel, Manager Analytical Integration Characterization Program

Westinghouse Hanford Company

WHC-SD-WM-DP-058

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Internal Memo

From: Analytical Integration

Phone:

373-3579 R2-18

Date:

May 12, 1994 a los estas market a

Subject: LETTER OF INSTRUCTION FOR TANK 241-T-111 ARCHIVE SAMPLES -

REVISION 2

To:	J. G. Kristofzski	T6-06		
	cc: H. Babad	R2-78	K. L. Kocher	T6-06
	G. S. Barney	T5-12	R. P. Marshall	T6-14
	D. B. Bechtold	T6-09	L. M. Sasaki	R2-12
	C. DeFigh-Price	R2-31	P. Segall	H4-19
	D. B. Engelman	R1-49	B. C. Simpson	Ř2−12
	J. M. Grigsby	H4-62	D. A. Turner	R2-78
	C. S. Haller	R2-12	T. E. Whelan	S1-57
	J. M. Kier-	T3-01	LMS File/LB	

- Internal Memo, D. R. Bratzel to J. G. Kristofzski, References: (1) "Revised Letter of Instruction for Tank 241-T-111 Archive Samples." dated May 2,1994.
 - (2)--Internal Memo, D. R. Bratzel to J. G. Kristofzski, "Letter of Instruction for Tank 241-T-111 Archive Samples," dated April 29, 1994.

This letter of instruction (LOI) provides a second revision to the referenced LOIs. This revision eliminates the additional work scope requested in Reference 1. It has been determined that additional differential scanning calorimetry and thermal gravimetric analysis tests on wet (as received) archive sample from tank 241-T-111, core 33, segment 2 are not needed at this time. The final letter report should be issued by May 19, 1994.

If there are any questions or comments regarding this letter, please contact -L.-M. -Sasaki -at-373-1027-or B. C. Simpson at 373-5915.

D. R. Bratzel, Manager Analytical Integration Characterization Program

pkc

TO: JG Kristofzski, JM Frye, KL Kocher

WHC-SD-WM-DP-058

FROM:

TL Welsh, RD Cromar, R Jeppson

REV. O-B

DATE: SUBJECT: May 5, 1994

Statistical Analysis of Core Segments

Performing a one-way analysis of variance (ANOVA) on segment two (core 33 only), we found that the standard deviation of segment two is 211.631 J/g dry wt. The ANOVA also leads us to conclude that some of the means are different. A multiple range comparison test showed that the means of the 1993 T-111 data and the present data are not statistically different, with a significance level of 0.05. These two means differ from the mean of the 1993 PNL data and also from the mean of the original data. The 1993 PNL mean and the mean of the original data also differ from each other.

Performing a one-way analysis of variance on segment one (core 33 only), the standard deviation of segment one is 169.621 J/g dry wt. The ANOVA leads us to conclude that the mean of the original data and the mean of the 1993 T-111 data are different. Using the error from segment two to test segment one, the means are also statistically different, with a significance level of 0.05.

Summary Statistics for Core Samples
(in J/o dry wt)

	Confidence Interval		Mean	\\Standard	RSD
	L.L.	υ.L.]	Deviation	
Core 33, 1994	*	*	280.9	81.29	0.2894
Secment 1	441.50	3088.17	1764.83	532.67	0.3018
Segment 2	-164.55	2671,11	1253.28	1142.06	0.9113
Segment 3	-14518.56	17932.56	1707	1805.95	1.0580

Segment 1

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Analysis of variance

Source of variation level	Sum of Squares	d.f.	Mean square	F-ratio	Sig.
Between groups	911070-25-	1	911070.25	31.666	.0302
Within groups	57542.50	2	28771:25	·	
Total (corrected)	968612.75	· 3			

O missing value(s) have been excluded.

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Segment 2

REV. 0-15

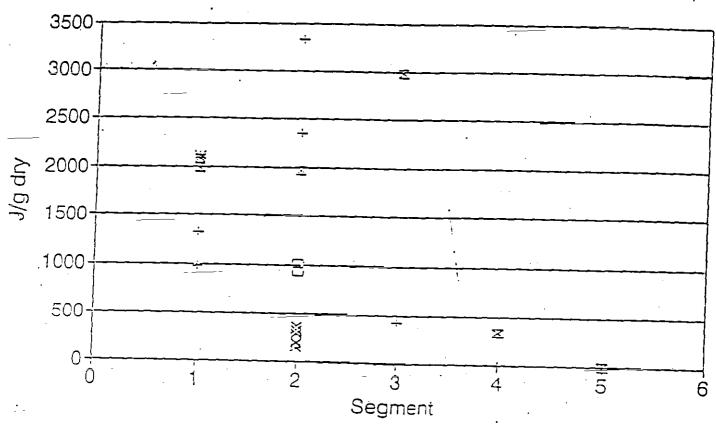
Analysis of variance

Source of variation level	Sum of Squares	d.f.	Mean square	F-ratio	Sig.
Between groups	11656441	3	3885480.2	86.753	.0000
Within groups	537452	12	44787.7		
Total (corrected)	12193893	15			

O missing value(s) have been excluded.

WHC-SD-WM-DP-058 REV. 0-13

T-111 DSC



Core 31 (1991) Core 33 (1991) -- * Core 33/(1993)

Core 33 (1993) P. × Core 33 (1994)

WHC-SD-WM-DP-058

REV. 0-13

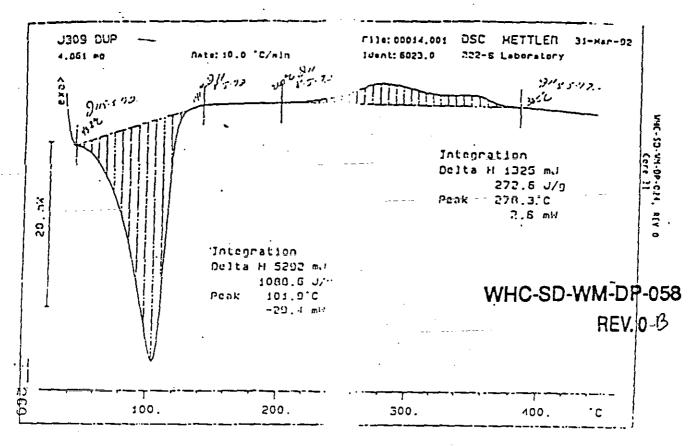
Preliminary Data for re-analysis on tank T-111 Core 33 Segment 2, May 2, 1994 All data is reported on a archive sample which was vacuum dried 60 degrees C. on 12/93. The analyses were performed under a nitrogen purge. This preliminary data fulfills the requirements per the letter or instruction (LOI) 7E720-94-119 and the revised LOI 7E720-94-120. The DSC values are not corrected and do not take into account the water content of the samples.

	- ₁				
Laboratory Id. number	Sample #	Analysis	Sample Size (mg)	Result 1 J/g	Result 2 J/g
J1668-5711 J1668-5811	Sample 1 Sample 1 dup	DSC	16.957 13.267	251.2	269.0
J1669-5711 J1669-5811	Sample 2 Sample 2 dup	DSC	12.596 18.182	309.2	287.5
J1670-5711 J1670-5811	Sample 3 Sample 3 dup	DSC	5.897 6.179	180.2	187.1
J1671-5711 J1671-5811	Sample 4 Sample 4 dup	DSC	6.695 7.023	162.7	175.3
J1672-5711 J1672-5811	Sample 5 Sample 5 dup	DSC -	7.740 6.154	336.3	335.9
Laboratory Id. number	Sample #	Analysis	Sample Size (mg)	Result 1 (% H2O)	Result 2 (% H2O)
J1668-5712 J1668-5812	Sample 1 Sample 1 dup	TGA	11.928 15.604	11.68	10.06
J1669-5712 J1669-5812	Sample 2 Sample 2 dup	TGA	17.548 19.760	10.66	9.90
J1670-5712 J1670-5812	Sample 3 Sample 3 dup	TGA	6.338 8.043	11.35	11.99
J1671-5712 J1671-5812	Sample 4 Sample 4 dup	TGA	7.786 5.750	12.07	12.08
J1672-5712 J1672-5812	Sample 5 Sample 5 dup	TGA	7.101 9.313 .:_	11.21	11.10

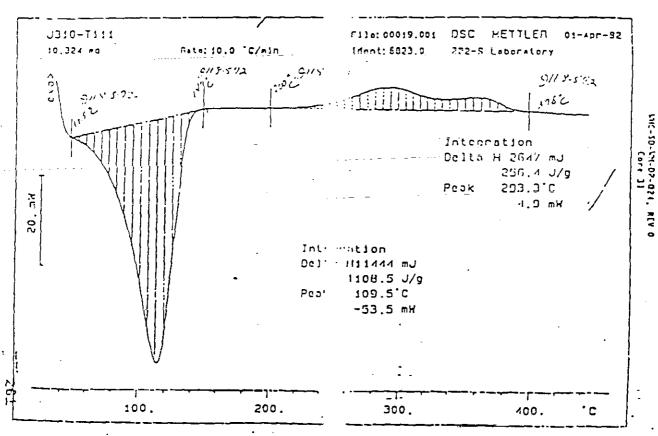
WHC-SD-WM-DP-058 REV. 0-B

Original T-111 DSC Data 4/92

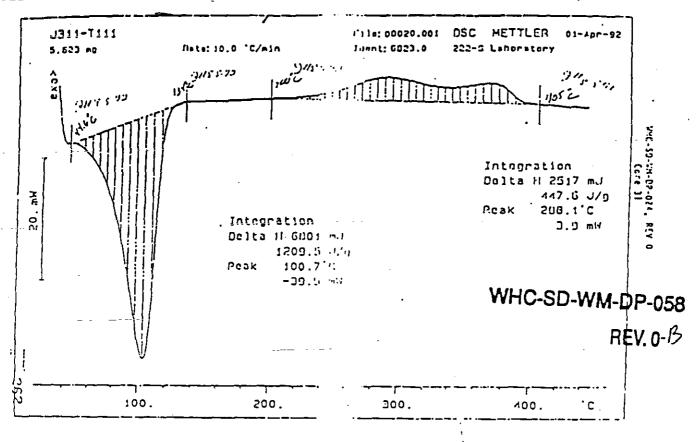
Core 31 seg], wet sample run under static air 4/92



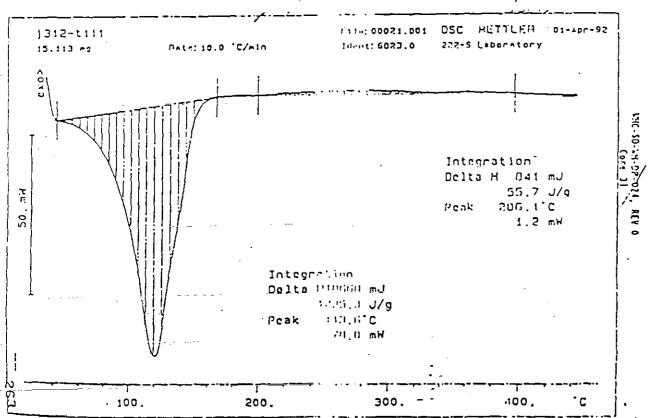
Core 31 seg 2, wet sample run under static air 4/92



Core 31 seg 3, wet sample run under static air 4/92

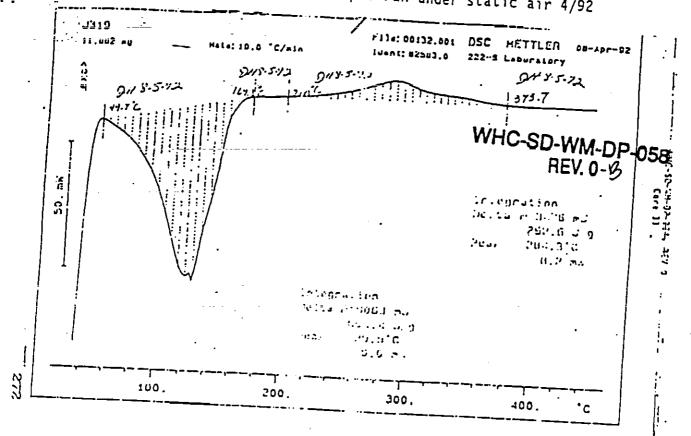


Core 31 seg 4, wet sample run under static air 4/92

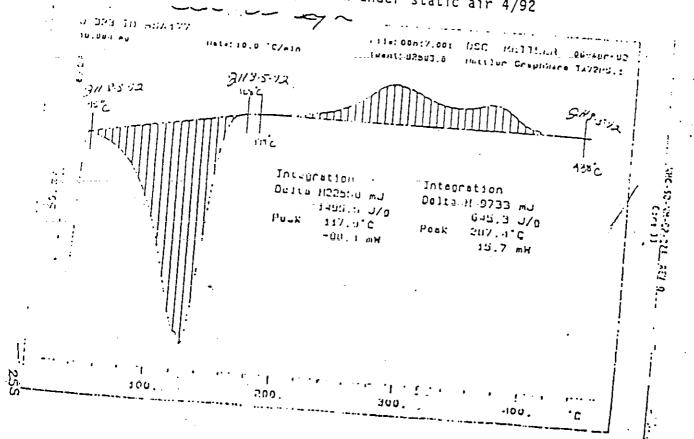


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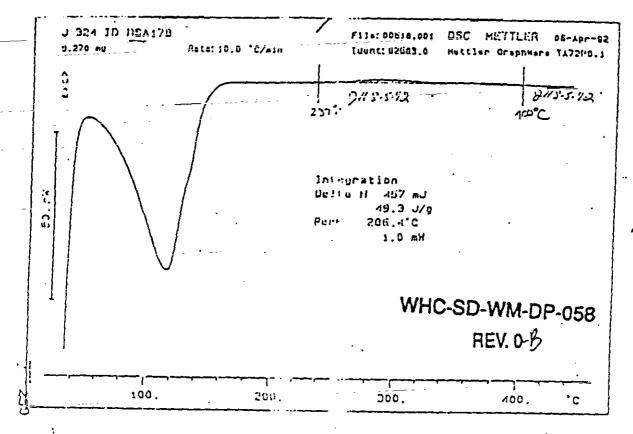
- WHC-EP-0806 Core 33 seg 1, wet sample run under static air 4/92



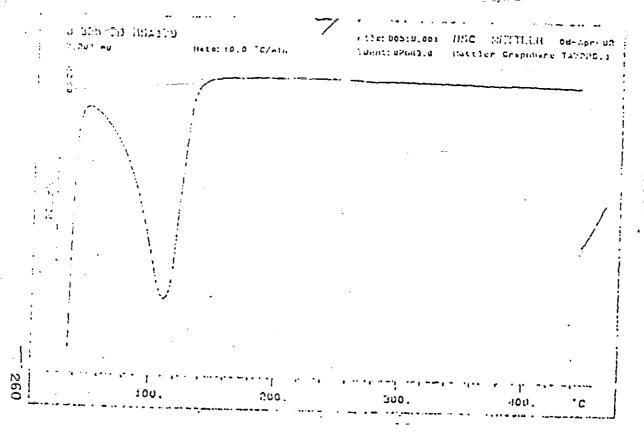
Core 33 seg 2, wet sample run under static air 4/92



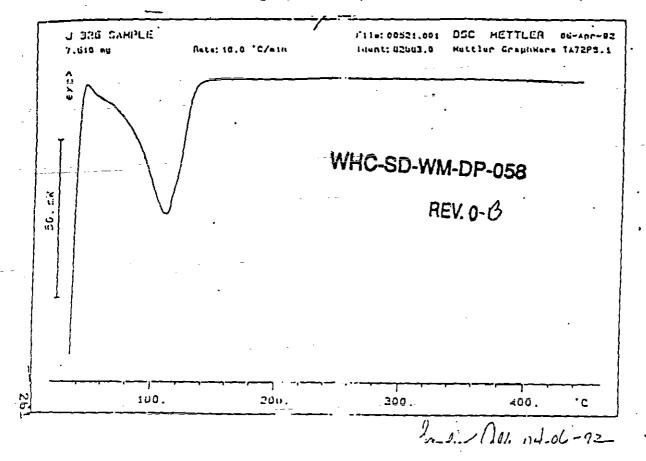
Core 33 seg 3, wet sample run under static air 4/92



Core 33 seg 4, wet sample run under static air 4/92

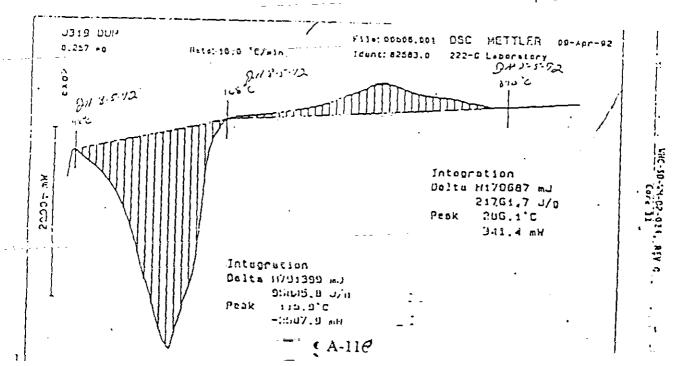


Core 33 seg 5, wet sample run under static air 4/92



4/92

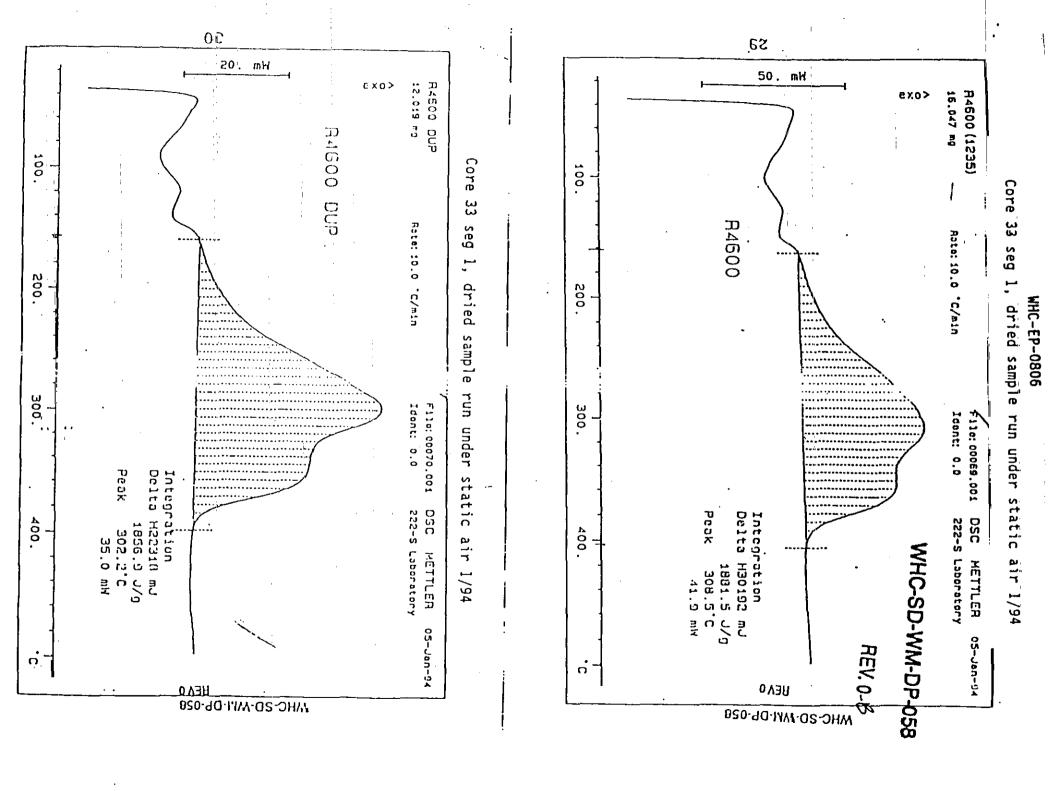
Core 31 seg I, wet sample run under static air 4/92



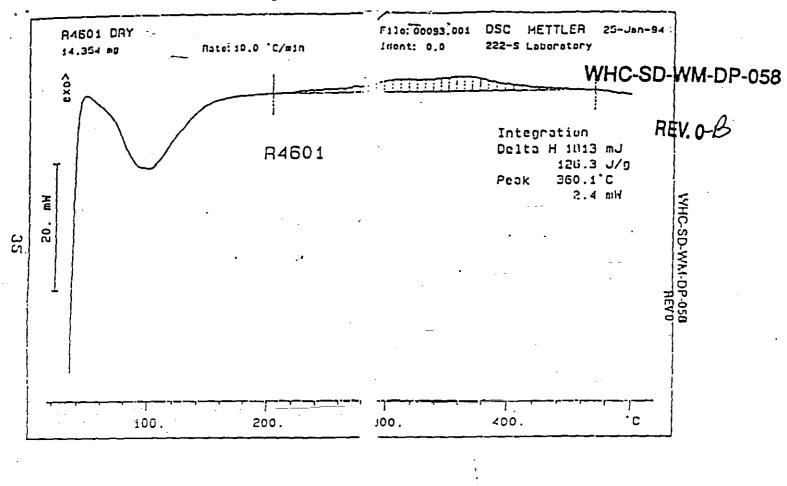
WHC-SD-WM-DP-058

REV. O-B

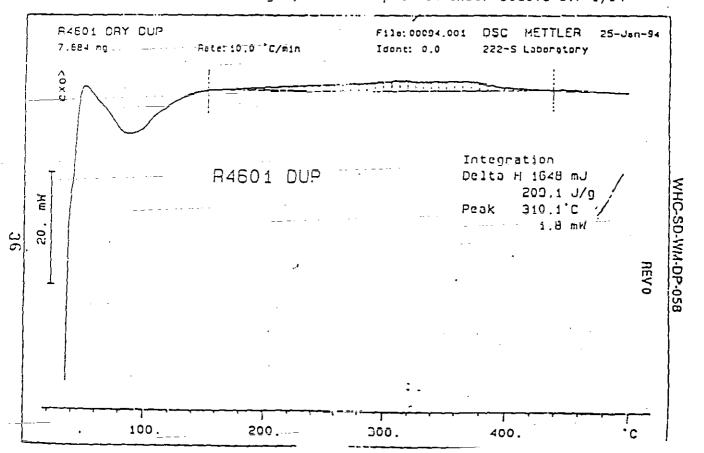
Re-analysis T-111 DSC Data 1/94



Core 33 seg 2, dried sample run under static air 1/94



Core 33 seg 2, dried sample run under static air 1/94

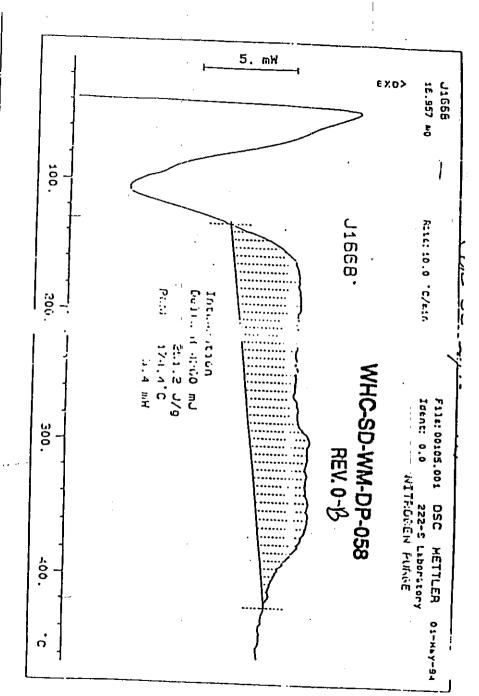


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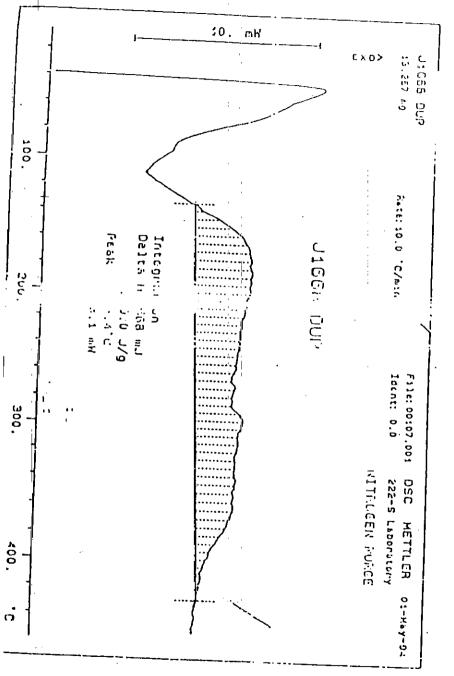
REV. O-B

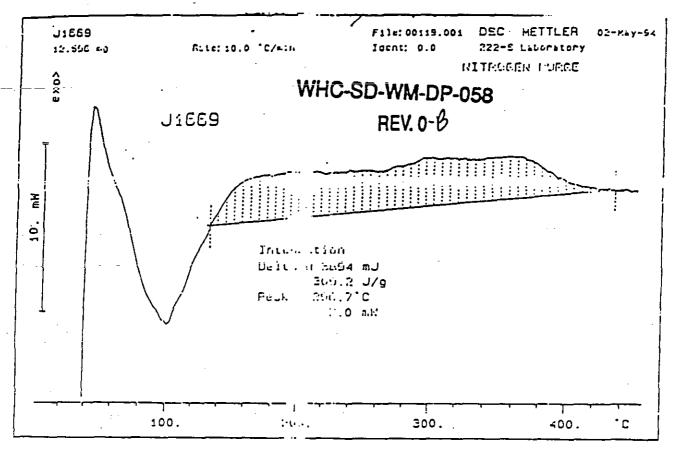
Re-analysis T-11-1 DSC Data 5/94

Core ű seg dried sample run under þ nitrogen purge 5/94

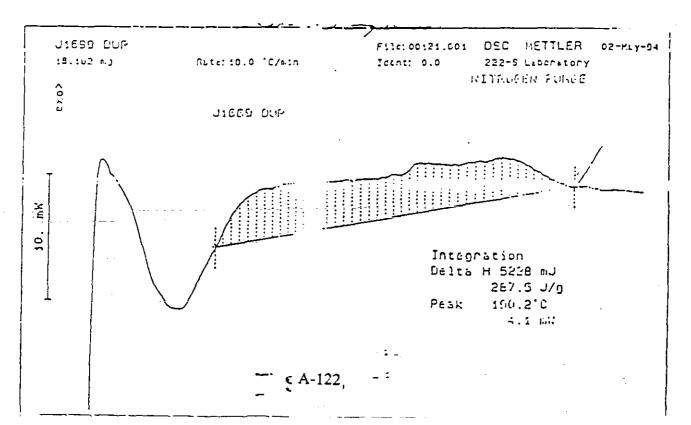


Core ယ seg ŝ dried sample run under ρ nitrogen purge 5/94



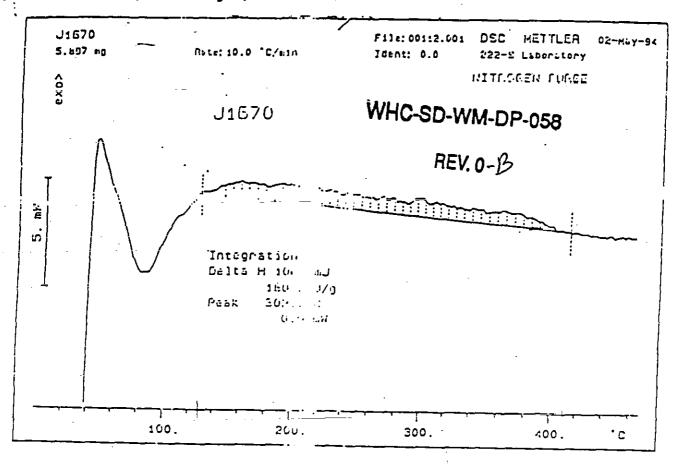


Core 33 seg 2, dried sample run under a nitrogen purge 5/94

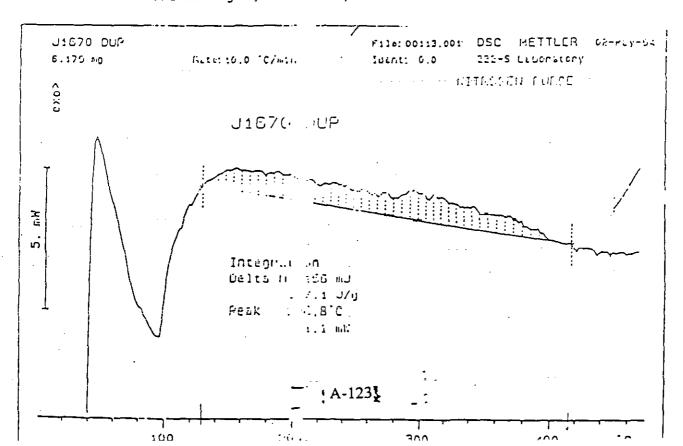


WHC-EP-0806

Core 33 seg 2, dried sample run under a nitrogen purge 5/94

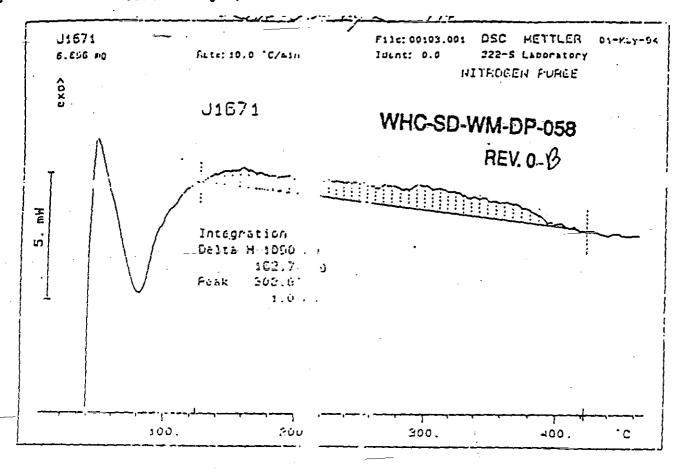


Core 33 seg 2, dried sample run under a nitrogen purge 5/94

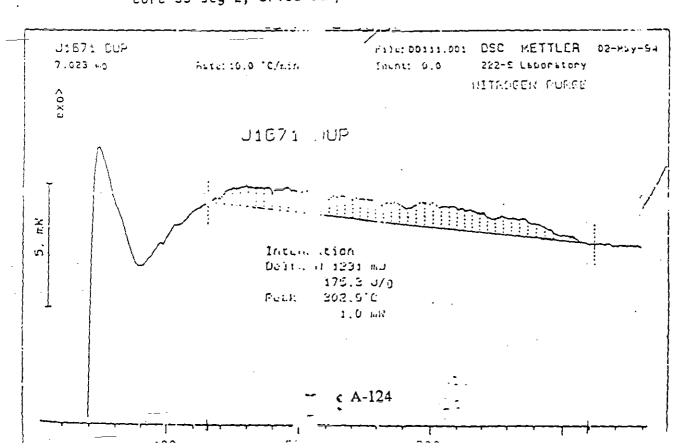


WHC-EP-0806

Core 33 seg 2, dried sample run under a nitrogen purge 5/94

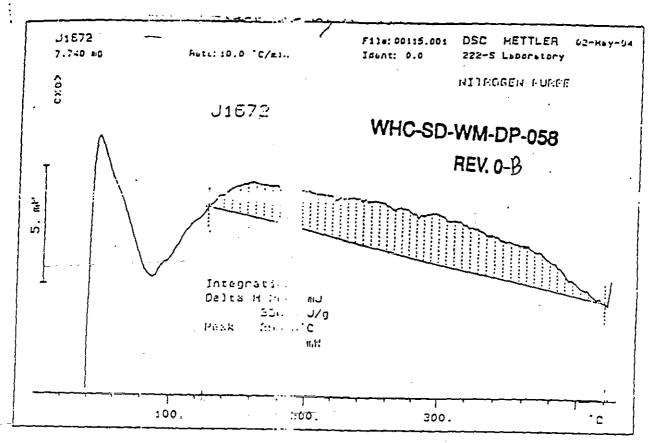


Core 33 seg 2, dried sample run under a nitrogen purge 5/94

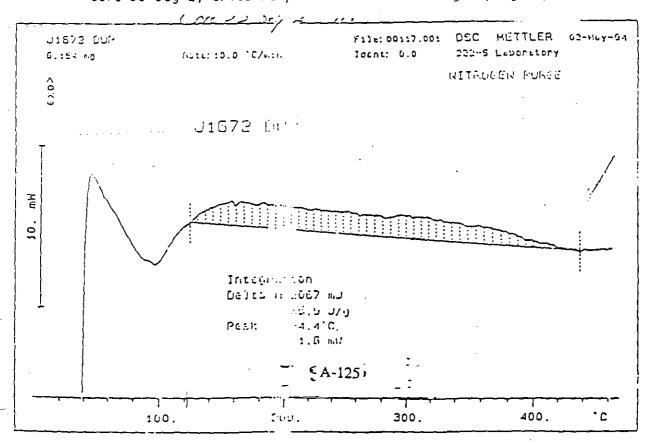


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Core 33 seg 2, dried sample run under a nitrogen purge 5/94



Core 33 seg 2, dried sample run under a nitrogen purge 5/94



WHC-SD-WM-DP-058

REV. 0-B

PNL Re-Analysis 12/93

Thermal Analysis Data

WHC-SD-WM-DP-058 REV. O.B

Thermal analyses of tank waste material from Tank 241-T-111 Core 33, Segment 2 were performed in duplicate. The thermal analysis includes both differential scanning calorimetry (DSC) and scanning thermogravimetry (TGA). DSC analyses were performed on dried sample material, but the TGA analyses were performed on the as-received sample. These analyses were performed according to technical procedure PNL-ALO-508, "Laboratory Procedure for Operation of the Differential Scanning Calorimeter (DSC), Thermogravimetric Analyzer (TG), and High Temperature Differential Thermal Analyzer (DTA) and DSC."

DSC analyses of sample 92-05856-N and it's duplicate were performed on a Perkin Elmer Model 2 Differential Scanning Calorimeter. The sample was prepared by transferring 0.4896 grams of sample from 92-05856 (labeled T-111, Core 33, Segment 2) to a glass vial. This sample was placed in a vacuum oven at 60°C and 28 in. Hg (711 mm Hg) at 1315 hours on December 9, 1993. The sample was removed from the oven at 1300 hours on December 10, 1993 and allowed to cool prior to weighing. The sample weight after this drying period was 0.0721 grams (a loss of 0.4175 grams or 85.3% mass loss). The sample was placed back in the oven at 60°C and 28 in. Hg at 1310 hours on December 10, 1993 and removed at 1500 hours on the same day. The sample was reweighed and a loss of only 0.0005 grams was observed; therefore, the sample was sealed and placed in a desiccator. The sample was analyzed on December 13, 1993 using a nitrogen gas environment. Samples of 11.19 and 15.72 milligrams were used for Runs 1 and 2, respectively. A temperature range from 35 to 550°C was scanned at 5°C/min. The results of these analyses are given in Table 1-1.

Table 1-1: Core 33, Segment 2, Energetics

RUN	RANGE (°C)	ONSET (°C)	ENTHALPY (J/g)
11	107 - 381	201	-898
·. 2 —	113 - 394	196	-836

WHC-SD-WM-DP-058

REV. O-B

An indium and a Zinc standard was run prior to analyzing the sample. The indium standard had an onset temperature of 156.8°C with an enthalpy of 28.3 J/g. The literature values for indium are 156.6°C and 28.45 J/g. The zinc standard had an onset temperature of 417.1°C with an enthalpy of 98.7 J/g. The literature values for zinc are 419.47°C and 108.37 J/g.

Thermogravimetric analysis was performed in triplicate on an aliquot of asreceived sample material using a Perkin Elmer Model 2 Thermogravimetric Analyzer.

A nitrogen gas environment was used to analyze these samples. Samples of 11.32,
17.11, and 19.40 milligrams were used for Runs 1, 2, and 3, respectively.

Run 1 was performed on Decamber 28, Run 2 on Decamber 29, 1993, and Run 3 on
January 3, 1994. A temperature range from 35 to 550°C was scanned at 5°C/min for
Runs 1 and 2. The temperature range for Run 3 was from 35 to 200°C at a scan
rate of 5°C/min. This shorter scan range was chosen, because no significant
weight loss was observed above 150°C. The results of these analyses are given
in Table 1-2.

<u>Table 1-2</u>: Core 33, Segment 2, Mass Loss as a Function of Temperature

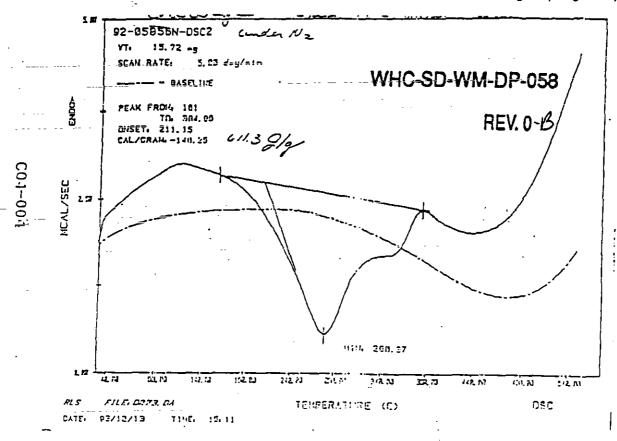
RUN	RANGE (· C)	WEIGHT LOSS (wt%)
l	31 - 103	67.3
2	31 - 116	80.3
3	30 - 130	79.7

Run 1 appeared to be on a sample that was significantly drier than the other samples. Runs 2 and 3 compare well with the gravimetric/determinations weight percent solids measurements. It was also noted in Runs 1 and 2 that no significant weight loss was observed in the temperature range of the exothermic reaction.

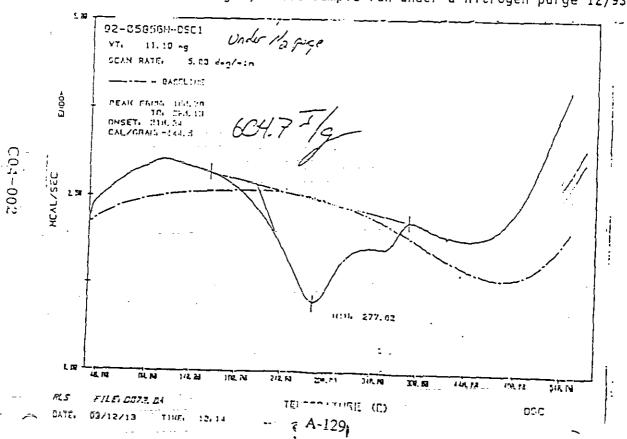
Alumel and perkalloy curie point standards were run prior to analyzing the sample. The alumel standard had an onset temperature of 162.35°C, and the perkalloy standard had an onset temperature of 595.73°C. The literature values for alumel and perkalloy are 163°C and 596°C, respectively.

Acres 1 September 1 September 1982

Core 33 seg 2, dried sample run under a nitrogen purge 12/93



Core 33 seg 2, dried sample run under a nitrogen purge 12/93



2

WHC-SD-WM-DP-058

Rev. O-NB

|@@:ppcalibration

mG

17-FEB-94

8:45

START TEMP. ٥C 35

END TEMP. ٥C 800

FILE NO.

00130.001

IDENT. NO.

WEIGHT

6.7520

TEMPERATURE °C

WEIGHT

GAIN---->

5.0000 mG _200..0 400.0 5(0) Ū.

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TEMP		

°C

143.67

TEMP.

٥C

349.67

°C

739.00

A - PT100 \mathbf{B} PT100 C PT100

22341 -70428

- 05962

OCALIBRATION

31-JAN-94

16:16

START TEMP. END TEMP.

35 750.

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WHC-SD-WM-DP-058

Rev. 0-XB /24/44

FILE NO. IDENT. NO.

mG

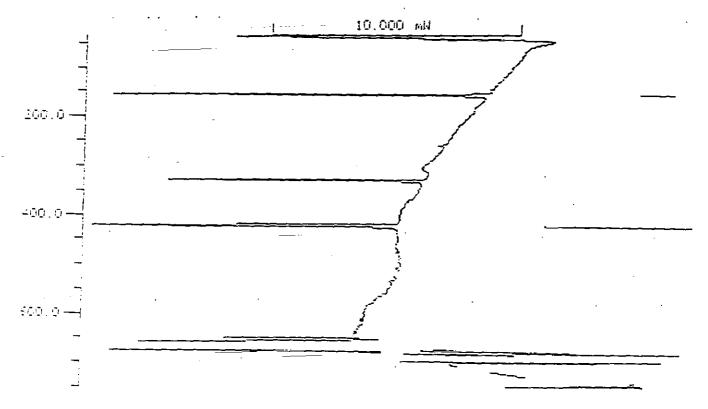
00120.001

7.57

TEMPERATURE °C

WEIGHT

HEAT FLOW EXOTHERMAL-->



Α	PT100			.21401
В	PT100			.75726
C	PT100			
	INDIUM	٠.	•	12421 310.23

. A-1315

WHC-SD-WM-DP-058 REV. 0-B

ADDITIONAL DATA SUBMITTED ON 05/18/94

WESTINGHOUSE HANFORD COMPANY

WHC-SD-WM=DP-058

222-S LABORATORY

REV. 0-XB	ANALYTICAL BATCH			
Lab Segment Serial No. 1 3/24/44 J1668-72	Customer ID: 1240-D-A,B,C,D,E			
Analysis:	Sample Prep: DIRECT			

Instrument: WC16134	Procedure/ Rev: LA 514-113/B-0	
Technologist: R. WENDLAND	Date: 05/03/94 5/2/94	
Starting Time: 0600	Temperature N/A	
Ending Time: 2300	Chemist: J. FRYE	

Lab ID

- J1672-5811

Comments: _____

	Description	Lab ID		Description
1	LMCS	J1667-5511	11 0	UP
2	SAMPLE	J1668-5711	12	
_3	DUP	J1668-5811	13	
4	SAMPLE	J1669-5711	14	
5	DUP	J1669-5811	15	
6	SAMPLE	J1670-5711	16	
7	DUP	J1670-5811	17	
8	SAMPLE	J1671-5711	18	
9	DUP	J1671-5811	19	
10	SAMPLE	J1672-5711	20	

Standard Type	Primary Book No. and Aliquot Vol.	Second Book No. and Aliquot Vol.	Third Book No. and Aliquot Vol.	Final Vol. of Standard
LMCS	12N14-A			
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WHC-SD-WM-DP-058 REV. 0-FB HW 924/94

. 34-400
DON'T SAY IT Write It!
DON'T SAY IT Write It! To Sata Handling FROM J. M. Fuye
Battet 3364
The DSC exotherms in these T-111 complex
were very broad The right end of the exother
was easily seen but the left end merce
was easily seen but the left end merged into the water endotherm. The plots
provided are optimized ploto the summe
provided are optimized plots. The suns
Jann M. Feyr 5-4/54
5-4/54
TO MAKE LIFE LAST PLIT SAFETY FIRST

WHC-SD-WM-DP-058

REV. 0-1/13

Jun 5/24/91

DIFFERENTIAL SCANNING CALORIMETRY ANALYSIS - UNDIGESTED SAMPLE

-	9	SUPER RI	JSH	Bilah	3364	
	Seral No	Sample Point 511 T111-C	Da [*]	5- 2-94	Time Issued	Priority 14
	Determination DEC	Melhod Standard LA-514-11	Result Units 3 % REC		Sarge Code 154D2	Aeruns O
	Sample Size 6. 680	ru/			ID	
	Femalis. Calculations. Judium 57	Results:	.9	5.3%	Recovery	
†	Roy No	39/2 Now	96	.0-95	Recovery	-
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-	mrs.	:	****3	H-3	1013	7
	5-2-94	Z3:00	7.m. J	en e E	2/2	
	Derenmination DISC Sample Size	5711 T111- Method Standard LA-514-1	033-32 *********	5- 2-8- HERMS	1 11:57 Crarge Code N54D2- Customer 15 1240-D-	Priority 14 Ferura
	Rev N.	251.Q	J/g s1	co Therm	<u>.</u>	
	Anatyst - 1	Analysi - 2	Analyst - 3	Analyst - 4	Anaiyst -	5
	Date	H's	H-3	H1·s		HITS
-	5-2-94	230b	am-1		M	
	BEST	AVAILA	ABLE (OPY	54-6E00-6	61 (A-10-22)

WHC-SD-WM-DP-058 REV. 0-16 JUN 1/24/41 DIFFERENTIAL SCANNING CALORIMETRY ANALYSIS - UNDIGESTED SAMPLE

		וסווס	LD L	WSF				
	Seral No	Salffpie Point		j P	14		Time issued	Priority
	J 16685		C33-		5- 2-9		11:57	14
	Determination DSC	LA-514-1	12	EXOTH		1 '	4D2	Returns
	Sample Size	TW-914-1	13	EVOIL	EM75		mer ID	0
	13	. 267 mg		·		1 - '	40-D-A	
	DUPLICATE							
		269.0J	19 "	rota	am			
	Rober We	ndiand.			-	-		
	Araiyat - 1	A-alyst - 2	Analys	: - 3	Analysi - 4	·	Aralyst - 5	
	mer \$	7.1		tar s	Hrs		Mrs	
	5-Z-93	Time Co-14:00	Les Un	11 J.	اسميه	Mi	3	
	5-2-94 B	2W	11		,		54-6800-061 .	A-10-231
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	J 198957	11 T111-			5-,2-9	4	12: 0	14
	<u>Garanthich</u>	Welfor Stancard		Result Units		Charge	e Cose	Reruns
	ISC Sample Sile	LA-514-1	13	EXOTH	ERMS	N5-		0
	1	12.596 mg				1	Ter (D	ļ
•	Remains Carcinistions					1 129	10-D-B	
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WHC-SD-WM-DP-058

REV. 0-XB KIND

STATE

DIFFERENTIAL SCANNING CALORIMETRY ANALYSIS - UNDIGESTED SAMPLE

J 16695	SUPER TILL	C33-S2	Date	5- 2-9	4 1	2: 0	14
DEC	LA-514-1		XOTHE	 RMS	N54		Reruns
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iemarks, Calculations					<u> </u>		
DUPLICATE	SAMPLE						. [
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		00	7				
KOEEY NE	والمنتال						
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	Time Completes	125 U21			0.		
<u>5-2-94</u>	12300	J. //	11. In		f/	54-6800-061	(A - 10 - 83)
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eursa T 1876 _8:	Sample Foint				- 1	te lissed	Priority
erenminasion	711 <u>1111-</u>	Ξe:	IUT UNITS	5 <u>- 2-</u> 9	Charge		14 Feruns
DSC Impersion	LA-514-1		XOTHER	<u>ws</u>	N54		0
erairs. Calculations,	5.897 mg				1240	0-D-C	
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Hrs.	ri's	bi ₁ ;		H-1		Mrs	
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5-Z-94	2300	9.71	7. Ju	<u></u>	1/4	9	
			1		,	54-6800-061	A . 10 . E 71

A-137

WHC-SD-WM-DP-058 Rev. 0-XB JW S/2YFY DIFFERENTIAL SCANNING CALORIMETRY ANALYSIS - UNDIGESTED SAMPLE

Serial No 3	111/11/11	$X \times XI$	ISH				7-
J 167058	Sample Point B11 T111-(C33-S		5- 2-9		ime Issued	14
Determination	Method Standard	T	Result Daris			e Cope	Retun
DSC	LA-514-12	13	EXOTH	ERMS		4D2	0
Sample Size	179 mg					10-D-C	
Bemails Calculations. DUPLICATE							
	187	./ (7/9	لاوم	2er	- 1	
Kossy h	FOLTON						
Analyst - 1	Analyst - 2	Analysi	- 3	Araiyst - 4		Analyst - 5	
	H-15		HIS	H		Pitt	
5-Z-94	Time Contrates	1.05 Un	77. 3		M	5	5
100-11	12000		-	Just 1	177	54-6500-061	. (A - 10 - E)
			ZIII.				
Serai No	CIIDE SALULIANE				1 7	me lissed	Property
J 1671.−57	,	33~5	2	5~ 2-9	ľ	2: 2	14
Determination	LA-514-11		esull Units		1	Cose	Resuns
しまださい。	DR 014-11		EXOTH:	15005	N5		0
DSC Sample Site					1	0-D-D	
Sample Size	696 mg				144	マーレーシー	
Sample Size					12~	<u> </u>	
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Sample Site	Résulta	91	g e	xo Hen	<u> </u>	<u> </u>	
Sample Site	Résulta	 I J (g e	xo Hen	<u> </u>	<u> </u>	
Sample Site	Résulta	 I J (g e	go Hu	<u> </u>	<u>0-0-9</u>	,
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Sample Site C. (Remarks, Calculations, R Analyst - 1	162. 7 NO AND Arahan-2	Analyst - I	3	,	<u> </u>		,
Sample Size C. () Remarks, Calculations, F	162. 7 NOVAND	Analyst - I	 _	,	<u> </u>		,
Sample Size C. () Fernancis, Calcillations, F Cossy Washington Cossi Washington Cossy Wa	MD AND Analysi - 2 Hrs Time Competed	Analyst - I	1478	Analyst - 4	<u> </u>	Analysi - S	,
Sample Size C. () Fernancis, Calcillations, F Cossy Washington Cossi Washington Cossy Wa	162. 7 ND AND Aralysi 2	Analyst - :	1478	Analyst - 4	<u> </u>	Analysi - S	,
Remarks, Carculations, F	MD AND Analysi - 2 Mrs Time Compeled 2300	4 nalysi - :	i J	Argiyst - 4	m Oh	Analysi - S	(-10-e3)
Remarch Carculations R Robby Washingtons Analyst - 1 Pris	MD AND Analysi - 2 Hrs Time Competed	4 nalysi - :	i J	Argiyst - 4	m Oh	Analysi - 5	1-10-231

WHC-SD-WM-DP-058 Rev. O-KB JAW 5/24/5U DIFFERENTIAL SCANNING CALORIHETRY ANALYSIS - UNDIGESTED SAMPLE

24 .54	SUPF	RRII	ึ่งบ				-
J 167158	Sample Point 11 T111-C		Date	5- 2-9		Time Issued 12: 2	14
Determination DSC	Method-Standard LA-514-11		ज पंत्रज्ञ COTHE	RMS		90 Cose 4D2	Reruns O
Sample Size	023 mg				I	40-D-D	
Remains, Calculations, F DUPLICATE	esults						
	175,	3 J/g		xo fla	m		
KORSY NA	Cim Class						
Anaryst - 1	Analysi - 2	Analyst - 3		Analyst • 4		Analyst - 5	
weg	m·s	H'S		Hrs		"	•
5-2-94	Time Completed Z300	97	7. 1	ine	/h		
			Ĺ	/	ν :		1 -A-10-E31
Serial No	CHĐEĐ						
J_1672571	1 T111-C3	3-82	5		1		Priority 14
	eines Sianzars LA-514-113	EXO			N54		Reruns
5. Fe 5 : e 7, 7	40 mg			F 1	124	.,,,,, 0-D-Ξ	
Remarks, Calculations, Res		 -				<u> </u>	
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ROBBY WEN	DUAND						,
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	2300 L	2.7M	Lu		A	1	
BEST	AVAIL	ABL	E (COP	Y	54-8800-061 (N-10-231

WHC-SD-WM-DP-058

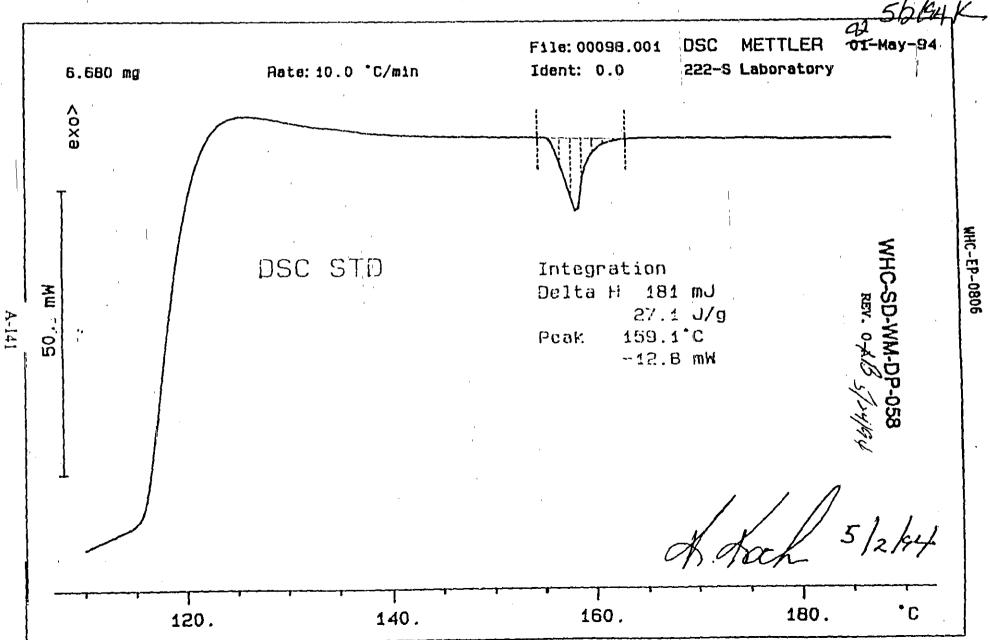
REV. 0-XB JUN Septent Sample

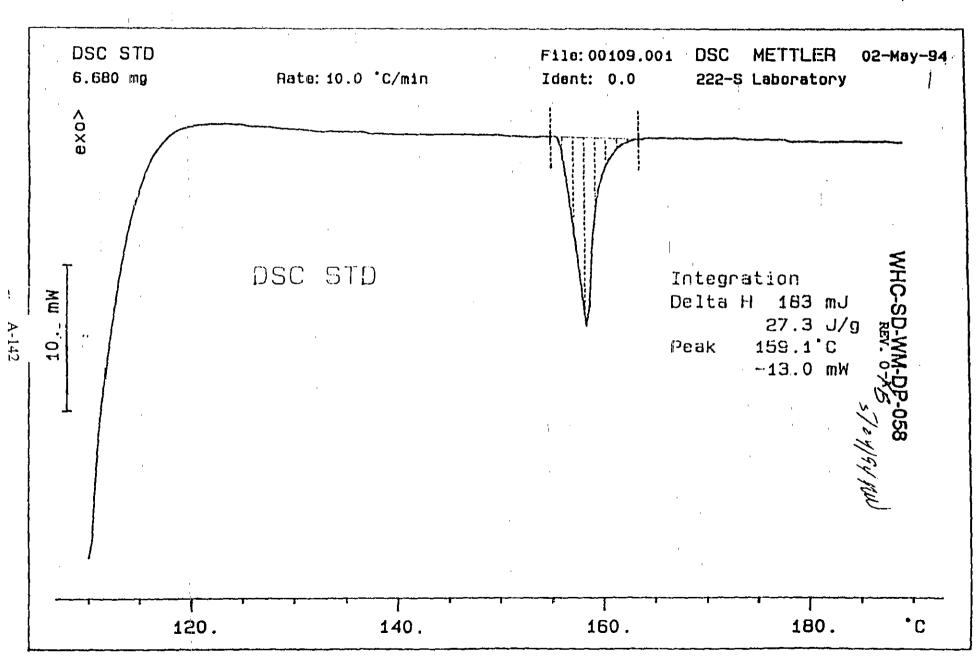
DIFFERENTIAL SCANNING CALORIMETRY ANALYSIS - UNDIGESTED SAMPLE

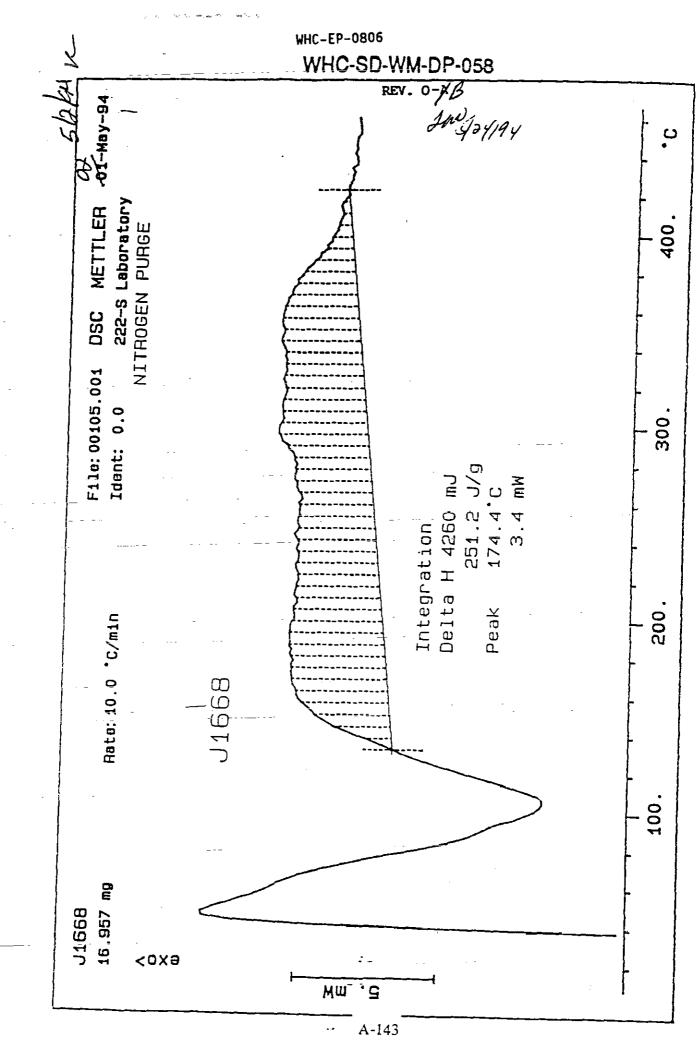
Serial No. J 16725		-C33-S2	5- 2-9	4 12: 3	Priority 14
DSC DSC	LA-514-		THERMS	Charge Code N54D2	Returns O
Sample Size	6.154 m	g-		Customer 10 1240-D-E	
Remarks, Carculation DUPLICATE					
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ROBBY (330,		axotherm	Analysi - 5	
Rosey 1			exotherm		

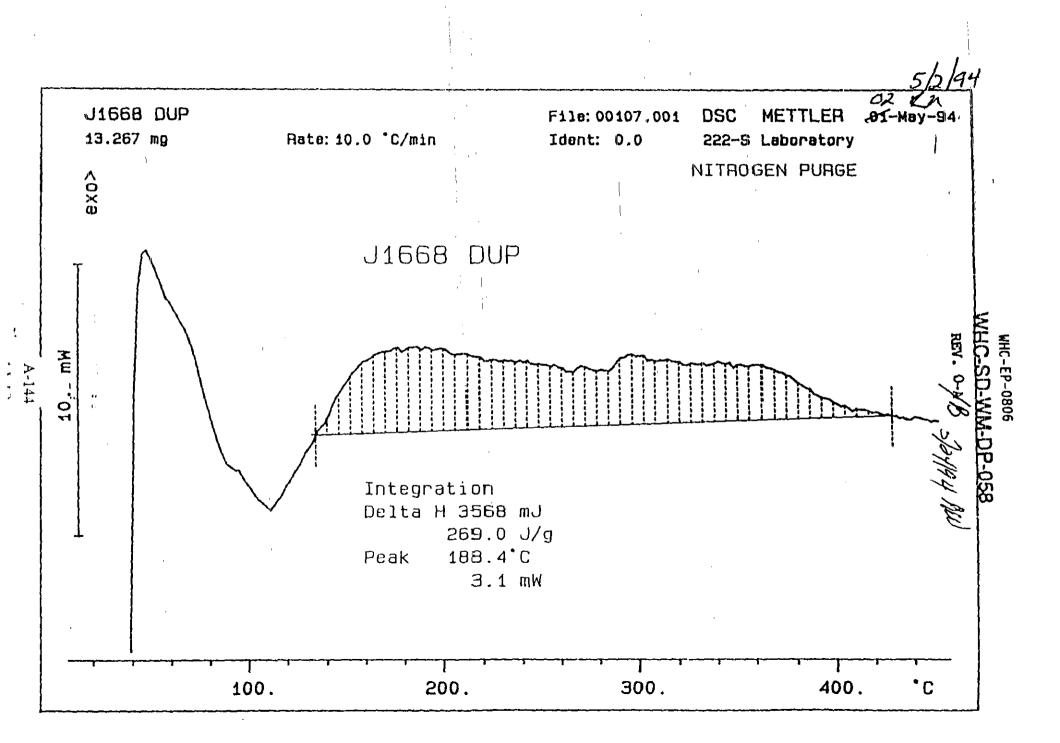
BEST AVAILABLE COPY

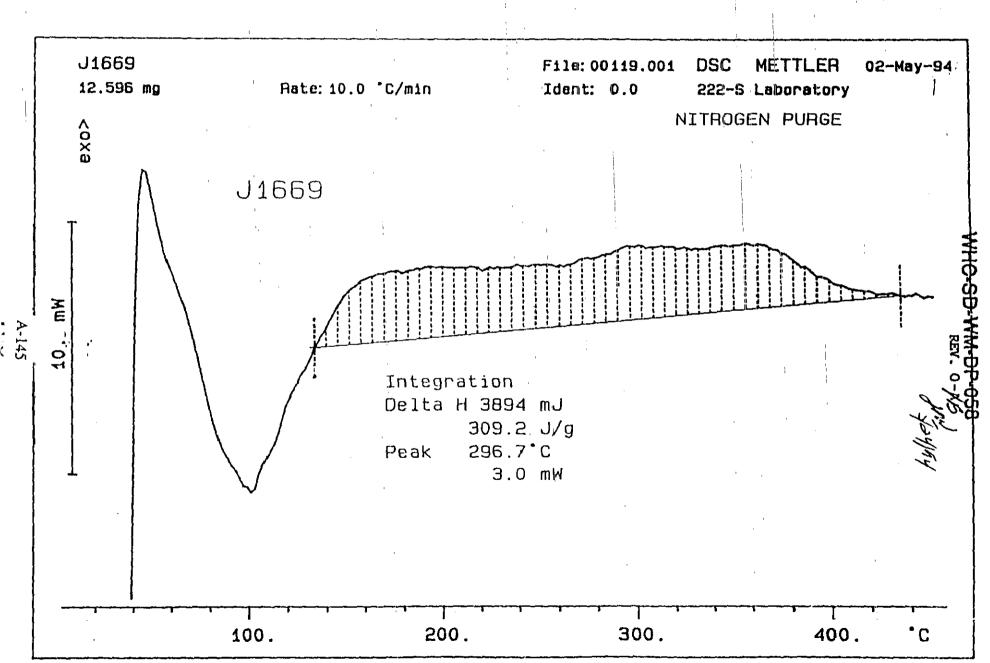
Signature below represents chemical technologist/chemist that completed/verified the calibration/analysis on pages _105 to 116.

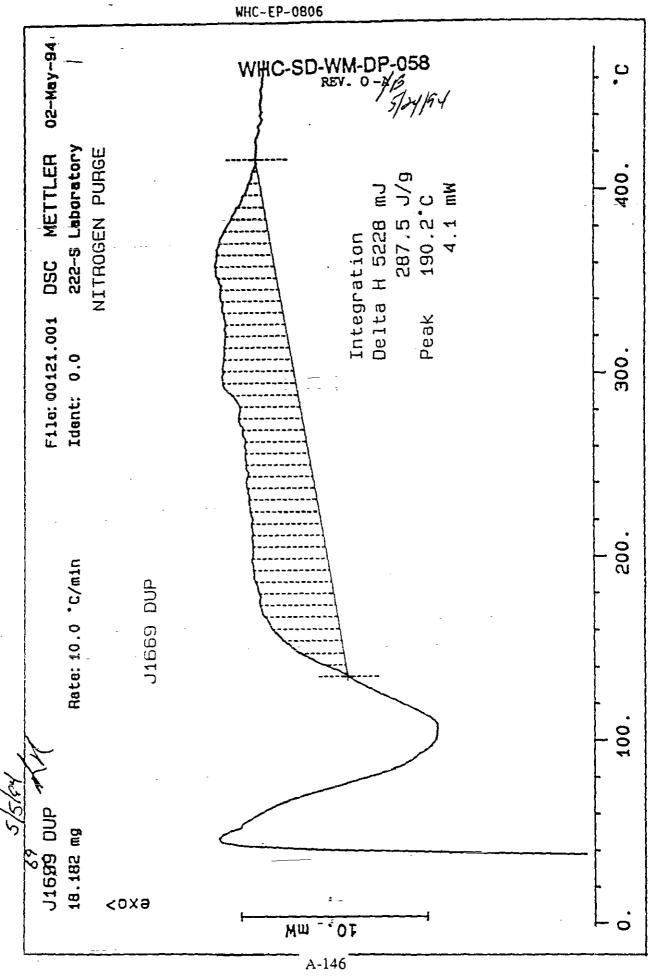


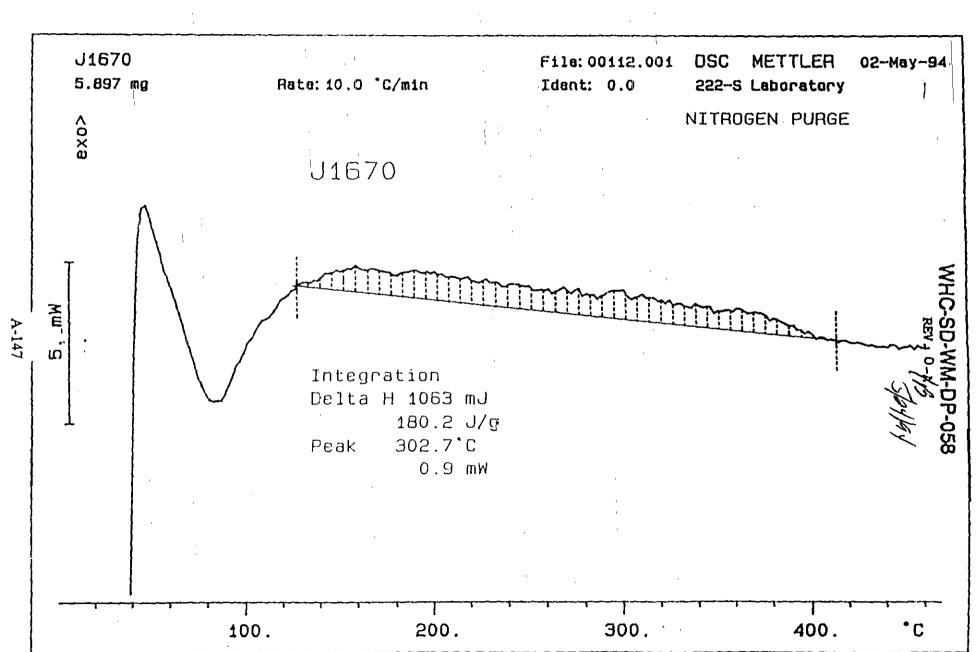


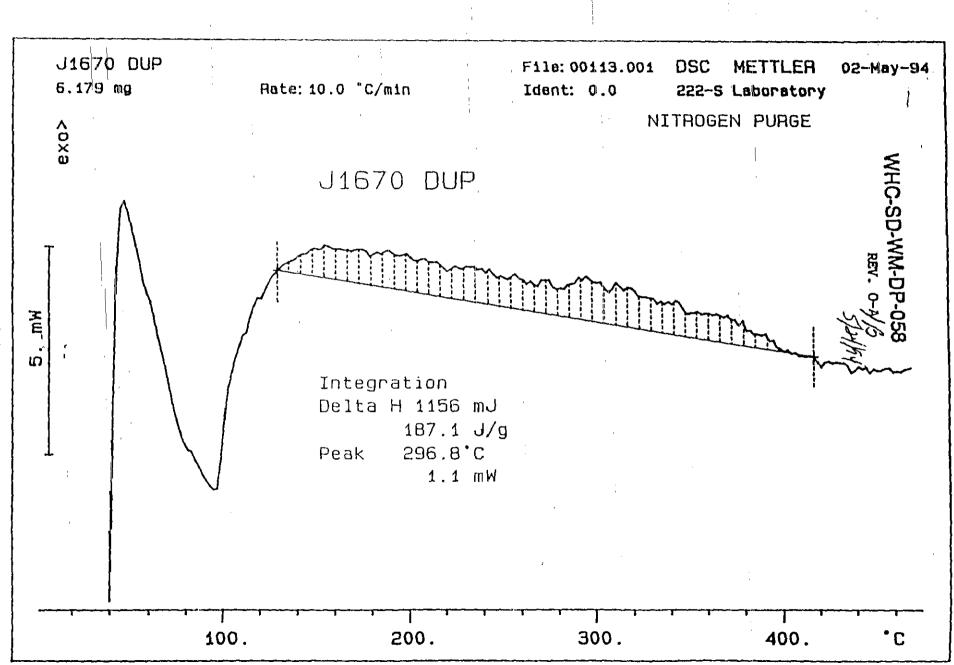


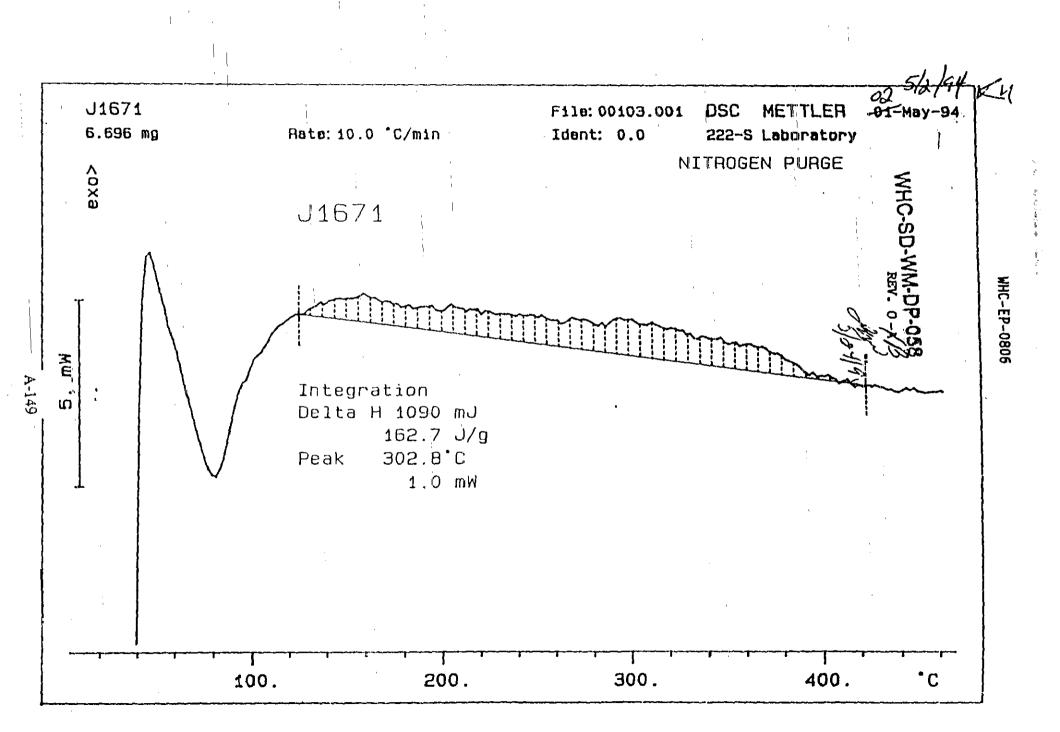


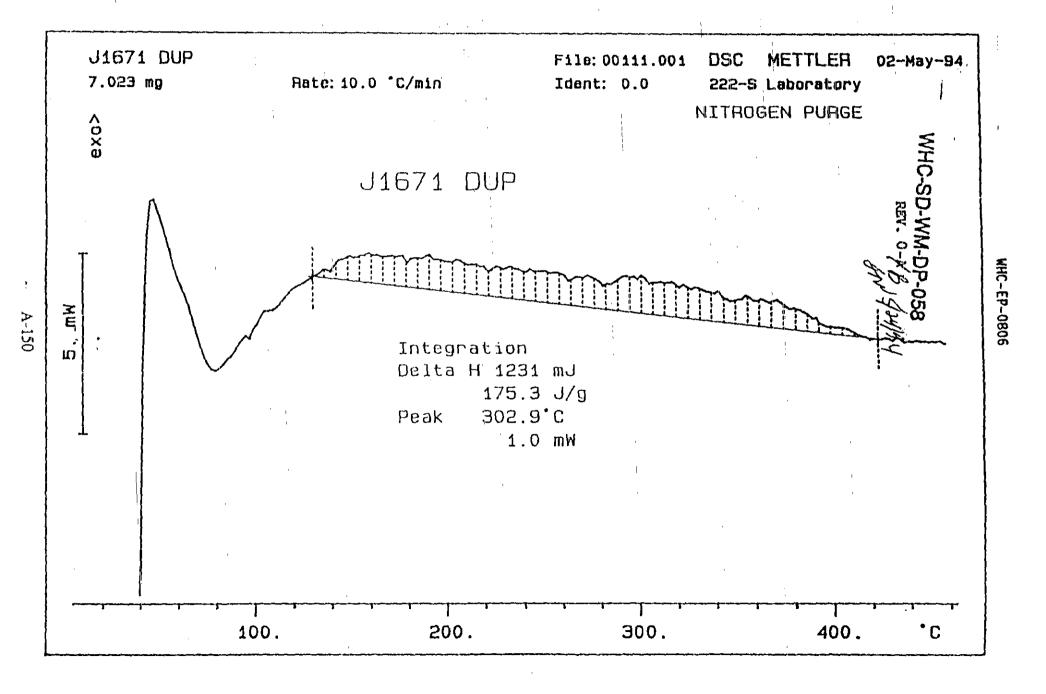


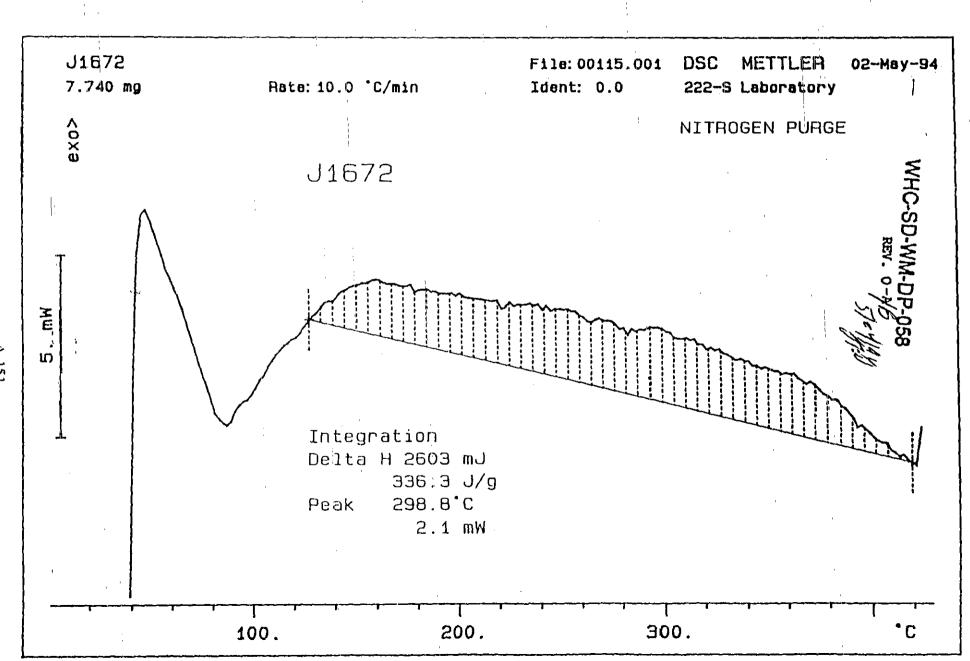


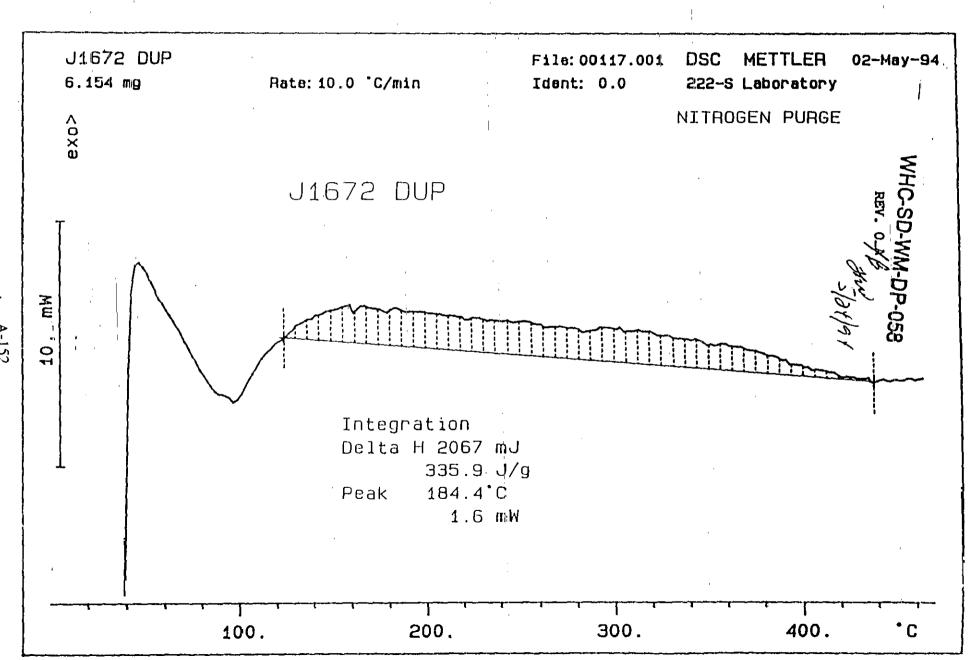












WESTINGHOUSE HANFORD COMPANY

WHC-SD-WM-DP-058 222-S LABORATORY

7 DUP

9 DUP

10 SAMPLE

8 SAMPLE

REV. 0- JB JW ANALYT	ICAL BATCH

Customer ID: 1240-D-A,B,C,D,E
Sample Prep: DIRECT
Procedure/ Rev: LA 560-112/A-1
Date: 05/03/94 56/94
Temperature N/A
Chemist: J. FRYE

	Description	Lab ID			Description	· Lab ID
1	LMCS	J1667-5512	11	DUP		J1672-5812
2	SAMPLE	J1668-5712	12	2		
3	DUP	J1668-5812	13	3	_	
4	SAMPLE	J1669-5712	14			
5	DUP	J1669-5812	15			
6	SAMPLE	J1670-5712	16			

17

18

19

20

J1670-5812

J1671-5712

J1671-5812

J1672-5712

Standard Type	Primary Book No. and-Aliquot-Vol.	Second Book No. and Aliquot Vol.	Third Book No. and Aliquot Vol.	Final Vol. of Standard
LMCS	24N8-A			·····
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		A 152		

WHC-SD-WM-DP-058

REV. 0-413 5/24/94

THERMAL GRAVIMETRIC ANALYSIS - UNDIGESTED SAMPLE

		-5512 T11			5- 2-9		1
į	Determination TGA	LA-560		Aesult Units		N54D2	Serur O
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1-2-94 1-6-957 1-6-957	2300 CIIDER 12 T111-0 LA-560-1: 548 29 FOR 10. 668	RU 333-5: 12 5:	7/7. Z	5- 2-9	Coste	12: 0 e Cere 1D2 mer 10 (0-D-B	14 Feruns
166957	2300 CIIDER Single Form 12 Till-0 IM-580-1: 548 mg Ferris 10.66%	RU 333-5: 12 5:	7/7. Z	5- 2-9 Analysi-4	Coste	12: 0 F COSTE 1D2 THE 10 10-D-B	14 Feruns
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WHC-SD-WM-DP-058

REV. 0-X S S J 4/4 4 S THERMAL GRAVIMETRIC ANALYSIS - UNDIGESTED SAMPLE

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J 16695		<u> </u>	SZ Result Un	5- 2-		12: 0	14
Determination	LA-560-1	10	% H2		N5-		0
TGA Sample Size	LA-380-1		A HZ			mer ID	
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19. Remarks, Calculation	s. Results				<u>.i</u> _		<u> </u>
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6.33 (-23) Calculations	// 35-, ENDLAND			Analysi - 4	1240		
C. 33 Remarks, Calculations. Robby L. Ralyst - 1	MINITING COTORIED		3		1240	Analyst - 5	
emarks, Calculations,	//. 35-, END AND Analysi 2	Analysi -	3		1240	Analyst - 5	

WHC-SD-WM-DP-058, REV. 0-4/5 JAN 3/34/64 THERMAL GRAVIMETRIC ANALYSIS - UNDIGESTED SAMPLE

Serial No			Date	5- 2-9	4	e issued 2: 1	14
J 167058	312 T111-	·C33-	SZ Result Units	0- 2-3	Charge		Returns
Determination TGA	LA-560-1	.12	% H2O		N54	02	0
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5-2-94	7500 2300	Lab	m -1			1/2	
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J 167157	12 T111-0	033-9	I .	5- 2-9		: 2	14
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Determination	Meinod Stancard		Result Unit			e Cose	Returns
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Senal No	Sample Point	<u> </u>)		Tin	re issued	Priority
J 1672	-5712 T111=	033-8	,2	5- 2-9	4 1	2: 3	14
1	Method Standard		Result Units % H20		Charge N54		Beruhs
Derermination	114-560-1		76 1:20		Custom	_	
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Derermination TGA Sample Size	101 mg	- , !				D-D-E	
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Semental Ch TGA Sample Size 7. Sements Calculate	101 Mg	· ·		Analyst - 4			
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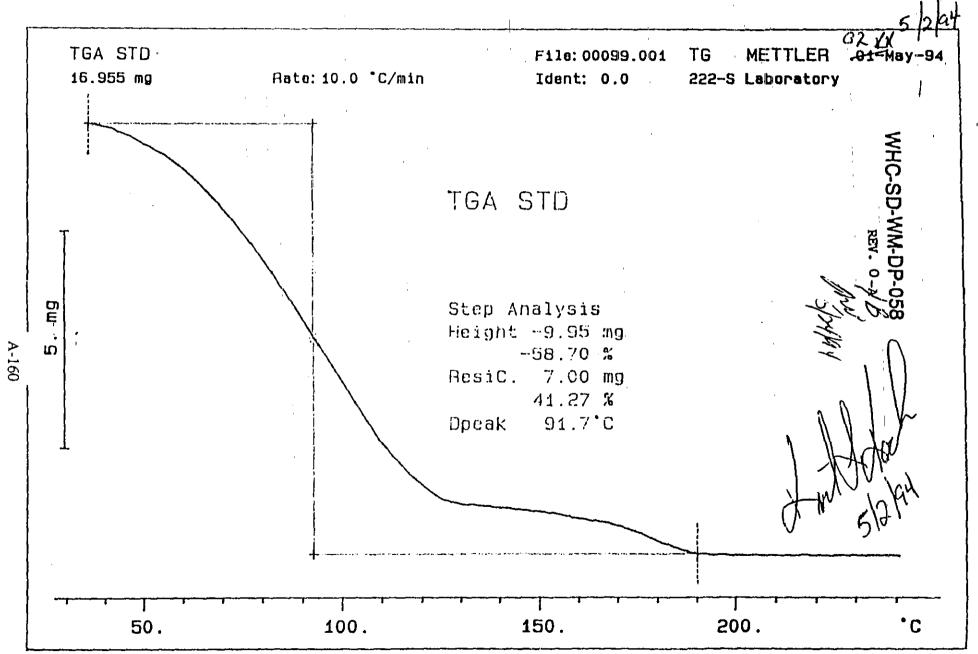
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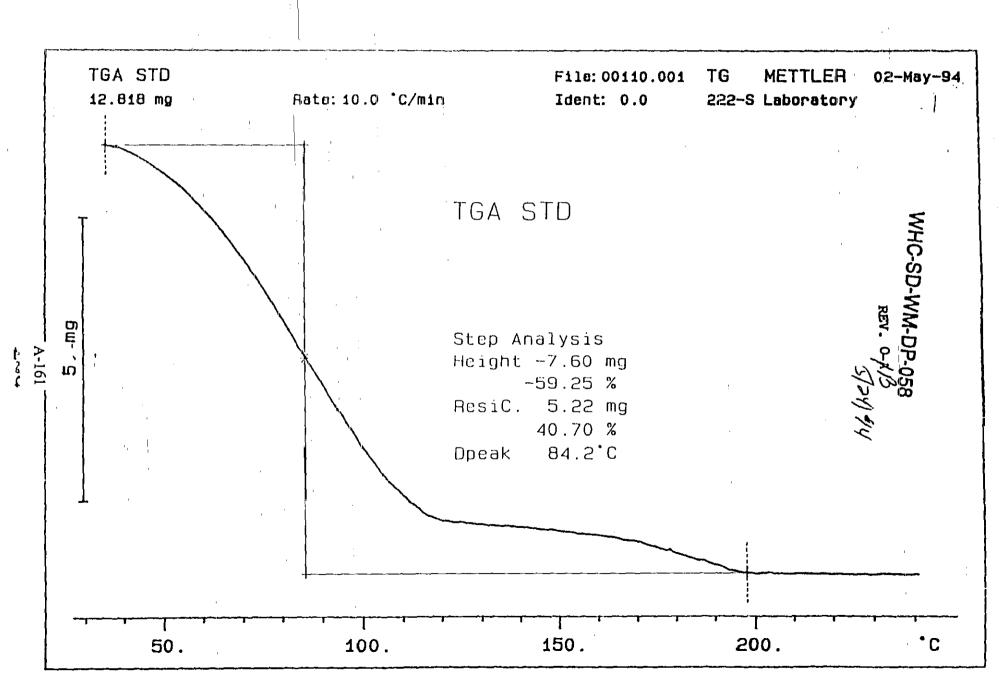
REV. 0-16 STORY THERMAL GRAVINETRIC ANALYSIS - UNDIGESTED SAMPLE

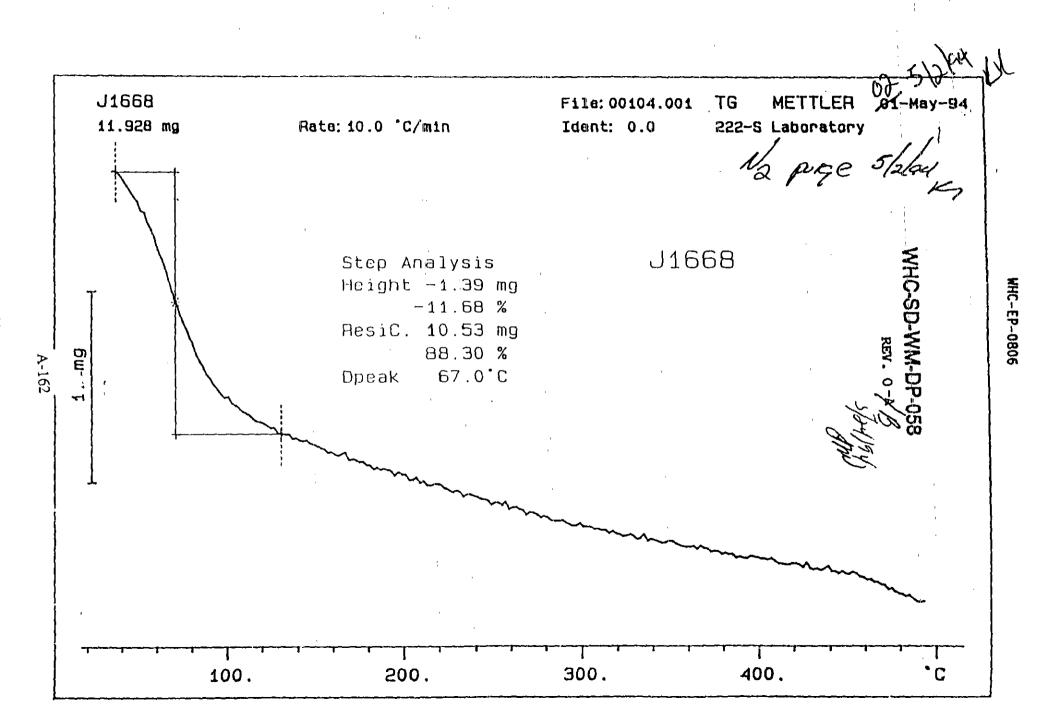
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Determination.	Memod Stancard	Result L	PKS	Char	ge Cope	Returns
TGA	LA-560-11	2 % 光	20	N5	4D2	0
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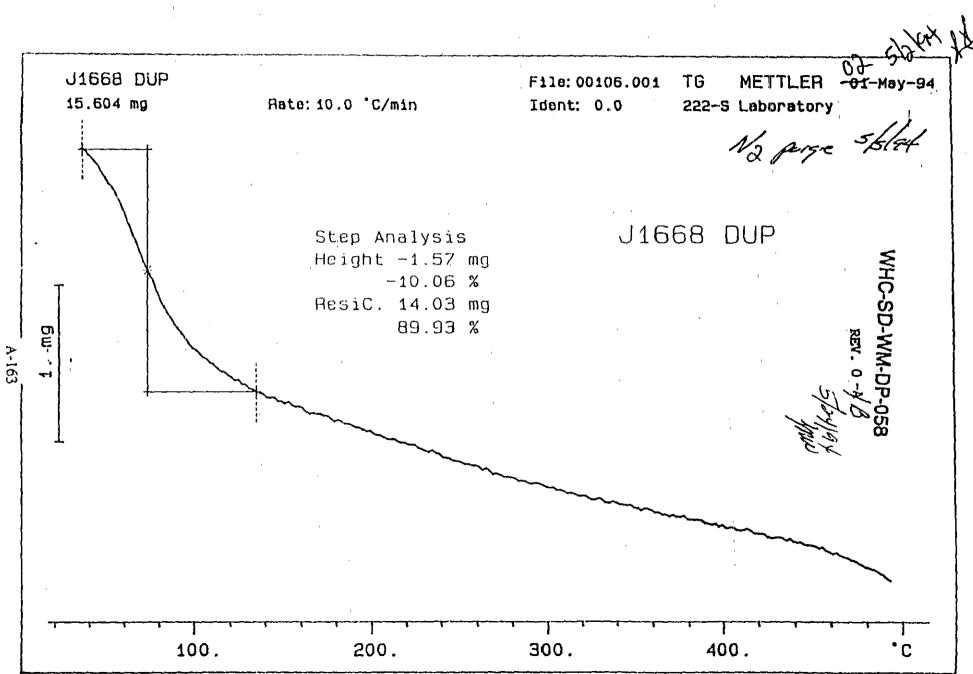
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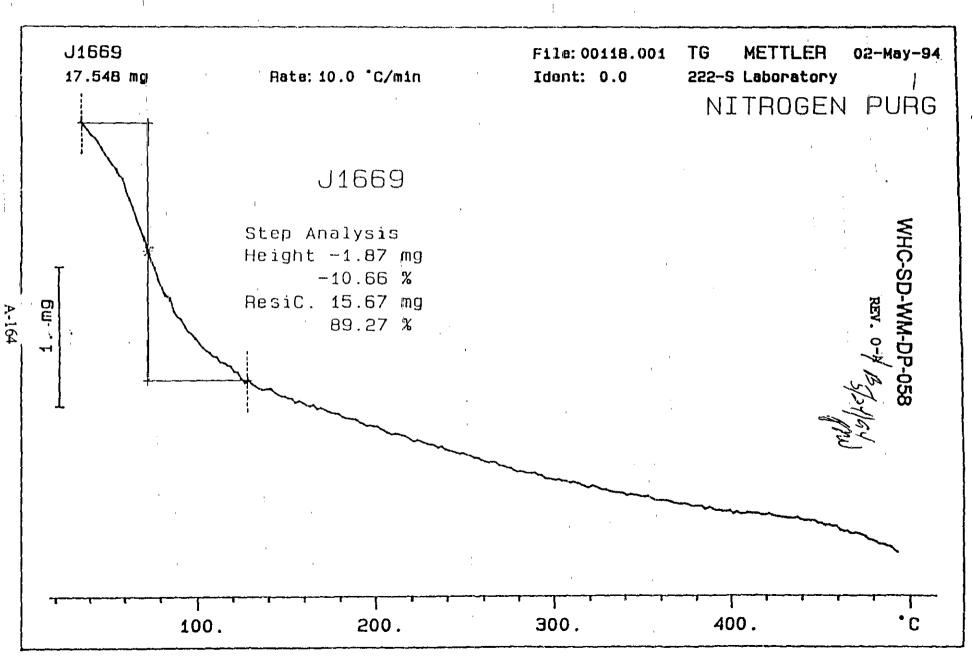
Signature below represents chemical Technologist/chemist that completed/verified the calibration/analysis on page 124 to 135

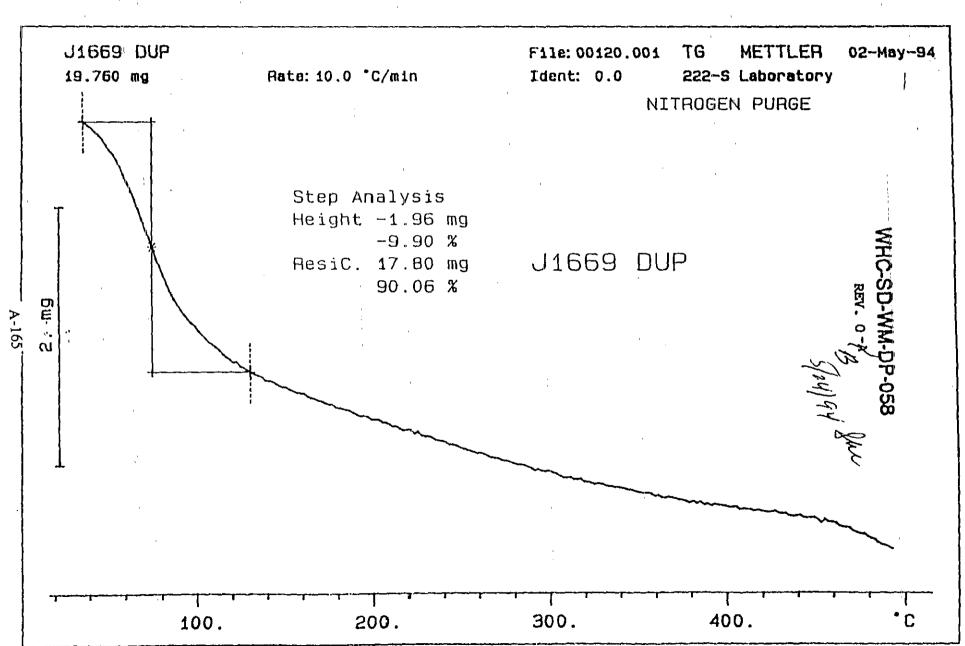




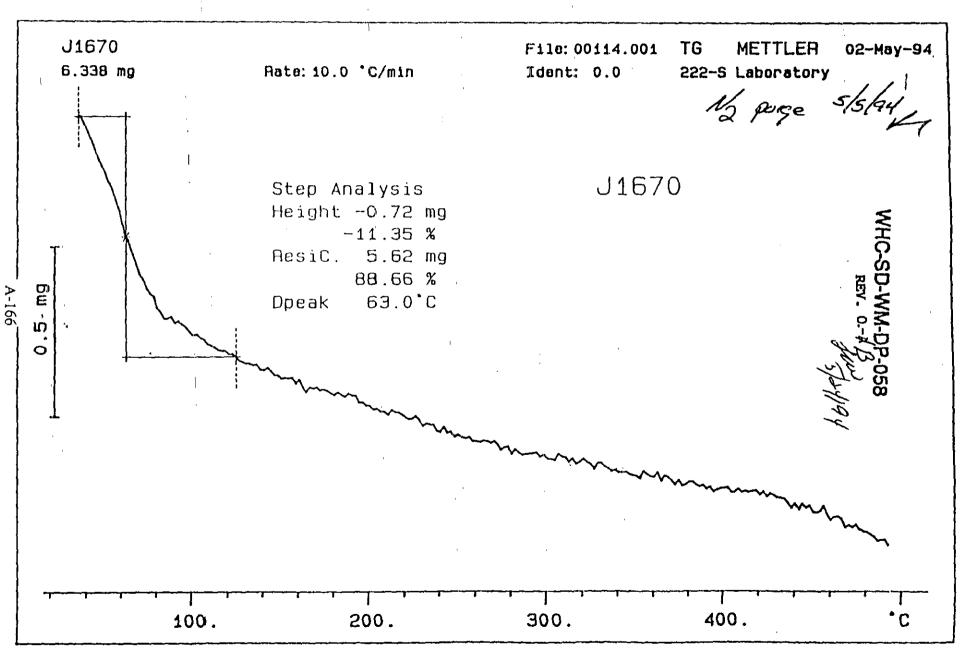


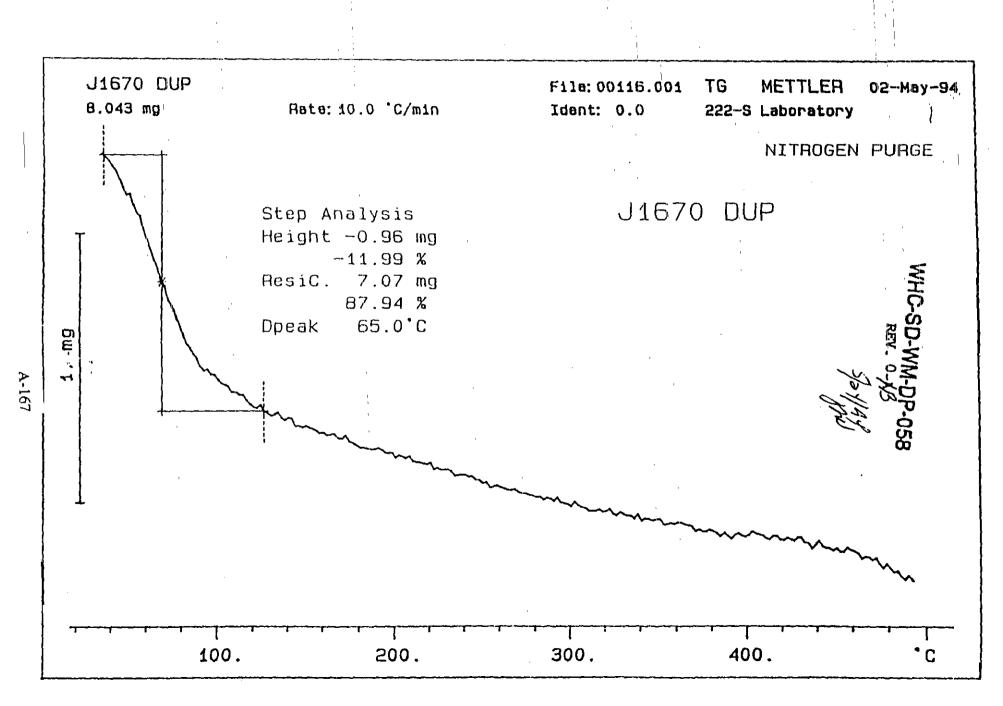


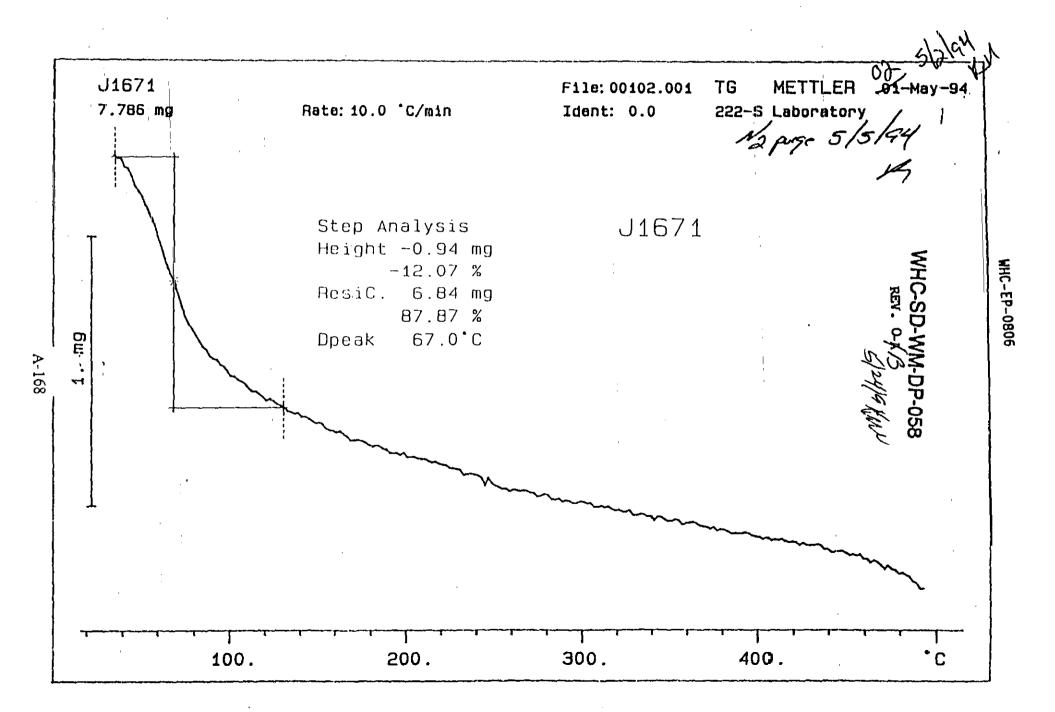


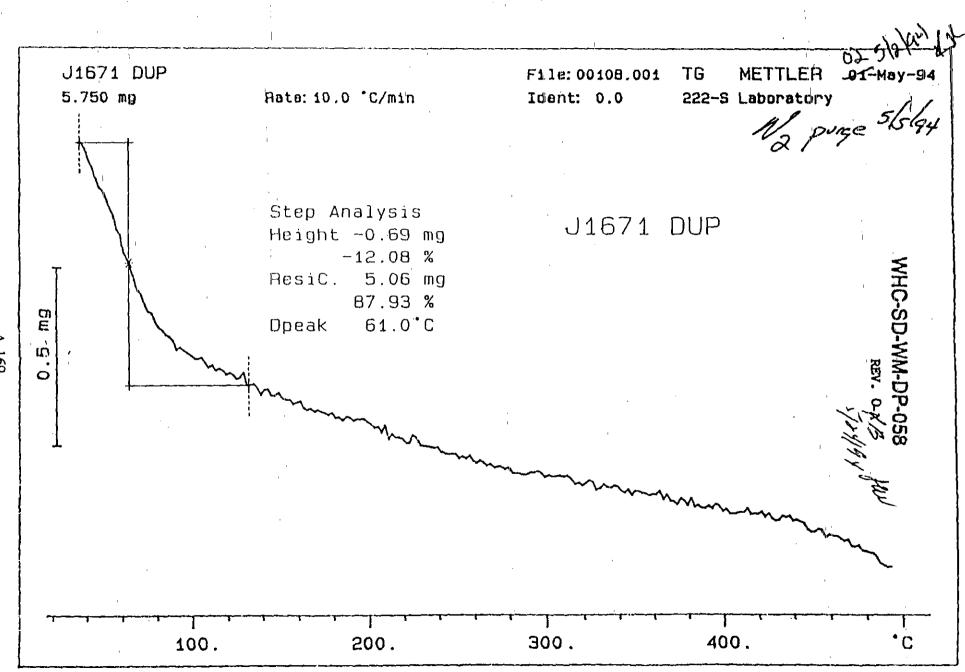


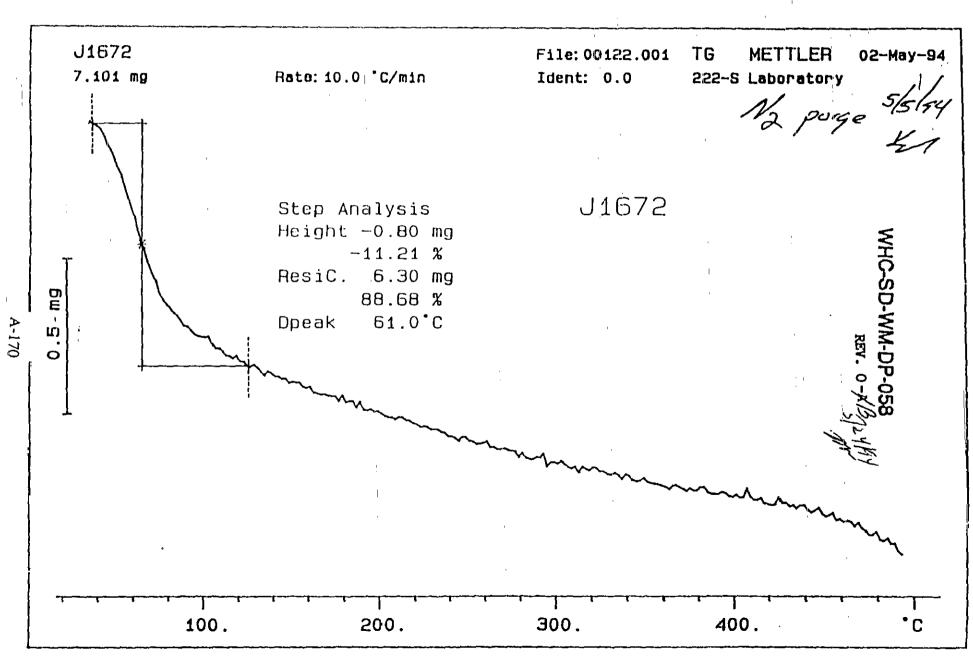


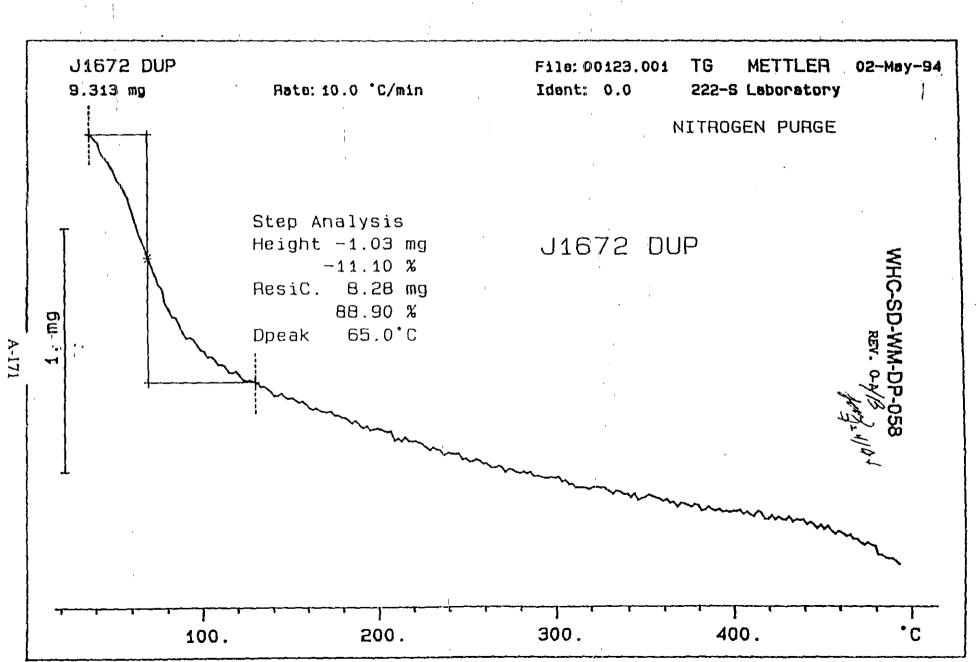












Appendix B: Statistical Interpretation

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------------------SUPPORTING DOCUMENT

1. Total Pages 61

2. Title 3. Number 4. Rev No. Statistical Characterization Report for WHC-SD-WM-TI-650 0 Single-Shell Tank 241-T-Ill

5. Key Words

Single-Shell Tank, Waste, Characterization, Sampling, Sample Homogenization, Core Composite Samples

6. Author

Name: R. D. Cromar/S.R. Wilmarth

Name: Louis Jensen

APPROVED FOR % PUBLIC RELEASE

Organization/Charge Code

8E100/N4D2G

5/20194

This report documents the statistical analyses performed on core data from Single-Shell Tank 241-T-111

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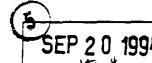


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9. Impact Level

NA



ENGINEERING DATA TRANSMITTAL

Page 1 of _/ 1. EDT 600493

2. To: (Receiving Organization) Distribution	3. From: (Originating Organization) Process Laboratories & Technology	4. Related EDT No.: 600490		
5. Proj./Prog./Dept./Div.: WM	6. Cog. Engr.: L. Jensen	7. Purchase Order No.: NA		
8. Originator Remarks: Release of document "Statis	9. Equip./Component No.: NA			
Single-Shell Tank 241-T-111		10. System/Bldg./Facility: NA		
11. Receiver Remarks:		12. Major Assm. Dwg. No.: NA		
		13. Permit/Permit Application No.: NA		
_		14. Required Response Date: 9-30-94		

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2-1.	Riser	location		• •													•	2-2
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LIST OF TERMS

222-S	Process and Analytical Laboratories
95% LL	lower limit of a 95% confidence interval on the mean
95% UL	upper limit of a 95% confidence interval on the mean
ANOVA	analysis of variance
T111	
CI	-confidence interval
df	degrees of freedom
.dir	direct
DL	detection limit
GEA	gamma_energy analysis
IC	ion chromatography
IC.w.	
ICP	inductively coupled plasma analysis
ICP.a.	inductively coupled plasma analysis of an acid digested sample
ICP.f.	inductively coupled plasma analysis of a fusion digested sample
	inductively coupled plasma analysis of a water leached sample
.w.Spec	spectroscopic method of a water leached sample
KOH/Ni	Potassium Hydroxide/Nickel
Mean/DL	the mean of two sample results (sample and duplicate) divided by
1	the detection limit
ML NA	milliliters
	not available
REML	radiochemistry analysis of a fusion digested sample
SST	restricted maximum likelihood estimation method single-shell tank
	thermal gravimetric analysis
_ <u>cu</u> n	weighted mean of the concentration data
ÿ ∂²(ÿ) μCi/I	estimated variance of \bar{y}
μCi/L	microcuries per liter
μg/g	micrograms per gram
μCi/g	microcuries per gram

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STATISTICAL CHARACTERIZATION REPORT FOR SINGLE-SHELL TANK 241-T-111

1.0 SUMMARY

This report contains the results of the statistical analysis of data from two core samples obtained from Single-Shell Tank 241-T-111 (T111).

Section 2 contains a description of the core samples and the chemical analyses done on the core samples.

Section 3 contains_mean_concentration-estimates_and_associated_95% confidence intervals (CI) on the mean for each of the analytes in Tlll.

Section 4 contains estimates of the spatial variability (variability between cores) and estimates of the "analytical error" from the core composite data.

Two types of analytical error were estimated from the core composite data: (1) compositing variability (variability between composite samples within the same core) and (2) analytical measurement error (variability between the primary and duplicate analyses within each core composite sample). Estimates of the analytical measurement error were used as the reference value to test the significance of the spatial and compositing variability. Spatial variability was significantly different than zero for 39 out of 85 analytes in T111. The compositing variance was significantly different than zero for 39 out of the 85 analytes.

Significantly different than zero at the $\alpha = 0.05$ level.

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2.0 INTRODUCTION

Two core samples (cores 31 and 33) were taken from T111. Cores 31 and 33 were obtained form risers 6 and 3, respectively. Figure 2-1 (p. 2-2) gives the locations of the risers. The segment recoveries for each core are given in Table 2-1. Core 31 had nearly 100% segment recovery for all segments except segments 1 and 6. There was nearly 100% segment recovery for all of the segments in Core 33.

Table 2-1. T-111 Core Recoveries.

Core	Segment Number									
	1	2	3	4	5	6	7	8	9	
31	27%	80- 100%	95- 100%	80- 100%	100%	0%	90- 100%	100%	100%	
33	100%	100%	87- 100%	75- 85%	88%	100%	100%	100%	100%	

Due to the incomplete core sample recovery, the concentration estimates given in this report are biased. The magnitude of the bias is unknown.

Two core composite samples were made for each core sample from the homogenized solid segment waste. Primary and duplicate results were obtained from each core composite sample.

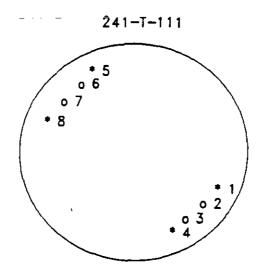
The chemical analysis of the primary and duplicate samples consisted of the following analytical methods and sample preparations (the notation for the method and preparation is also listed):

- inductively coupled plasma analysis (ICP) of an acid digested sample (ICP.a),
 - ICP of a potassium hydroxide/nickel (KOH/Ni) fused sample, (ICP.f)
 - ICP analysis of a water leached sample (ICP.w),
 - ion chromatography (IC) of a water leached sample (IC.w),
 - radio-chemistry, and
 - gamma energy analysis (GEA).

Occasionally, special <u>analytical methods are used on prepared</u> and <u>unprepared</u> waste samples. The notation for the results from such methods is given in the tables of Appendix A and the tables in Sections 3 and 4.

Whenever possible, the results and data given here are identified by the analysis method, the type of dissolution and analyte. For example, the notation ICP.a.Al refers to the Al concentration from an ICP analysis of an acid digested sample.

Figure 2-1. Riser location.



- o 12" Riser

 * 4" Riser

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The core composite sample results are contained in Table A-1 (Appendix A). Ratios of the mean of each primary and duplicate result divided by the detection limit for that pair are also included. The ratios (Mean/DL) are provided to show how large the analyte concentrations are relative to the DL. The data packages for tank Tlll (Kocher, 1993 and McKinney, 1993) contain a complete report of the sample results along with the laboratory quality control data. The core composite data for each analyte are illustrated in the figures provided in Appendix B.

Analysis of variance (ANOVA) models were fit to the data for all analytes that did not have any "less than values". All of the data available for statistical analysis, including the analytes with at least one "less than value," are reported in Table A-1. The ratios (Mean/DL) of all of the data are also reported in Table A-1.

From a quality of data perspective, it is desirable that the ANOVA models be applied to analyte concentrations greater than 10 times the detection limit. However, Analytical Evaluation and Reporting personnel, within the TWRS Information Management Systems, identified a list of "special analytes," given in Table 2-2. For this list, they requested that the ANOVA models be fit to the data provided the concentration values were greater than 3 times the detection limit or less than 10 times the detection limit.

Whenever possible the ANOVA models were fit using all of the data, regardless of the value of the detection limit. To warn the reader of potential detection limit problems the analytes in Tables 3-1, 4-1 and A-1 are denoted by a "•" or a "*" whenever a concentration is less than 3 times or 10 times the detection limit respectively. The results for such analytes are to be used with caution.

In Table A-1, some analytes are marked with a "*." Due to the dilution factors, the analytical result values are so close to the detection limit (DL), that the concentration values are questionable. For these analytes, ANOVA models were not fit to the data.

Table 2-2. Special Analyte List.

	
Aluminum	Nitrate
Bismuth	Nitrite
Calcium	Phosphate
Chromium	Carbonate
Iron	Fluoride
Silicon	Chloride
Sodium	Total Organic Carbon
Zirconium	Cyanide

A close examination of the figures in Appendix B reveals several "outlier data points" in the core composite data:

• ICP.a.Co The primary result for core 31, composite 2 of 11.7 μ g/g is over three times as large as the duplicate result. The other results for ICP.a.Co all fall in the range 2.7 to 3.8 μ g/g. The detection limit for ICP.a.Co is 0.8 μ g/g.

- ICP.a.Cu The duplicate result for core 31, composite 2 of 127 $\mu g/g$ is about four times as large as the primary result of 31.7 $\mu g/g$. The detection limit for ICP.a.Cu is 0.4 $\mu g/g$.
- NO₂ Both the primary and duplicate results from core 31, composite 2 are about one half the results for core 31, composite 1. The average of the results for core 31, composite 1, is 952 μ g/g, the average for composite two is 525 μ g/g. The detection limit for NO₂ (by water digestion spectrophometric analysis) is 50 μ g/g.

There is no direct evidence that the results noted above are due to analytical measurement errors. Consequently, the statistical analysis was performed on the data as it is reported in Table A-1. The potential anomalies in the data noted above can be clearly seen in the figures in Appendix B.

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3.0 MEAN CONCENTRATION ESTIMATES

One of the tasks outlined in the Tank Waste Characterization Plan, Section 5.1.1.2 (Bell, 1993), is to estimate the constituent inventories in the waste. The inventories are estimated by computing mean concentrations and 95% confidence intervals (CIs) on the mean concentrations for each constituent. The estimate of the inventory and CI on the inventory of an analyte in the tank are equal to the corresponding mean concentration estimates and CI multiplied by the volume of waste in the tank. Therefore, estimates of tank inventory are not given in this document.

Table A-1 (Appendix \hat{A}) contains the core composite data used to compute the mean concentration estimates and associated CIs.

3.1 STATISTICAL METHODS

The concentration estimates are given in the form of 95% CIs on the mean concentration. It is assumed that each primary sample and its duplicate are analyzed independently of one another. The two analytical results are used to estimate the analytical measurement error. Due to the hierarchical structure of the data, the analytical measurement error alone is not the appropriate error term to use in computing the CIs. A linear combination of the analytical measurement variance and the spatial variance is the appropriate variance of the mean for the CIs (Snedecor and Cochran, 1980). Appendix C contains a description of the statistical model and formulas used to calculate estimates of the mean, variance of the mean and the CI on the mean.

3.2 STATISTICAL RESULTS

Table 3-1 contains the summary statistics, by analyte, for ICP acid digestion, ICP water leach, ICP KOH/Ni fusion dissolution, radio chemistry and IC analyses. The summary statistics are defined as follows:

 \bar{y} mean of the concentration data, $\hat{\sigma}^2(\bar{y})$ estimated variance of \bar{y} .

df degrees of freedom,

95% LL ---- lower limit to the 95% CI on the mean and

95% UL upper limit to the 95% CI on the mean.

For some analytes the lower confidence limit (95% LL) was negative.

Since concentrations are greater than or equal to zero, any negative 95% LL

values were set equal to zero.

The confidence intervals in Table 3-1 are wide relative to the range of the data because only two cores were used to estimate the <u>spatial variability</u>. Two core samples is the <u>minimum</u> number of core samples needed to estimate a tank's spatial variability.

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Table 3-1. Concentration Estimate Statistics (Units μ g/g Except Radionuclides μ Ci/g). (sheet 1 of 3)

	Radionuclide	$s \mu (1/g)$. (sheet	1 of 3)	
Analyte	ý	δ ² (ÿ)	df	95% LL	95% UL
ICP.a.Ag	1.26E+02	7.86E+03	1_1_	0.00	1.25E+03
ICP-a.Al	5.41E+02	1.06E+04	1_1_	0.00	1.85E+03
1CP.a.8	2.80E+01	7.56E+00	1	0.00	6.30E+01
ICP.a.Ba	6.90E+01	6.46E+01	1	0.00	1.71E+02
ICP.a.Bi	2.59E+04	6.38E+06	1	0.00	5_80E+04
ICP.a.Ça	1.88E+03	2.12E+05	1	0.00	7.725+03
ICP.a.Cd+	5.80E+00	3.03E+00	1	0.00	2.79E+01
ICP.a.Ce•	3.37E+01	9.69E+00	1	0.00	7.33E+01
ICP.a.Co+	4.30E+00	1-63E+00	1	0.00	2.05E+01
ICP.a.Cr	1.98E+03	1.63E+04	1	3.57E+02	3.60E+03
ICP.a.Cu	3.35E+01	3.53E+02	1	0.00	2.72E+02
ICP.a.Fe	1.85E+04	1.21E+06	1	4.55E+03	3.25E+04
ICP.a.K	1.14E+03	2.24E+03	1	5.34E+02	1.74E+03
ICP.a.La	4.22E+03	3.00E+05	1_1_	0.00	1.12E+04
ICP.a.Mg	_ 3.77E+02	6:36E+03	1	0.00	1.39E+03
ICP.a.Mn	6.33E+03	2.68E+04	1	4.25E+03	8.41E+03
ICP.a.Na	3.69E+04	1.56E+06	1	2.10E+04	5.27E+04
ICP.a.Ni	1.32E+02	5.12E+02	1	0.00	4.19E+02
ICP.a.P	1.035+04	1.21E+05	1	5.90E+03	1.47E+04
ICP.a.Pb	3.47E+02	2.64E+04	1	0.00	2.41E+03
ICP.a.S	1.21E+03	1.06E+03	1_1_	8.00E+02	1.63E+03
ICP.a.Sb.	3.14E+01	1.65E+01	1	0.00	8.30E+01
ICP.a.Si	4.69E+02	9.17E+02	1	8.40E+01	8.54E+02
ICP.a.Sr	3.00E+02	3.75E+02	1 1	5.39E+01	5.46E+02
ICP.a.Ti	1.958+01	1.39E+02		0.00	1.69E+02
ICP.a.V	1.45E+01	6.58E+00	1	0.00	4.71E+01
lĉP.a.Zn	6.50E+01	6.46E+02	1	0.00	3.88E+02
ICP.f.Ag	1.286+02	8.05E+03	1	0.00	1.27E+03
ICP.f.Al	5.70E+02	9.70E+03	_ 1		1.82Ē+Ū3
ICP.f.Ba	6.46E+01	2.45E+01	1	1.73E+00	1.28E+02
ICP.f.Bi	2.36E+04	9.08E+06	1	0.00	6-18E+04
ICP.f.Ca+	2.42E+03	8.27E+04	1	0.00	6.07E+03
ICP.f.Cd+	8.12E+00	1.765+00	1	0.00	2.50E+01
ICP.f.Co+	1.15E+01	1-16E+00	1	0.00	2.52E+01
ICP.f.Cr	1.80E+03	1.56E+03	1	1.30E+03	2.30E+03
e: Analytes with a					

Analytes with a portion of the data below 3 times the DL.
 Analytes with a portion of the data below 10 times the DL.

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Table 3-1. Concentration Estimate Statistics (Units μ g/g Except Radionuclides μ Ci/g). (sheet 2 of 3)

Analyte ICP.f.Cus ICP.f.Fe ICP.f.La ICP.f.Mg ICP.f.Mn ICP.f.Na	2.39E+01 1.80E+04 4.11E+03 3.55E+02 6.28E+03 3.70E+04 8.14E+03	3.56E+01 4.05E+06 3.08E+05 7.31E+03 1.88E+04 6.00E+06	df 1 1 1 1 1 1 1 1	95% LL 0.00 0.00 0.00	95% UL 1.05E+02 4.36E+04 1.12E+04 1.44E+03
ICP.f.Fe ICP.f.Mg ICP.f.Mn ICP.f.Na ICP.f.Ni	1.80E+04 4.11E+03 3.55E+02 6.28E+03 3.70E+04	4.05E+06 3.08E+05 7.31E+03 1.88E+04	1 1	0.00 0.00 0.00	4.36E+04 1.12E+04
ICP.f.Ng ICP.f.Nn ICP.f.Na ICP.f.Ni	4.11E+03 3.55E+02 6.28E+03 3.70E+04	3.08E+05 7.31E+03 1.88E+04	1	0.00	1.12E+04
ICP.f.Mg ICP.f.Mn ICP.f.Na ICP.f.Ni	3.55E+02 6.28E+03 3.70E+04	7.31E+03 1.88E+04	1	0.00	
ICP.f.Na ICP.f.Ni	6,28E+03 3,70E+04	1.88E+04			1.44E+03
ICP.f.Ni	3.70E+04		1 1	ļ . .	
ICP.f.Ni		6.00E+06		4.54E+03	8.02E+03
	8.14E+03		1_1_	5.82E+03	6.81E+04
R I		6.41E+06	1	0.00	4.03E+04
ICP.f.P	1_04E+04	8.42E+05	1	0.00	2.21E+04
ICP.f.Pb+	3.65E+02	9.38E+03	1	0.00	1.60E+03
ICP.f.S	1.23E+03	1.13E+04	1	0.00	2.58E+03
ICP.f.Si	5.67E+03	5.41E+04	1	2.71E+03	8.62E+03
ICP.f.Sr	2.98E+02	6-24E+01	1	1.97E+02	3-98E+02
ICP.f.Ti	4.79E+01	6.09E+02	1	0.00	3.62E+02
ICP.f.V+	1.47E+01	8.32E-01	1	3.06E+00	2.62E+01
ICP.f.2n+	1.06E+02	7.17E+00	1	7.2 <u>2E</u> +01	1.40E+02
ICP.w.Al+	1.096+01	5.75E+00	1	0.00	4.14E+01
ICP.W.B*	4.07E+00	6.77E-01	1	0.00	1.45E+01
ICP.w.Bi	2.02E+02	2.40E+03	1	0.00	8.245+02
ICP.w.Ca+	6.16E+01	3.32E+01	1	0.00	1.35E+02
ICP.w.Cr	2.18E+02	2.45E+01	1	1.55E+02	2.81E+02
ICP.w.Fe	1 28E+02	3.11E+02	1	0.00	3.52E+02
ICP.W.K	7.19E+02	1.54E+03	1	2.21E+02	1.22E+03
ICP.W.La+	1.10E+01	1.42E+01	1_1_	0.00	5.89E+01
ICP.w.Mg+	3.64E+00	5.45E-02	1	6.75E-01	6.61E+00
ICP.w.Mn	2.47E+01	2.36E+01	1_1_	0.00	8.65E+01
ICP_W.Na	3.30E+04	2.44E+06	1	1.31E+04	5.28E+04
ICP.w.P	5.68E+03	3.24E+04	1	3.39E+03	7.97E+03
ICP.w.s	1.15E+03	2.38E+03	1	5.29E+02	1.77E+03
ICP.w.Si	5.72E+02	5.35E+03	1	0.00	1.50E+03
ICP.w.Sr*	1.96E+00	8.65E-02	1	0.00	5.70E+00
IC.w.Cl	4.50E+02	1.11E+03	1	2.56E+01	8.74E+02
IC.w.F	2.30E+03	6.46E+05	1	0.00	1.25E+04
IC.w.N03	4.12E+04	7.77E+06	1	5.82E+03	7.67E+04
IC.w.P04	1.55E+04	1.53E+06	1	0,00	3.13E+04
IC.w.so4	3.54E+03	2.85E+04	1	1.40E+03	5.69E+03

^{*:} Analytes with a portion of the data below 3 times the DL. *: Analytes with a portion of the data below 10 times the DL.

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Table 3-1. Concentration Estimate Statistics (Units μ g/g Except Radionuclides μ Ci/g). (sheet 3 of 3)

		<u> </u>	7011000	01 07	
Analyte	ÿ	δ ² (ÿ)	df	95%_LL	95% UL
GEA.Am-241	4.24E-02	2.61E-06	1	2.19E-02	6.29E-02
GEA.Cs-137	1.66E-01	3.35E-03	11	0.00	9.02E-01
Gross.alpha	3.73E-01	1.965-04	1 4 -	1.95E-01	5.51E-01
Gross, beta	1.51E+01	3.48E+01	1	0.00	9.00E+01
TGA.%.H20	7.65E+01	2.23E+01	1	1.64E+01	1.37E+02
Am-241±	4.26E-02	6.65E-25	1	4.26E-02	4.26E-02
Hg÷	1,43E+00	1.53E-01	•	0.00	6-40E+00
NO2.H.Spec	7.93E+02	8.76E+03	1	0.00	1.98E+03
Percent.H20	7.60E+01	5.81E-01	1	6.63E+01	8.57E+01
Pu-239/40	1.39E-01	9.19E-06	1	1.00E-01	1.77E-01
\$r- 9 0	5.41E+00	3.53E+00	1	0.00	2.93E+01
TOC*	3.12E+03	3.83E+05	1	0.00	1.10E+04
Tc-99•	7.92E-03	8.90E-06	1	0.00	4.58E-02
U+	2.796+03	2.01E+05	1	0.00	8.50E+03
PH	9.98E+00	7.79E-03	1	8.86E+00	1.11E+01

^{#:} Analytes with a portion of the data below 3 times the DL. #: Analytes with a portion of the data below 10 times the DL.

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4.0 COMPARISON OF THE VARIANCE COMPONENT ESTIMATES

Using the hierarchical structure of the core composite data, estimates of the between core spatial variability $(\sigma^2(S))$, the compositing variability $(\sigma^2(C))$ and the analytical measurement variability $(\sigma^2(A))$ can be obtained. The spatial variance is a measure of the variability between cores. The compositing variance measures the variability between composite samples within the same core. The analytical measurement variance includes, among other things, the sample handling error and the chemical analysis error. This variance is a measure of the difference between the analytical results from the primary and duplicate samples.

The estimate of the variance of the mean is a linear function of the spatial, compositing and analytical measurement variances. To help evaluate the magnitude of these three variance components, estimates of each variance component are given in Table 4-1.

4.1 STATISTICAL METHODS

Estimates of the spatial variance $(\hat{\sigma}^2(S))$, compositing variance $(\hat{\sigma}^2(C))$ and analytical measurement variance $(\hat{\sigma}^2(A))$, were obtained for each analyte using Restricted Maximum Likelihood Estimation (REML) methods. Restricted maximum likelihood estimation is discussed by Harville (1977).

To test the significance of the variance components, an analysis of variance (ANOVA) was calculated using the hierarchical statistical model described in Appendix C. The mean square error terms in the ANOVA tables were used to perform F-tests on the spatial variability and F-tests on the composite variability. These tests determine whether or not $\sigma^2(S)$ and $\sigma^2(C)$ are significantly different than zero. The p-values given in Table 4-1 were derived from the results of these tests.

4.2 STATISTICAL RESULTS

The REML estimates of each component of variability along with the p-values (significance level) from the F-tests are given in Table 4-1. A p-value less than 0.05 indicates that $\sigma^2(S)$ or $\sigma^2(C)$ is significantly different than zero at the 0.05 level.

The p-values from the tests on $\sigma^2(S)$ were less than 0.05 for 39 out of the 85 analytes in Tlll. This indicates that the differences between the two cores were statistically significant for these 39 cases.

The p-values from the tests on $\sigma^2(C)$ were less than 0.05 for 39 out of the 85 analytes in T111. This indicates that, relative to the analytical error, differences between composite samples were significantly different than zero in 39 cases. This means, for these 39 analytes, the two composite samples were statistically different from each other. Conversely, for 46 out of 85 cases, differences between composite samples were not statistically significant.

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Table 4-1. Variance Component Estimates. (sheet 1 of 3)

	lable 4	-1. vallance	Component	Lacimerea.	(sneet 1 of	3/
Analyte		δ ² (\$)	Test: \(\sigma^2(\sigma) = 0	δ ² (C)	Test:	σ ² (A)
ICP.e.Ag		1.56E+04_	0.001	1.94E+02	0.000	3.16E+00
ICP.a.AL		1.89E+04	0.055	4.79E+03	0.000	1.06E+01
ICP.a.B		1.25E+01	0.093	9.28E-01	0.387	8.59E+00
ICP.a.Ba		6.91E+01	0,263	1.206+02	0.000	3.54E-01
ICP.a.Bi		1.27E+07	0.000	6,25E+03	0.360	3.75E+04
ICP.a.Ca		4.10E+05	0.010	2.17E+04	0.074	9.95E+03
ICP.a.Cd+		5.96E+00	0.004	1.90E-01	0.039	5.34E-02
ICP.a.Ce•		1.69E+01	0.064	9.66E-01	0.380	7.87E+00
ICP.a.Cox		8.92E-01	0.376	8.08E-01	0.389	7.84E+00
ICP.a.Cr		3.17E+04	0.007	1.36E+03	0.081	6.75E+02
ICP.a.Cu		3.39E+02	0.290	1.67E+02	0.369	1.14E+03
ICP.a.Fe		2.33E+06	0.013	1.55E+05	0.063	6.25E+04
ICP.a.K		2.45E-24	0.911	8.86E+03	0.000	1.75E+02
ICP.a.La		5.91E+05	0.003	1_63E+04	0.019	2.88E+03
ICP.a.Mg		1.25E+04	0.005	5.18E+02	0.001	1.58E+01
ICP.a.Mn		2.97E+04	0.254	4.29E+04	0.030	1.00E+04
ICP.a.Na		2.76E+06	0.057	6.73E+05	0.016	1.05E+05
ICP.a.Ni		1.02E+03	- 0.000	6.13E+00	0.091	3.38E+00
ICP.a.P		2.37E-22	0.608	4.77E+05	9.000	1.25E+04
ICP.a.Pb		5.20E+04	0.003	1.41E+03	0.000	1,91E+01
ICP.a.S		1.11E+03	0.268	1.94E+03	0.003	1.25E+02
ICP.a.Sb-		1.118-20	0.739	1.44E-14	0.578	1.32E+02
ICP.a.Si		2.57E-31	0.946	7.61E-14	0.418	7.34E+03
ICP.a.Sr		6.38E+02	0.079	2.18E+02	0.004	1.54E+01
ICP.a.Ti		2.75E+02	0.001	4.69E+00	0.000	7.83E-02
ICP.a.V		1.83E+00	0.423	2.25E+01	0.000	2.306-01
ICP.a.Zn		1.22E+03	0.021	1.35E+02	0.008	1.42E+01
ICP.f.Ag		1.61E+04	0.000	6.23E+00	0.295	1.83E+01
îcp.f.AL		1.90E+04	0.005	6.78E+02	0.030	1.57E+02
ICP.f.Ba		3.99E+01	0.102	1.71E+01	0.012	2.19E+00
ICP.f.Bi		1.81E+07	0.001	2.88E+04	0.392	2.94E+05
ICP.f.Ca+		1.51E+05	0.008	2.07E-19	0.804	5.86E+04
ICP_f_Cd+		2.17E+00	0.126	1.21E-20	0.649	5.38E+00
ICP.f.Co.		1.44E+00	0.078	1.01E-29	0.755	3.48E+00
ICP.f.cr		3.72E-15	0.994	5.63E+03	0.013	1.21E+03

^{*:} Analytes with a portion of the data below 3 times the DL. *: Analytes with a portion of the data below 10 times the DL.

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Table 4-1. Variance Component Estimates. (sheet 2 of 3)

	Table 4-1.	Val Talle	Component	L3 Cilila ces.	(Sneet 2 of	
Anmiyte	∂ ² (\$)	Test: o ² (\$)=0 p-value	8 ² (C)	Test: \(\sigma^2(C) \neq 0 \\ p-value	δ ² (A)
ICP.f.Cu+		6.97E+01	0.005	2.00E+00	0.119	1.43E+00
ICP.f.Fe		8.00E+06	0.003	1.68E+05	0.079	8.13E+04
ICP.f.La		5.96E+05	0.010	3.82E+04_	0.012	4.98E+03
ICP.f.Mg		1.46E+04	0.000	5.15E-44	0.913	1.60E+02
ICP.f.Mn		_1.66E-11	0.553	7.02E+04	0,011	9.70E+03
ICP.f.Na		1.17E+07	0.007	5.43E+05	0.044	1.65E+05
ICP.f.Ni		1.17E+07	0,033	4.18E-12	0.498	4.39E+06
ICP.f.P		1.57E+06	0.028	1.958+05	0.049	6.45E+04
ICP.f.Pba		1.85E+04	0.003	4.35E+02	0.030	1.02E+02
ICP.f.S		2.15E+04	0.017	2.00E+03	0.008	2.13E+02
ICP.f.Si		1.06E+05	0.905	- 1.00E+03	0.369	6.53E+03
ICP.f.Sr		6.12E-16	0.661	1.15E+02	0.222	2.69E+02
ICP.f.Ti	<u> </u>	1.22E+03	0.000	2.66E-01	0.331	1.12E+00
ICP.f.V+		5.86E-19	0.905	2.17E+00	0.107	2-31E+00
ICP.f.Zn+	-	5.69E-23	0.440	1.10E-22	0.839	5.74E+01
ĪĒP.H.ĀL+		7.35E+00	0.208	6.93E+00	0.061	2.72E+00
ICP.w.B+		1.14E+00	0.082	3.99E-01	0.008	3.99E-02
ICP.w.Bi		3.90E+03	0.102	1.30E+03	0.123	9.66E+02
(CP.w.Ca+ -		1-86E-11	0.144	1.69E-22	0.862	2.66E+02
ICP.W.Cr		5.66E-20	0.979	9.52E+01	0.001	5.6ZE+00
ICP.w.Fe		6.16E+01	0.436	9.05E+02	0.077	4.29E+02
ICP.W.K		2.28E+03	0.148	1.57E+03	0.001	4.73E+01
1CP.w.La+		2.71E+01	0.015	1.34E+00	0.226	2.23E+00
ICP.w.Mg+		3.04E-21	0.597	1.61E-01	0.098	1.14E-01
ICP.w.Mn		2.51E+01	0.266	3.19E+01	0.130	2.51E+01
ICP.w.Na		4.53E+06	0.030	6.84E+05	0.002	2.88E+04
ICP.w.P		3.98E+04	0.222	3.94E+04	0.089	2.13E+04
ICP.w.S		3.95E+03	0.091	1.50E+03	0.014	2.13E+02
ICP.M.Si		8.53E+03	0.113	1.98E+03	0.272	4.72E+03
ICP.w.Sr+		6.35E-02	0.340	1.83E-01	0.061	7.19E-02
IC.w.Cl		1.99E+03	0.043	1.15E-10	0.509	9.66E+02
IC.w.F		1.28E+06	0.001	1.37E+04	0.102	8.44E+03
IC.w.NO3		1.42E+07	0.038	2.46E+06	0.013	3.44E+05
IC.w.P04	malytes with a por	2.59E+06	0.082	8.54E+05	0.030	1.99E+05

Analytes with a portion of the data below 3 times the DL.
 Analytes with a portion of the data below 10 times the DL.

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Table 4-1. Variance Component Estimates. (sheet 3 of 3)

140	16 4-1.	ar rance	Component	E3 C Fille CC3 :	(2)1665 2 01	<u> </u>
Analyte	δ ² (\$)		Test: σ^2 (S)=0 p-value	δ ² (c)	Test: σ ² (C)=0 p-value	∂ ² (A)
IC.w. so 4	5	.27E+04	0.032	7.04E+03	0.071	3.14E+03
GEA - Am - 241		.99E-32	0.786	8.13E-06	0.062	4.59E-06
GEA.Ca-137	6	-61E-03	0.003	1.87E-04	0.000	1.88E-06
Gross.elpha	3	.42E-04	0.065	7.74E-05	0.098	4.58E-05
Gross.beta	6	.93E+01	0.000	3.14E-01	0.026	6.62E=02
TGA.X.H20	3	.17E+01	0.008	3.78E-33	0.935	5.18E+01
Am-241±	6	.65E-25	0.976	7.41E-06	0.173	1.66E-05
Hg∗	1	.53E-01	0.022	1.36E-02	0.115	9.38E-03
NO2. H. Spec	1	.92E-14	0.789	3.46E+04	0.000	9.50E+02
Percent H20		.99E-01	0.128	2.55E-01	0.259	5.46E-01
Pu-239/40	2	.49E-28	0.736	2.01E-05	0.171	3.33E-05
\$r- 9 0		.04E+00	0.000	1.79E-02	0.209	2.61E-02
TOC+	6	.10E+05	0.113	2.98E+05	0.006	2.72E+04
Tc-99•		. 76E-05	0.001	2.63E-07	0.027	5.75E-08
U+	6.	.24E-06	0.831	7.91E+05	0.001	2.99E+04
pH	7.	.11E-28	0.601	3.06E-02	0.001	1.19E-03

^{•:} Analytes with a portion of the data below 3 times the DL. *: Analytes with a portion of the data below 10 times the DL.

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APPENDIX A

.... CORE COMPOSITE DATA

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Analyte	Core 31 Comp. 1	Mean/DL	Core 31 Comp. 2	Heary/DL	Core 33 Comp. 1	Mean/DL	Core 33 Comp. 2	Mean/DL
ICP.a.Ag	2.02E+02 2.03E+02	4.05E+02	2.25E+02 2.28E+02	4.53E+02	4 .48E+01 4 .39E+01	8.87E+01	2.81E+01 3.19E+01	6.00E+01
ICP. o.Al	5.80E+02 5.88E+02	2.43E+02	7.03E+02 7.07E+02	2.94E+02	4.71E+02 4.73E+02	1.97E+02	4.05E+02 4.04E+02	11.69E+02
ECP. E.As	#3.13E+00 3.17E+00	NA	*3.02E+00 *2.81E+00	MA	3.39E+00 3.50E+00	1.15E+00	3.23E+00 *2.90E+00	MA
ICP.a.B	2.34E+01 3.08E+01	4.52E+01	2.53E+01 2.16E+01	3.91E+01	2.92E+01 2.96E+01	4.90E+01	3.20E+01 3.23E+01	5.36E+01
ICP.a.Ba	5.66E+01 5.75E+01	1.90E+02	6.53E+01 6.46E+01	2.17E+02	6.67E+01 6.70E+01	2.23E+02	8.67E+01 8.79E+01	2.91E+02
ICP.a.Be	*1.04E-01 1.06E-01	NA NA	*1.01E-01 *9.38E-02	HA.	1.13E-01 1.20E-01	1.17E+00	1.08E-01 *1.00E-01	, NA
ICP.e.Bi	2.34E+04 2.37E+04	3.14E+03	2.33E+04 2.32E+04	3.10E+03	2.84E+04 2.86E+04	3.80E+03	2.86E+04 2.82E+04	3 - 79E+03
ICP.a.Ca	2.13E+03 2.26E+03	4 .99E+02	2.36E+03 2.61E+03	5.65E+02	1.49E+03 1.50E+03	3.40E+02	1.35E+03 1.34E+03	3-06E+02
ICP.a.Cdx	7.19E+00 7.25E+00	1.81E+01	7.78E+00 7.94E+00	1.97E+01	4.09E+00 4.70E+00	1.10E+01	3.64E+00 3.80E+00	9.30E+00
ICP.a.Ce•	3.39E+01 3.13E+01	3.23E+00	2.90E+01 2.83E+01	2.84E+00	4.00E+01 3.57E+01	3.75E+00	3.28E+01 3.89E+01	3.55E+00
ICP.a.Co*	3.42E+00 3.38E+00	4 25E+00	3.79E+00 1.17E+01	9.68E+00	3.15E+00 3.10E+00	3.91E+00	3.13E+00 2.70E+00	3.64E+00
ICP.s.Cr	1.63E+03 1.89E+03	2.07E+03	1.84E+03 1.84E+03	2.04E+D3	2.05E+03 2.08E+03	2.29E+03	2.16E+03 2.13E+03	2.38E+03
ICP.a.Cu	2.48E+01 2.57E+01	6.31E+01	1.27E+02 3.17E+01	11.98E+D2	1.64E+01 1.65E+01	4.11E+01	1.29E+01 1.30E+01	3.24E+01
CP.a.Fe	1.89E+04 1.95E+04	1.92E+04	2.01E+04 2.00E+04	2:_01E+04	1.74E+04 1.76E+04	1.75E+04	1.75E+04 1.72E+04	1.74E+04
ICP.a.K	1.08E+03 1.11E+03	9.78E+01	1.22E+03 1.20E+03	1.08E+02	1.21E+03 1.22E+03	1.08E+02	1.02E+03 1.02E+03	9.11E+01

NA: Not Available.

*: Value less than DL or less than 5 percent above the DL.

•: Analytes with a portion of the data below 3 times the DL.

*: Analytes with a portion of the data below 10 times the DL.

B-20

1.42E+01

1.04E+01

Not Available.

1.31E+01

2.15E+01

^{*:} Value less than DL or less than 5 percent above the DL.

^{*:} Analytes with a portion of the data below 3 times the DL.

^{*:} Analytes with a portion of the data below 10 times the DL.

Table A-1.: Core Composite Data (Units $\mu g/g$ except Radionuclides $\mu Ci/g$). (sheet 3 of 9) Hean/DL Mean/DL Core 33 Core 31 Core 31 Mean/DL Core 33 Mean/DL Analyte Comp. 1 Comp. 2 Comp. 2 Comp. 1 3.38E+02 4.36E+01 1.47E+02 3.45E+01 1.17E+02 7.68E+01 2.65E+02 9.68F+01 ICP.a.Zn 4.47E+01 3.54E+01 8.20E+01 1.06E+02 8.61E-01 8.34E-01 MA 8.07E-01 A.K 9.05E-01 NA MA ICP.a.Zr 7.51E-01 9.20E-01 7.70E-01 8.47E-01 8.86E+01 3.89E+01 1.36F+01 3.69E+01 1.286+01 8.56E+01 2.17E+02 2.10E+02 ICP.f.Ag 2.26E+02 4.00E+01 3.74E+01 2.18E+02 5.37F+01 5.78E+01 4.83E+02 4.03E+01 4.59E+02 3.83E+01 ICP.f.AL 6.32E+02 6.80E+DZ 4.85E+02 4.59E+02 7.06E+02 6.56E+02 *1.50E+01 NA *1.50E+02 MA ICP .f .As *1.50E+01 N'A *1.50E+01 MA *1.50E+01 *1.50E+02 *1.50E+01 *1.50E+01 *2.99E+00 NA *5.3&E+00 NA *4.58E+00 ĦΑ *3.01E+00 NA ICP.f.B *2.99E+00 *4.32E+00 *5.10E+00 *3.00E+00 5.72E+01 3.92E+01 5.96E+01 4.04E+01 6.60E+01 4.36E+01 7.45E+01 4.92E+01 ICP.f.Ba 6.48E+01 7.30E+01 6.04E+01 6.15E+01 *4.99E-01 MA *5.00E-01 NA *4.99E-01 ICP.f.Be *5.01E-01 NA MA *5.00E-01 *4.99E-01 *5.00E-01 *4.99E-01 2.73E+04 2.02E+04 5.37E+02 2.66E+04 7.05E+02 7.12E+02 2.05E+04 5.59E+02 ICP.f.BI 2.63E+04 2.61E+04 2.01E+04 2.14E+04 ICP.f.Ca* 2.58E+03 1.25E+02 2.49E+03 1.21E+02 2.50E+03 9.23E+00 2.18E+03 8.54E+00 2.83E+03 1.93E+03 1.92E+03 2.93E+03 7.18E+00 5.32E+00 6.761-00 3.21E+00 6.86E+00 3.59E+00 7.34E+00 4.13E+00 ICP.f.Cd* 1.41E+01 6.0BE+00 7.48E+00 9.16E+00 *5.04E+01 *5.04E+01 *5.05E+01 *5.06E+01 NA NA ICP.f.Ce NA NA *5.05E+01 *5.04E+01 *5.05E+01 *5.04E+01

2.70E+00

3.78E+02

1.71E+01

1.37E+01

1.30E+01

1.81E+03

1.76E+03

2.16E+01

2.26E+01

3.34E+00

3.97E+02

9.82E+00

8.90E+00

1.48E+01

1.82E+03

1.82E+03

2.30E+01

2.61E+01

2.96E+00

4.04E+02

1.09E+01

NA: Not Available.

ICP.f.Co.

ICP.f.Cr

ICP.f.Cua

* : Value less than DL or less than 5 percent above the DL.

9.70E+00

1.05E+01

1.86E+03

1.92E+03

3.68E+01

3.59E+01

. : Analytes with a portion of the data below 3 times the DL.

2.53E+00

4.20E+02

1.82E+01

1.11E+01

1.05E+01

1.67E+03

1.73E+03

3.41E+01

3,42E+01

*: Analytes with a portion of the data below 10 times the DL.

Ariallyte	Core 31 Comp. 1	Mean/DL	Core 31 Comp. 2	Mean/DL	Core 33 Comp. 1	Mean/DL	Core 33 Comp. 2	Mean/DL
ICP.f.Fe	2.02E+04 2.08E+04	4.10E+03	1.95E+04 1.97E+04	3.926+03	1.62E+04 1.57E+04	6.16E+02	1.61E+04 1.61E+04	6.22E+02
CP.f.K	*5.61E+01 *5.60E+01	NA	*5.59E+01 *5.59E+01	HA	*5.59E+01 *5.60E+01	NA	*5.60E+01 *5.59E+01	MA
ICP. f.La	3.63E+03 3.75E+03	5.27E+02	3.38E+03 3.45E+03	4.88E+02	4.58E+03 4.45E+03	6.45E+02	4.84E+03 4.78E+03	6.87E+02
ICP.f.Mg	4.24E+02 4.52E+02	2.92E+02	4.31E+02 4.56E+02	2.96E+02	2.78E+02 2.58E+02	3.46E+01	2.74E+02 2.69E+02	3.50E+01
ICP.f.Mn	6.29E+03 6.47E+03	6,38E+03	5.86E+03 6.02E+03	5.94E+03	6.29E+03 6.15E+03	4.01E+(32	6.59E+03 6.59E+03	4.258+02
ICP.f.Na	3.96E+04 4.01E+04	2.57E+03	3.85E+04 3.94E+04	2.51E+03	3.41E+04 3.36E+04	5.34E+()1	3.51E+04 3.52E+04	5.54E+01
ICP.f.Ni	5.94E+03 5.61E+03	6.79E+02	7.49E+03 3.40E+03	6.41E+02	9.49E+03 9.09E+03	8.68E+01	9.81E+03 1.43E+04	1.13E∳02
ICP.f.P	1.12E+04 1.19E+04	3.986+02	1.11E+04 1.11E+04	3.83E+02	8.99E+03 9.15E+03	1.04E+02	9.91E+03 9.91E+03	1.13E+02
ICP.f.Pb+	4.27E+02 4.53E+02	1.42E+01	4.82E+02 4.86E+02	1.56E+01	2.62E+02 2.72E+02	8.61E+90	2.72E+02 2.67E+02	8.69E+00
ICP.f.S	1.35E+03 1.36E+03	1.00E+02	1.29E+03 1.33E+03	9.70E+01	1.08E+03 1.08E+03	8.00E+01	1.16E+03 1.16E+03	8.59E+01
ICP.f.Sb	*8.87E+01 *8.85E+01	NA	*8.83E+01 *8.83E+01	NA	1.29E+02 *8.85E+01	HA	*8.85E+01 *8.83E+01	NA :
ICP. f. Se	*3.81E+01 *3.80E+01	NA	*3.79E+01 *3.79E+01	NA	*3.79E+01 *3.80E+01	NA	*3.80E+01 *3.79E+01	NA
ICP.1.SI	5.88E+03 6.04E+03	9.17E+02	5.78E+03 5.89E+03	8.98E+02	5.52E+03 5.39E+03	7.43E+0‡	5.41E+03 5.41E+03	7.37E+01
ICP.f.Sn	*8.02E+00 *8.00E+00	NA	*7.98E+00 *7.98E+00	NA	*7.98E+00 *8.00E+00	NA	*8.00E+00 *7.98E+00	NA
CP.f.Sr	3.08E+02 2.98E+02	2.02E+02	2.70E+02 2.91E+02	1.87E+02	2.97E+02 2.84E+02	6.65E+01	2.98E+02 3.36E+02	7.258+01

MA: Not Available.

*: Value less than DL or less than 5 percent above the DL.

: Analytes with a portion of the data below 3 times the DL.

: Analytes with a portion of the data below 10 times the DL.

Table A-1. Core Composite Data (Units µg/g except Radionuclides µCi/g). (sheet 5 of 9)

Analyte	Core 31 Comp. 1	Mean/DL	Core 31 Comp. 2	Menn/DL	Core 33 Comp. 1	Mean/DL	Core 33 Comp. 2	Mean/DL
ICP.f.Ti	7.32E+01 7.25E+01	3.64E+0.1	7.13E+01 7.34E+01	3.62E+01	2.25E+01 2.22E+01	1.12E+01	2.51E+01 2.31E+01	1.218+01
TCP.f.V+	1.11E+01 1.32E+01	4.86E+00	1.55E+01 1.74E+01	6.58E+00	1.38E+01 1.69E+01	6.14E+00	1.42E+01 1.51E+01	5.86E+00
ICP.f.Zn#	9.71E+01 1.11E+02	6.94E+01	1.11E+02 1.00E+02	7.03E+01	1.10E+02 1.00E+02	7.66E+00	1.02E+02 1.19E+02	8.07E+00
1CP.f.Zr	4.01E+00 4.00E+00	NA	3.99E+00 3.99E+00	NA	*3.99E+00 *4.00E+00	NA NA	*4.00E+00 *3.99E+00	WA
[CP.W.Ag	7.80E-01 *5.00E-01	NA	1.26E+00 1.07E+00	2.33E+00	*4 .99E -01 6 .08E -01	NA	*4.99E-01 *5.00E-01	NA
ECP.W.AL	6.99E+00 6.43E+00	2.50E+00	1.04E+01 1.02E+01	4.29E+00	1.34E+01 1.77E+01	6.48E+00	1.19E+01 1.02E+01	4.60E+00
I CP. H.AS	*3.00E+00 *3.00E+00	NA	*3.00E+00 *3.00E+00	NA	*2.99E+00 *2.99E+00	NA NA	*2.99E+00 *3.00E+00	NA
ICP.W.B+	3.50E+00 3.11E+00	5.51E+00	3.27E+00 3.12E+00	5.33E+00	5.54E+00 5.54E+00	9.23E+00	4.44E+00 4.06E+00	7.08E+00
ICP.w.Ba	*3.09E-01 *3.00E-01	NA .	*3.00E-01 5.32E-01	MA	378E-01 715E-01	1.82E+00	5.32E-01 5.01E-01	1.72E+00
ICP.w.Be	*1.00E-01 *9.99E-02	NA .	*9.99E-02 *9.99E-02	МА	*9.98E-02 *9.98E-02	RA	*9.98E-02 *9.99E-02	MA
JCP.N.Bi	8.34E+01 1.47E+02	1.54E+01	1.76E+02 2.05E+02	2.54E+01	2.05E+02 2.58E+02	3.09E+01	2.73E+02 2.67E+02	3.60E+01
ICP.W.Ca+	5.05E+01 5.12E+01	1.16E+01;	5.43E+01 6.89E+01	1.40E+01	7. 16E+01 6. 13E+01	1.51E+01	9.35E+01 4.15E+01	1.53E+01
ICP.w.Cd	*4.00E-01 *4.00E-01	MA	*4.00E-01 *4.00E-01	NA	*3.99E-01 *3.99E-01	NA	*3.99E-01 *4.00E-01	NA
ICP.w.Ce	*1.01E+01 *1.01E+01	НА	*1.01E+01 *1.01E+01	NA	*1.01E+01 *1.01E+01	MA	*1.01E+01 *1.01E+01	MA
CP.W.Co	*8.00E-01 *7.99E-01	MA	*7.99E-01 8.51E-01	NA	*7.98E-01 8.43E-01	NA	8.50E-01 *7.99E-01	NA

NA: Not Available.

^{*:} Value less than DL or less than 5 percent above the DL.

: Analytes with a portion of the data below 3 times the DL.

: Analytes with a portion of the data below 10 times the DL.

Table A-1. Core Composite Data (Units $\mu g/g$ except Radionuclides $\mu Ci/g$). (sheet 6 of 9)

				2/2				
Analyte	Core 31 Comp. 1	Mean/DIL	Core 31 Comp. 2	Hean/DL	Core 33 Comp. 1	Mean/DL	Core 33 Comp. 2	Mean/DL
ICP_w.Cr	2.11E+02 2.07E+02	2.32E+02	2.28E+02 2.30E+02	2.54E+02	2.22E+02 2.26E+02	2.49E+02	2.12E+02 2.09E+02	2.34E+02
1CP_w.Cu	*4.00E-01 *4.00E-01	MA	*4.00E-01 *4.00E-01	NA	*3.99E-01 *3.99E-01	NA '	*3.99E-01 *4.00E-01	NA
ICP.w.Fe	5.70E+01 1.02E+02	7.95E+01	1.30E+02 1.51E+02	1.41E+02	1.16E+02 1.47E+02	1.32E+02	1.60E+02 1.58E+02	1.59E+02
ICP.W.K	7.40E+02 7.28E+02	6.55E+01	7.83E+02 7.83E+02	6.99E+01	7.04E+02 7.19E+02	6.35E+01	6.47E+02 6.50E+02	5.79E+01
ICP.W.Law	5.00E+00 7.05E+00	4.30E+00	7.69E+00 9.35E+00	6.09E+00	1.22E+01 1.55E+01	9.89E+D0	1.57E+01 1.58E+01	1.13E+01
ICP.W.Mg#	2.67E+00 3.22E+00	9.82E+00	3.66E+00 4.23E+00	1.32E+01	3.60E+00 4.08E+00	1.28E+01	3.95E+00 3.72E+00	1.28E+01
ICP.w.Mn	1.01E+01 1.93E+01	7.35E+01	2.31E+01 2.70E+01	1.25E+02	2.04E+01 3.04E+01	1.27E+02	3.34E+01 3.42E+01	1.69E+02
ICP.W.Na	3.41E+04 3.39E+04	1.10E+04	3.51E+04 3.50E+04	1.13E+04	3.06E+04 3.09E+04	9.92E+03	3.19E+04 3.22E+04	1.03E+04
ICP.w.Ni	*1.70E+00 *1.70E+00	NA	*1.70E+00 *1.70E+00	NA	*1.70E+00 *1.70E+00	NA	*1.70E+00 *1.70E+00	HA
ICP.W.P	5.63E+03 5.89E+03	9.93E+02	5.81E+03 6,11E+03	1.03E+03	5.26E+03 5.34E+03	9.14E+02	5.66E+03 5.74E+03	9.836+02
ICP.W.Pb	7.91E+00 *6.19E+00	NA	6.93E+00 8.92E+00	1,285+00	*6.19E+00 *6.28E+00	NA NA	*6.19E+00 *6.19E+00	NA NA
ICP.W.S	1.21E+03 1.18E+03	4.43E+02	1.19E+03 1.21E+03	4.44E+02	1.05E+03 1.07E+03	3.93E+02	1.14E+03 1.14E+03	4.22E+02
ICP.w.Sb	*1.77E+01 *1.77E+01	NA !	*1.77E+01 *1.77E+01	NA	*1.77E+01 *1.77E+01	NA NA	*1.77E+01 *1.77E+01	NA
ICP.w.Se	*7.60E+00 *7.59E+00	NA	*7.59E+00 8.44E+00	NA	*7.58E+00 *7.58E+00	NA	*7.58E+00 *7.59E+00	MA
ICP.W.Si	3.45E+02 5.30E+02	3.37E+02	5.30E+02 5.89E+02	4.30E+02	6.68E+02 6.71E+02	5.15E+02	6.22E+02 6.18E+02	4.77E+02

NA: Not Available.
*: Yalue less than DL or less than 5 percent above the DL.

Analytes with a portion of the data below 3 times the DL.
 Analytes with a portion of the data below 10 times the DL.

Analyte	Core 31 Comp. 1	Mean/DL	Core 31 Comp. 2	Mean/DL	Core 33 Comp. 1	Mean/DL	Core 33 Comp. 2	Mean/DL
ICP.w.\$n	*1.60E+00 *1.60E+00	NA NA	*1.60E+00 *1.60E+00	NA	*1.60E+00 *1.60E+00	NA	*1.60E+00 *1.60E+00	NA
ICP.w.Sr*	9.37E-01 1.48E+00	4.03E+00	1.97E+00 2.29E+00	7.10E+00	1.97E+00 2.38E+00	7.25E+00	2.39E+0() 2.29E+0()	7.806+00
ICP.W.Ti	*4.00E-01 *4.00E-01	NA	*4.00E-01 *4.00E-01	NA	*3.99E-01 *3.99E-01	NA	*3.99E-0 *4.00E-0	NA
ICP.W.V	*500E -01 *500E -01	NA NA	5.94E-01 8.20E-01	1.41E+D0	*4.99E-01 *4.99E-01	NA	9.21E-01 6.76E-01	1.60E+00
ICP.w.Zh	*3.00E-01 *3.00E-01	NA NA	*3.00E-01 *3.00E-01	MA	*2.99E-01 *2.99E-01	NA	*2.99E-01 *3.00E-01	NA
ICP.w.Zr	8.00E-01 7.99E-01	HA	7.99E-01 7.99E-01	N/A	7.98E-01 7.98E-01	NA NA	7.98E-01 7.99E-01	NA
IC.w.cl	4.73E+02 4.66E+02	4.70E+01	5.18E+02 4.75E+02	4.97E+01	4.40E+02 3.62E+02	4.01E+01	4.40E+02 4.23E+02	4.32E+01
IC.W.F	3.14E+03 3.03E+03	3 .09E+02	3.16E+03 3.09E+03	3 . 13E+02	1.47E+03 1.26E+03	1.37E+02	1.67E+03 1.59E+03	1.63E+02
C.W.NO2	<1.10E+03 <1.10E+03	NA	8.71E+02 <1.10E+03	NA	8.42E+02 7.04E+02	7.73E+00	7.59E+02 7.04E+02	7.32E+00
IC.W.N03	4.45E+04 4.41E+04	4.43E+02	4.36E+04 4.39E+04	4.38E+02	3.76E+04 3.61E+04	3.69E+02	3.98E+04 4.03E+04	4.01E+02
IC.W.P04	1.56E+04 1.67E+04	1.62E+02	1.71E+04 1.77E+04	1.74E+02	1.36E+04 1.35E+04	1.36E+02	1.51E+04 1.50E+04	1.51E+02
C.W.S04	3.69E+03 3.69E+03	3.69E+01	3.72E+03 3.75E+03	3.74E+01	3.23E+03 3.34E+03	3.29E+01	3.41E+03 3.52E+03	3.47E+01
EA . Am-241	4.37E-02 4.80E-02	4.02E+01	4.02E-02 4.16E-02	3.59E+01	3.95E-02 3.79E-02	3.39E+01	4.24E-02 4.61E-02	3.88E+02
EA. Co-60	<4.21E-04 <3.39E-04	NA	<3.85E-04 <3.75E-04	NA	<3.45E-04 <3.70E-04	NA	<3.29E-04 <3.45E-04	: NA
EA.Cs-137	2.12E-01 2.11E-01	5.72E+02	2.36E-01 2.38E-01	6.41E+02	1.15E-01 1.12E-01	3.07E+02	1.03E-01 1.04E-01	2.806+02

NA: Not Available.

*: Value less than DL or less than 5 percent above the DL.

*: Analytes with a portion of the data below 3 times the DL.

*: Analytes with a portion of the data below 10 times the DL.

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Table A-1.	Core Composite	Data (Unit:	s μg/g except	Radionuclides	μ Ct/g).	(sheet 8 of 9)	

Analyte	Corre 31 Comp. 1	Hean/DL	Core 31 Comp. 2	Mean/DL	Core 33 Comp. 1	Mean/DL	Comp. 2	Hean/DL
GEA.Eu-154	*1.09E-03 *1.07E-03	на	<1.06E-03 3.24E-03	NA	<1.11E-03 <1.01E-03	WA.	<1.05E-03 <9.64E-04	NÀ
GEA.Eu-155	<2.08E-03 <2.09E-03	NA	<2.13E-03 <2.12E-03	NA	2.97E-03 3.16E-03	5.15E+00	<1.49E-03 <1.49E-03	MA
Gross alpha	3.59E-01 3.58E-01	5.05E+01	3.50E-01 3.69E-01	5.06E+01	3.78E-01 3.76E-01	7.116+02	3.97E-01 3.97E-01	7.496+02
Gross.beta	2.04E+01 2.07E+01	2.20E+02	2.16E+01 2.13E+01	2.29€+02	9.86E+00 9.32E+00	1.56E+02	8.95E+00 8.71E+00	1.44E+02
TIC.W.CO3	<5.00E+02 6.50E+02	NA	6.49E+02 9.99E+02	1.65E+00	7.49E+02 8.98E+02	1.65E+00	7.99E+02 1.40E+03	2 20E+00
TGA.X.H20	6.13E+01 8.53E+01	NA	6.92E+01 7.12E+01	WA	8.10E+01 8.22E+01	MA	7.86E+01 8.30E+01	NA
Am-241#	4.43E-02 3.85E-02	7.26E+00	4.66E-02 3.96E-02	7.56E+00	3.71E-02 3.94E-02	6.59E+00	4.44E-02 5.11E-02	7.96E+00
C-14	3.80E-04 <2.25E-04	KA	<2:.23E-04 <2:.25E-04	NA	<2.25E-04 <2.25E-04	AM	<2.25E-04 <2.25E-04	HA
CN	<4.50E+00 <4.00E+00	NA	<3.41E+00 <3.57E+00	МА	<4.90E+00 <4.81E+00	NA	<4.76E+00 <4.61E+00	NA
н3	<3.15E-04 <3.15E-04	NA	<3.12E-04 <3.15E-04	NA	<3.15E-04 <3.15E-04	MA	<3.15E-04 <3.15E-04	NA
Hg+	1.70E+00 1.48E+00	1.27E+01	1 - 79E+00 1 - 88E+00	1.47E+01	1.18E+00 1.22E+00	9.60€+00	1.02E+00 1.15E+00	B.68E+90
I - 12 9	<1.74E-02 <1.72E-02	NA	<240E-02 <1.92E-02	NA	<2.15E-02 <1.44E-02	NA	<1.75E-02 <2.17E-02	NA
NH3	<4.50E+03 <4.50E+03	NA	<4.50E+03 <4.50E+03	NA	<4.50E+03 <4.50E+03	MA	<4.50E+03 <4.50E+03	MA
NO2.w.Spec	9.49E+02 9.55E+02	1.90E+01	5.22E+02 5.27E+02	1.05E+01	8.72E+02 8.84E+02	1.76E+01	7.74E+02 5.60E+02	1,63E+01
Np-237	<3.25E-02 <3.24E-02	NA	<3.24E-02 <3.24E-02	NA	<3.24E-02 <3.24E-02	NA	<3.24E-02 <3.24E-02	MA

NA: Not Available.

*: Value less than Dt or less than 5 percent above the Dt.

*: Analytes with a portion of the data below 3 times the Dt.

*: Analytes with a portion of the data below 10 times the Dt.

Analyte	Core 31 Comp. 1	Medin/DL	Core 31 Comp. 2	Mean/DL	Core 33 Comp. 1	Mean/DL	Core 33 Comp. 2	Mean/DL
Percent.H20	7.48E+01 7.44E+01	HA	7.59E+01 7.59E+01	MA	7.72E+01 7.57E+01	NA '	7.78E+01 7.64E+01	NA
Pu-238	<1.04E-02 <1.07E-02	NA	<1.03E-02 <1.04E-02	MA	<1.05E-02 <1.00E-02	MA	<1.13E-02 <1.04E-02	NA
Pu-239/40	1.35E-01 1.41E-01	3.83E+01	1.34E-01 1.37E-01	3.76E+01	1.39E-01 1.29E-01	3.83E+01	1.53E-01 1.62E-01	4.21E+01
Se	<6.05E-05 <6.11E-05	NA	6.17E-05 6.31E-05	1.11E+00	<1.23E-04 <1.26E-04	NA '	<1.29E-04 <1.26E-04	NA
Sr-90	6.97E+00 7.34E+00	1.01E+03	7.55E+00 7.31E+00	1.62E+03	3.67E+00 3.62E+00	1.70E+03	3.37E+00 3.48E+00	1.59€+03
TOC+	3.30E+03 3.68E+03	6.98E+00	4.12E+03 3.85E+03	7.97E+00	2.00E+03 2.00E+03	4.00E+00	3.00E+03 3.00E+03	6.00E+00
Tc-99•	4.95E-03 5.33E-03	6.05E+00	4.90E-03 4.56E-03	5 .561:+00	1.12E-02 1.16E-02	2.486+00	1.03E-02 1.05E-02	2.26E+00
U+	2.14E+03 2.21E+03	8.70E+00	4.00E+03 3.75E+03	1.556+01	3.01E+03 3.34E+03	9.34E+00	2.07E+03 1.82E+03	5.72E+00
pH	1.02E+01 1.02E+01	N/A.	9.91E+00 9.94E+00	MAL	1.01E+01 1.00E+01	NA	9.72E+00 9.81E+00	NA

MA: Not Available.

*: Value less than Dt. or less than 5 percent above the DL.

*: Analytes with a portion of the data below 3 times the DL.

*: Analytes with a portion of the data below 10 times the DL.

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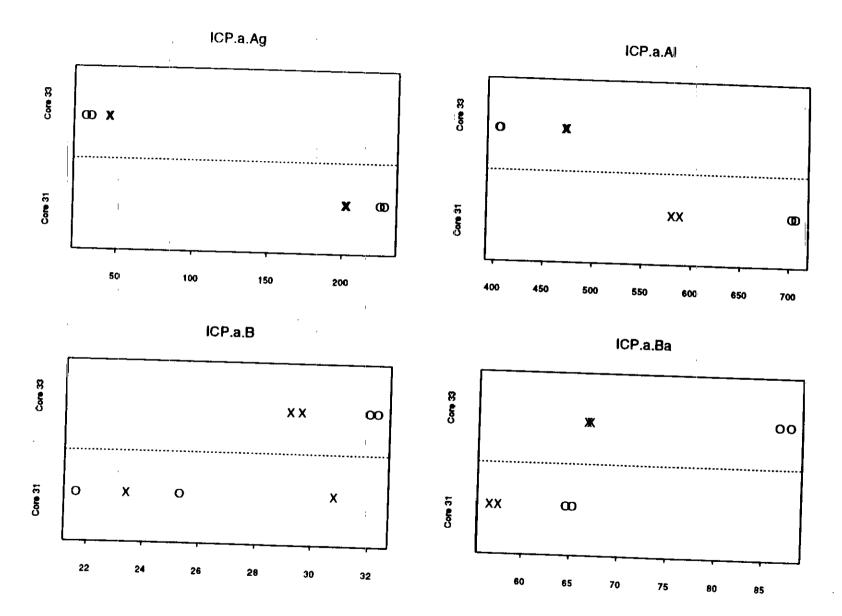
WHC-EP-0806 WHC-SD-WM-TI-650, Rev. O

APPENDIX B

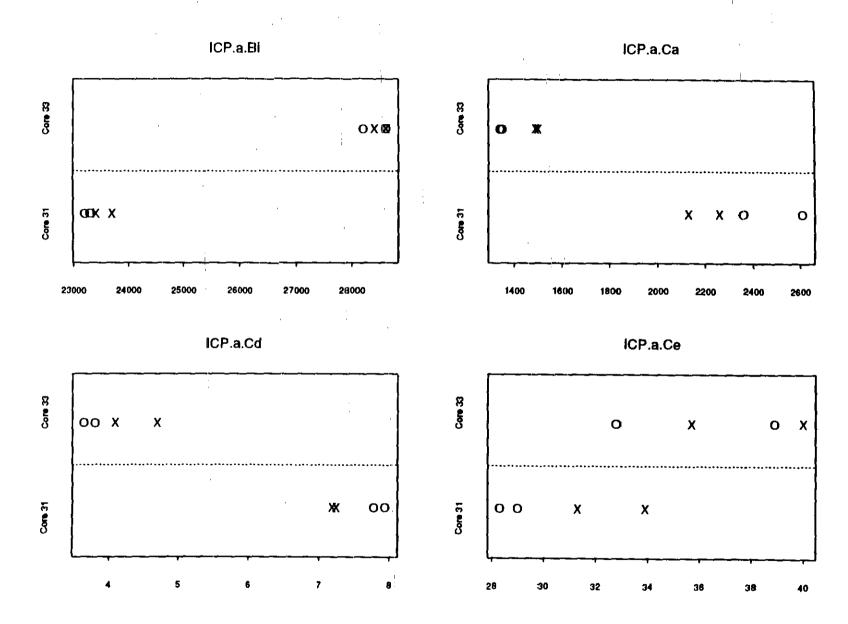
CORE COMPOSITE DATA PLOTS

WHC-EP-0806 WHC-SD-WH-TI-650, Rev. 0

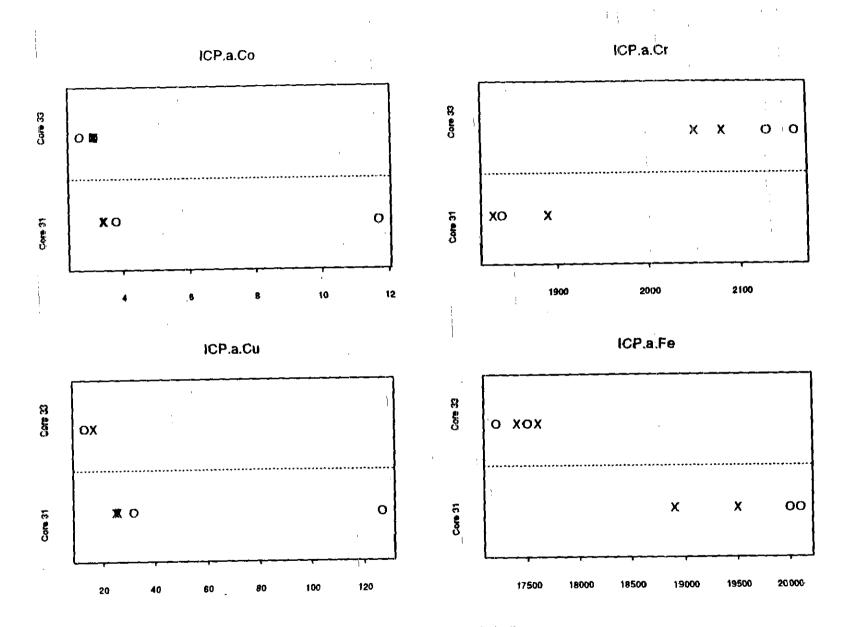
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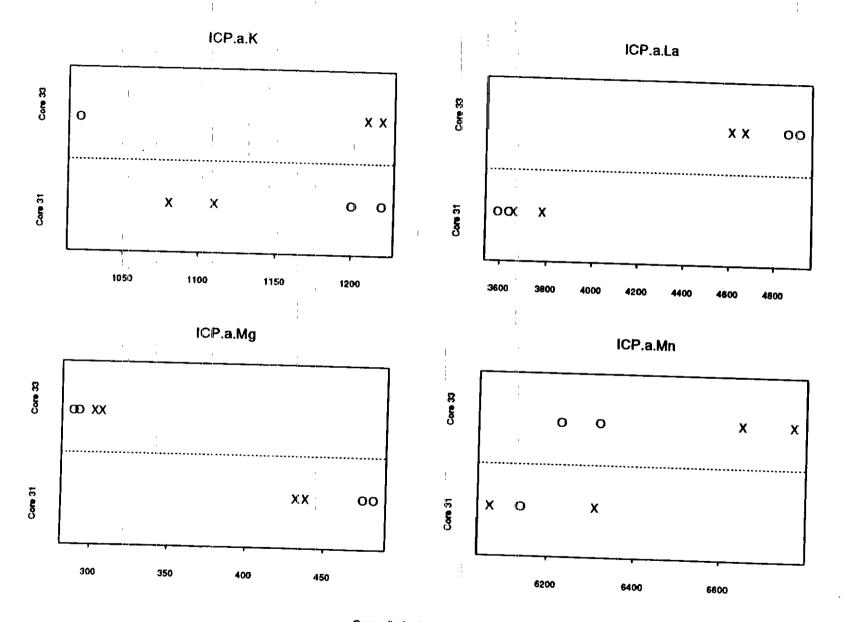
Composite 1 = 0, Composite 2 = X



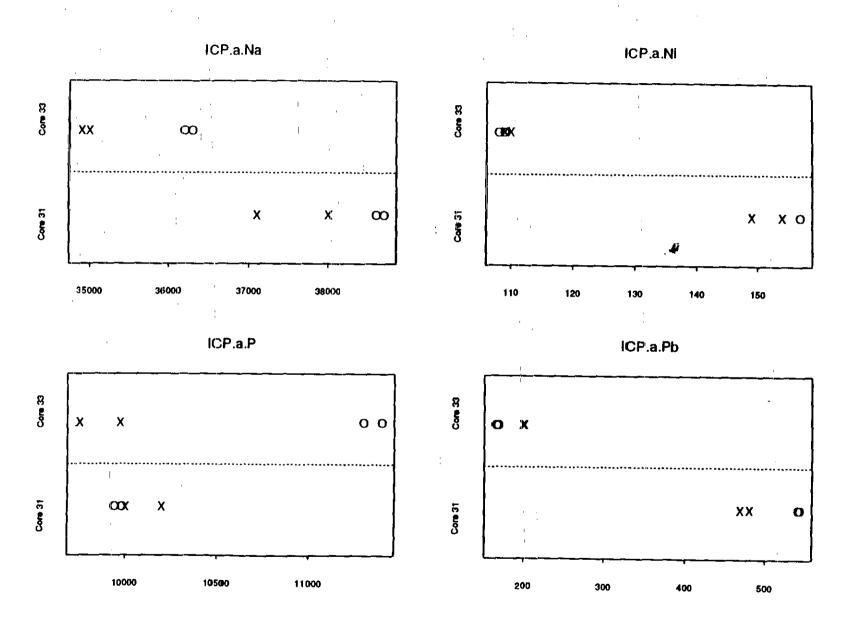
Composite 1 = CI, Composite 2 = X



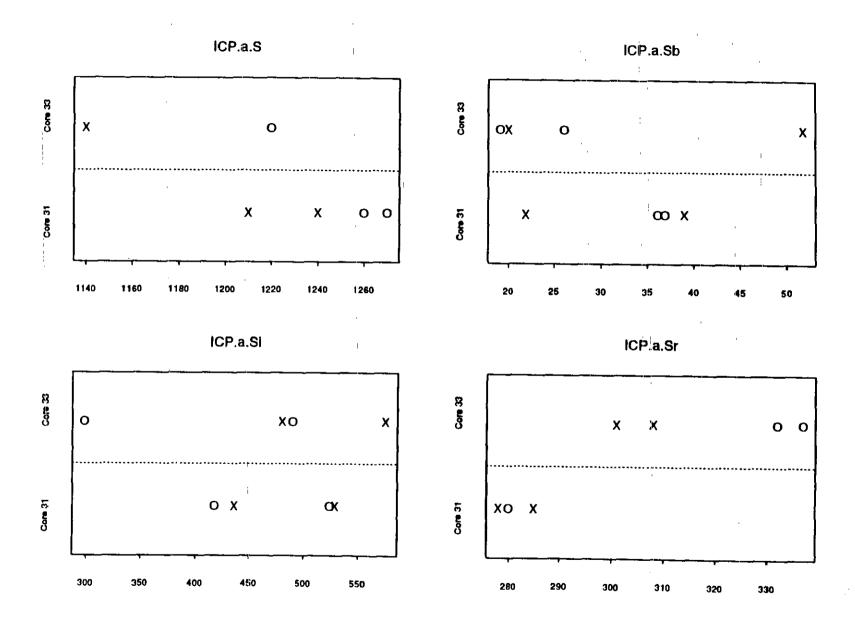
Composite 1 = 0 , Composite 2 = X



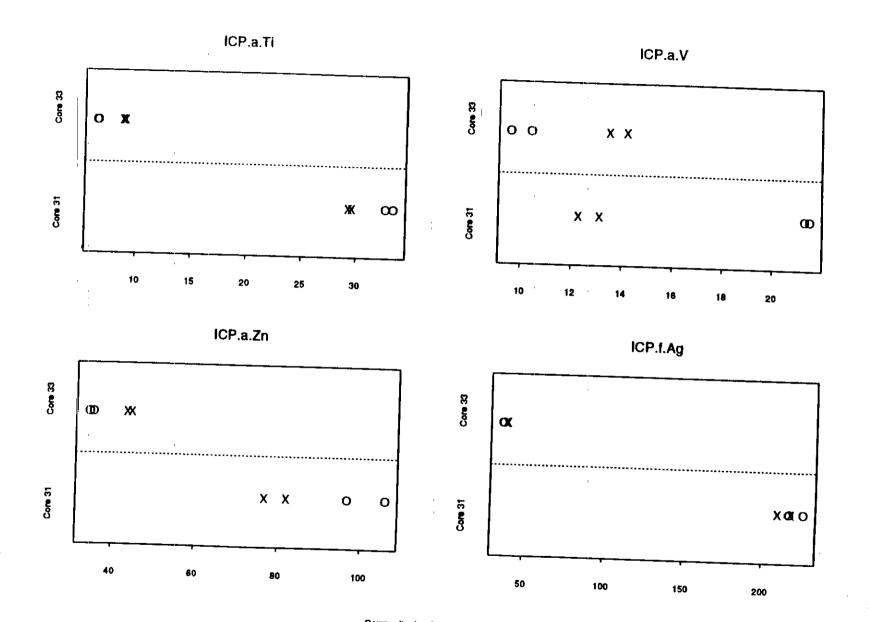
Composite 1 = 0, Composite 2 = X



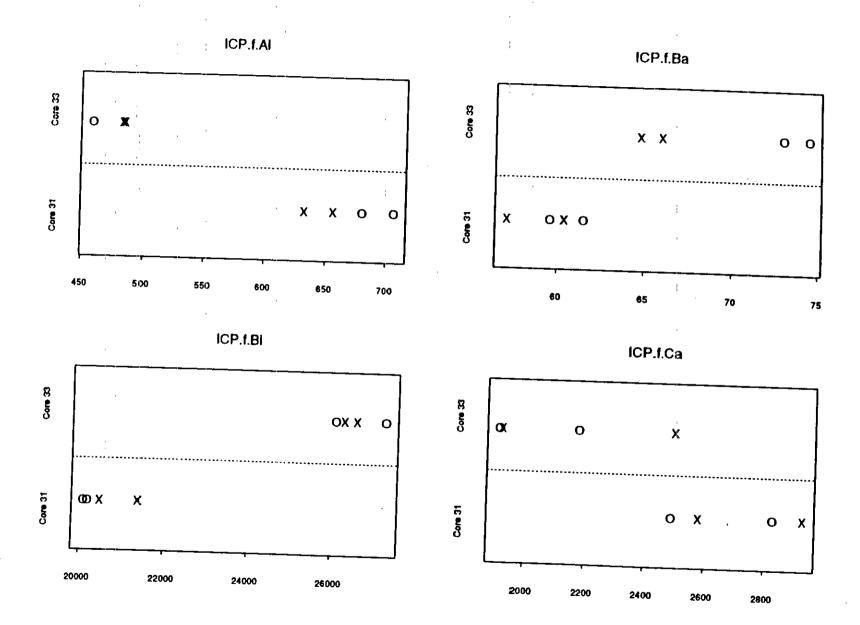
Composite 1 = 0, Composite 2 = X



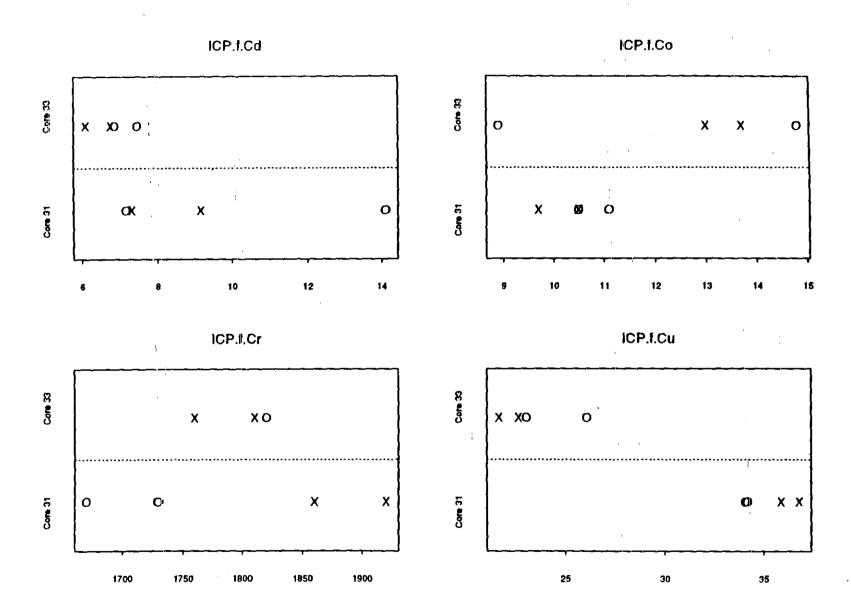
Composite 1 = 0 , Composite 2 = X



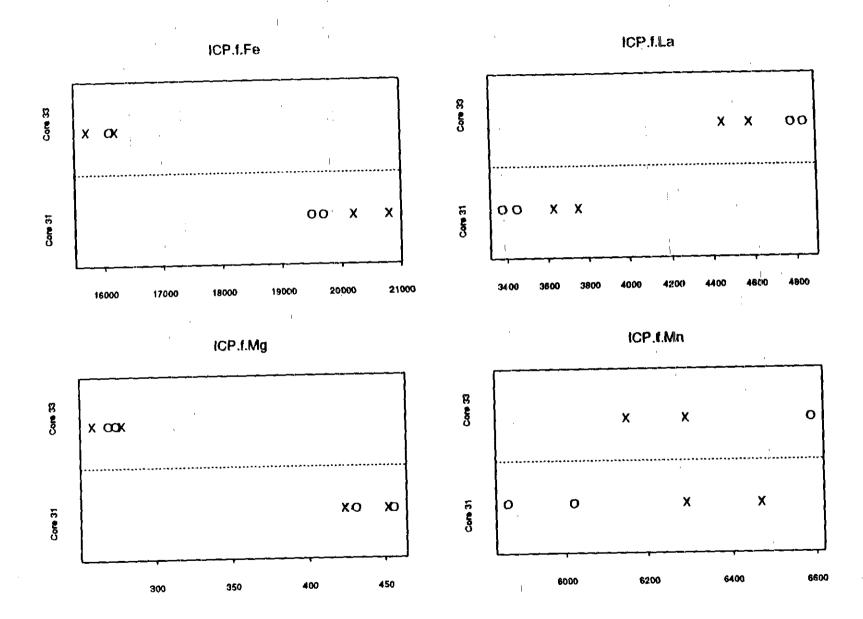
Composite 1 = CI, Composite 2 = X



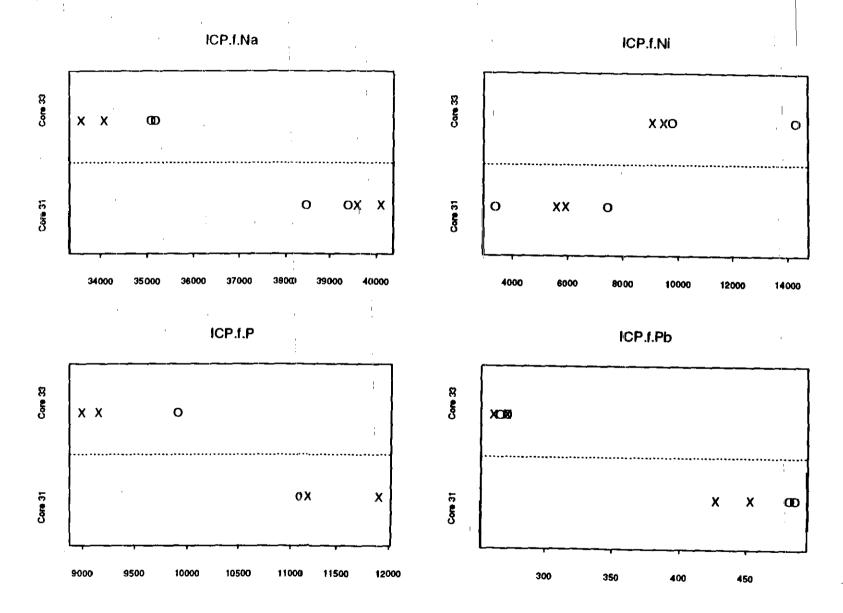
Composite 1 = 0, Composite 2 = X



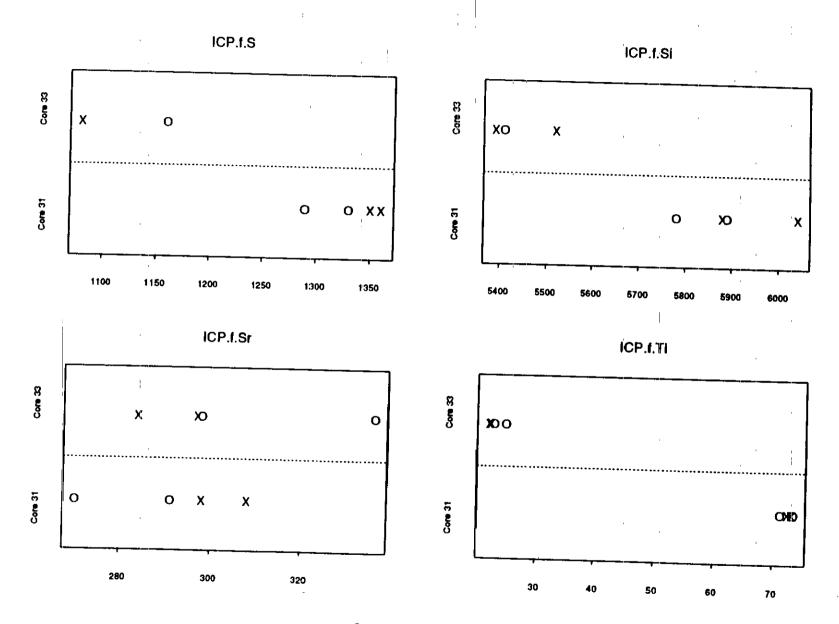
Composite 1 = 0, Composite 2 = X



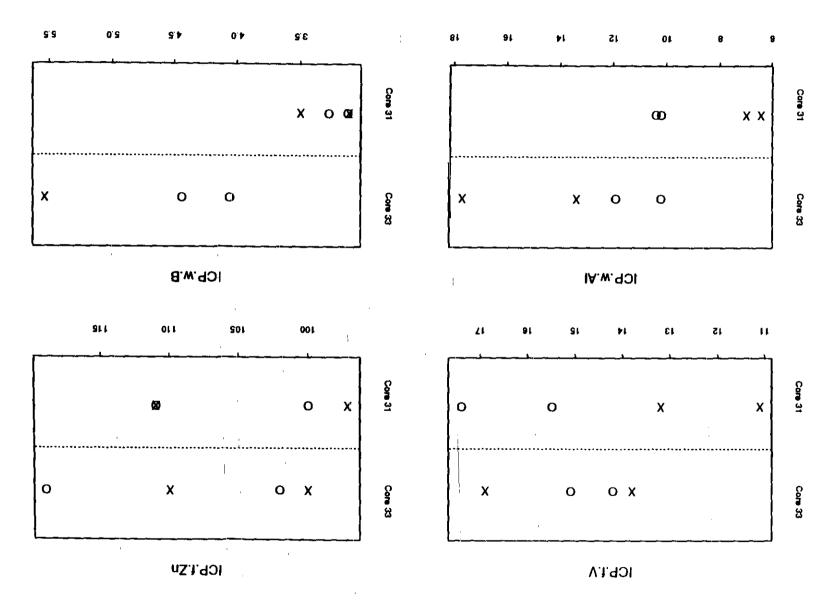
Composite 1 = 0 , Composite 2 = X



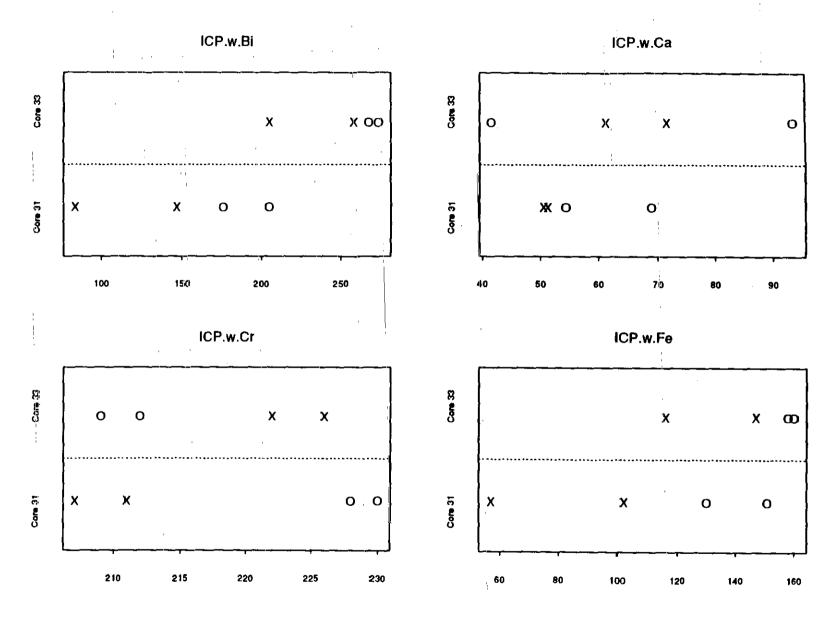
Composite 1 = 0, Composite 2 = X



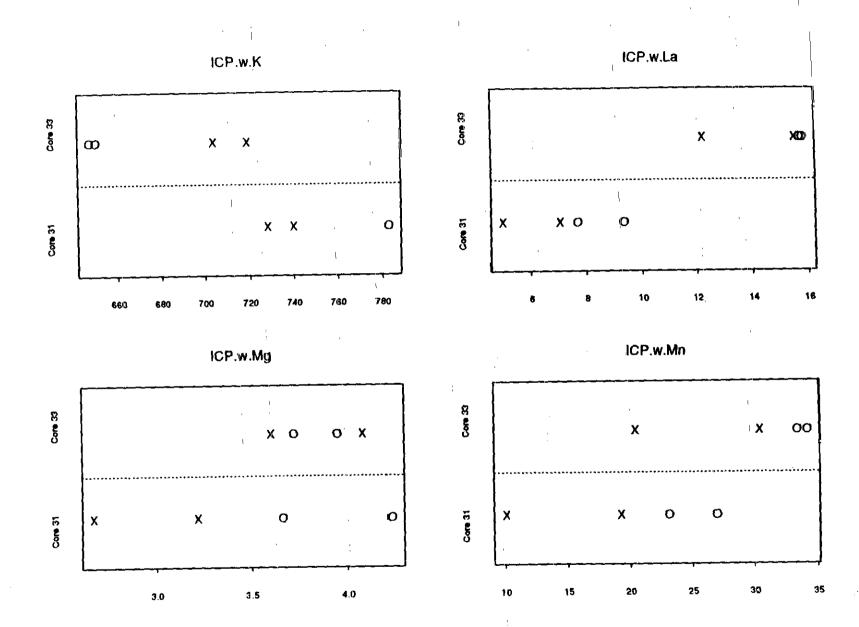
Composite 1 = 0, Composite 2 = X



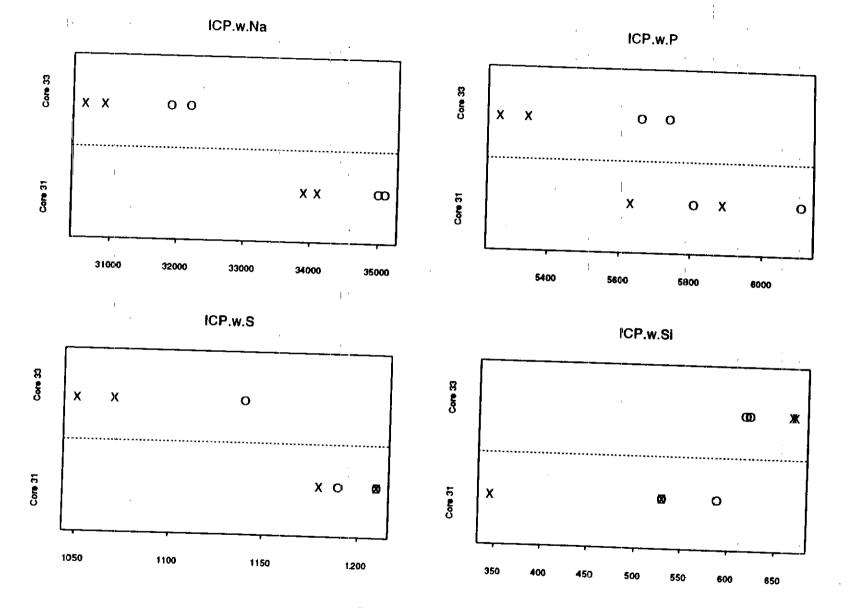
X = S etheograph , $O \simeq I$ etheograph S



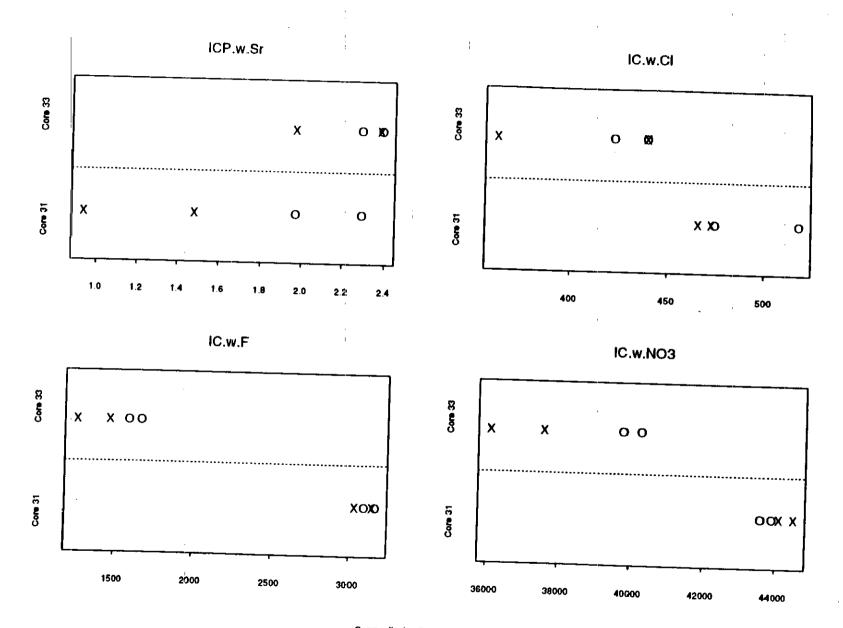
Composite 1 = 0, Composite 2 = X



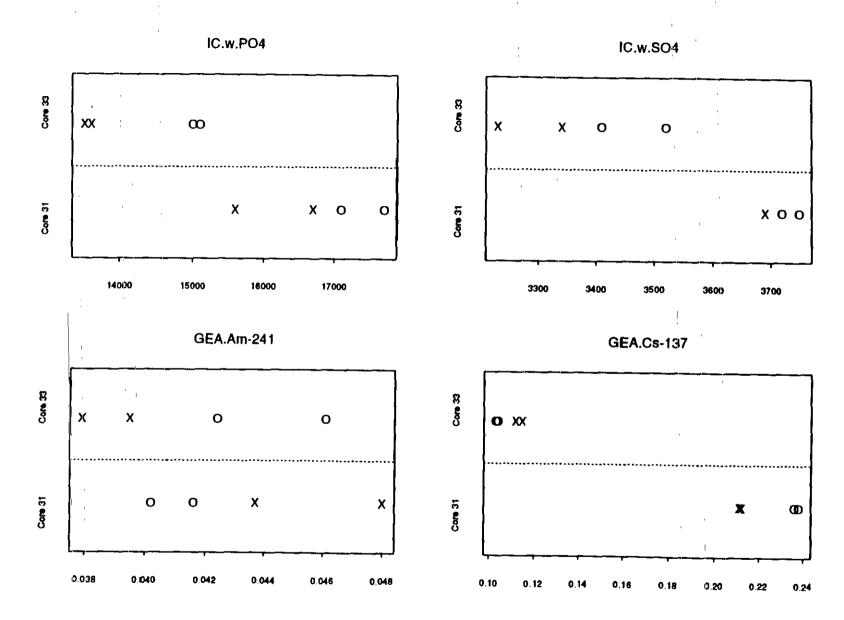
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Composite 1 = 0, Composite 2 = X



Composite 1 = 0 , Composite 2 = X



Composite 1 = 0, Composite 2 = X

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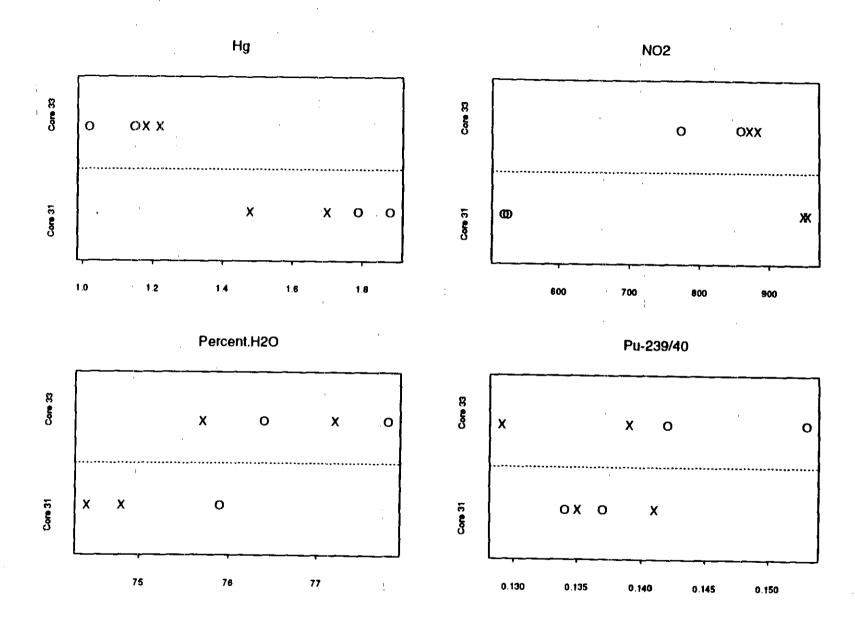
65

70

75

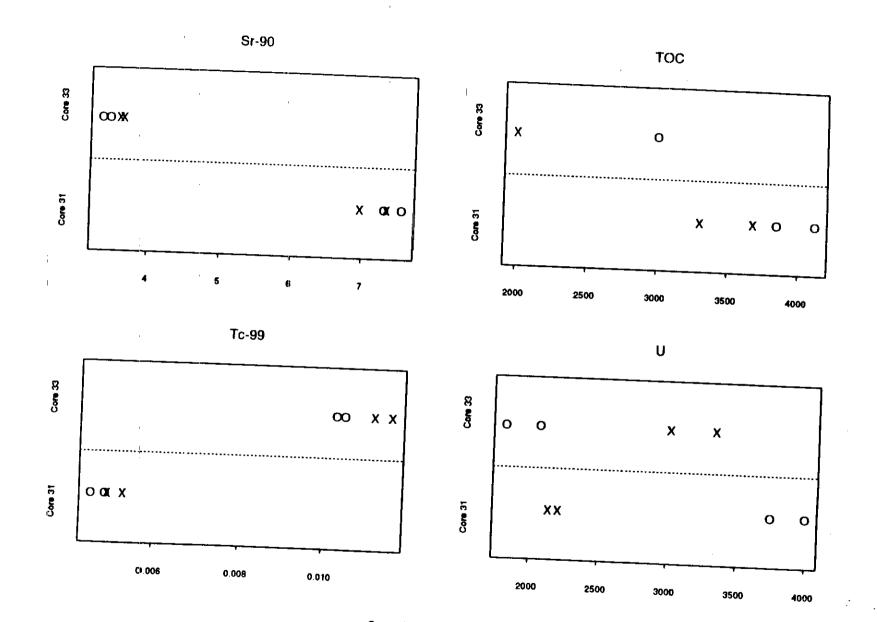
Composite 1 = 0, Composite 2 = X

85

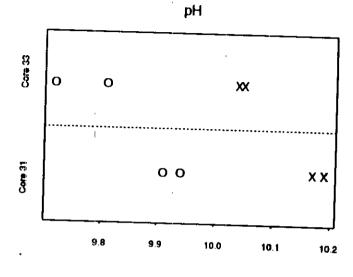


B-22

Composite 1 = 0, Composite 2 = X



Composite 1 = 0, Composite 2 = X



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APPENDIX C MEAN CONCENTRATION CALCULATION METHODS

WHC-EP-0806

WHC-SD-WM-TI-650, Rev. 0

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WHC-SD-WM-TI-650, Rev. O

This appendix includes a description of the statistical model that describes the structure of the data from core samples taken from Tank 241-T-111. Equations are also presented for estimates of the mean concentration, the variance for the mean concentration, and the CI on the mean concentration.

The statistical model that describes the structure of the core composite data is

$$y_{ijk} = \mu + S_i + C_{ij} + A_{ijk}, i = 1, ... a, j = 1, ... b_i, k = 1, ... n_{ij},$$
 (1)

where

 y_{ijk} = laboratory results from the k^{th} duplicate of the j^{th} composite of the i^{th} core from the tank,

 μ = the grand mean,

 S_i = the effect of the ith core (spatial effect),

 C_{ij} = the effect of the jth composite sample from the ith core,

 A_{ijk} = the analytical error associated with the k^{th} duplicate in the j^{th} composite from the i^{th} core,

a = the number of cores.

 b_i = the number of composite samples in the i^{th} core and

 n_{ij} = the number of analytical results from the jth composite sample in the ith core.

For cores 31 and 33 there are two core composite samples (i.e., $b_i = 2$).

The variables S_i and C_{ij} are treated as random effects. It is assumed that S_i , C_{ij} and A_{ijk} are each distributed normally with mean zero and variances $\sigma^2(S)$, $\sigma^2(C)$ and $\sigma^2(A)$, respectively. Estimates of $\sigma^2(S)$, $\sigma^2(C)$ and $\sigma^2(A)$ were obtained using Restricted Maximum Likelihood Estimation (REML). This method applied to variance component estimation is described by Harville (1977).

WHC-SD-WM-TI-650, Rev. 0

The mean concentration of each analyte of interest in the tank was calculated using the following equation:

$$\overline{y} = \frac{1}{a} \sum_{i=1}^{a} \overline{y}_{i++} = \frac{1}{a} \sum_{i=1}^{a} \frac{\sum_{j=1}^{b_i} \sum_{k=1}^{n_{ij}} y_{ijk}}{n_{i+}} = \frac{1}{a} \sum_{i=1}^{a} \frac{\sum_{j=1}^{b_i} \sum_{k=1}^{n_{ij}} (\mu + S_i + C_{ij} + A_{ijk})}{n_{i+}}.$$
 (2)

where

$$\overline{y}_{i++} = \frac{\sum_{j=1}^{b_i} \sum_{k=1}^{n_{ij}} y_{ijk}}{n_{i+}} \quad \text{and} \quad n_{i+} = \sum_{j=1}^{b_i} n_{ij}.$$
 (3)

This mean gives the results from each core the same weight regardless of the unbalance that may exist for a particular analyte.

The variance of \bar{y} is

$$V(\bar{y}) = C_1 \sigma^2(S) + C_2 \sigma^2(C) + C_3 \sigma^2(A)$$
 (4)

where

$$C_1 = \frac{1}{a}, \quad C_2 = \frac{1}{a^2} \sum_{i=1}^{a} \left[\frac{1}{n_{i+}} \right]^2 \left[\sum_{j=1}^{b_i} n_{ij}^2 \right], \quad C_3 = \frac{1}{a^2} \sum_{i=1}^{a} \left[\frac{1}{n_{i+}} \right].$$
 (5)

Using $\hat{\sigma}^2(S)$, $\hat{\sigma}^2(C)$, and $\hat{\sigma}^2(A)$ (REML variance component estimates), an estimated variance of \bar{y} -is

$$\hat{\sigma}^2(\overline{y}) = C_1 \hat{\sigma}^2(S) + C_2 \hat{\sigma}^2(C) + C_3 \hat{\sigma}^2(A). \tag{6}$$

The approximate degrees of freedom used for $\hat{\sigma}^2(\bar{y})$ is the number of cores with data minus one.

The lower and upper 95% CI limits (95% LL and 95% UL respectively) on the mean concentration are

95% LL =
$$\bar{y}$$
 - t_{.975} $\sqrt{\hat{\sigma}^2(\bar{y})}$ and 95% UL = \bar{y} + t_{.975} $\sqrt{\hat{\sigma}^2(\bar{y})}$ (7)

where t₉₇₃ is the 0.975 quantile from a Student's t-distribution with the approximate degrees of freedom associated with $\hat{\sigma}^2(\bar{y})$.

Appendix C: Calculation and Miscellaneous Data

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		· ESTIMATED WASTE	TANK INVENTORIES TOTA	AL 1 ST QUARTER, 1981	l	
COMPONENT	TANK T-10		TANK T-110 TANK MULES CURIES MOLES		TANK T-201 Moles curies	TANK T-202 Moles curies
Ac 225 Ac 227 Am 241 Am 242 Am 242m	2.E-04 2.1 1.E-14 2.1	E-00 5.E-18 6.E-00 E-06 5.E-10 4.E-08 E-01 3.E-05 2.E-02 E-08 1.E-15 3.E-07 E-08 1.E-10 3.E-07	1.E-18 2.E-11 3.E-18 2.E-10 3.E-08 4.E-10 9.E-03 7.E+00 2.E-02 4.E-14 8.E-08 1.E-11 4.E-08 8.E-06 9.E-07	8.E-08 1.E-09 2.E-05 2.E+01 4.E-02 3.E+01 2.E-03 2.E-12 3.E-04	0. 0. 0. 0. 0.	7.E-22 9.E-15 4.E-17 6.E-13 6.E-05 5.E-02 0. 0. 0.
Am243 A1217 Ba137m B1210 B1211	2 E-23 0. 2 E-00 1. 2 E-10 4	E-08 1,E-08 0.E-07 E-09 2.E-23 6.E-08 E+03 8.E-08 0.E+03 E-12 7.E-20 2.E-12 E-06 9.E-17 8.E-06	5.E-07 2.E-05 3.E-08 5.E-26 2.E-11 1.E-25 4.E-18 3.E-07 0. 4.E-19 1.E-11 6.E-19 3.E-17 3.E-06 7.E-17	4.E-11 2.E-25 6.E-11 0. 0. 0.	0. 0. 0. 0. 0. 0. 0. 0.	0. 2.E-29 8.E-15 0. 0. 8.E-22 2.E-14 7.E-24 6.E-13
81213 81214 C14 Cm242 Cm244	2 E-21 2. 2 E-03 1. 2 E-12 2.	E-00 1.E-18 0.E-09 E-11 6.E-22 6.E-12 E-01 2.E-03 1.E-01 E-06 3.E-13 2.E-07 E-04 1.E-09 2.E-05	4 E-21 2 E-11 1 E-20 5 E-21 4 E-11 8 E-21 3 E-14 2 E-12 0 . 8 E-12 7 E-06 2 E-08 4 E-20 8 E-16 0 .	7.E-11 5.E-20 5.E-10 0. 6.E-30 4.E-37	0. 0. 0. 0. 0. 0. 0. 0.	2.E-24
Cm245 Cu135 Cu137 Fr221 Fr223	4.E-02 7. 1.E-01 2. 2.E-10 0.	E-00 7.E-12 3.E-10 E-03 6.E-01 1.E-01 E+03 6.E-01 7.E+03 E-00 2.E-10 6.E-00 E-08 1.E-17 1.E-07	8.E-22 4.E-20 0. 1.E-11 2.E-12 0. 2.E-11 3.E-07 0. 4.E-22 2.E-11 1.E-21 4.E-18 4.E-08 1.E-11	0. 0. 0. 0. 8.E-37 1.E-37 0. 1.E-35 1.E-31 4.E-11 1.E-21 8.E-11 8.E-08 3.E-17 3.E-07	0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0	0. 0. 0. 0. 0. 0. 2.E-25 9.E-15 1.E-24 9.E-15
1120 Nb03m N163 Np237 Np230	7.E-08 2. 8.E-04 3. 1.E-02 2.	E-04 1.E-01 3.E-03 E-01 1.E-08 4.E-02 E+00 8.E-03 2.E+01 E-03 4.E-02 7.E-03 E-06 1.E-14 8.E-07	4.E-12 8.E-14 0. 9.E-06 2.E-01 8.E-06 1.E-14 5.E-11 2.E-04 3.E-04 5.E-05 5.E-04 4.E-13 2.E-05 2.E-13	8.E-01 2.E-03 9.E+00 9.E-05 1.E-03 2.E-04	0. 0 0. 0 0. 0	0. 0. 0. 0. 2.E-06 3.E-07
Pa231 Pa233 Pa234m Pb200 Pb210	4.E-10 2. 4.E-13 1. 4.E-11 4.	E-00 1.E-00 1.E-05 E-03 1.E-00 7.E-03 E-01 2.E-13 4.E-02 E-00 0.E-18 0.E-00 E-12 1.E-18 2.E-12	6.E-07 7.E-08 2.E-06 1.E-11 5.E-05 2.E-1 1.E-12 2.E-01 4.E-1 2.E-20 2.E-11 5.E-26 6.E-18 8.E-12 8.E-16	2	0. 0. 0. 0. 0. 0.	2.E-13 2.E-12 5.E-14 3.E-07 0. 0.E-24 0.E-15 1.E-18 2.E-14
Pb211 Pb214 Pd107 Po210 Po213	1.E-02 7. 4.E-18 4.	E-00 2.E-15 8.E-08 E-11 9.E-22 6.E-12 E-04 5.E-02 3.E-03 E-12 2.E-18 2.E-12 E-09 2.E-27 6.E-08	5.E-18 3.E-08 1.E-12 6.E-21 5.E-11 1.E-2 2.E-12 1.E-13 0. 9.E-18 9.E-12 1.E-1 6.E-30 2.E-11 2.E-2	0. 2.E-37 1.E-38 7 1.E-11 0.E-17 0.E-11	0. 0.	1.E-22

SSUMPTIONS USED IN THIS HODEL ARE STATED AT END OF REPORT

			ESTIMATED	WASTE TANK 1	NVENTOR I	ES T	D'TAI_	1 ST QU/	ARTER, 19	81	CONTINUE	E D		
	COMPONENT	TANK T-10 MOLES CL	18 TAN	K T-109 ES CURIES	TANK T		TANK T		TANK T-	112	TANK T		TANK T-	
	Po214 Po215 Po218 Pu238 Pu239	2 E-22 1 2 E-22 2 2 E-06 8	E-11 1.E- E-06 1.E- E-11 1.E- E-03 4.E- E+00 3.E-	21 8 E-06 22 6 E-12 07 2 E-03	8 E-2B 4 E-22 7 E-22 1 E-04 1 E+01	6 . E - 1 1 3 . E - 06 5 . E - 1 1 4 . E - 0 1 2 . E + 0 2	1 .E-27 1 .E-21 1 .E-21 2 .E-04 8 .E+00	9 .E-11 6 .E-06 7 .E-11 7 .E-01 1 .E+02	3 E-21 8 E-21 3 E-03	6.E-10 2.E-05 5.E-10 1.E+01 2.E+02	0 . 0 . 0 . 0 . 0 .	0. 0. 0. 0.	2 . E - 30 1 . E - 28 2 . E - 24 5 . E - 07 2 . E - 02	1 E-13 6 E-13 1 E-13 2 E-03 3 E-01
	Pu240 Pu241 Ra223 Ra225 Ra226	3 E-05 8 1 E-13 1 7 E-16 6	E-01 7 E- E-01 3 E- E-06 7 E- E-09 7 E- E-11 3 E-	06 9 E-02 13 8 E-06 16 6 E-09	3 E-01 3 E-03 2 E-13 2 E-18 2 E-13	2.E+01 6.E+01 3.E-06 2.E-11 5.E-11	4 E-01 6 E-03 6 E-13 5 E-18 3 E-13	2 E+01 2 E+02 6 E-06 4 E-11 7 E-11	1.E-02 2.E-12 6.E-18	4 .E+01 4 .E+02 2 .E-05 6 .E-11 5 .E-10	0. 0. 0. 0.	0. 0. 0. 0.	1 . E - 03 2 . E - 05 5 . E - 20 1 . E - 21 5 . E - 16	6.E-02 5.E-01 6.E-13 9.E-15 1.E-13
_	Ru 106 Sb 126 Sb 126 in Se 79 Sa 151	3 E-09 3. 3 E-12 3. 2 E-03 9.	E-04 5 E- E-02 3 E- E-02 4 E- E-03 1 E- E+02 3 E-	10 3 E-03 13 3 E-03 02 5 E-02	3 E-11 4 E-09 4 E-12 5 E-13 3 E-02	1 E-05 4 E-02 4 E-02 3 E-12 1 E+02	1 E - 10 4 E - 09 4 E - 12 0 2 E - 02	4 E-05 4 E-02 4 E-02 0 E+01	9.E-10 5.E-09 6.E-12 5.E-38 2.E-02	3.E-04 6.E-02 6.E-02 3.E-37 8.E+01	0 . 0 . 0 . 0 .	0. 0. 0. 0.	0 0 . 0 . 0 .	0. 0. 0. 0.
<u>`</u> -2	Sh126 Sr90 Tc99 Th227 Th229	4 E-01 4 2 E-01 3 2 E-13 1	E-02 I E- E+03 7 E- E-01 I E+ E-06 I E- E-09 I E-	03 9 E+01 00 2 E+00 12 8 E-06	1 E-02 4 E-01 6 E-11 4 E-13 3 E-13	4 E-02 4 E+03 1 E-10 3 E-06 2 E-11	1 E-02 3 E-01 0 9 E-13 9 E-13	4 E-02 4 E+03 0 6 E-06 4 E-11	2 . E - 02 5 . E - 01 3 . E - 36 3 . E - 12 1 . E - 12	6 .E-02 6 .E+03 5 .E-36 2 .E-05 6 .E-11	0. 0. 0. 0.	0 . 0 . 0 . 0 . 0 .	0. 0. 0. 8.E-20 2.E-16	0. 0. 0. 6.E-13 9.E-15
	Th230 Th231 Th234 T1207 U233	5 E-11 6. 2 E-08 1. 3 E-17 1	E-09 2 E- E-03 1 E- E-01 7 E- E-06 2 E- E-06 2 E-	11 2 E-03 09 4 E-02 16 8 E-06	2 . E - 09 8 . E - 11 4 . E - 08 7 . E - 17 3 . E - 09	1 .E-08 1 .E-02 2 .E-01 3 .E-06 7 .E-09	4 E-09 2 E-10 1 E-07 2 E-16 8 E-09	2.E-08 3.E-02 6.E-01 6.E-06 2.E-08	3 . E - 0 8 7 . E - 10 4 . E - 07 5 . E - 16 1 . E - 08	1.E-07 9.E-02 2.E+00 2.E-05 3.E-08	0. 0. 0. 0.	0. 0. 0. 0.	6.E-12 6.E-17 0. 2.E-23 5.E-12	3.E-11 8.E-09 0. 6.E-13 1.E-11
	U234 U235 U238 Y90 Zr93	1 .E+01 6 . 2 .E+03 1 . 9 .E-05 4	E-05 3 E- E-03 3 E+ E-01 5 E+ E+03 2 E- E-01 1 E-	00 2 E-03 02 4 E-02 06 9 E+01	5.E-05 2.E+01 3.E+03 9.E-05 1.E+00	7.E-05 1.E-02 2.E-01 5.E+03 3.E-01	8.E-05 5.E+01 8.E+03 8.E-05 1.E+00	1 E-04 3 E-02 6 E-01 4 E+03 3 E-01	7.E-04 2.E+02 3.E+04 1.E-04 1.E+00	1.E-03 9.E-02 2.E+00 6.E+03 3.E-01	0. 0. 0. 0.	0 . 0 . 0 . 0 . 0 .	1.E-07 2.E-05 0. 0.	2.E-07 8.E-09 0. 0.
	Ag AI Ba Bi C2H3O3	2 E-08 0 1 E+03 0 3 E+00 0 7 E+03 0	2 E+	04 0 00 0	2 E-17 0 . 2 E-03 1 E+07	0 . 0 . 0 . 0 . 0 .	0 0 0 1 E+07	0 0 0 0	0. 0. 6.E-03 1.E+07	0. 0. 0. 0.	0 0 0 0	0. 0. 0. 0.	0. 0. 0. 3.E+02	0 . 0 . 0 . 0 . 0 .

ASSUMPTIONS USED IN THIS MODEL ARE STATED AT END OF REPORT

			ESTI	MATED WAS	TE TANK	INVENTORI	ES	TOTAL	1 ST Q	UARTER, 1	981	CONTINU	ED		
	COMPONENT	TANK T MOLES	-108 CURIES	TANK T MOLES	-109 CURIES	TANK T MOLES	-110 CURIES	TANK T Moles	-111 CURIES	TANK T Moles	-112 CURIES	TANK T MOLES	-201 CURIES	TANK T	-202 CURIES
	G6H5O7 CO3 Ga Ce C1	0. 1.E+04 6.E-33 5.E-01 4.E-06	0 . 0 . 0 . 0 .	0. 3.E+05 6.E-34 2.E+01 1.E-04	0 . 0 . 0 . 0 .	0. 0. 0. 0.	0 . 0 . 0 . 0 . 0 .	0 . 0 . 0 . 0 . 0 .	0 0 0 0	0. 0. 0. 0.	0. 0. 0. 0.	0. 0. 0. 0.	0 . 0 . 0 . 0 .	0. 0. 0. 0.	0. 0. 0. 0.
	Cr EDTA F. Fe Fe(CN)6	1 .E+03 0 3 .E+04 2 E+04 3 .E+01	0 0 0 0	1 .E+02 0 . 1 .E+04 2 E+03 3 .E+00	0 0 0 0	2.E+04 0. 8.E+03 4.E+05	0 . 0 . 0 . 0 . 0 .	3 E+04 0 3 E+03 4 E+05	0. 0. 0. 0. 0.	3.E+04 0. 3.E+03 5.E+05 0.	0. 0. 0. 0.	0. 0. 0. 0.	0 . 0 . 0 . 0 . 0 .	2.E+02 0. 2.E+04 0.	0. 0. 0. 0.
_	HĖDTA K La Mn NO2	0 . 3 . E + 02 0 . 0 . 3 . E + 02	0 0 0 0	0 3 E+01 0 0 2 E+04	0 0 0 0 0	0 . 0 . 0 . 0 . 0 .	0 . 0 . 0 . 0 . 0 .	0 0 8 E+03 2 E+04	0 . 0 . 0 . 0 . 0 .	0. 0. 8.E+02 2.E+03 0.	0. 0. 0. 0.	0 . 0 . 0 . 0 . 0 .	0. 0. 0. 0.	0 2.E+04 1.E+02 3.E+02 0	0. 0. 0. 0.
'n	NO3 Na Ni OH PU4	2 .E+04 4 .E+05 6 .E-03 6 .E+04 1 E+05	0 0 0 0	3 E + 05 1 E + 06 6 E - 04 8 E + 03 5 E + 04	0 0 0 0	3.E+04 8.E+04 0. 1.E+08 1.E+07	0 . 0 . 0 . 0 . 0 .	0 0 0 2 E+06 1 E+07	0 0 0 0 0 0	4 .E-28 5 .E-07 0 . 2 .E+06 1 .E+07	0 . 0 . 0 . 0 .	0 . 1 .E+01 0 . 1 .E+01	0. 0. 0. 0.	1 E+05 2 E+05 0 . 5 E+04 3 E+03	0. 0. 0. 0.
	Pb Si03 S04 Sr Zr0	4 E-11 7 E+01 3 E+03 0 2 E+03	0 . 0 . 0 . 0 .	3 E - 10 4 E + 03 8 E + 04 0 . 3 E + 02	0. 0. 0. 0.	6 .E-11 2 .E-05 2 .E-03 0 . 4 .E-01	0 . 0 . 0 . 0 .	1 E-10 0 0 0 3 E-01	0 . 0 . 0 . 0 .	4 . E - 10 6 . E - 30 6 . E - 03 0 . 4 . E - 01	0. 0. 0. 0.	0. 0. 0. 0.	0 - 0 - 0 - 0 - 0 -	9.E-18 0. 3.E+02 0.	0. 0. 0. 0.

WHO ENVIRONMENTAL INDUSTRIAL HYGIENE FIELD MONITORING Page 1 of 2									
1. Work Package No.: ZW-94-00244-W 2. Contact: BUTCH HALL									
3. Building/Facility	3. Building/Facility: 272.WA TANK FARM 4. Area: 2-W 5. Monitoring Date: 3-5-94								
5. Room/Location/Site 40. T- FARM									
7. Monitoring By: Even D. Clerch of									
3. PPE (Resp. Prot.	& Prot. Clothing	g)	9. Engineer	ing Controls/Work Pract	ices:				
RWP-	CS-009		ALARA						
-10. [nstrument Type		11. Instrument ID Wos.	•	12. Calibration Data	. (Date, Time)				
Mx25	1	9308082-09	74	3-5-94 0700					
OVM 580 B		37025-25.	5	3-5-94	0700				
13. Monitoring Type	14. Fime	15. Location	16. Activit	Y	17. Results				
02	0940	111-	SUPERI	VATANT SAMP	Es 20.9%				
LEL		111-T	Supplus	ITANT Sample	5 0.0%				
ORGANICS	1040	111-T		TANT SAMPLES					
CRGANICS	0900 - 10:40	. 111-T		HING ZONE	0.0 Ppm				
		_			-				
		1/1 -			1-				
		MA			100				
· -									
	-								

EIH-FIELD MONITORING FORM (REV. 3, 9-24-91)

1. Menitering Type	14. Time (cont.)		ont inuetion Page	Page Z of
cont.		<u> </u>		17. Results (cont.)
				. /
	-			
- ·		λ	1/1	
			P	

18. comments: * ALL WORKER INSIDE 111-T SALTWELL PIT WAS ON SCBA
READING DIDN'T EXCEED 9.2 PPM T.O.C., BUT WAS A 3 MINUTE CONTINUOUS
HIT.

Westinghous Hanford Company

Internal Niemo

94-003

From:

West Tank Farms Operations

Phone: Date:

372-3919 T4-08 March 3, 1994 TANK T-111 Subject:

To:	D. A. Turner	R2-78		
cc:	H. Babad	R2-78	J. W. Lentsch	-78
	D. C. Board	S1 - 57	D. M. Lucoff	is. 36
	V. C. Boyles	R1-49	G. J. Miskho	R2- 2
	D. A. Bragg	R1-49	D. J. Newland	R2- 5
	aR. G. Brown	R2-11	R. Ni	S 5- 07
	R. J. Cash	R2-78	J. W. Osborne	R2-78
	G. M. Christensen	H4-21	М. А. Раупе	R2-31
	C. DeFigh-Price		M. Plys	H2-62
	A. J. Duckett	HO-39	A. Postma	H4-61
MAR-9 1994 N. W. Killy	G. T. Dukelow		T. E. Rainey	R4-02
MAH	J. C. Fulton	R2-31		Rz -4
المراجع المراجع المراجع المراجع المراجع المراجع المراجع المراجع المراجع المراجع المراجع المراجع المراجع المراجع	J. M. Grigsby			R2-
N. M. Barren	G., A. Hanson	R2-08	M. H. Shannon	H4-c.
	J. O. Honeyman	R2-52	M. J. Sutey	R1-49
•	M. N. Islam	R2-08	O. S. Wang	R2-78
	G. D. Johnson	R2-178	N. W. Kirch	R2-I1/
	R. D. Wojtasek	R2-36	J. L. Lee	R2-36
	JHW File/LB			,,,,

References: (1) Internal Memo, Organic Tank Safety to J. H. Wicks, et al., "Tank T-111" dated February 28, 1994

In accordance with your recommendation in reference 1, Tank T-111 has been placed on the organics tanks Watch List. Operational Specifications for -- Watch-List-Tanks (OSD-T-151-00030) has been changed to add tank T-111 to the Organic Salt List in Appendix A. Notification of this change was included in the Daily Operating Report (DOR) on 3/3/94.

If you have any questions, please call David P. Reber (373-5385).

J. H. Wicks, Manager

West Tank Farm Operations



To:

Internal Memo

7A700-94.004

From: Organic Tank Safety
Phone: 373-2238 R2-78

rnore: 373-2238 Date: February

February 28, 1094

of the the tenth of the law to

Subject: TANK T-111

J.	Н.	Wic	ks	20		
cc	:		Babad	R2-78 S1-57	J. W. Lentsch D. M. Lucoff	R2-78 N1-36
			C. Board C. Boyles	R1-49	G. J. Miskho	R2-12
		-	A. Bragg	R1-49	D. J. Newland	R2-36
		R.	G. Brown	R2-11	R. Ni	S5-07
		R.	J. Cash	R2-78	J. W. Osborne	R2-78
		G.	M. Christensen	H4-21	M. A. Payne	R2-31 MCK
		С.	DeFigh-Price	R2-31	M. Plys	H2-62
		A.	J. Duckett	H0~30	A. Postma	H4-61
			T. Dukelow	R2-78	T. E. Rainey	R4-02
	-	J., :	€Fulton ''-	R2-31 / 19 -	Ř. E. Raymond	R2-54
			M. Grigsby	H2-62/14/25	記記P. Sederburg	R2-11
			A. Hanson	R2-08/	M. H. Shannon	H4-61
			O. Honeyman	R2-52	M. J. Sutey	R1-49
	1		N. Islam	R2-08	O. S. Wang	R2-78
			D. Johnson	R2-78	J. H. Wicks	T4-07
			W. K <u>irch</u>	R2-11_	R. D. Wojtasek	R2-36
		J.	L. Lee	R2-36	DAT File/LB	

References:

- (1) Internal Memo, D. A. Turner to H. Babad, et al., "Reactive Component in Tank 241-T-111," dated January 17, 1994.
- Dry/Wet Preliminary TOC Results from Two Methods," dated January 11, 1994.
- "Tank T-111, Core 33, Segment 2 Preliminary Data," dated December 20, 1993.
 - (4) Internal Memo, A. J. Duckett to D. B. Bechtold, et al., "Tank Earms_PRC_Meeting Minutes for December 1, 1993 Potential Reactive Component in Tank 241-T-111," dated December 2, 1993.
 - (5) WHC-EP-0681, UC-600, "Interim Criteria for Organic Watch List Tanks at the Hanford Site," dated September 1993.

J. H. Wicks, et al. Page 2

February 28, 1994

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Waste Tank Safety Program recommends that tank T-111 be placed on the organic tanks Watch List, effective immediately.

The Tank Farms Plan Review Committee (PRC) made a decision on December 1, 1993, that tank T-111 was to be treated as an organics Watch List tank (Reference 4). This decision was prompted by the fact that portions of tank T-111 core material retrieved in 1991 and analyzed in 1992 exhibited significant exothermic activity in Differential Scanning Calorimetry (DSC) tests. The PRC also recommended that additional analyses be performed on existing samples (Reference 4).

PNL has rerun a number of key analyses on tank T-111 Core 33, Segment 2 samples (References 2 and 3). Reference I was issued after review of the PNL analytical results to put the tank T-111 issue into perspective, based on what we know at this time. Two conclusions were reached in Reference 1:

- No imminent safety hazard exists relative to the reactive component identified in tank T-111 waste because of the high (*80 wt.%) moisture content of the waste.
- Additional information on the active component in tank T-111 waste is required. The observed exothermic activity can not be fully accounted for based on reported total organic carbon (TOC) levels.

The recommendation to place tank T-lll on the organics tank Watch List at this time is based on the following rationale:

- Tank T-Ill Core 33, Segment 2 sample analyses performed by PNL were conducted on material dried at 60°C under vacuum in an effort to reduce the scatter previously encountered in analytical data.
- PNL analytical results—indicate exothermic activity in the dried sample of 215 cal/g (dry basis) from DSC tests (Reference 3). TOC content was determined to be 4.1 wt.% (dry basis) using the Furnace Total Combustion Method on dried sample material (Reference 2). Waste moisture content was confirmed to be approximately 80 wt.% (Reference 3).
- The attachment derives the energy equivalent of a 5 wt.% TOC (dry ____basis)_organic-nitrate/nitrite mixture based on sodium acetate as _____the organic-waste surrogate: 5 wt.% TOC (dry basis) is shown to be equivalent to exothermic activity of 151 cal/g.
 - Reference 5 establishes interim criteria for organic Watch List tanks at the Hanford Site. Single-shell tank (SST) waste is to be classified as "conditionally safe" if its organic content exceeds 5 wt.% TOC (dry basis) and its moisture content is >17 wt. %.

J. H. Wicks, et al. Page 3 February 28, 1994 7A700-94.004

- If, as shown above, 151 cal/g represents the energy equivalent of a 5 wt.% TOC (dry basis) organic-nitrate/nitrite mixture, then the waste in tank T-111 would be classified as "conditionally safe" in view of its 215 cal/g exotherm and 80 wt.% moisture content.
- SSTs with waste classified as "conditionally safe" are to be placed on the organic tanks Watch List (Reference 5).
- Tank T-III should, therefore, be placed on the organic tanks Watch List.
- It should be noted that PNL's analytical results indicate a TOC content of only 4.1 wt.% (dry-basis) (Reference 2). This TOC content, by itself, would not qualify tank T-111 for the Watch List. However, what matters from the standpoint of continued safe interim storage (e.g. safety) is waste energy content (cal/g). Tank T-111 clearly qualifies for the Watch List on this basis.

It appears that tank B-202 exhibits similar characteristics to those of tank T-111. However, it is not the recommendation of the Waste Tank Safety Program to place tank B-202 on the organic tanks Watch List at this time. The preliminary and incomplete analytical data have yet to be confirmed and further evaluation is necessary before making a final decision about tank B-202.

Please call (373-2238) if additional information or discussion would-be --beneficial.

D. A. Turner, Manager Organic Tank Safety

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Attachment

Modeling T-111

A simple model of T-111 was made on the ESP simulation program to attempt to shed some light on the vapor concentration issue. This simple model would mix about 1 liter of aqueous solution similar to T-111 and 1 liter of air. Some NPH type organics were also mixed. The vapor phase leaving the mixture was estimated.

Aqueous solution

The agueous solution used for this simulation was:

H2O 1000 g NaNO2 0.54 g NaNO3 30.7 g Na2CO3 0.5 g

The model calculated this to be 1.017 liters with a density of 1.026 compared with T-111 liquid density of 1.07.

AIR

One liter of air was introduce assuming only 02 and N2.

The model calculated that there was 1.08 liters of air

Organic

The organic chosen was of the NPH type.

C10H22 0.3 g C11H24 0.15 g C12H26 0.05 q

This mixture is just an estimation.

Mixture Conditions

The mixture conditions were 72 °F and 14.7 psia.

Case #1

A total of 0.5 grams of the organic mixture was added to the mixture.

This resulted in the following vapor composition:

Component	Vol %
H20	2.62 %
02	18.81%
N2	71.68%
C10H22	4.32%
C11H24	1.97%
C12H26	0.60%

The remainder of the organic was soluble in the liter of aqueous solution. There was no organic phase.

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Note that the vapor phase contained large amounts of the hydrocarbons. This is well above what was measured. Several orders of magnitude.

Conclusions:

....

T-111 does not need a separable phase organic layer to explain the observed vapor pressure of 9 ppm. Dissolved organics in an aqueous phase can explain this.

Had T-111 had a small amount of NPH floating on it, it probably would have evaporated by now. Notice that the ClOH22 is very much greater in concentration. This means that over time, the Cl2H26 would tend to be left behind. This has a smaller vapor pressure and is safer.

Distribution

LR Pederson	K2-44
MG Plys	H4-62
LL Burger	P7-25
RD Scheele	P7-25
GL Borsheim	R2-11
DM Camaioni	K2-38
MH Campbell	R3-77
BC Simpson	R2-12
WD Samuels	- K2-44
JA Campbell	P8-08
JP Sederberg	R2-11
H Babad	R2-78

% F	Battelle	ED	Project Number		
াক - Pa	cific Northwest Laboratories	RECEIVED	internal Distribution		
		JAN 2 U 1994 R.M. BEAN	JM Tingey MW Urie		
Date To -	January 14, 1994 RM Bean		RT Steele SG McKinley EW Hoppe		
From	DL Baldwin		File/LB		
Subject	FINAL T-111 (Core 33. Se	ament_2) DRY/AS-			

-----This is the Final report of results from TOC/TIC/TC measurements of the analyzed "dried" and "as-received" T-III sample, ALO No. 94-2631, analyzed by two methods, as follows:

1) Hot Persulfate method, in normal TIC/TOC mode, and TC mode,

RECEIVED TOC RESULTS FROM TWO METHODS, Rev. 2.

_____2) Eurnace Total Combustion method, in both TOC mode and TC mode.

The data on the drying of the sample, as performed by JM Tingey, shows the sample to contain 18.4% solids. In this report, no corrections are made for the "dried" samples results.

Description of Methods: Part 1 of this work is done by the hot persulfate oxidation method. Test Procedure PNL-ALO-381, rev. 0, "Determination of TC, TOC, and TIC in Radioactive Liquids, Soils and Sludges by Hot Persulfate Method". The M&TE No. for the carbon measurements is WCO1713, the balance M&TE No. is 360-06-01-016. The data is located on the accompanying data sheets, review reports or on file in the ALO-Records Office. TOC standard used is alpha-d-glucose, Kodak lot# BIF, and the TIC standard is reagent-grade Na.CO₃. Both materials are used in liquid form, dissolved in water, for system standards as well as matrix spikes. Normal operation is with a TIC step first, using acid only, followed by the TOC step on the same sample, adding the persulfate. TC is calculated by addition. A modified mode was used in which TC was determined, eliminating the TIC step. This should minimize possible loss of any unmeasured volatile organics. The estimated precision is ±10% and the estimated accuracy is ±15%.

Part 2 of this work is done by the furnace total combustion method, Test Procedure PNL-ALO-380, rev. 0, "Determination of Carbon in Solids Using the Coulometrics Carbon Dioxide Coulometer". The M&TE No. for the carbon measurements is WD04981, the balance M&TE No. is 360-06-01-023. The data is located on the accompanying data sheets, review reports or on file in the ALO Records Office. TOC standard used is alpha-d-glucose and the TIC standard is CaCO₃. Both materials are used in solid form for system standards as well as matrix spikes. As per normal procedure, TOC is determined at 600°C on a sample. Then TC is determined at 1000°C on a different sample. TIC is calculated by difference. The estimated precision is ±10% and the estimated accuracy is ±15%.

QC Narrative: This sample was analyzed on I-10-94. The QC for both methods all came within required MCS-033 limits, with no apparent outliers. For the

RM Bean January 14, 1994 Page 2

persulfate method, the TIC and TOC system standard recoveries ranged from 91.9% to 101.5%. The system blanks were consistent. The RPD's (Relative Percent Difference) for the dried sample, analyzed in duplicate, ranged from 3% to 10%. The matrix spike recoveries were 95% and 119%, respectively, for TIC and TOC, and 119% for the "crushed" sample TC spike.

For the furnace total combustion method, the TIC and TOC system standard recoveries ranged from 93.9% to 99.3%. The system blanks were consistent. The RPD's for the dried and as-received samples, analyzed in duplicate, ranged from 3% to 16%. The matrix spike recoveries, at 600°C, were 102% and 83%, respectively, for the dried and as-received samples.

Table 1: Hot Persulfate Method Results

Samole	sample wt (a)	TIC (%)	RPD (%)	TOC RPD (%)	TC RPD (%)
T-111 Dry-1 T-111 Dry-2	0.1304 0.1857	0.41 0.42	3	0.94 1.04 10 (avg=0.99%)	1.35 1.46 8 (avg=1.41%)
T-111 Dry-3 Spike T-111 Dry-4 (crushed)	95% 0.1638	TIC Rec 0.52	overy	119%	TOC Recovery 2.25
T-111 Dry-A (crushed) T-111 Dry-B (crushed)	•			(2000)	1.90 2.09 10
7-111 Dry-C (crushed) Sp	ike			(avg=2.00%) 119%	TC Recovery

Yot≇:

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Undissolved small black chunks of material were observed in reaction flask, indicating incomplete reaction.

The dried material was composed of large chunks, therefore it was crushed, in order to allow better dissolution. Improved dissolution was poserved, but still black undissolved fines were seen.

³⁾ Samples I through 4 were analyzed as per-normal TTC/TOC procedure. Sample A through 6 were analyzed by modified TC procedure. In this case, the persulfate and acid were acced at the same sime to the sample, deleting the TLC step, minimizing possibility of loss of volatile organic material.

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Table 2: Furnace Method Results

Sample	sample wt (g)	TIC (%)	RPD (%)	TOC (%)	RPD (%)	TC (%)	RPD (%)
T-111 Dry-1 (600°C) T-111 Dry-2 T-111 Dry-3, Spike	0.01495 0.01864 0.02345	102%	Recovery	3.76 4.41 (avg=4	16 .09%)		
T-111 Dry-1 (1000°C) T-111 Dry-2	0.03158 0.02333						6.2
T-111 As-Recd-1 (600°C) T-111 As-Recd-2 T-111 As-Recd-3, Spike	0.04890 0.10883 0.09272		ecovery	0.90 0.87 (avg=0	•	(avg=4.	. 84 <i>%</i>)
T=111 "As-Recd=1"(1000°C)	0.04227					0.98	

Note: 1) In the furnace method, TGC only is considered to be released at 600°C. At 1000°C, TC is considered to be released. Note that different weighed samples are used at the two different temperatures.

CONCLUSIONS:

- The Hot Persulfate method alone provides one conclusion, that there
 is no loss of volatile organic material in the TIC step. The results
 appear to agree well with the earlier reported "as-received" sample
 results, after accounting for the water weight loss.
- 2) The Furnace method results, in comparison, indicate that there is additional organic material present in these samples. This additional organic material was, for some reason, not fully-oxidized by hot persulfate, but required oxidizing with some stronger oxidant, (e.i., oxygen) at elevated temperatures.
 - In the dry material, from the Furnace method, the total TOC found was 4.09%, the total TC found was 4.84%. This compares, from the hot persulfate method, with total TOC found of 0.99% and total TC found

....E54-3000-101 (10/89)

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> of 2.0%. Both method results show good agreement between "dried" and -"As-Recieved" results, after accounting for water weight loss.

_____4) The missing, additional organic material, in the dried so e, amounts to:

-TC comparison: (4.84% - 2.00%) = 2.8%TOC comparison (4.09% - 0.99%) = 3.1%

Concur by: File: T-111.RMB

System File: Toc 11094

Date: 1/14/94

DON'T SAY IT --- Write It! DATE: June 9, 1994

Jan the transfer on the second

TO: Distribution

FROM: C. H. Delegard

Telephone: 373-4658

SUBJECT: CENTIFUGATION AND ANALYSES OF SLUDGE FROM TANK 241-T-111

Attached please find Internal Memo 8E110-PCL94-046, "Centrifugation and Analyses of Sludge From Tank 241-T-111." The attached IM provides recent thermal analyses of the centrifuged sludge and completes the reporting of the tank 111-T centrifugation tests. The attached IM replaces IM 8E110-PCL94-043, same subject. Please discard the IM 8E110-PCL94-043.

From:

Process Chemistry Laboratories

8F110-PCL94-046

Phone:

373-4658 T6-09

Date:

June 9, 1994

Subject: CENTRIFUGATION AND ANALYSIS OF SLUDGE FROM TANK 241-T-111

To:	D. B. Engelman	:-49
	CC: H. Babad G. S. Barney D. R. Bratzel J. M. Frye J. R. Jewett J. G. Kristofzyki L. M. Sasaki C. S. Simmons B. C. Simpson J. P. Sloughter C. S. Sutter D. T. Vladimiroff CHD File/LB	R2-78 T5-12 R2-12 T6-30 T6-09 T6-06 R2-12 K6-77 R2-12 T6-07 T5-12 R2-12

References:

- (1) Internal Memo, _G. S. Barney-and C.-H. Delegard to D. B. Engelman, "Test Plan for Centrifuge Drainage Tests of Tank 241-T-111 Sludge-Samples," dated May 3, 1994.
- (2) Internal Memo, G. S. Barney to D. W. Jeppson, "Results of Centrifuge Drainage Tests for Simulated Infarm-2 Ferrocyanide Sludge," dated March 28, 1994.

INTRODUCTION AND SUMMARY OF RESULTS

In Reference 1, a laboratory test procedure was described to determine the liquid retention capacity of sludge from tank 241-T-111 (111-T). The test procedure was based on a study, described in Reference 2, conducted to measure sludge drainage rates. In the present studies, tests were designed to simulate the drainage of interstitial liquid from the existing 14-foot (4.3 meter) overburden of sludge in 111-T should salt well pumping at the tank bottom take place.

Pressure-enhanced drainage was imposed in the laboratory by removing interstitial solution from samples of genuine sludge by filtration through coarse-frit glass in a centrifugal field of 500 times gravity (500 G). This field simulates the 4.3-meter sludge hydrostatic pressure at the tank bottom. The weights of liquids extracted and sludge retained were measured intermittently over a period of about five days. By knowing the water concentration of the original sludge and the expressed liquid, the amount of liquid remaining in the sludge cake could be determined and compared with

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water contents of the sludge cake measured directly at the end of centrifugation.

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Four sludge samples from lll-T were tested in duplicate by this procedure. Results showed the water content of the initial sludges ranged from about 69 to 80 weight percent (wt%). Centrifugation filtering produced compressed, but still damp, sludge materials and quantities of expressed liquid. The rate of liquid removal from the sludge cake was rapid in the initial hours of centrifugation and slowed considerably after two days' centrifugation. The water content of the sludges after two days' centrifuging ranged from 58 to 66 wt%; after five days, the water content decreased a further 1.6 wt% for each sludge on average.

The experiments and measurements performed to determine the solution retention of the centrifuged sludge samples are described in this report. Ancillary chemical and thermal analyses of the intact and separated sludge and solution fractions also are reported.

EXPERIMENTAL MATERIALS AND METHODS

Archive sludge samples taken in 1991 were used in these tests. The sample materials were derived from segments 3 and 7 of core 31 and from segments 1 and 7 from core 33. The segment numbering proceeds, in order, from top (segment 1) to bottom (segment 9) in 19-inch (0.5-m) increments corresponding to the length of the segments.

The experimental approach used in the centrifugation tests generally followed the methods described in reference 1. The experimental observations were recorded in notebook WHC N 656 $^\circ$

Coarse-frit glass lilters (40 to 60 μm pore size) fused into 1-cm diameter glass tubes (about 8 cm long) were used as the filtration medium. Samples of well-mixed sludge were introduced to the top of frit filter surface by plunger-type polyethylene thief samplers. Tare and gross weight measurements (\pm 0.00005 grams using a five-place balance) of the tubes before and after introduction of sample were obtained. The glass tubes were placed in screw-top polypropylene 50-mL capacity centrifuge tubes and reweighed. Sample weights ranged from about 1.8 to 3.4 grams. Most sample weights were about 2.3 grams.

The tubes with samples were spun in a centrifuge located in an open-face hood in the 222-S Laboratory at a target velocity of 1680 revolutions per minute (RPM). This velocity corresponded, at the 15.8 cm distance of the sludge from the axis of rotation, to a field of 500 G at the filter frit disc. Measurements of the rotational velocity of the centrifuge were obtained during the course of the experimentation using an optical tachometer. The nine velocity measurements ranged from 1617 to 1720 RPM (i.e., 464 to 524 G) and averaged 1684 RPM (501 G). The temperature within the centrifuge was about 1°C above ambient lab temperature. The lab temperature ranged from about 19 to 24°C during the experiments.

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The centrifuge was stopped periodically, the glass tubes and centrifuge cones were weighed and the weights of centrifuged liquid and sludge were determined by difference. The liquid was sampled for analyses of water content, density, thermal behavior and concentration of total organic carbon (TOC). After 113 hours, the centrifugation was stopped and the sludge analyzed for water content and thermal behavior. Samples of the original sludge also were analyzed for water content and thermal behavior. Sludge—water content was determined by weight loss on oven heating to constant weight at 116°C. For the water content determinations, original sludge material sample sizes were 1 to 2 g; about 0.2 to 0.5 g samples were used for the centrifuged sludge materials.

Densities of the original and centrifuged sludge were determined. The densities of original sludge were determined by weighing sludge samples into volume-calibrated graduated centrifuge tubes and noting the total sludge solution plus settled solids volume after centrifugation.

The compressed centrifuged sludges from the solution retention tests were not sufficiently plastic to slump in the volumetric centrifuge tubes used for density measurement. Therefore, the densities of the compressed sludge were determined via a displacement technique using an immiscible liquid of known density (n-hexane). Samples of the compressed sludge were added to a tare-weighed 5-mL volumetric flask, the flask reweighed, the flask filled to the mark with n-hexane, and the flask again reweighed. The volume of added n-hexane could be determined by weight and density. The volume and weight of compressed sludge then was calculated by difference and the sludge density calculated.

All chemical, physical and thermal analyses (except the water concentrations of the centrifuged sludges and the densities of the original and centrifuged sludges) were determined by 222-S Analytical Operations. The densities of the expressed liquids were determined by weighing known volumes of solution held in volume-calibrated pipets. The TOC concentrations were found by acidified sample pyrolysis followed by coulometric titration of the product carbon dioxide collected in a scrubber trap. Water content in the expressed ---solution was determined by weight loss of sample due to heating in an oven.

The thermal analyses included differential scanning calorimetry (DSC) and thermogravimetric analyses (TGA). All sludge and expressed solutions were analyzed by DSC/TGA. The DSC determined if a heat-producing (exothermic) event occurred. An example of an exothermic reaction would be the oxidation of organic carbon by nitrate. The TGA showed fractional weight changes (e.g., weight loss by water evaporation) as a function of increasing temperature. Thus water concentration values were obtained via TGA for all samples.

RESULTS AND DISCUSSION

All the original sludge samples were dark brown in color with a pasty but slightly gritty consistency. Sludge samples from the segment 1 and 3 archives showed no natural segregation whereas the segment 7 samples had about 10% supernatant solution. The entire contents of each archive sample

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8E110-PCL94-046

were mixed prior to drawing subsamples for testing. Subsampling from the same archive vials two days later showed separation of supernatant solution had recurred in the segment 7 samples but again no supernatant solution was found for the segment 1 and 3 samples. Centrifugation resulted in the drainage (through the glass frit filters) of yellow liquids from all sludge samples and the production of a sludge cake that was smaller both in height and diameter than the original sludge (i.e., the sludge compressed as well as shrunk away from the glass tube wall).

The water content of the centrifuged sludge was calculated by weight loss of the sludge as centrifugation proceeded, the water content of the original sludge, and the water content of the expressed liquid. The water content of the centrifuged sludge, after 113 hours' centrifugation at 500 G. was also determined directly via oven weight loss. The water content data, presented in the Figure, show the original sludge was about 80 wt% water for the segment 1 and 3 samples and about 74 wt% for the segment 7 samples taken near the tank bottom. Though the archive-segment 7 samples had supernatant liquid while the segment 1 and 3 samples did not, the segment 7 samples had lower water content than the segment 1 and 3 samples.

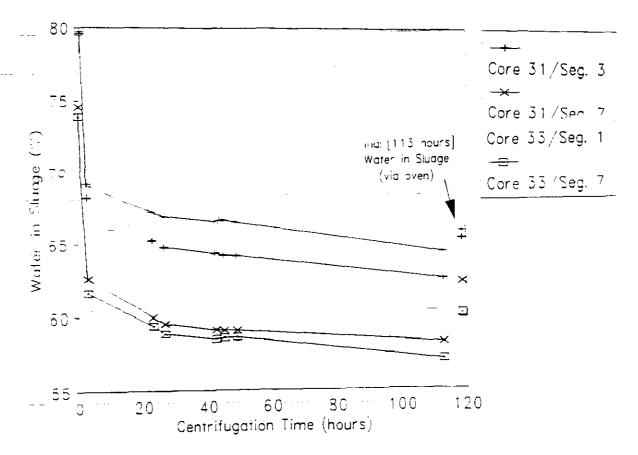


Figure. Water Concentration in Centrifuged 111-T Sludge.

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The sludge samples lost solution (and water content) quickly in the initial three hours of centrifugation. The rate of weight loss decreased with time of centrifugation. After 113 hours, the water contents of the segment 1 and 3 samples were about 63 wt%; the segment 7 samples were about 58 wt%. The water contents of the centrifuged sludges determined by direct measurement of oven weight loss were about 65 wt% and 61 wt%, respectively.

The water content data of the individual replicate sludge samples are given in Table 1. The data show precision in the replicate measurements of about 0.5% relative or 0.8% absolute in the oven measurements and about 1.3% relative or 1.9% absolute for the calculated values based on the sludge weight losses. The oven values are about $2.8 \pm 1.0\%$ higher, on average, than the calculated values. The bias probably is due to the cumulative effects of water vapor loss from the centrifuged solution during each weighing operation. For this reason, the oven wt% water values are judged to be more reliable than the calculated values.

Core/	Centrifugation Time (hours)								
Seg/ Dup	0	3_	23	26.5	42.5	45	49	_113	113 (Oven)
31/3/1	- 79.54	-68.6	-65.8	65.4	64.7	- 64.5	64.6	63.0	65.14
31/3/2		67.7	64.4	63.9	63.6	63.5	63.4	61.5	64.78
31/7/1	74.52	63.1	60.6	60.2	59.7	59.8	59.8	59.1	62.28
31/7/2		62.0	59.2	58.7	58.3	58.2	58.2	57.1	61.83
33/1/1	79.71	68.7	67.1	66.8	<u>6</u> 6.4	66.5	66.3	64.4	65.74
33/1/2		69.3	67.0	66.7	66.3	66.4	66.2	63.8	65.23
33/7/1	73.87	61.8	59.9	59.4	59.1	59.2	59.3	57.3	59.66
33/7/2		61.3	58.8	58.2	57.7	57.8	57.8	56.6	60.23

Table 1. Water Content of Sludge During Centrifugation.

The chemical and thermal analyses of the original sludge materials are summarized in Table 2. Exotherms were found for all sludge samples. The exotherms decreased as the sludge sampling depth increased (i.e., exotherms were greatest at the top). The wt% water values of the original sludge materials were determined both by weight loss, using 1 to 2 g samples heated to dryness in an oven, and by TGA, using 30 to 50 mg samples.

As shown in Table 2, the oven values for wt% water were higher than the TGA values. The values derived by oven weight loss are judged to be more reliable than the TGA values because they showed higher precision (better reproducibility in their duplicates), were obtained from larger, more representative samples, and thus were less subject to evaporative weight losses prior to initial weighing. The smaller TGA samples would be apt to lose more weight, on a relative basis, than the larger oven samples because they have a higher surface area to volume ratio. The oven-determined wt% water duplicate values for particular core/segment samples were averaged for

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use in calculation of the water content of the corresponding centrifuged sludge.

Table 2. Analyses of the Original Sludge.

	Wt% \	Vater			
Core/Seg/Dup	Oven	TGA	Density (g/mL)	Exotherm (J/g)	
31/3/1	79.47	76.33		191	
31/3/2	79.60	77.10	1.24	112	
31/7/1	74.28	72.99		33.1	
31/7/2	75.16	75.12	1.19	10.2	
33/1/1	79.64	77.12		249	
33/1/2	79.48	79.04	1.16	254	
33/7/1	73.93	81.95		41.4	
33/7/2	74.20	74.24	1.20	37.5	

The chemical, physical and thermal analyses of the solution samples are summarized in Table 3. Though the samples (especially the segment 1 and 3 materials) contained organic carbon, no exotherm was noted in the DSC of any of the solutions. Again, both oven and TGA values of wt% water were determined and, for the same reasons, the oven values judged to be more reliable. The average value of the oven-determined wt% water duplicate analyses for particular core/segment samples was used to calculate the water content of the corresponding centrifuged sludge.

Table 3. Analyses of Expressed Solution.

	Wt% W	later			
Core/Seg/Dup	Oven	TGA	Density (g/mL)	[TOC] (mg/L)	
31/3/1	92.27	86.80	Insuff. sample	1620	
31/3/2	91.65	85.13	1.040	979	
31/7/1	88.87	80.80	1.065	210	
31/7/2	88.52	85.40	1.087	140	
33/1/1	95.41	87.02	0.990	1010	
33/1/2	94.65	92.06	1.028	1000	
33/7/1	90.02	85.50	1.089	180	
33/7/2	89.44-	86.66	1.101	200	

The chemical and physical analyses of the centrifuged sludge samples are given in Table 4. The wt%s water derived from the TGA are significantly lower than those found by oven drying. Substantial air-drying of the centrifuged sludge apparently occurred prior to the TGA determinations. The DSC analyses showed that on a dry basis, centrifuged sludges had exotherms somewhat larger than the original sludges. For example, for sludge from

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segment I of core 33 (which exhibited the highest exotherm of the four samples studied), the centrifuged material had an exotherm of about 1700 J/g dry sludge as compared to 1140 J/g dry sludge for the original material. For segment 3 of core 31, the centrifuged sludge gave about 1150 J/g dry sludge (the 1525 J/g value was not included in the calculation) while the original sludge gave about 640 J/g dry sludge. The centrifuged segment 7 samples showed no measurable exotherms; exotherms for the original segment 7 sludges were about 100 to 200 J/g dry sludge.

Table 4. Analyses of Centrifuged Sludge.

	Wt%	Water		
Core/Seg/Dup_	Oven	TGA	Density (g/mL)	Exotherm (J/g)
31/3/1	65.14	53.80	1.05	546.9, 465.3
31/3/2	64.78	56.92	1.12	1525.3, 544.9
31/7/1	62.28	58.94	1.16	0, 0
31/7/2	61.83	52.71	1.24	0
33/1/1	65.74	51.11	1.21	838.1
33/1/2	65.23	51.62	1.16	822.4
33/7/1	59.66	44.82	1.24	0
33/7/2	<u>. 60.23</u> .	45.48	- 1.33	0

Please call should you have questions or comments on this work.

C. H. Delegard, Principle Senior Scientist Process Chemistry Laboratories

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SAMPLE STATUS REPORT FOR T 546. T-111 T-111 #1 TIME: 3/ 9/94 9:31 DISPATCHED: 3/5/94 10:31 SAMPLE HAS NOT BEEN SLURPED RECEIVED: 3/5/94 13:15

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1001	APPR/OTR	YELLOW CLEAR				••	N54D2
5706	SPG	1.03610E 00	NONE		N	Y	N54D2
5711	DSC	OUT FOR RERUN					N54D2
5711	DSC	INCOMPLETE					N54D2
5712	TGA	INCOMPLETE 1.15700E 01	NONE		37	17	N54D2
5713 5714	pH * H2O	OUT FOR RERUN	NONE		N	Y	N54D2
5714 5734	% H2O % H2O	INCOMPLETE					N54D2
5714 5720		2.21000E-01	nci /MT		37	32 –	N54D2
5720	TB				N	Y -	N54D2
5725	AT TOC	2.31000E-03			N	Y	N54D2
5726 5727	TOC TIC	4.73000E 02			N	Y	N54D2
5727 5728	NH4	8.00000E 02			N	Y	N54D2
5728 5729	OH LIQ	3.61000E 02			Ŋ	Y	N54D2
5729 5730	GEA	3.54000E 03 8.96000E-02		00-127	N		N54D2
5750 5750	ICP-LIQ			Cs-137	N		N54D2
5750 5750	ICP-LIQ	2.04000E 04		Na	N		N54D2
			uG/ML	Al	N		N54D2
5771		1.95997E 03		F-	Ŋ		N54D2
5771	IC	4.77260E 02		CL-	N		N54D2
5771	IC	1.33501E 03	uG/ML	NO2-	N		N54D2
5771	IC	2.91022E 04		NO3-	N		N54D2
5771	IC	8.06644E 03		PO4-	N		N54D2
5771	IC	2.78205E 03		S04 -	N		N54D2
5778	CN LIQ	2.71000E 00	uG/ML		N		N54D2
5781		OUT FOR RERUN					N54D2
5781	Pu239/40	INCOMPLETE	= 1 45				N54D2
5782	Am241	< 3.97000E-05			N	Y	N54D2
5783	Np237	< 1.38000E-05	uCi/ML		N	Y	N54D2
_5786 .		OUT FOR RERUN					N54D2
5786	Sr90	7.09000E-04	uci/ML		N	Y	N54D2
9114	ARCHIVE	INCOMPLETE					N54D2

END OF REPORT

SAMPLE STATUS REPORT FOR T 548. T-111 T-111 #2 TIME: 3/ 9/94 9:32 DISPATCHED: 3/ 5/94 10:48 SAMPLE HAS NOT BEEN SLURPED RECEIVED: 3/ 5/94 13:16

					OUT	OF	GOOD	CHARGE
EXT.	DETER.	RESULTS OR ST	ATUS		RANC	E?	ANS?	CODE
* * * *	****	*****	*****	******	**	t	***	****
1001	APPR/OTR	CLEAR YELLOW	< 1% SOLIDS					N54D2
5706	SPG	1.03760E 00			N		Y	N54D2
5711	DSC	OUT FOR RERUN						N54D2
5711	DSC	INCOMPLETE						N54D2
5712	TGA	INCOMPLETE						N54D2
5713	РĦ	1.15900E 01	NONE		N		Y	N54D2
5714	% H2O	9.29000E 01	% H2O		N		Y	N54D2
5720	$_{\mathtt{T}}\mathtt{B}$	2.31000E-01	uCi/ML		N		Y	N54D2
5725	AT	2.32000E-03	uCi/ML		N		Y	N54D2
5726	TOC	4.18000E 02	uG C/ML		N		Y	N54D2
5727	TIC	7.90000E 02	ug C/ML		N		Y	N54D2
5 728	NH4	3.70000E 02	uG/ML		N		Y	N54D2
5729	OH LIQ	2.70000E 03	uG/ML		N		Y	N54D2
5730	GEA	9.21999E-02		Cs-137	N		Y	N54D2
5750	ICP-LIQ	2.43000E 04		Na	N		Y	N54D2
5750	ICP-LIQ		uG/ML	Al	 N		Y	N54D2
5771	IC · ·	2.15977E 03		F-	N			N54D2
5771	IC	4.97640E 02		CL-	N		Y	N54D2
5771	IC	1.37823E 03		NO2-	N		Y	N54D2
5771	IC	3.00138E 04	uG/ML	NO3 -	N		Y	N54D2
5771	IC	8.24791E 03	uG/ML	PO4 -	N			N54D2
5771	IC	2.85180E 03	uG/ML	SO4-	N			N54D2
5778	CN LIQ	1.84000E 00	uG/ML		N		Y	N54D2
5781	Pu239/40							N54D2
5781	Pu239/40	INCOMPLETE						N54D2
5782	Am241	< 2.76000E-05			N		Y	N54D2
	Np237	< 2.89000E-05	uCi/ML		N		Y	N54D2
	Sr90	OUT FOR RERUN						N54D2
5786	Sr90	3.03000E-04	uCi/ML		 N	-	¥	N54D2
9114	ARCHIVE	INCOMPLETE						N54D2
			the state of the s					

END OF REPORT

T-111 #3 TIME: 3/ 9/94 9:32 SAMPLE STATUS REPORT FOR T 550. T-111

DISPATCHED: 3/5/94 10:49 SAMPLE HAS NOT BEEN SLURPED RECEIVED: 3/5/94 13:16

of the management of the the

					A		6412262
			> 071C				CHARGE
EXT.	DETER.	RESULTS OR ST	ATUS		RANGE?	ANS?	CODE
***	****			****	**************************************	***	
1001	APPR/OTR	CLEAR YELLOW				••	N54D2
57.06	SPG	1.03310E 00	NUNE		N	Y	N54D2
5711	DSC	OUT FOR RERUN					N54D2
5711	DSC	INCOMPLETE					N54D2
5712	TGA	INCOMPLETE					N54D2
5713	рН	1.17800E 01			N	Y	N54D2
5714	% H2O	9.29000E 01			N	Y	N54D2
5720	TB	2.48000E-01			N	Y	N54D2
5725	AT _	2.49000E-03			N	Y	N54D2
5726	TOC	3.80000E 02	ug C/ML		N	Y	N54D2
5727	TIC	OUT FOR RERUN					N54D2
5727	TIC	4.07000E 02			N		N54D2
5728	NH4	5.22000E 02	ug/ML		N		N54D2
572 9	OH LIO				N		N54D2
5730	GEA	8.78000E-02		Cs-137	N		N54D2
5750	ICP-LIQ	2.60000E 04		Na	N		N54D2
5750	ICP-LIQ			Al _	N		N54D2
5771	IC	2.18774E 03		F-	N		N54D2
5771	IC	5.11930E 02	uG/ML	CL-	N		N54D2
5771	<u>IC</u>		ug/ML	NO2-	N		N54D2
5771	IC	3.16720E 04		моз –	N		N54D2
5771	IC	8.83930E 03		PO4-	N		N54D2
577-1	IC	3.14581E 03		SO4-	N		N54D2
5778	CN LIQ	2.39000E 00	uG/ML		N	Y	N54D2
5781	Pu239/40	OUT FOR RERUN					N54D2
5781	Pu239/40	INCOMPLETE					N54D2
5782	Am241	< 2.81000E-05			N	Y	N54D2
5783	Np237	< 2.68000E-05	uCi/ML		N	Y	N54D2
5786	Sr90	OUT FOR RERUN	- • · · ·				N54D2
5786	Sr90	1.21000E-03	uCi/ML		N	Y	N54D2
9114	ARCHIVE	INCOMPLETE					N54D2

END OF REPORT

Estimate of plutonium concentration that would place 241-T-111 out of specification with regard to established inventory limits.

50 kg ^{239/240}Pu is the standard inventory limit for single-shell tanks¹.

Therefore, $\frac{50.000 \text{ g}}{2.171\text{E}+09 \text{ g}} = \frac{2.303\text{E}-05 \text{ g}}{2.303\text{E}-05 \text{ g}} = \frac{239/240 \text{Pu}}{2.303\text{E}-05 \text{ g}} = \frac$

Approximate proportion of Pu-239 = 96%Approximate proportion of Pu-240 = 4%

Specific activity of $Pu-239 = 6.2E-02 \text{ Ci/g}^2$ Specific activity of Pu-240 = 2.27E-01 Ci/g

Specific activity for the plutonium in the waste: 0.96*(6.2E-02 Ci/g) + .04*(2.27E-01 Ci/g) = 0.0686 Ci/g

The estimates of historical inventory values for selected analytes were derived from Agnew³. The estimates for 2C concentrations were done by taking the average concentrations of the two flowsheet formulations, 2C 44-51 and 2C 52-56. The 224 waste concentrations were used as presented. These values were multiplied by their respective volumes to develop the bulk inventory estimates. The buld concentrations presented for the tank were derived by multiplying each analyte's concentration by the fruition its waste contributes to the total inventory and adding them together, i.e. 0.526*(2C_{avg} concentration) + 0.474*(224 concentrations) = Bulk analyte concentration

¹Boyles, V. C. 1992, Operating Specifications for Single-Shell Waste Storage Tanks, OSD-T-151-00013, Rev. D-1, Westinghouse Hanford Company, Richland, Washington.

²Van Tuyl, H. H., 1962, Fission Product Generation and Decay Calculations, HW-75978, General Electric Company, Richland, Washington.

³Agn= , S. F. 1994, Hanford Defined Wastes: Chemical and Radionuc_ ie Compositions, LA-UR-2657, Los Alamos National Laborator Los Alamos, Mew Mexico.

															LAUR	-94-2657
	<u>;</u>					ļ]	1		1	 		Ī	:		
	MW 5 1		MW 52-	1044- 51/CW	1C52- 56/CW	2C44-5	1 2C52-5(S#224_	UR/TBP	PFeCNT	PFeCH2	TFeCN	1CFeCN		IR', 59- I67	:CWR/AI
	<u> </u>		<u> </u>	1C44-	1C52-									•	IR', 59-	-
hancs	Hainw	0		151/CW	-,	2044-51			UR/TBP	· - · · · · · · · · · · · · · · · · · ·	0.13541		1CF+CN		67 0 0.280	50-61
NaNO2		- ö			0 0): 0) 0	•			
NeCi				<u> </u>]			0							
N==		- 0	(0.170	8 0.1763	0.1058	0	0.356	5	0.0634	0.0634	2	0.1639			
					- 0	 	7 0		0); 0´	1	0	<u> </u>			
Ne2CO3	-	0.1	0.		0	0			!		1	<u> </u>	0		0 0	<u>) </u>
Na3PO4	0.20	873	0.2842	0.5587	0.5891	0.3119	0.9047	<u>.</u> 0	0	0	0	0 0	0.583	i (ò <u>o</u>) 0
Na2SO4	0.46	3274	0.4741	i d	0	0	. 0	0	0.1475	0.35513	0.35514	. 0	0	. (0 0). 0
Na2SIO3		0	C	0.5024	0.5058	0.4683	0.231	1			0.43805		0.4983	1	1 1	0.3509
AI203.3H20		0.6	0.6		+								0.07	0.6		
NEAIO2 FeO(OH)	<u> </u>	0 t	1	0.937			0.8814			0.85951				<u>`</u>		
Cr(OH)3		1	<u>-</u>													
MnO2		1	1		+											
BiPO4		1:		0.8987	0.8994	0.8525	0.7866	0.7587			0.88653					
Pb(OH)2		1 1	1	1 1	<u> </u>	1	1	1		11	1	1	1	1	1	1
La203 Ce203					 			0.6012								
Ce203 Na3cit.5H2O		01	0	· 0	0	0	01	<u>0,7</u>					0:			
Na Acetate	-	0	<u>0</u>										0:	0	<u>-</u>	
Na 2 Oxalate		0	ō					0.25					0	- 0		
Ne3HEDTA		0	. 0	· · -0	0	0	-0	0	1, 100	0		oi	0.			
Ne4EDTA		0	0			0 :	0 1	0		0:	0	0	0 :	0	0	
CaCO3.6H2O												+	1.	. 1	1	1
NI(OH)2 ZrO(OH)2				0.2909									0.6569	1		
Na2NiFe(CN)6.6	H2O		<u>-</u>	. 0.2909 1		1	1		1:	1 1		11	0.2851	1		
Pu		0	0.5			0	0	0			0;		0.5291	0	 -	0.8964
UO2(OH)216H2C)0	.75	0.75	0.9413	0.9811	0.9508	0.9289	0				0.4872		0.1873		0.7899
<u> </u>		0	- 0	0	07	6	0	0			1	11	1	0		
Sr		0.5	-0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.1		
ored. sludge	<u> </u>			1044-	1C52-	:								R'. 52- :	R', 59-	CWR.
opm	MW		:	51/CW	56/CW	2C44-51	2C52-56	#224 J	UR/TBP	PFeCN1	PFeCN2	TFeCN 1	CFeCN			50-61
<u> </u>	7442			84906		91756	37520	85054		108828	112478	59935	149106	27267	30241	39088
\		- 0 -		26886 8261.2		14346	0.	0!	0	47000 1	0	00	0	80416	56395	
		-			66.147		21538	35 236	0	17206.4			17854		13519	
31		0			8542.2				- 0				19.034 25403	12451	86255 0	0
<i>-</i>		0		0		_ 0		26692	0	Q	- 0	0:	0	- 0		
<u> </u>		<u> </u>		0		0	0	0		0	0	0	0	0	0:	
r0(0H)2		<u> </u>			506.03		0:	<u> </u>	0_	<u> 0:</u>	0	0	1051.7	0	0	0
0		- }-		0.		0	0	0		5200 80	0	0.	0_	0		0
Sr		-		- 0	0			0:	0	5296.88: 0-	3929.25	24844	4711.7 0	<u> </u>	0 !	0
An		0		0		0		190.28	<u></u> _	0.	- 0	0		0	0	0
<u> </u>		0		0.		0	0	0	- 0	0	0	0			0.	
<u> </u>		_0_		0	0 :		0	7789.5	0_	0:	0	0.	0	0	0	0
lensity	1.480			1 2050	1 2640	4 0070	4 4 2 2 2									
ol%solids		12		13.7	24.9	6.8	3.4	3.9	1.5282	1.49782	3.2					
old frac.	0.50			0.6026			0.9658	0.9251	0.5526	0.56193	0.49288	0.9942	0.2781	8.9	2.3	8.1
л.% H2O	38.34			65.494	73.61	70.904	81,073	68.962	30.449	32.84931	28.0527	81.588	57.788	56.029	38.078	0.675 51.253
00 wt.%0		0		01		_ 0	0	0.4152	0	0.64959	0.36973	1.8521	0.3474		0.	0
	_ 1413		<u></u>	28.312	73:634	193.87	218.81	468.48	60.802	†52.367°	67.4196	701.21	33.653	295.87	64.443	91.914
)H- 103-	7032			594491	40478 19717	13325	19938	468.46	99214	18341.8	21283.4	26242	17599	168648	220413	
102-	441	0		3656.1		42965	42545 0	61618	65015	67454.7°				67381	60603	20060
03-	3198			0		٥.	0		4363.5	0:	0	5600.1 0	1431.2′ 0		<u> </u>	
04	3038			95307		86076	26565			24511.2		8841.4			0	0
04	5647			2632.7		2778.5	2751.3		112329		131968	1100.7		680.86	848.28	0
103-	40.62		<u>·</u>	3048.1	2049.7	5981.9	4834.6	0	0	7981.57		0	5231	0		2350.8
<u>.</u> !-		0		5507.9	4694	7386.1	2494.2	46554	0	5989.52	6428 00	^	7514.8	^		
		n		Δ.	n.									0	0_	
6H5O7-		0		0:	0	0 -	0	0	296.26	0.	0	0	0	0	0	0

	i I MW 44 - 15 1	MW 52-		1C52-	2044-5	 2C52-56	##224	UR/TBP	'PFeCN1	PFeCN2	TFeCN	ICFoCN	R', 52-	R', 59-	:GWR/AI,
HEDTA-	0		0): 0): 0
·· NTA-	0		. 0								, 0) 0) () (
glycolate-	0		0						<u>. </u>						
acetate-	0		· 0												<u>_</u>
TBP	<u></u> 0		. 0						 			·			
NPH	ŏ		0												
CCM	0		0	- 0	- O	Ō	0	. 0	0	0					
hexone	0	1	0									0	0	0	
NIFe(CN)6-	0		0						24415.6					. 0	, 0
Pu (μCl/g)	0.00476	<u> </u>	0,0045			321.82				0.00076					
U (μανα) Ce (μCVg)	0.62888	0				0.2258				19212.1 8.12054					
Sr (µCVg)	52.661		0.3635			-0.3188			29.4438				41.255	كالش أرار المستعدد	
							<u>~</u>		25.4400	00,0707	v	4.23/3	. 34.403	3466.5	14.487
pred. su. ppm	MW	,		1C52- 56/CW	2044-51	2C52-56	#224	UR/TEP	PFeCN1	PFeCN2	TFeCN	1CF#CN	,	IR', 59-	CWR.
Na	41131.1		33049		41478	32592	31582	78745	75838	75558.1	60515	32694	53095		
A:	0		1252.8		0		0	0			0		6454.4		
Cr	0.		105.23	105.46	104.12	105.75		97.166			0			91.354	0
BI	01		103.63 295.33				297.34	01	274,174	274.271				1276	- 0
La	0		0	01	0		790.29	01		2/4.2/1	0	296.6	<u>0</u> ;		- 0
Ca.	0			0	0	0	0	- 0		01	0	0			- 0
Zr	0 :		257.82	258.4	0	0	0	0		01	o	258.93	01		
Pb	0:		0	0	0	0	0	0	Ō	01	0	0	0		0
Ni	0!	-	0	0.	0.	0 -	01	0	0:		95.186	99,991	0	0	0
Min	0		0	01	0	01	01	0 :		0	0	0	0		
Ca	- 0		0	- 0	- 0		240.37	01	0	0!	0	01	0	0 :	
K	0		0	0:	0	<u>_</u>	9839.9		0,	01	0	0	01	0	
balance							0,000.0		<u>_</u>					<u> </u>	
density	1.06901		1.0614	1.059	1.0733	1.05671	1,0545	1.1495	1.14291	1.1425	1.1102	1.0569	1.1161:	1.2223	1.2643
vol%solids	12'		13.7	24.91	6.8	3.41	3.9	1.2	3.7	3,2	1.4	6.5	8.9	2.3	8.1
wt.%-H2O	0.5032			0.7555	0.7506				0.56193		0.9942	0.2781	0.7056	0.6688	0.575
TOC wt.%C	09.3249		90.06	90.2461	86,928	90.017	87.115i 0.0514i	73.256	74.1115	74,141	98.76	99.415	79.053	69.708	71.7
species						<u>~</u> _	0.03141			<u></u>	0	0:		01	
ОН	3889.47		61.7611	116.72	304.93	244.05	591.77	146.28	355.351	182.146	882.05	182.54	576.13	156.43	184.62
NO3	6167.45		30889	31170	67577	47453	77838	156414	157318	1573741					40293
NO2	. 0		7975.41-		0_	0.	<u>0</u>	0	0:		7044.5	7762.9	0:	0	52316
PO4	32235.2 12435.1		12586	12610	0	0:	0 i	10498	0:	0	0		0 :	0	0
S04	10781			5795.2°	12479		4019.1		11104.9	11109.1	111221	12636	0:	01	
Si	111.77			530.46			O	0	491 374	491 5481		531.56	1325.8		
F	<u>0</u>		3580.1	3588	3542.5	2782					0	3595.5	0.	<u>_</u>	444.37
a	0		0	ō	0	0		712.76	0	0	0	0	0:	0	
C6H5O7_	00		0	0	0.	0	0	0	0 :	0	0 ·	0	10	0	0
EDTA HEDTA	0		0	. 0:		0		0	0.	0	0	0	0	0	0
NTA	0		0	0	0	0	0	0	0.	0:	o:		0 '	0 !	
glycolate	0		-0		0	0	0	- 0	0	01	0:	0	0 -	0:	
acetate	0		0	0 -	0	0	0	o o	0!	0	. 0:	0	0.	0:	0
oxalate			0	0	0		1883.6	0	0 :	0	0	0	0	0	0
NPH	0		<u> </u>	0.	<u> </u>	0	0	0	0	0	0	0	0 ·	01	0
CCH	0		0	<u>0 :</u> .0⊣	. <u>0</u>	0:	<u>o</u> ,	0	0		01	0	- 0:-		Ō
hexone	0	-	0	···· 0	- <u>.0</u>	0. 0	0	0	0		0:	0	0 .	0	0
NIFe(CN)6-	0		0	01	0	01	0	0.	0,	0:	0	0	0 '	0:	<u> </u>
Pu (µCVL)			<u>-</u>			` _	<u>_</u>	<u> </u>	<u>v</u> :		UI	0	0:	0	
U (M)												1			
Cs (CVL)															
Sr (CVL)						·									
8			10303					5605.5							
8:								2443'2							
BY															
BY'															
<u>A1</u>			<u>-</u> _												
A1'															
T1			10303					5605.5							
								J003.5							i

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TANK 111-T STATUS

Liquid, Gal.	Sludge, Gal.	Sp. Gr. (g/cc)
82 x 10 ³	430 x 10 ³	1.032

Chemical Analyses

0.701N
0.022
15.8
21.0
2.47
9.5
N.D.
3.48×10^2
1.21 x 10 ³
N.D.
N.D.
1.15×10^3

Boil-Down Studies

Percent Boil-Down	 Ð	50	80	9 ō
Boiling Point (°C)	105	105	106	107
Percent Solids - at boiling point at room temperature	clr clr	clr	5	7 70

3**-**15*-*65

Atlantic Richfield Hanford Company



Date:

June 7, 1974

To:

R. L. Walser

From:

R. E. Wheeler of Mkein

Subject: ANALYSIS OF TANK FARM SAMPLES

SAMPLE: T-3304 111-T

Vis-OTR: Clear, Yellow, 30% solids like rust. Filtrate < 10 mrad/hr.

pH:

13.25

SpG:

1.018

0H:

0.254 M

A1:

< 6.48 x 10⁻⁴ M

Na:

0.446

 NO_2 :

 $4.39 \times 10^{-3} M$

NO₃:

0.083 <u>M</u>

Pu:

1.17 x 10⁻⁶ g/1

DTA:

No Exotherm

SO4:

 3.79×10^{-3}

PO4:

 $10^{-2} \frac{M}{2.16 \times 10^{-2}}$ M 3 56

F:

 $2.16 \times 10^{-2} \underline{M}$ $3.56 \times 10^{-2} \underline{M}$

CO₃:

0.024 M

GEA:

134 137Cs - 1.13 uCi/gal 137Cs - 3.69 x 10² uCi/gal 125Sb - 4.66 μCi/gal

Water:

98.24%

Cooling Curve: 10°C for 45 min. No solids.

5°C for 45 min. No solids.

30% Reduction - 27.5

Solids - 0

50% Reduction-50.0

Solids - 0.5%

REW:jd

** #370.350 43.44

-c 120-4

Attentio affield Hanford Company

of the transformation the man

Date: September 24, 1974

To:

R. L. Walser

From:

R. E. Wheeler A Military

Subject: AMALYSIS OF TANK FARM SAMPLES

SAMPLE: T-4893

Vis-OTR: Black. 90% solids. 10 mrad/hr, filtrate

pH:

12.9

SoG:

1.0202

OH:

0.206 M

Al:

7.75 x 10⁻⁴ M

ila:

0.188 M ACT

NOpt

 $5.17 \times 10^{-3} M$

1103:

0.109 M

Pu:

 $< 1.41 \times 10^{-6} \text{ g/l}$

DTA:

No exotherm below 200°C

SO::

 $4.43 \times 10^{-3} M$

20_:

 $2.33 \times 10^{-2} M$

F:

4.28 x 10⁻² M

00 = :

 $6.59 \times 10^{-3} \text{ M}$

 $GEA: - - \frac{137}{2}Cs - 5.72 \times 10^{2} \mu Ci/gal$

inter:

95,45%

Sisting Surve: 40°C for 45 min. 90% Solids.

35°C for 45 min. 90% solids.

30°C for 45 min. 90% solids.

25°C for 45 min. 90% solids.

20°C for 45 min. 90% solids. 15°C for 45 min. 90% solids.

10°C for 45 min. 90% solids. 5°C for 45 min. 90% solids.

REWijd

cc: UR Christensen

US Buckingham

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